

Sistemas Hidrotermais e Metalogênese

GSA 0407

MINERALIZAÇÕES EPITERMAIS

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PDAC

ABOUT COMMUNICATIONS CONVENTION MEMBERS PRIORITIES SEARCH

FRANK ARNOTT AWARD

FAA - NEXT GENERATION EXPLORERS AWARD

Em nome da Next Generation Explorers Award Association, temos o prazer de anunciar que as inscrições para o Prêmio Frank Arnett estão abertas.

O Prêmio Frank Arnett é uma competição internacional em que equipes de estudantes universitários de geociências, com ênfase na mineração, colaboram e inovam para transformar os conjuntos de dados de geociências em sua interpretação dos alvos subterrâneos de geologia e mineralização.

As equipes finalistas apresentarão seus resultados para um painel ao vivo de juízes e público, durante o PDAC de 2021. Esta é uma ótima oportunidade para obter experiências do "mundo real" com conjuntos de dados modernos, além de reconhecimento internacional.

Como estudantes, vemos que a participação nesta competição aumentará sua rede com a indústria, além de oportunidades de emprego. Para as universidades, a competição pode aumentar o seu reconhecimento como uma universidade líder na comunidade de geociências com ênfase em mineração e aumentar o potencial de atrair futuros alunos e financiamento adicional.

Para inscrever sua equipe ou saber mais sobre o Prêmio Frank Arnett, acesse www.frankarnottaward.com ou envie um e-mail para info@NextGenerationExplorersAward@gmail.com

Sinceramente,
Next Generation Explorers Award Committee

FAA - NEXT GENERATION EXPLORERS AWARD

Background Concept Award How to Enter Datasets Sponsorship Contact

Background

Frank Arnett - Next Generation Explorers Award is a follow up to the Frank Arnett Award (FAA) 2017, founded in honour of Frank Arnett (1951-2009), an exceptional exploration industry leader who championed innovative techniques that maximized the value of the multidisciplinary data underpinning exploration campaigns worldwide.

Recognizing that effective data integration and visualization of data remains one of the biggest challenges, the Frank Arnett Award 'collaborative' challenge, focused on innovation in data integration and visualization, was launched in 2016. Participants selected one of five high quality global datasets and submissions were entered from around the globe in April of 2017. Evaluated by independent judges, the top 3 teams in both the Experienced and Apprentice categories were selected and honoured during the prestigious Exploration 2017 Decennial Mining Exploration Conference (DMEC). Each finalist team gave a presentation of their project findings and were formally presented with their final ranking and award October 24th 2017 in Toronto, Canada.

Given the success of the Frank Arnett Award and the opportunity that it presented for students worldwide, with the combined efforts of various sponsors and end-users the challenge will continue.

Gold Deposits

Periodic Table of the Elements 2006

See "If's Elemental: The Periodic Table" <http://pubs.scs.org/cen/019/elements.html>

Molecular Research Institute

High Sulfidation Epithermal Gold Deposits

Periodic Table of the Elements 2006

See "If's Elemental: The Periodic Table" <http://pubs.scs.org/cen/019/elements.html>

Molecular Research Institute

Epitermais

Pórfiros

Fig. 7. Schematic cross-section through a typical coupled arc batholith-capped-volcanic system, with associated porphyry Cu-Au and linked high-sulfidation Cu-Au epithermal deposits. Also shown are the thermal structure, fluid flow pathways and characteristics during the main stage of hydrothermal activity, and overprinting hydrothermal alteration zones. Porphyry alteration by circulating heated groundwater can be assumed to affect all the supracrustal rocks in the field of view, with general intensity (igneous, arc-related) due to the isotherms, fading to background diagenesis. Modified from Richards (2007); sources: Wilford (1973, 2010), ENR (1987), Shiohara and Hedenquist (1997), Hedenquist et al. (1996), and Fournier (1999).

Estados de Oxidação do Enxofre

Valence of Sulfur	Gaseous Molecules	Molecules	Aqueous	Ions
+6	SO ₃ trioxide	H ₂ SO ₄ sulfuric acid	HSO ₄ ⁻ bisulfate	SO ₄ ²⁻ sulfate
+4	SO ₂ dioxide	H ₂ SO ₃ sulfurous acid	HSO ₃ ⁻ bisulfite	SO ₃ ²⁻ sulfite
+2	SO monoxide	—	HS ₂ O ₃ ⁻ thiosulfate	S ₂ O ₃ ²⁻ thiosulfate
0	S, S ₂ , S ₄ , S ₆ , S ₈ sulfur	—	—	—
-2/x	—	H ₂ S _x hydrogen polysulfide	HS _x ⁻ bipolysulfide	S _x ²⁻ polysulfide
-2	H ₂ S hydrogen sulfide	H ₂ S hydrogen sulfide	HS ⁻ bisulfide	S ²⁻ sulfide

PRINCIPAIS TIPOS DE MINERALIZAÇÕES EPITERMAIS

1) High-sulfidation

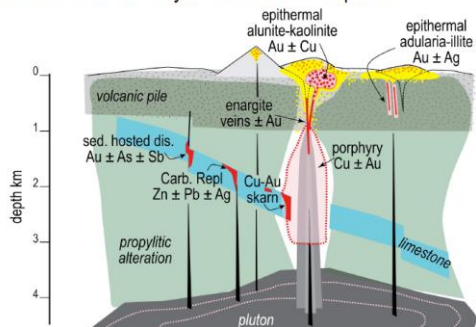
(fluidos ácidos oxidados produzidos durante a cristalização de magmas)

2) Intermediate-sulfidation

3) Low-sulfidation

(fluidos reduzidos, ~ neutros, com grande contribuição de água meteórica)

Intrusion-related hydrothermal ore deposits



Modified from Sillitoe, 2010

Mineralizações Magmáticas-Hidrotermais (alto conteúdo de Enxofre)

A happy family

Epitermais

Pórfiros oxidados

High-sulfidation

Intermediate-sulfidation

Low-sulfidation

Pórfiros reduzidos

Low-sulfidation

Acidez dos fluidos devida predominantemente a H₂SO₄

Sistemas com Baixo conteúdo de Enxofre

IOCGs

Low-sulfidation

Acidez dos fluidos devida predominantemente a HCl e HF

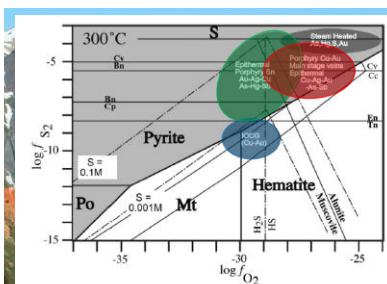


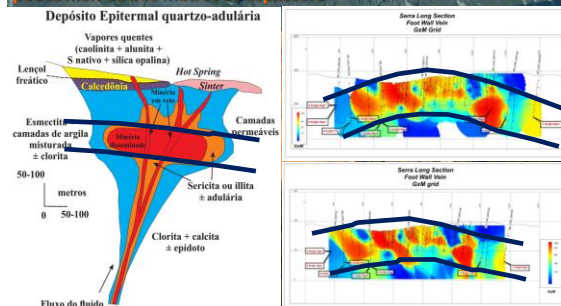
Fig. 8. Ore mineral stabilities plotted on f_2 vs. f_{O_2} diagram. Common environments with associated advanced argillic alteration are shown in broad fields. Note that many of these environments plot in the sulfate-dominated portion of the diagram, and the presence of alunite indicates oxidized conditions. The term sulfidation state does not apply to the low-sulfur systems dominated by hypogene iron oxides in the lower portion of the diagram. Diagram after Barton and Skinner (1979) calculated using SUPCRT (Johnson et al., 1992) with temperature fixed at 300 °C and pH = 5.4, buffered by muscovite-potassium feldspar. Abbreviations: Po = pyrrhotite; Mt = magnetite; S = native sulfur; Cv = covellite; Cc = chalcocite; Bn = bornite; Cp = chalcopyrite; En = enargite; Tn = tennantite.

Kroner, E-Barton (2017) Ore Geology Review

Mineralizações epitermais

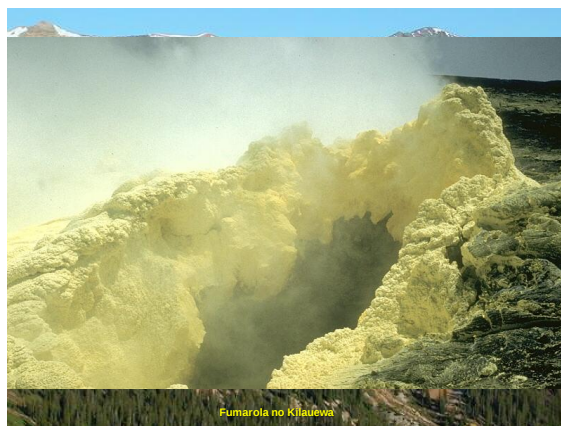
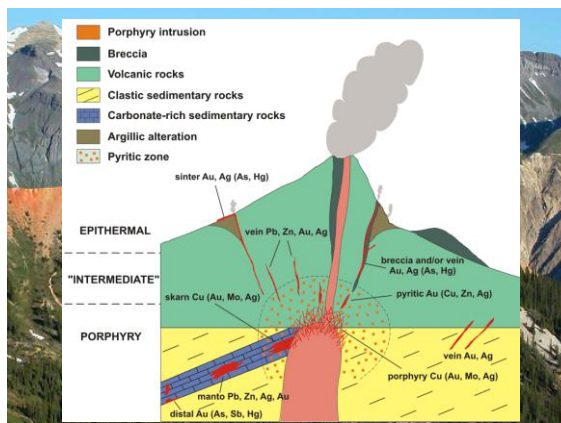
- 1) Forma-se em **temperaturas** entre 200 e 300 °C, com variações para < 150 a > 400 °C)
- 2) Formam-se em **profundidades rasas** ou **superficiais** (pressões baixas), alcançando até ~ 2 km (normalmente formam-se a poucas centenas de metros de profundidade)
- 3) Associam-se predominantemente a **vulcanismo intermediário e fêlsico** e arcos magmáticos continentais e insulares, sistemas oceânicos - principalmente em bacias de back-arc e em magmatismo intraplaca
- 4) Se relacionam com caldeiras vulcânicas

