

# MAT 2020

## MÉTODOS DE ANÁLISE TÉRMICA

DSC ... Metodologia e Aplicações

DQB... 2019-2020

# PHASE EQUILIBRIA: Mixtures

**Luís M. N. B. F. Santos**

**CIQUP, Departamento de Química e Bioquímica  
Faculdade de Ciências Universidade do Porto  
Porto, Portugal**

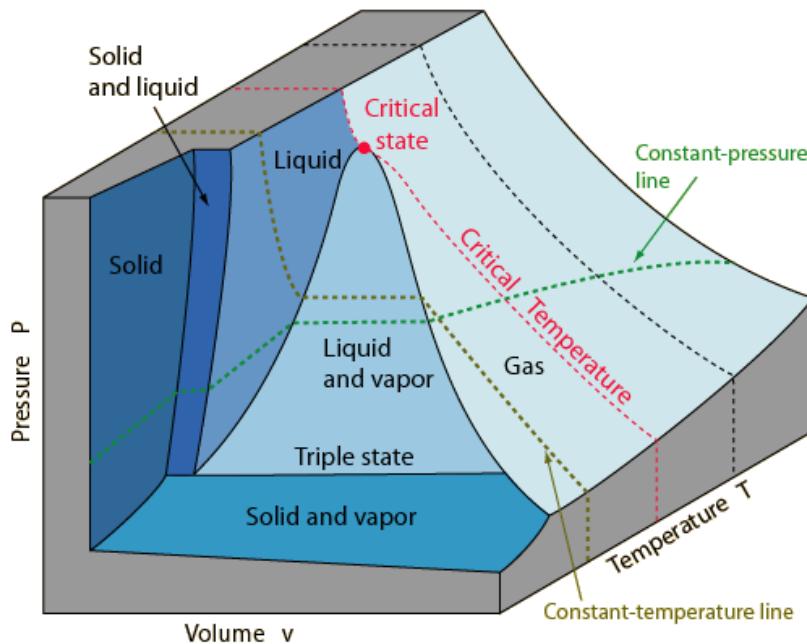
email: lbsantos@fc.up.pt

email: lbelchiorsantos@gmail.com

URL: <http://www.fc.up.pt/pessoas/lbsantos>

# Pure substances & Mixtures

## PVT diagram

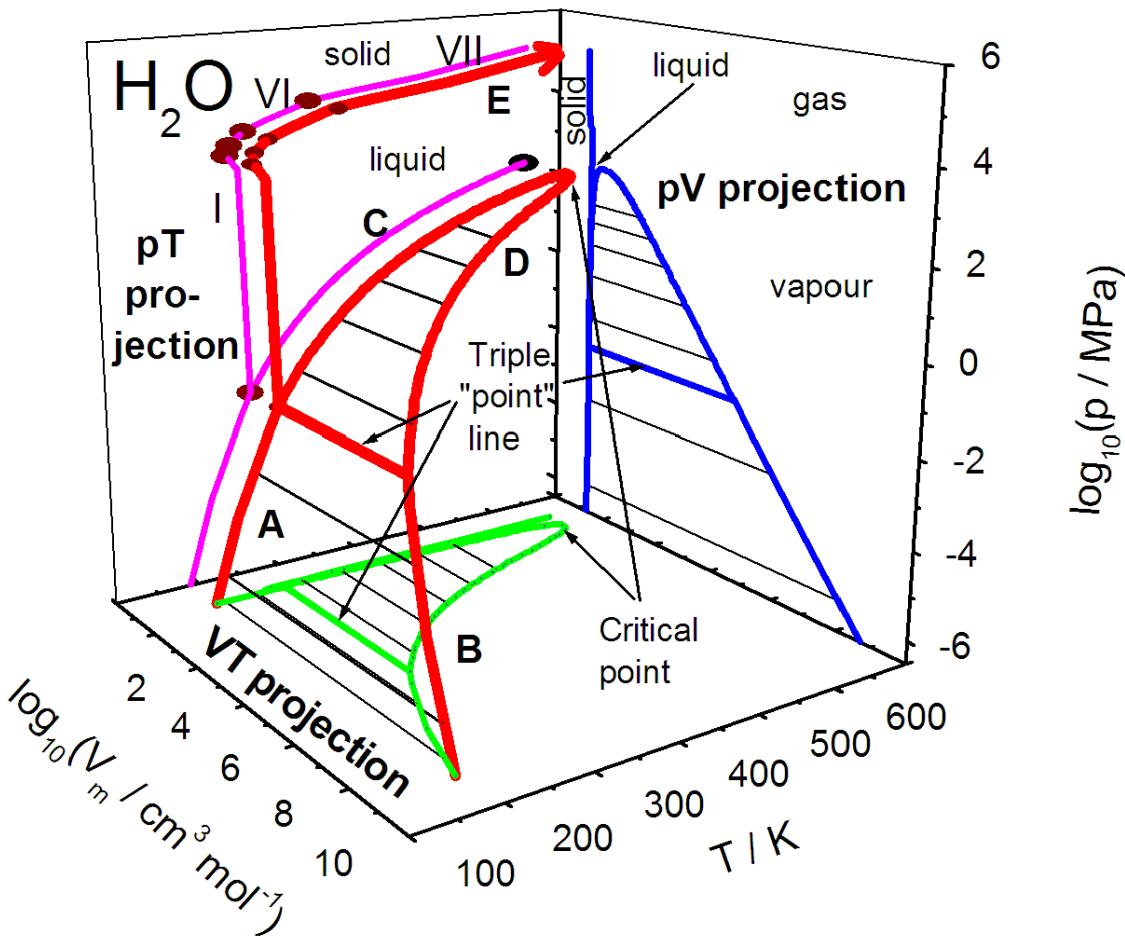


-Why each compound have a different PVT surface profile?