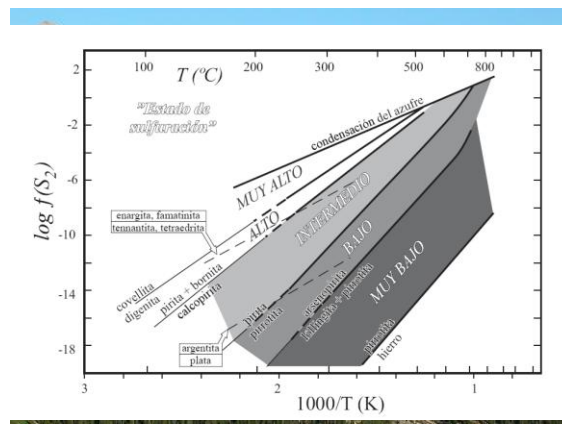
- Deposição do conteúdo metalífero dos fluidos nas zonas de **boiling**, indicando que os corpos mineralizados não tem continuidade em profundidade nas falhas
- A concentração dos corpos mineralizados estão numa zona de pouco mais de 300 metros de espessura



CARACTERÍSTICAS GEOLÓGICAS GERAIS DA MINERALIÇÃO EPITERMAIS

- **Rochas hospedeiras:** vulcânicas de composição cálcio-alcalina ou alcalina → andesitos, dacitos, riolitos (raramente basaltos) e → vulcanoclásticas piroclásticas a epiclásticas → rochas (meta)sedimentares → rochas do embasamento das vulcânicas
- Mineralização em **sistemas de fraturas ou falhas** → estruturas extensionais, às vezes associadas a estruturas de colapso em zonas de caldeira, brechas hidrotermais → circulação mais efetiva de fluidos.
- Mineralização em veios (1-10 m de espessura) de alto ângulo, sistema de veios ou stockworks e disseminada. Mineralização extensa ao longo da direção, mas **resistia em profundidade** (500 a 1500 m, média de 350 m).





Goldfield type	Hedenquist (2005), baseado em Sillito & Hedenquist (2003) e Enaudi et al. (2003)	Ramsone (1997)
Alunite-kaoilinite gold veins	Sericitic zinc-silver veins	Gold-silver-adularia veins Fluoritic tellurium-adularia gold veins
Gold-sulfite deposits	Gold-quartz veins in andesite	Lindgren (1933)
Secondary quartzite (Altered host rock)	Argentite veins Base metal veins	Fedorov (1992); Nakovnik (1993)
Acid	Alkaline	Sillito (1977)
Enargite-gold	Epithermal	Buchanan (1981)
High sulfur	Hot-spring type	Ashley (1982)
Acid sulfate	Low sulfur	Giles and Nelson (1982)
High sulfidation	Adularia-sericite	Bonham (1988, 1989)
Adularia-kaoilinite	Low sulfidation	Hayba et al. (1985), Heald et al. (1987)
High sulfidation	Adularia-sericite	Hedenquist (1987)
Adularia-kaoilinite	Adularia-sericite	Berger and Bentley (1993)
High sulfidation	Type 1 adularia-sericite	Albino and Margolis (1991)
High sulfidation	High sulfide + base metal, low sulfidation	Sillito (1993)
Lithocap (Altered host rock)	Low sulfide + base metal, low sulfidation	Sillito (1995)
High sulfidation	Western andesite assemblage, low sulfidation	John et al. (1999); John (2004)
High sulfidation (HS)	Intermediate sulfidation (IS)	Low sulfidation (LS)
		Hedenquist et al. (2000)

OS DOIS SUB-TIPOS PRINCIPAIS:

Associação mineral diagnóstica: **High-Sulfidation (HS)**
Quartzo + Alunite ± Pirofilita ± Dickita ± Caolinita

Indica: pH ácido e condições oxidantes (H_2SO_4)

← Sulfato

Associação mineral diagnóstica: **Low-Sulfidation (LS)**
Quartzo ± Calcita ± Adulária ± Illita (Sericita)

Indica: pH - neutro (ao redor de 7) a levemente alcalino e condições redutoras (H_2S)

← Sulfeto

Segundo as classificações de Hedenquist (1987) e White & Hedenquist (1990, 1995)



UM TERCEIRO SUB-TIPO IMPORTANTE:

Associação mineral diagnóstica:
Quartzo + Illita/Sericita + Sulfetos de metais de base relativamente abundantes (calcopirita, esfalerita, galena) + Carbonatos (Calcita manganosífera e/ou rodocrosita) ± Piemontita (Alunite, Pirofilita e Adulária ausentes ou ocasionais)

Indica: pH - neutro e condições levemente oxidantes

← Sulfetos

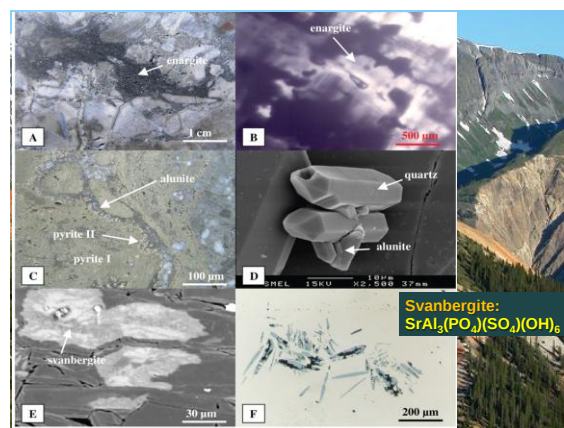
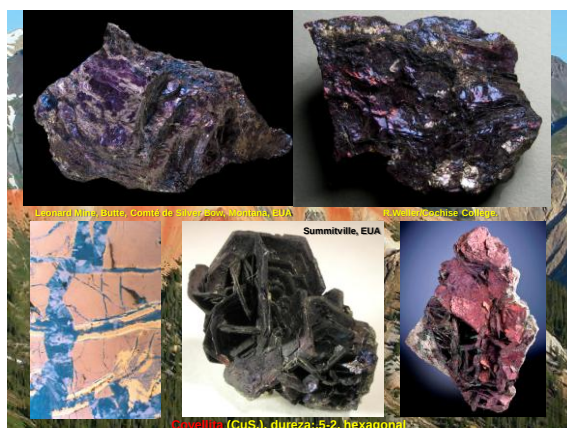
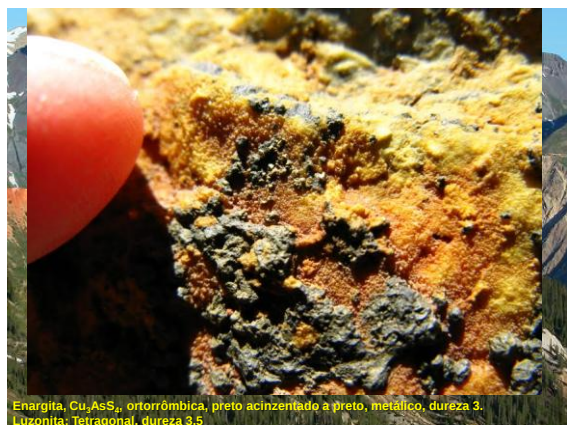
Intermediate-Sulfidation (IS) ou Base Metals Epithermal

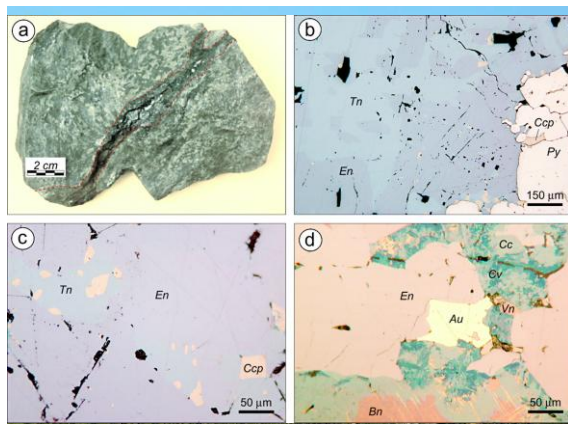
Segundo as classificações de Hedenquist (1987) e White & Hedenquist (1990, 1995)

Epithermal Indicator Minerals: Chemical Conditions

pH	sulfidation fS_2	redox
acid	high	oxidized
quartz alunite pyrophyllite dickite kaolinite	pyrite+bornite enargite, <u>laminitite</u> luzonite, covellite, orpiment	alunite hematite+magnetite
neutral	intermediate	reduced
quartz calcite adularia illite	low	magnetite+pyrite+pyrrhotite, chlorite+pyrite
	arsenopyrite+bellingite+pyrrhotite, Fe-rich sphalerite+pyrite, pyrrhotite	

	VERY LOW	LOW	INTERMEDIATE	HIGH	VERY HIGH		
	ldl+po+S ₂ +asp	ldl+S ₂ +asp+As	po+S ₂ +py mgt+S ₂ +hem+py	ten+S ₂ +enar cpy+py+S ₂ +bor	dig+S ₂ +cov		
SYSTEM	1	2	3	4	5	6	7
Fe-S	pyrrhotite (po-pj)	pyrrhotite	pyrrhotite	pyrite (py)	pyrite	pyrite	pyrite, <u>As</u>
As-S	As	As	As	As	<u>realgar</u> , <u>orpiment</u>	<u>orpiment</u> , AsS glass	AsS glass
Fe-As-S	<u>bellingite</u> , <u>luzonite</u>	bellingite, arsenopyrite (asp)	arsenopyrite	Arsenopyrite <u>px+asp</u> , <u>px+As</u>			
Cu-S	chalcocite	chalcocite	chalcocite, digenite	digenite	digenite	digenite	<u>covellite</u>
Cu-Fe-S	chalcocopyrite (cpy) + pyrrhotite	chalcocopyrite (cpy) + pyrrhotite	Chalcocopyrite +pyrrhotite	chalcocopy + pyrite	Chalcocopy +pyrite	bornite <u>bor+py</u> , digenite +pyrite	digenite +pyrite
Cu-Fe-As-S	<u>luzonite</u> , ldl + cpy	asp+cpy, ldl + cpy, asp+tennan	asp+cpy, asp + tennantite	asp+tennan, <u>asp+py</u>	enargite	enargite	enargite
Cu-Fe-S-O	magnetite (mgt)+ bornite, mgt+cpy	mgt+bor, mgt+cpy	mgt+bor, mgt+cpy	mgt+bor, mgt+cpy, <u>dig+asp</u>	mgt+cpy, hematite (hem) +py	hem+py	hem+py
OTHER	Bi, Ag, Hg, Sb, Sb+po, galena(gal)+hematite	Bi, Ag, Hg, Sb, Sb, po, galena(gal) +hematite	Bi, Ag, Hg, Sb, bismuthinit, stibnite, gal+hem, argentite + berthierite	Bismuthinit (bit), argentite + berthierite, cinnabar, stibnite(stb), gal+hem, stb+py	bit, arg, cin, stb, gal+hem, angite+py, stb+py, famatinite	bit, arg, cin, stb, stb+py, angite+py, famatinite	bit, arg, cin, stb, stb+py, angite+py, famatinite
modified in Sphalerite Coexisting with Fe-sulf	>40	40-35	35-20	20-5	5-0.1	0.1-0.01	<0.01