- C-H- $\pi$  and  $\pi-\pi$  .. interactions
- H-bond
- Electrostatic .. interactions
- Molecular shape
- ...???

# Properties of pure substances

## PVT diagram

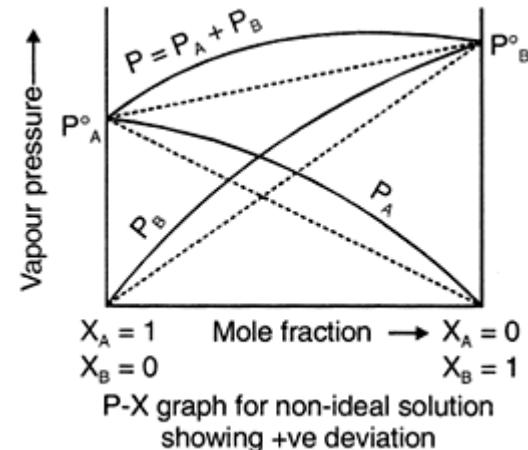
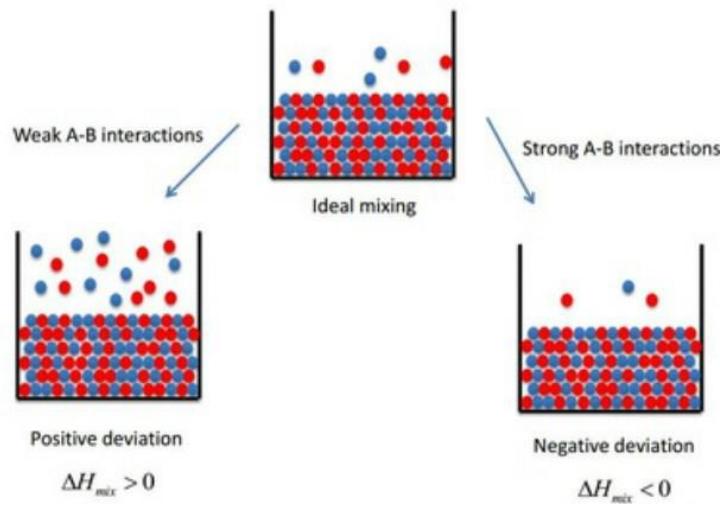


- PT projection :  
Complex and interesting profile!
- VT projection :  
Little dependence of the liquid / solid phases
- PV projection :  
Little dependence of the liquid / solid phases

# Liquid Solutions

On the basis of **Raoult's Law**, liquid-liquid solutions can be of two types.

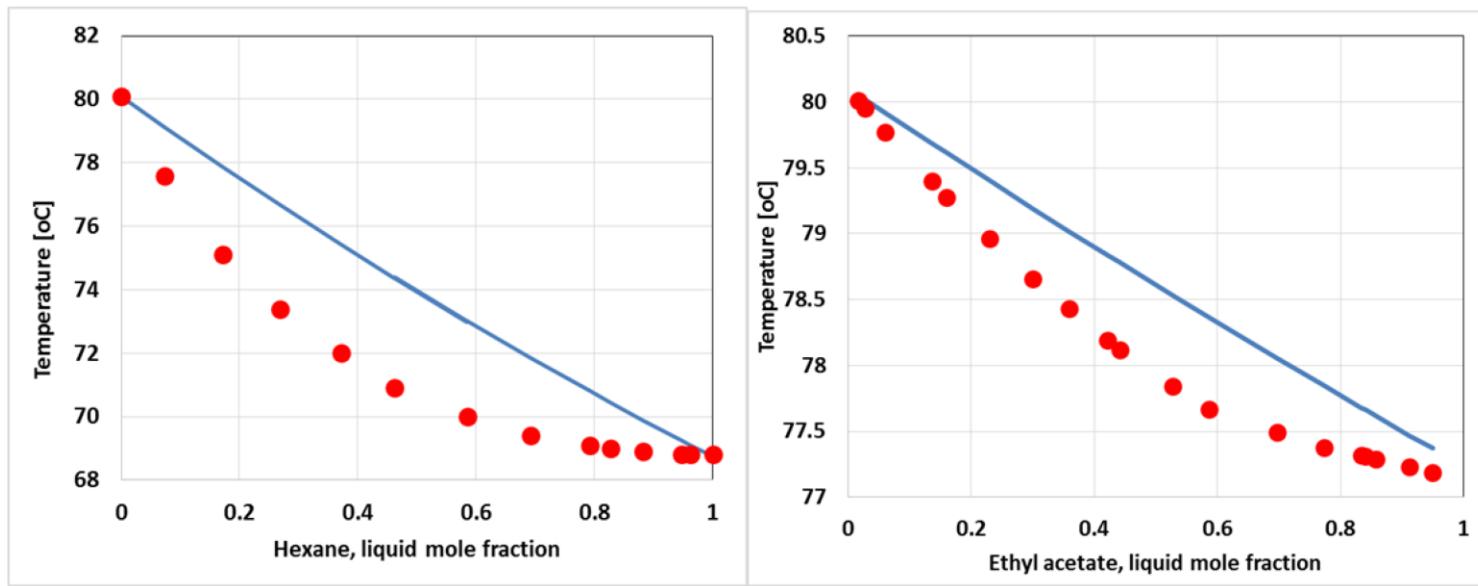
- *Ideal Solutions*
- *Non-ideal Solutions*



# Liquid Solutions

On the basis of **Raoult's Law**, liquid-liquid solutions can be of two types.