Löllingita (FeAs_2), ortorrômbico, cinza-azul a prateado, metálico, dureza 5,0 a 5,5



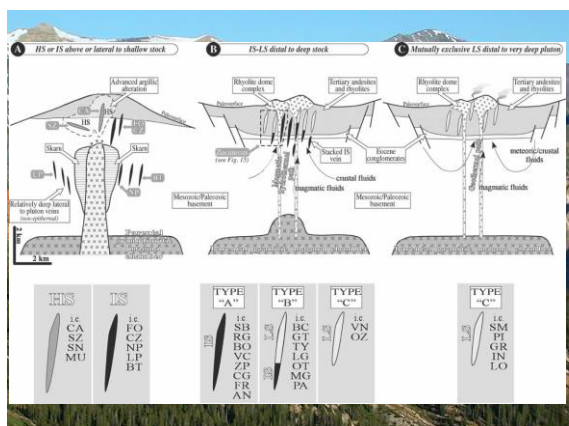
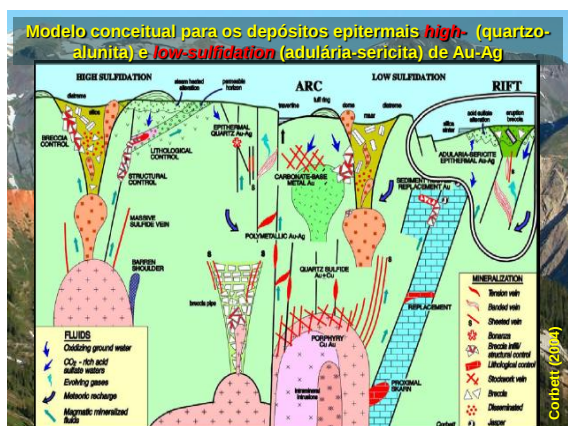
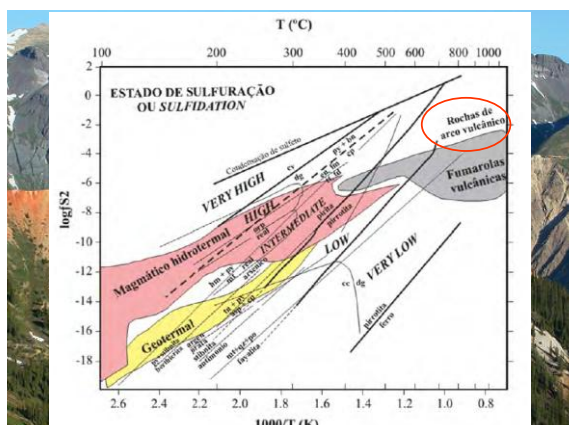
Alunita

$\text{KAl(SO}_4)_2 \cdot 12\text{H}_2\text{O}$

Trigonal

Dureza: 3,5 - 4,0

Cor: amarela, vermelha, marrom-avermelhado, branca, rósea, cinza, incolor



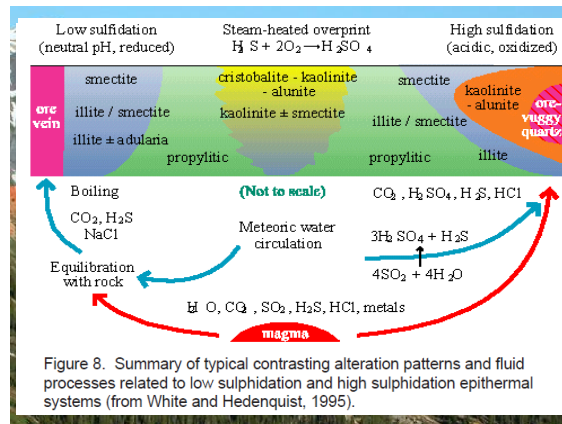
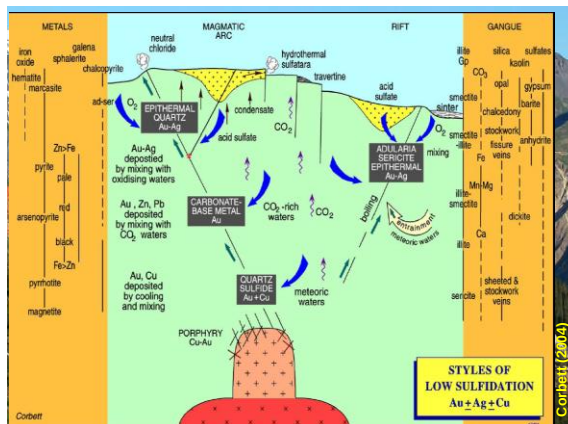
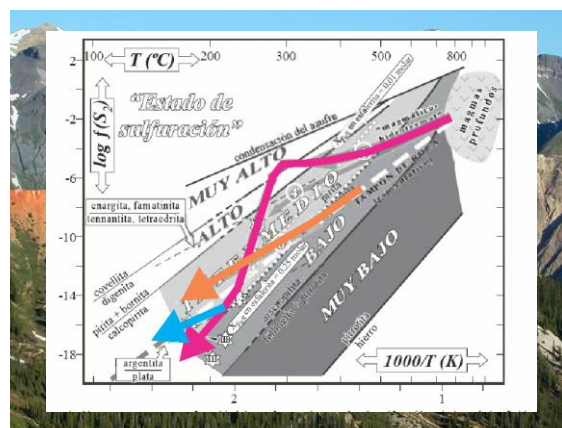
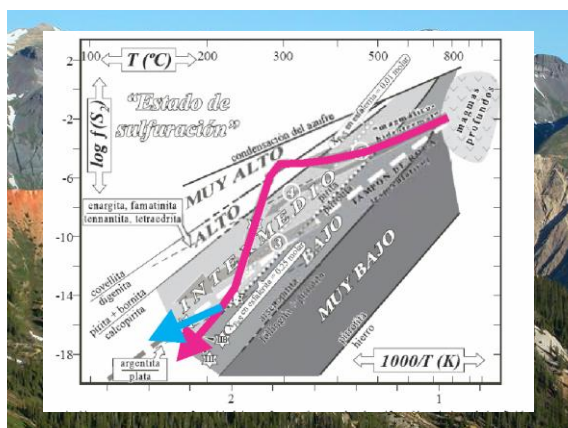
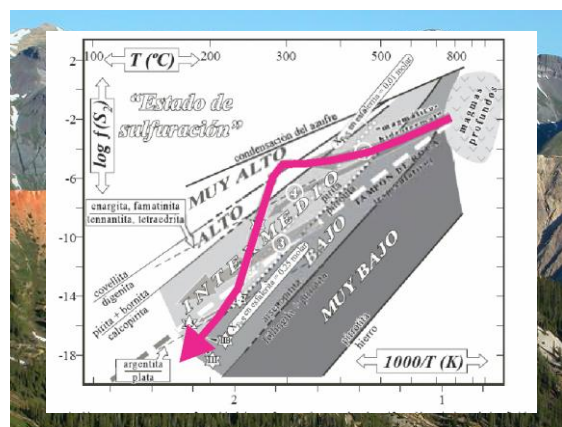
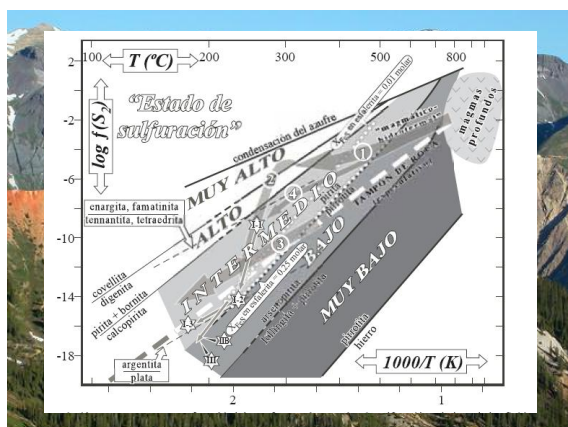
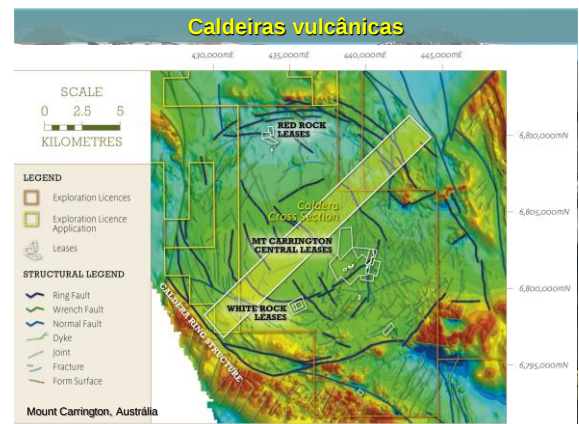
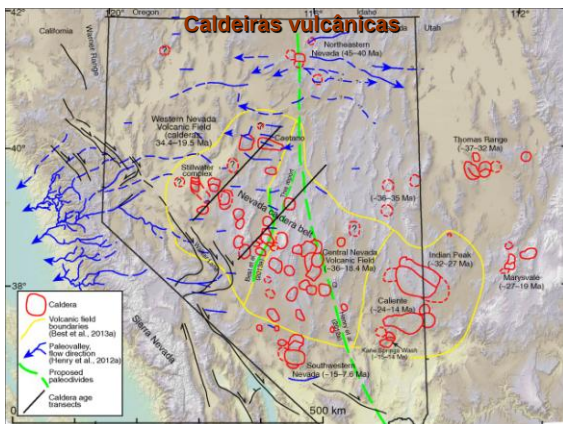
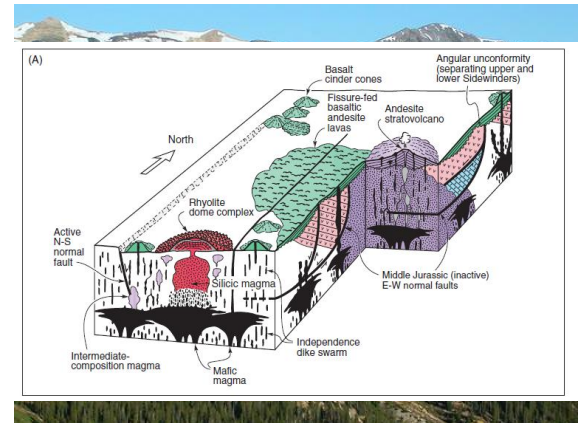
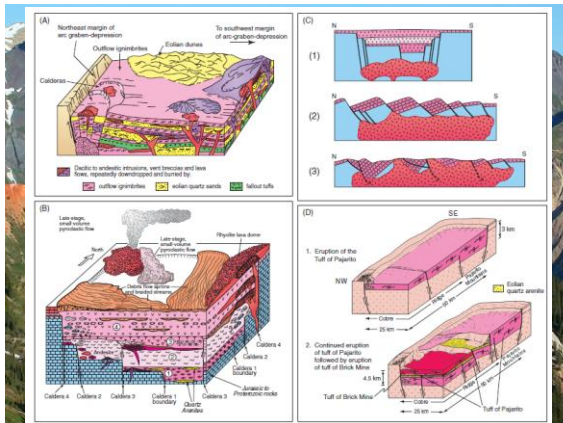
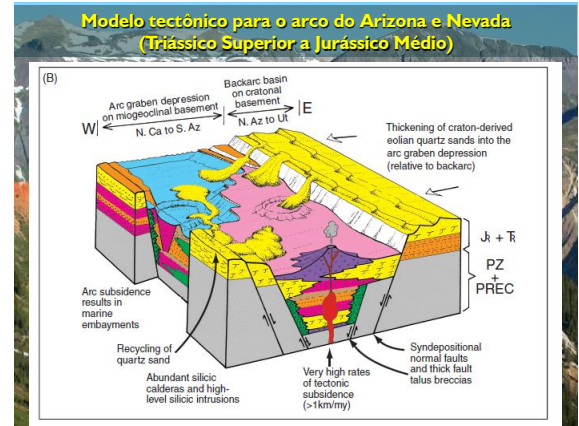
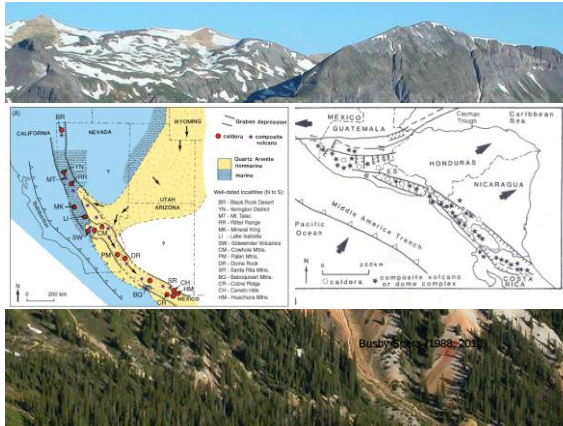
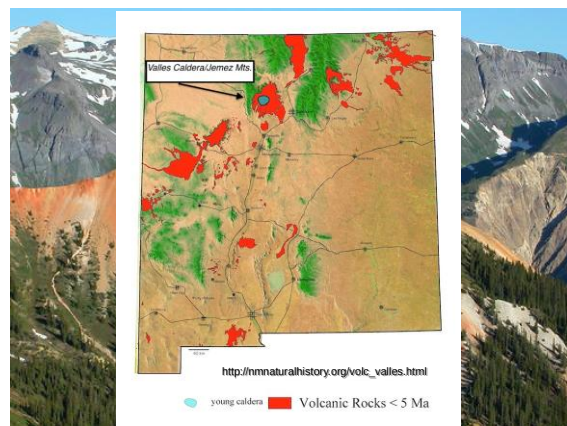
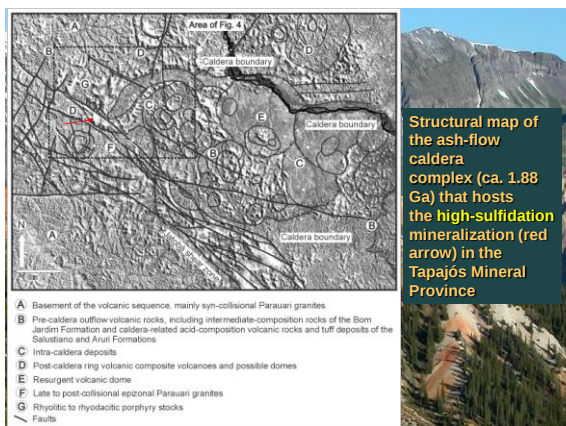
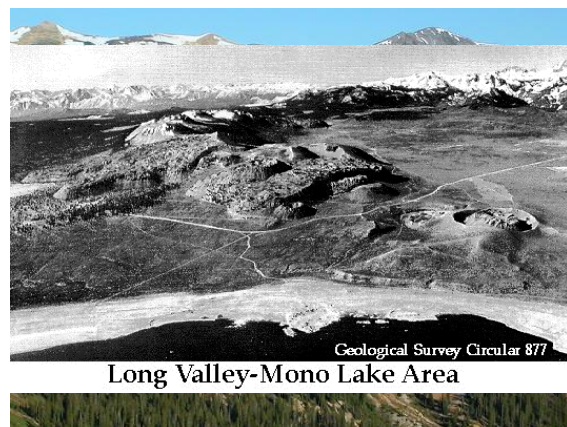
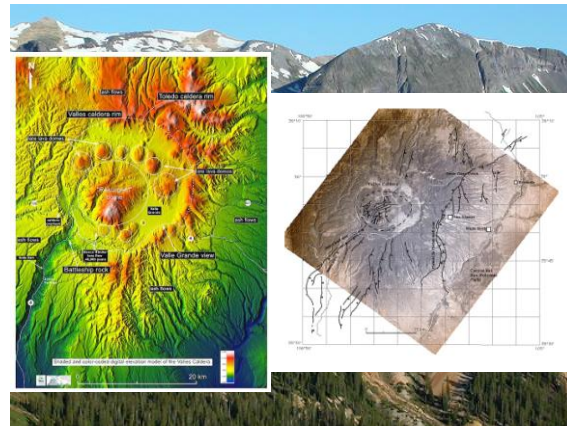
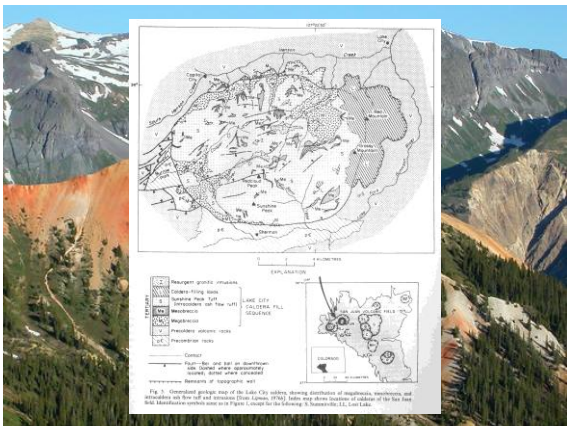
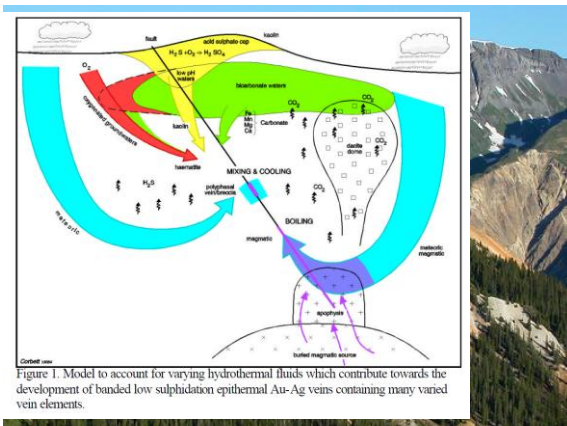
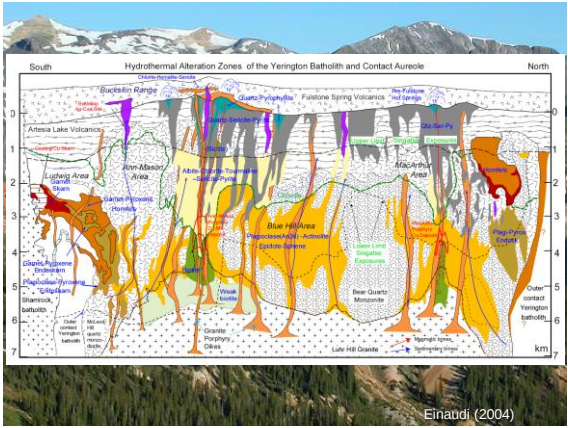