- *Ideal Solutions*
- *Non-ideal Solutions*

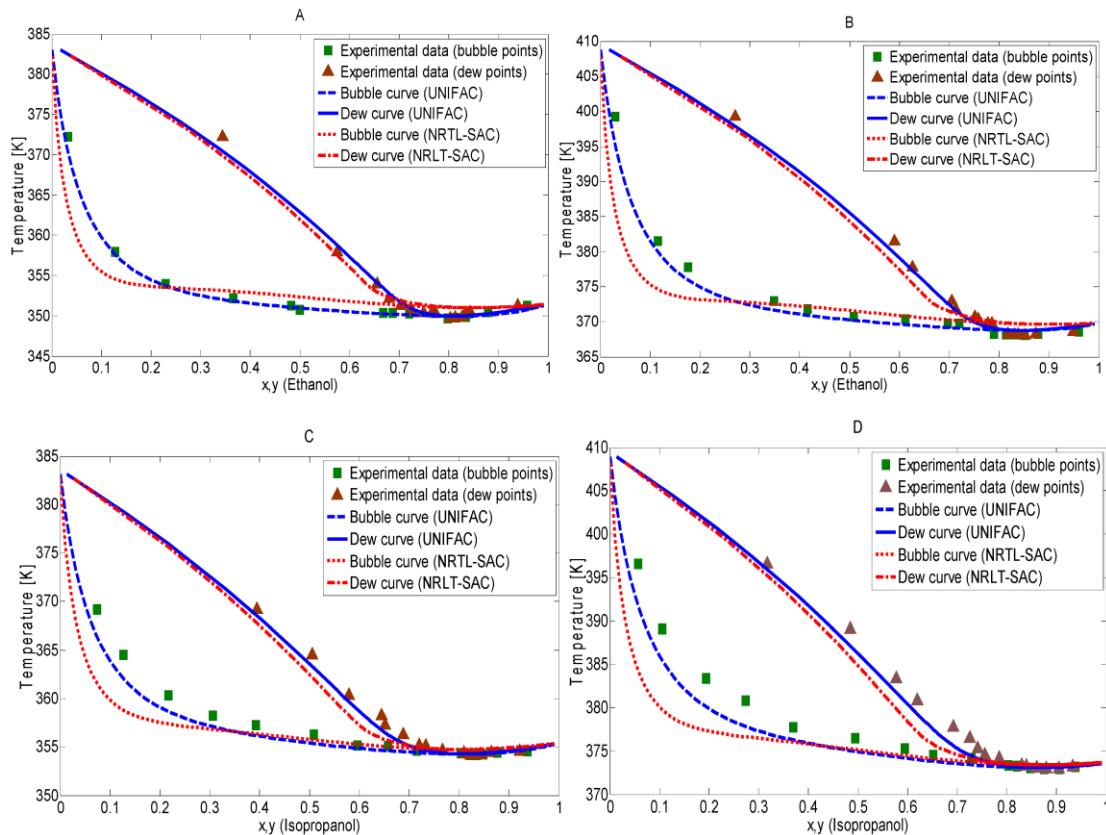


T-x-y diagram for the binary systems of hexane-benzene (left) and ethyl acetate-benzene (right) at a pressure of 101.33 kPa.

# Liquid Solutions

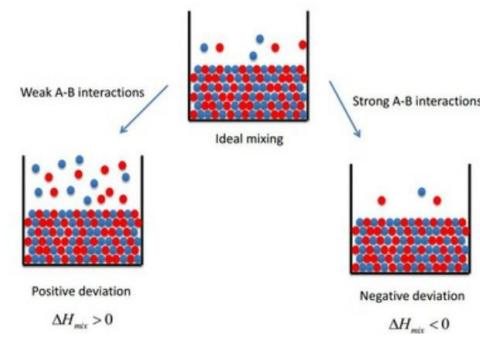
On the basis of **Raoult's Law**, liquid-liquid solutions can be of two types.

- *Ideal Solutions*
- *Non-ideal Solutions*



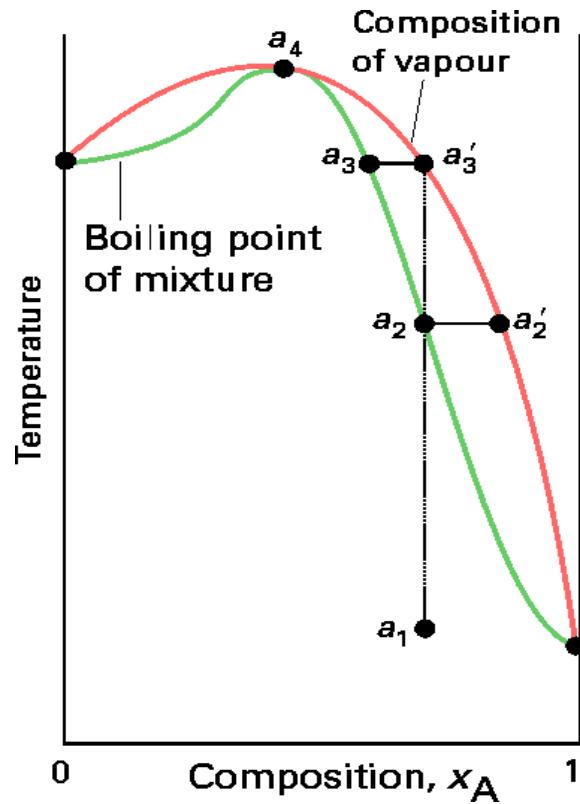
The results for the systems  
(A) **ethanol-toluene** at 101.3 kPa,  
(B) **ethanol-toluene** at 201.3 kPa,  
(c) **isopropyl alcohol-toluene** at 101.3 kPa,  
(D) **isopropyl alcohol-toluene** at 201.3 kPa