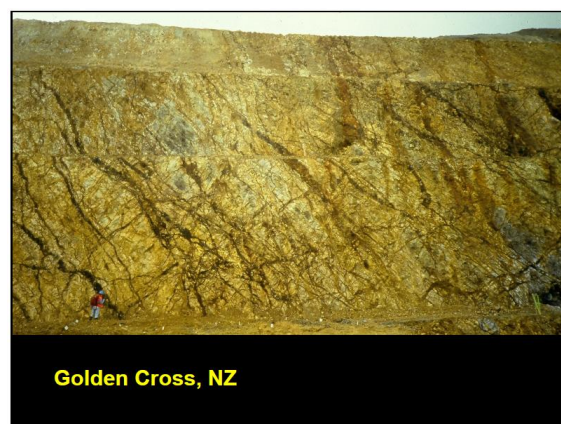
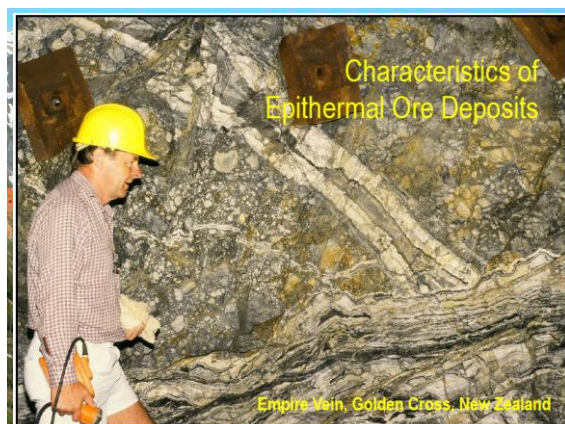
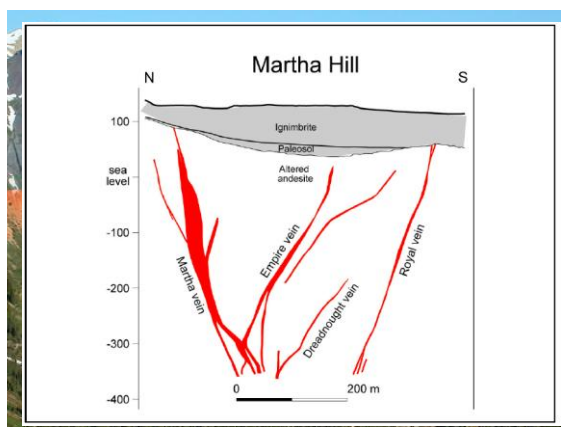
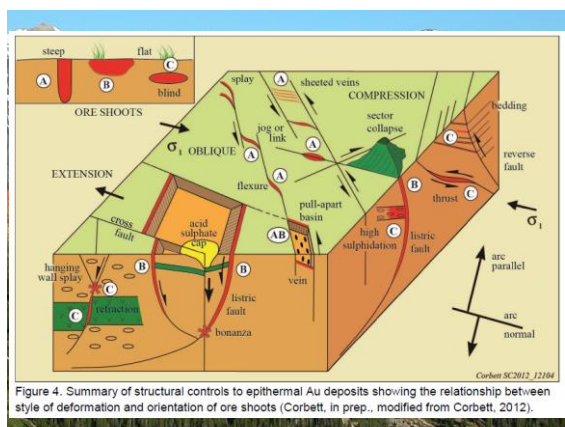
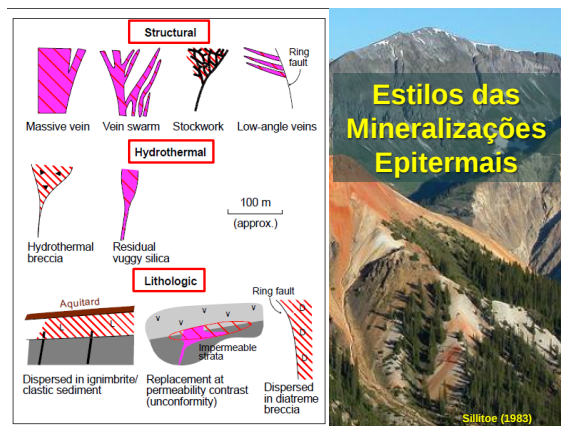
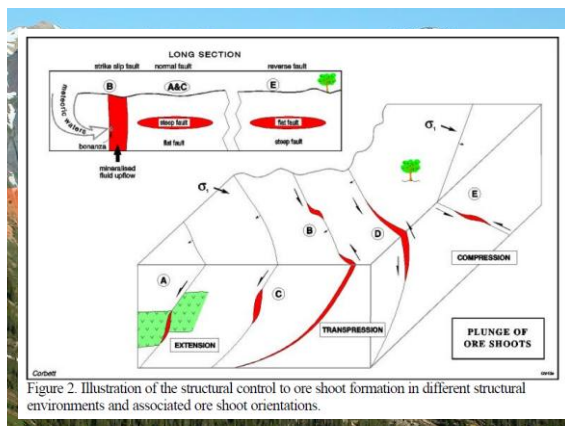
Figure 8. Summary of typical contrasting alteration patterns and fluid processes related to low sulphidation and high sulphidation epithermal systems (from White and Hedenquist, 1995).

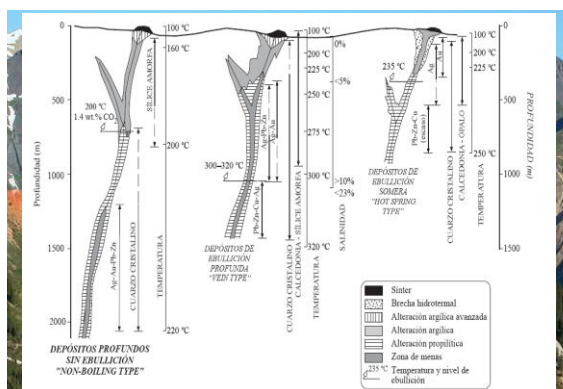
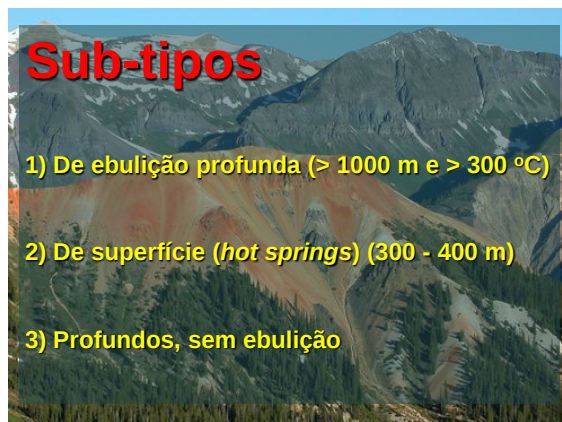
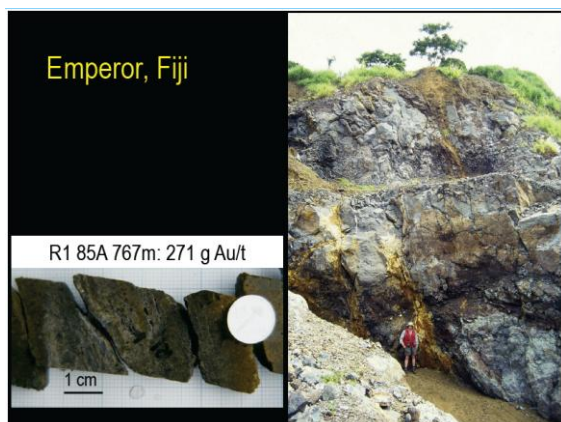




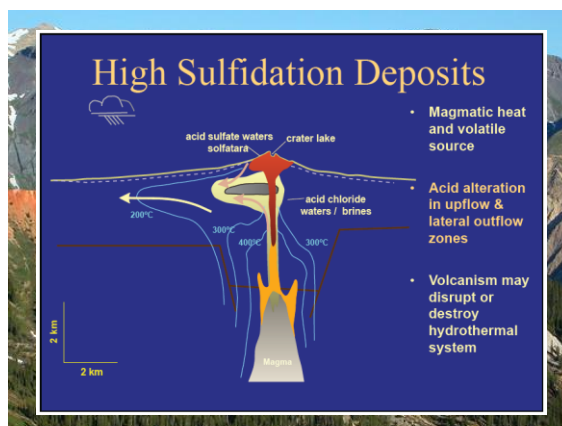
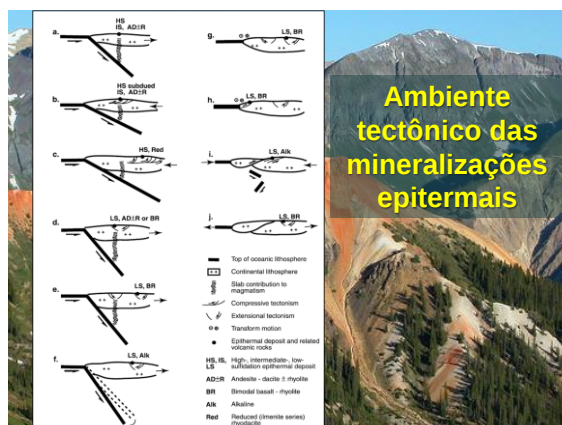


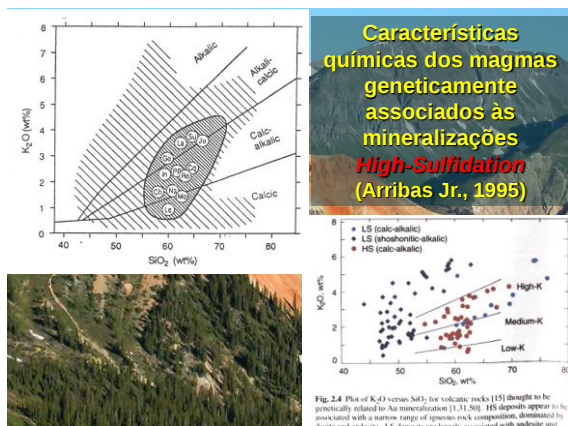






Tipos de depósitos epitermais alcalinos/intermedios (baja e intermedia sulfuração) presentes em México, según su profundidad de formación relacionado al inicio del proceso de ebullición de fluidos sodio-clorados ascendentes (simplificado de Albinsen et al., 2001).





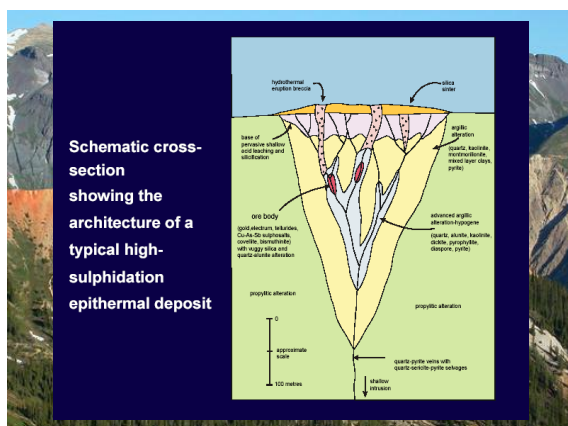
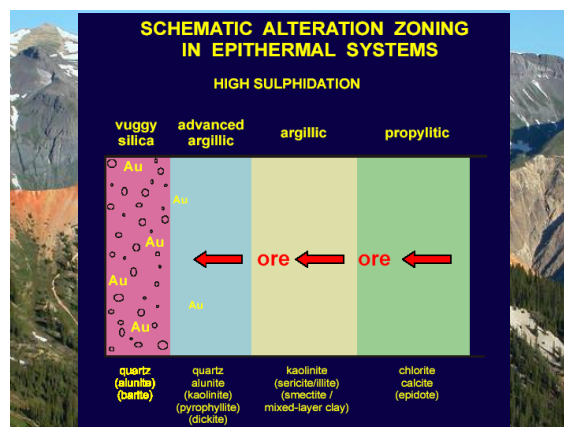
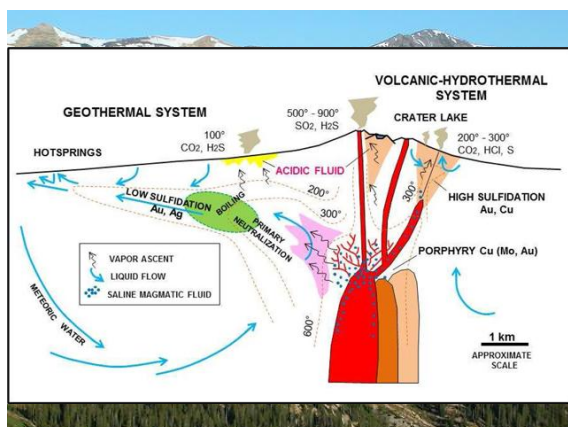
Nos sistemas *high-sulfidation* os vapores ricos em SO_2 e HCl são absorvidos pelas águas superficiais, fazendo com que haja a desproporcionalização em H_2SO_4 e H_2S .