# Liquid Solutions

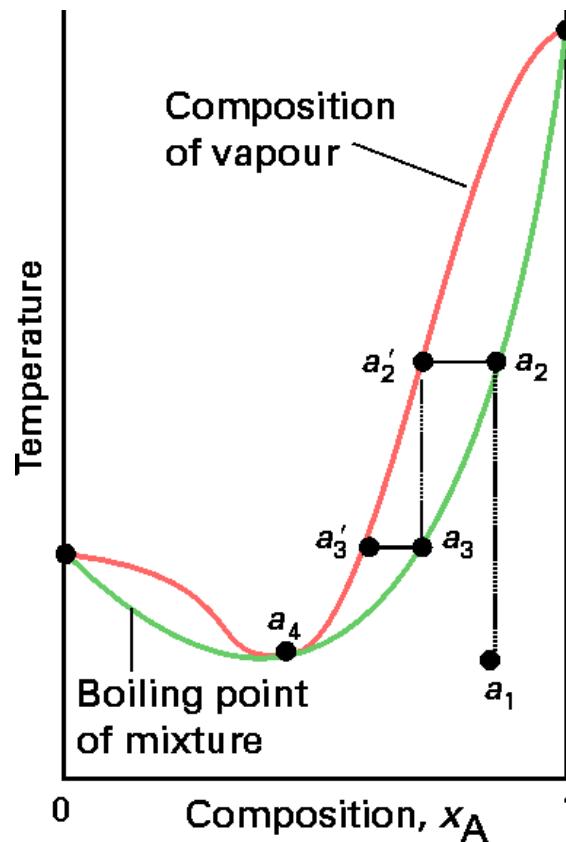


**High-boiling azeotrope,**  
e.g., nitric acid/water

*p* ...fixed



**Low-boiling azeotrope,**  
e.g., ethanol/water

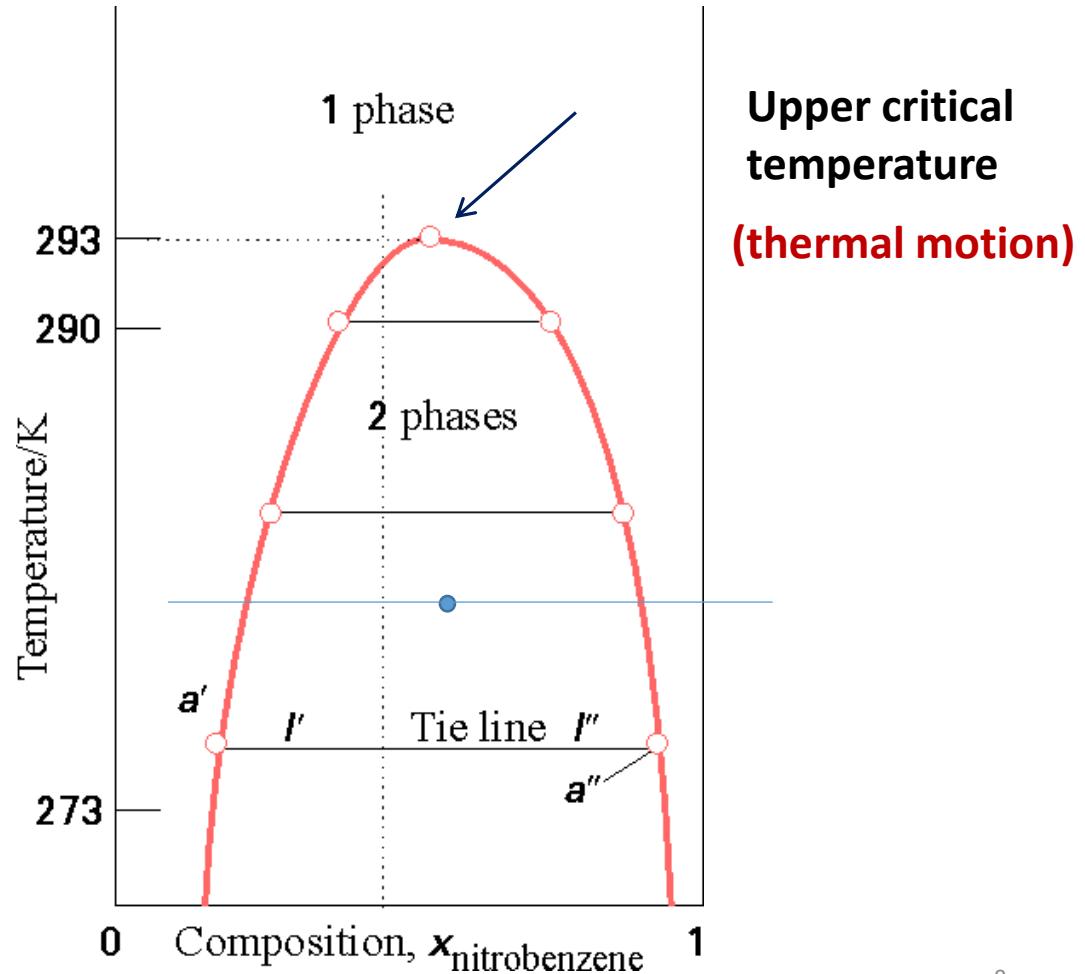
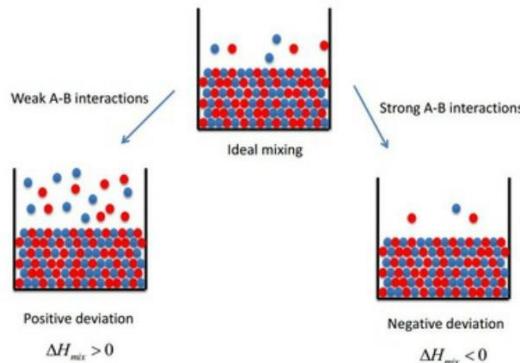


# Liquid Solutions

## Phase diagrams

### Phase separation

E.g., hexane/nitrobenzene:

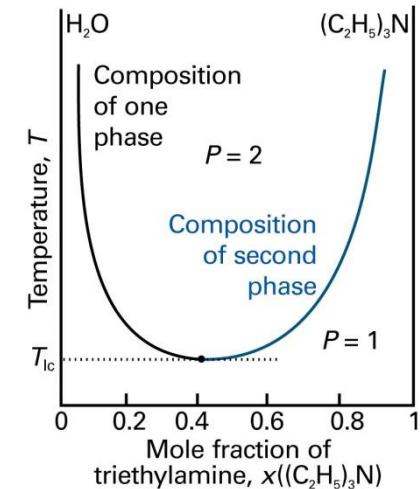
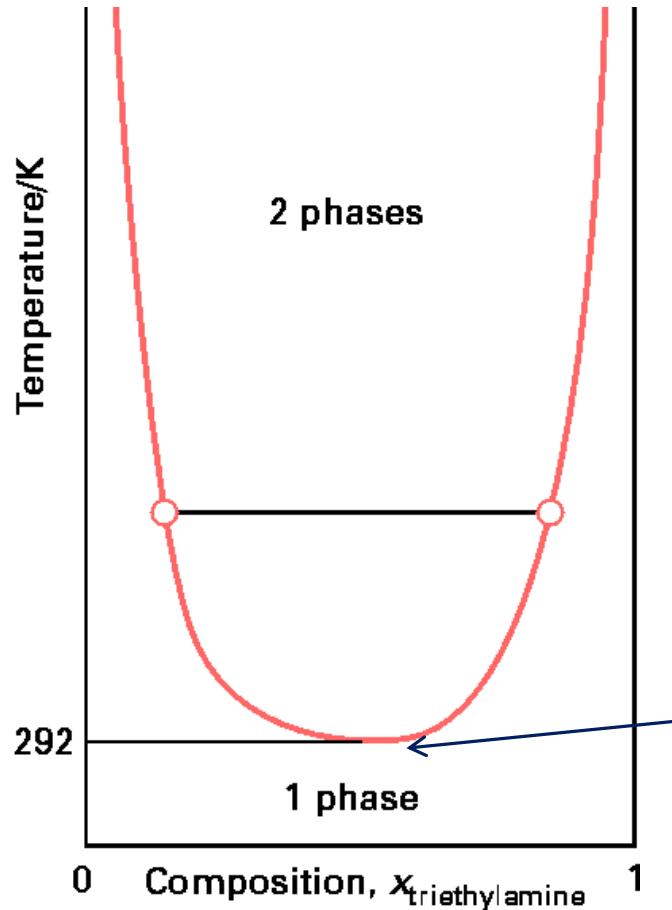
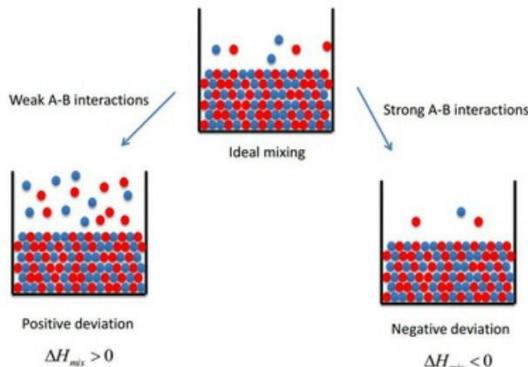