Esse fluido ácido ($\text{pH} = 0$ a 2) reage extensivamente com as rochas em profundidades baixas e as lixivia intensamente, resultando em **alteração argílica avançada** (caolinita, dickita, pirofilita, diásporo e alunita). Essa zona passa externamente para **alterações argílicas** com menor volume de argilas e com aumento proporcional da illita ou da esmectita e essa passa para zona de **alteração propílica** (albita, calcita, clorita e epidoto) e **clorítica**.

Ametista é comum.

O fluido ácido é progressivamente neutralizado pelas reações com as rochas hospedeiras em condutos.

White & Hedenquist (1995)



No topo do sistemas podem ser formados:

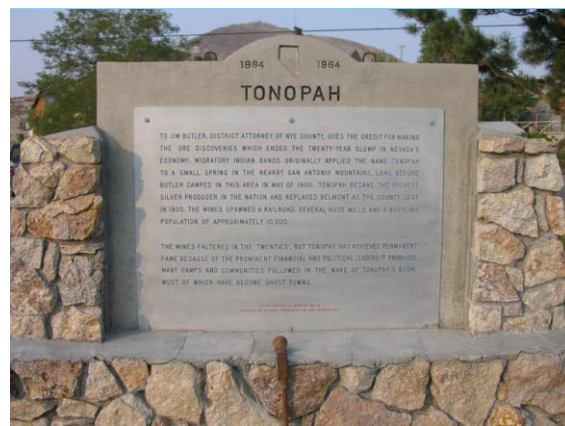
- Silica cap, com **vuggy silica**
- Corpos de argila
- Depósitos de **sinter** (especialmente nos sistemas **low-sulfidation**)

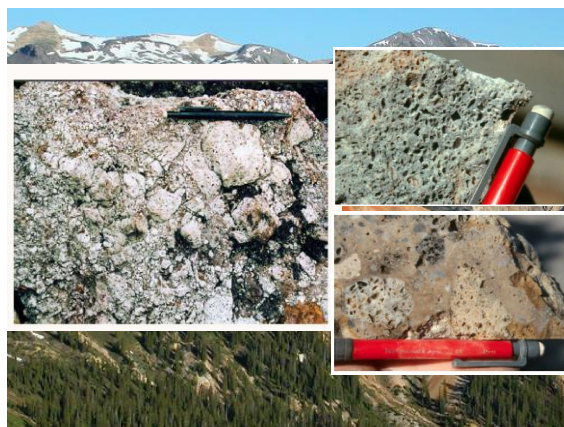
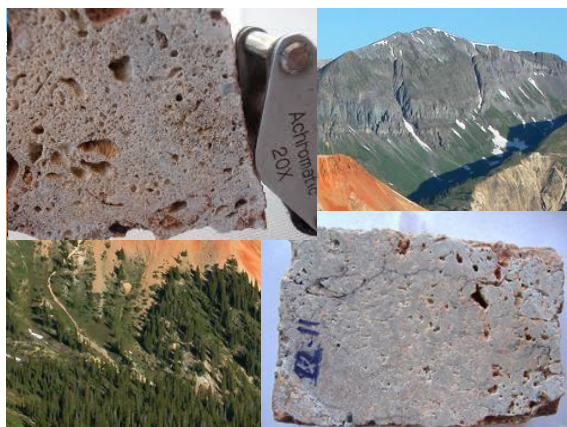
No núcleo do sistemas hidrotermais, principalmente nas zonas de falhas se forma veios bandados e laminados e zonas silicificadas, com **silica maciça**.

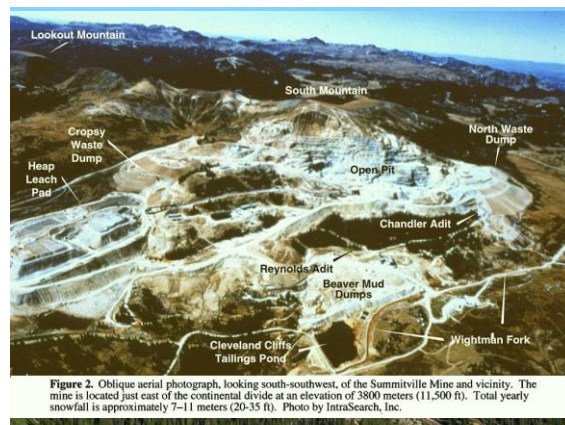
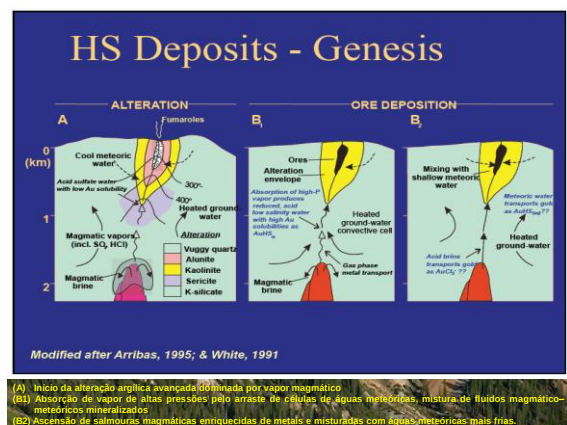
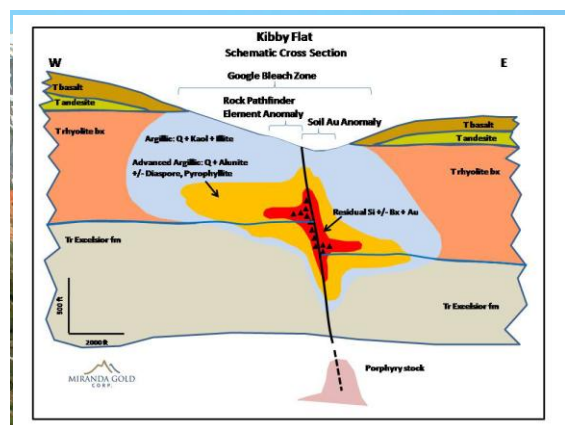
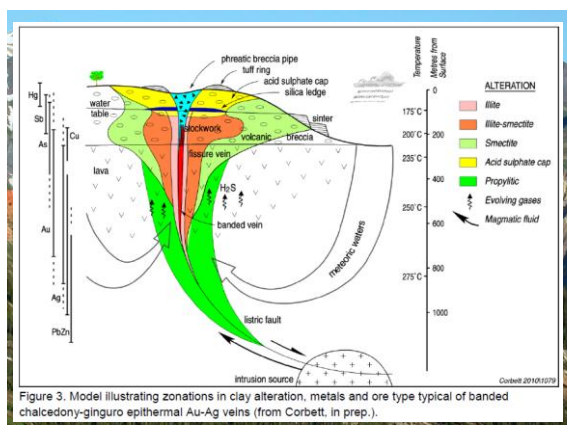
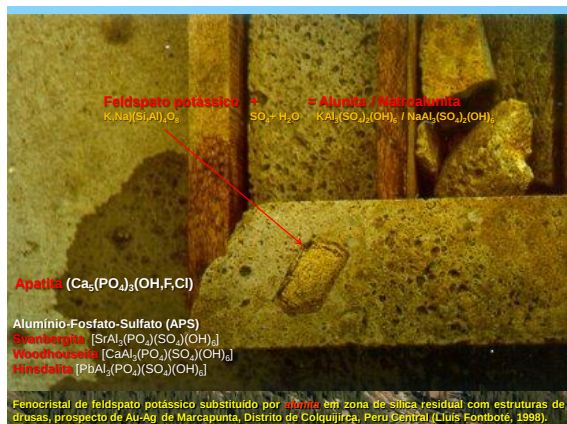
A mineralização ocorre como **vênulas de sulfetos** no núcleo silicificado ou em **sistemas de veios de quartzo microscristalino** ou de calcedônia.

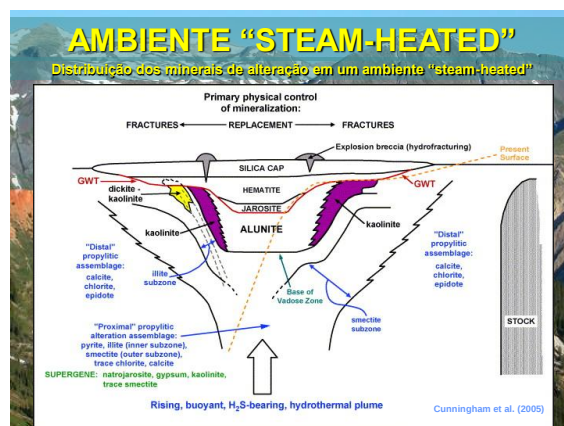
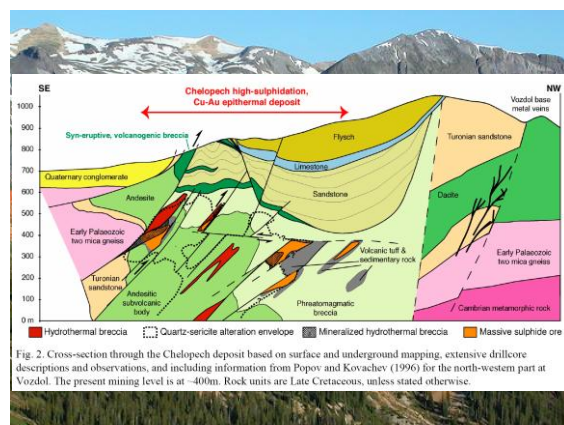
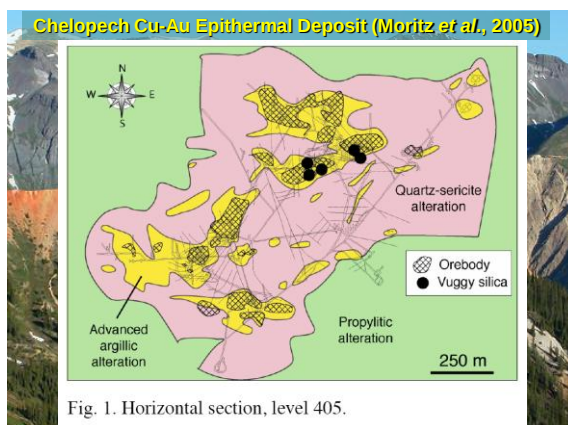
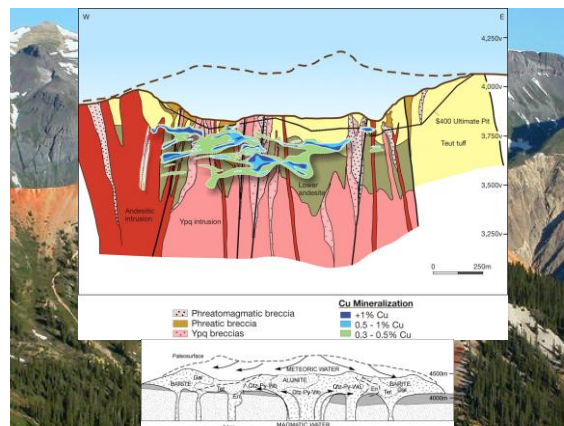
Calcita plácide e calcita plácide substituída por sílica.

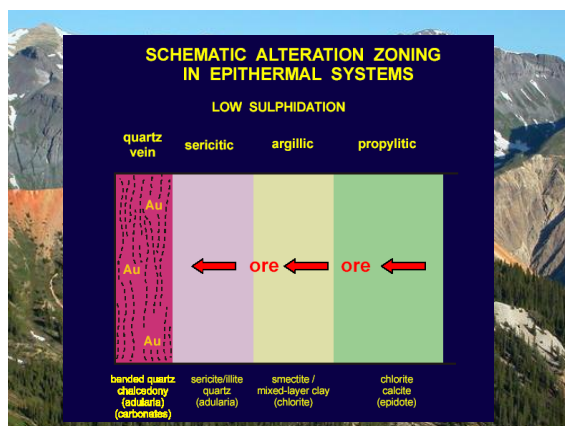
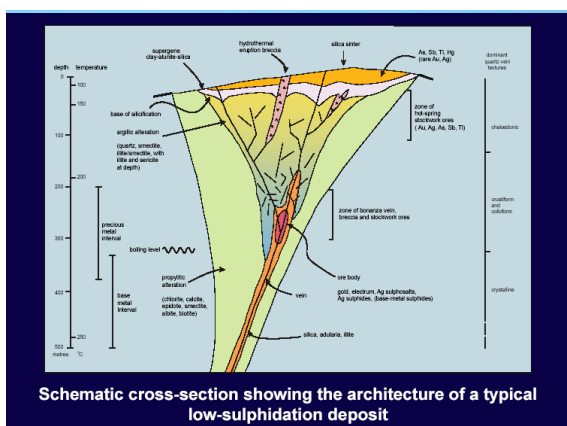
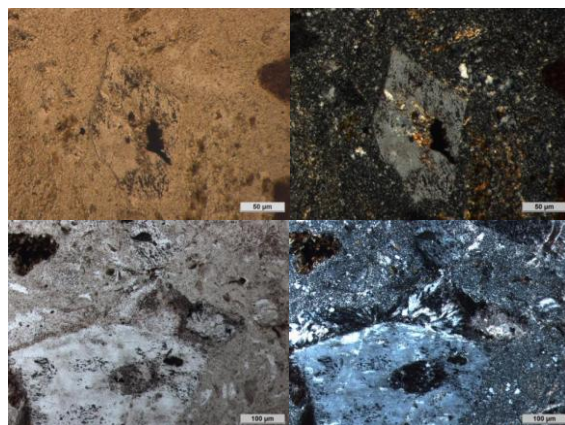
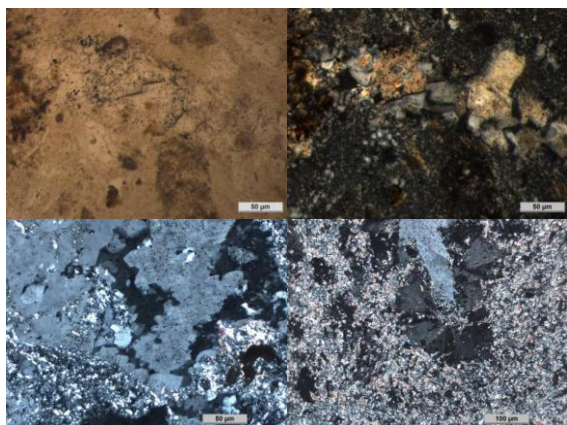
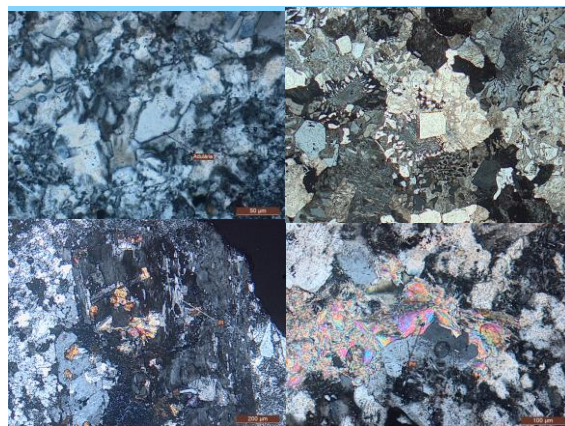
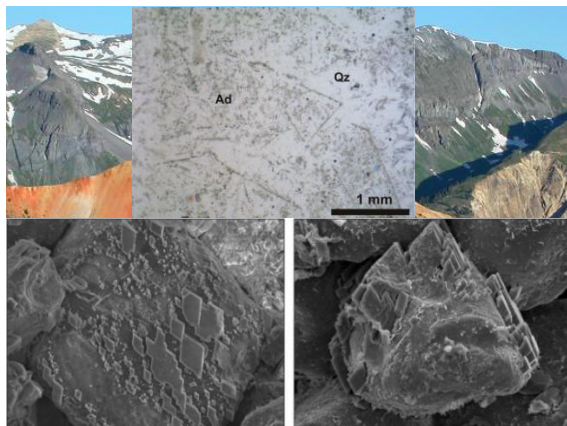
White & Hedenquist (1995)





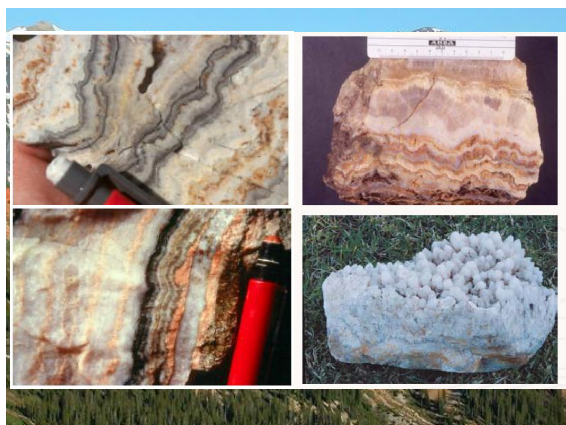
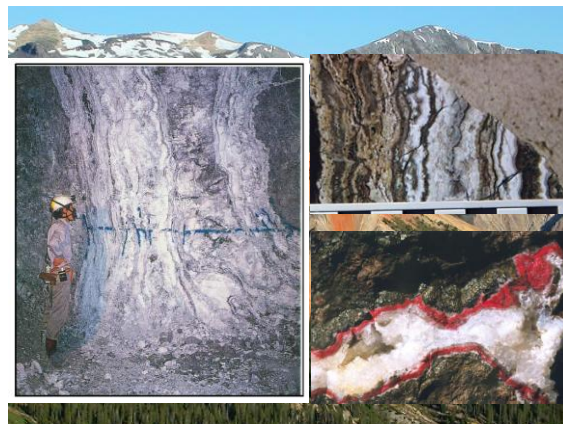






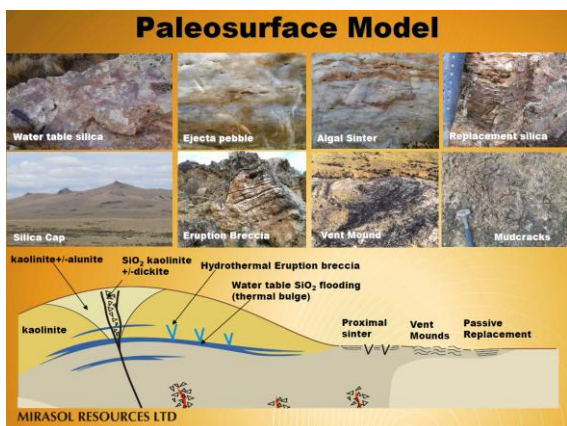
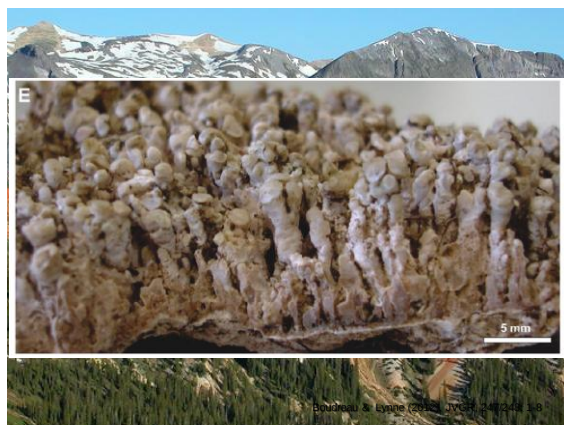
CARACTERÍSTICAS DOS DEPÓSITOS DE OURO EPITERMAIS LOW-SULFIDATION

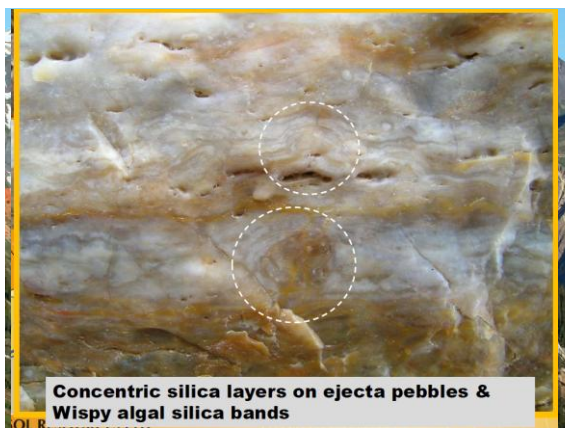
- 1) Arcos vulcano-plutônicos em margens continentais e de ilhas oceanicas, e em zonas de *back arcs*
- 2) Associam-se a sistemas de falhas regionais em níveis crustais rasos, grabens, caldeiras, estratovulcões, complexos de domos e diatremas
- 3) Hospedados em andesitos, riolacitos e riolitos
- 4) Depósitos em veios e *stockworks*, teores variando de <1 to >100 g/t e ouro muito fino (não garimpável)
- 5) Depósitos controlados por falhas, zonas de falhas, litotipos permeáveis e brechas de falhas e hidrotermais
- 6) Alteração variando de silicificação (proximal), sericitita com adularia, argilica e propilitica (distal)
- 7) Quartzo com texturas *open-fill*, crustiforme, coloforme bandada, *comb*, *cockade* e *bladed*
- 8) Ganga: quartzo, chalcedônia, ametista, carbonato, adularia, barita e fluorita
- 9) Minério com electrum, ouro, pirita, prata e esfalerita, galena, tetraedrita e calcopirita subordinados
- 10) Conteúdos de metais: Au, Ag (Zn, Pb, Cu, Mo, As, Sb, Te, Se, Hg, Ba, F , Mn)
- 11) Idades principalmente terciária e quaternárias

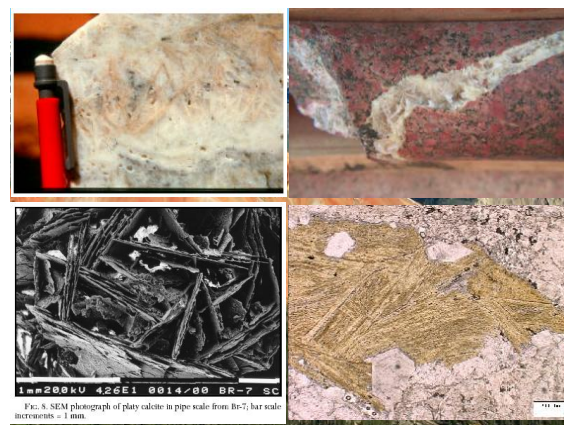
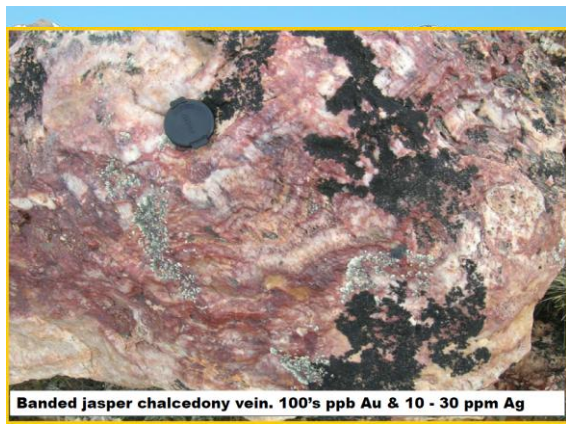


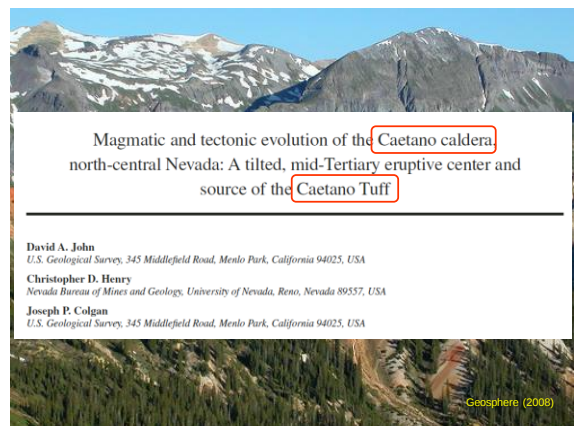
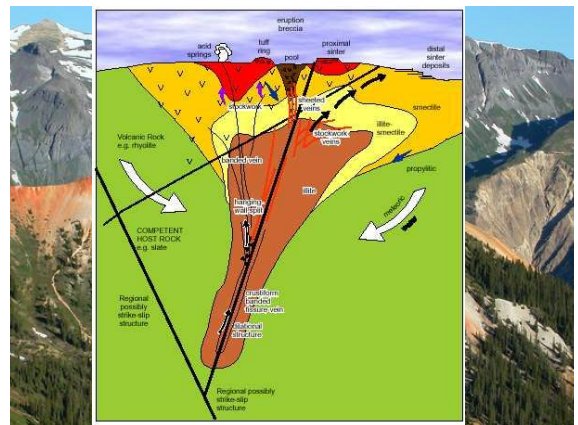
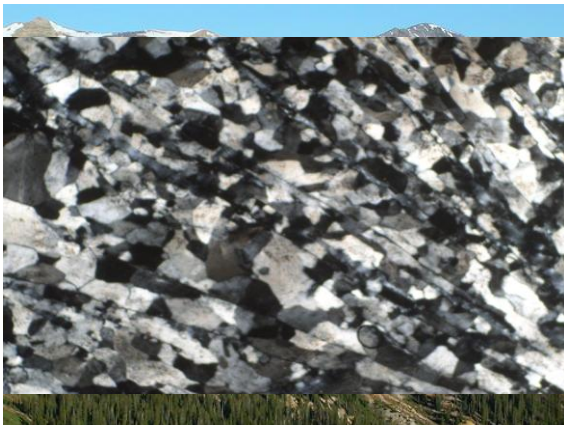
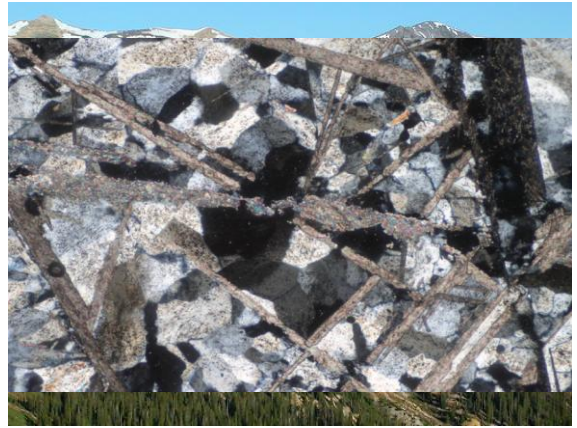
Gold-bearing quartz
stockwork in low-
sulphidation deposit,
Ecuador

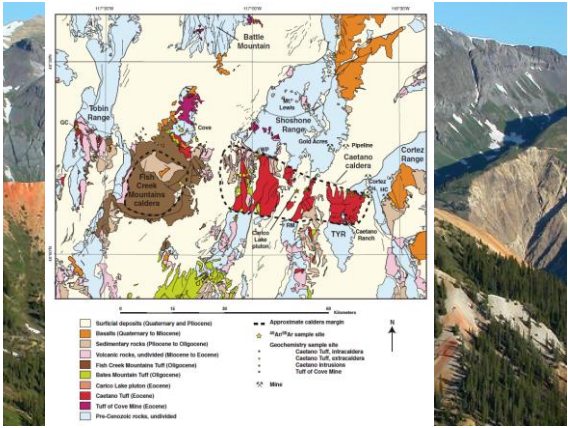
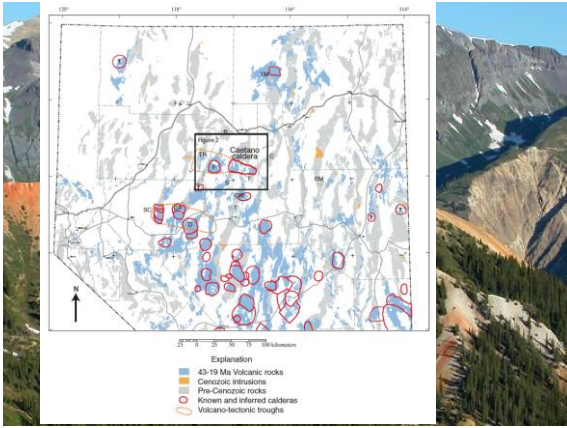






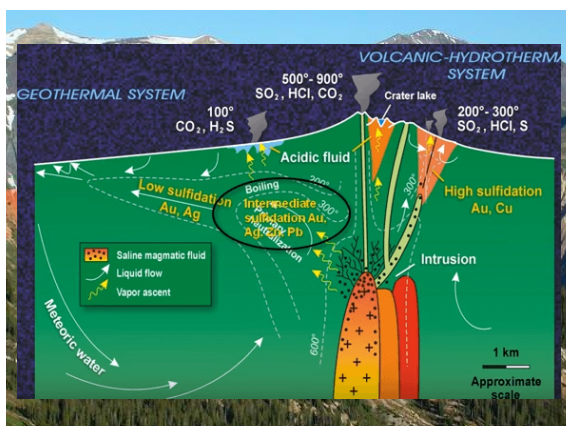


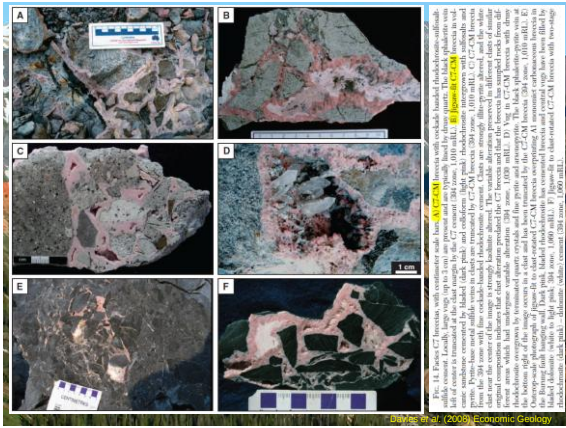












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A review of intermediate sulfidation epithermal deposits and subclassification

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