# Liquid Solutions

## Phase diagrams

### Phase separation

E.g., triethylamine /water:

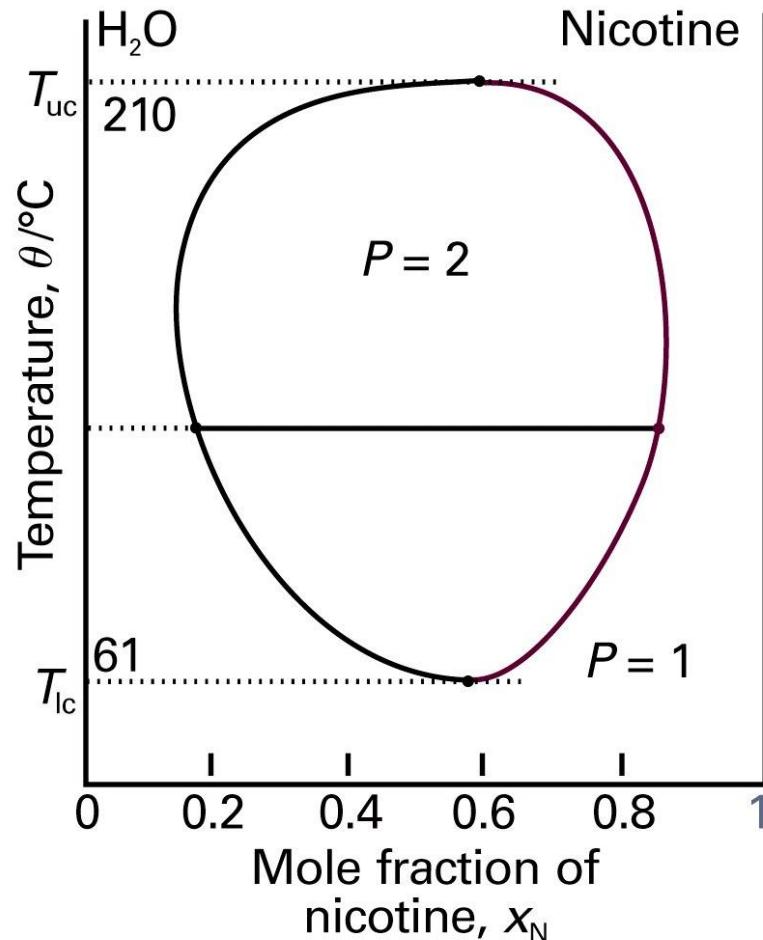


Lower  $T_{\text{critical}}$   
(complex formed)

# Liquid Solutions

## Phase diagrams

### Phase separation



Upper critical temperature  
(thermal motion)

Lower  $T_{critical}$   
(complex formed)

# Liquid Solutions

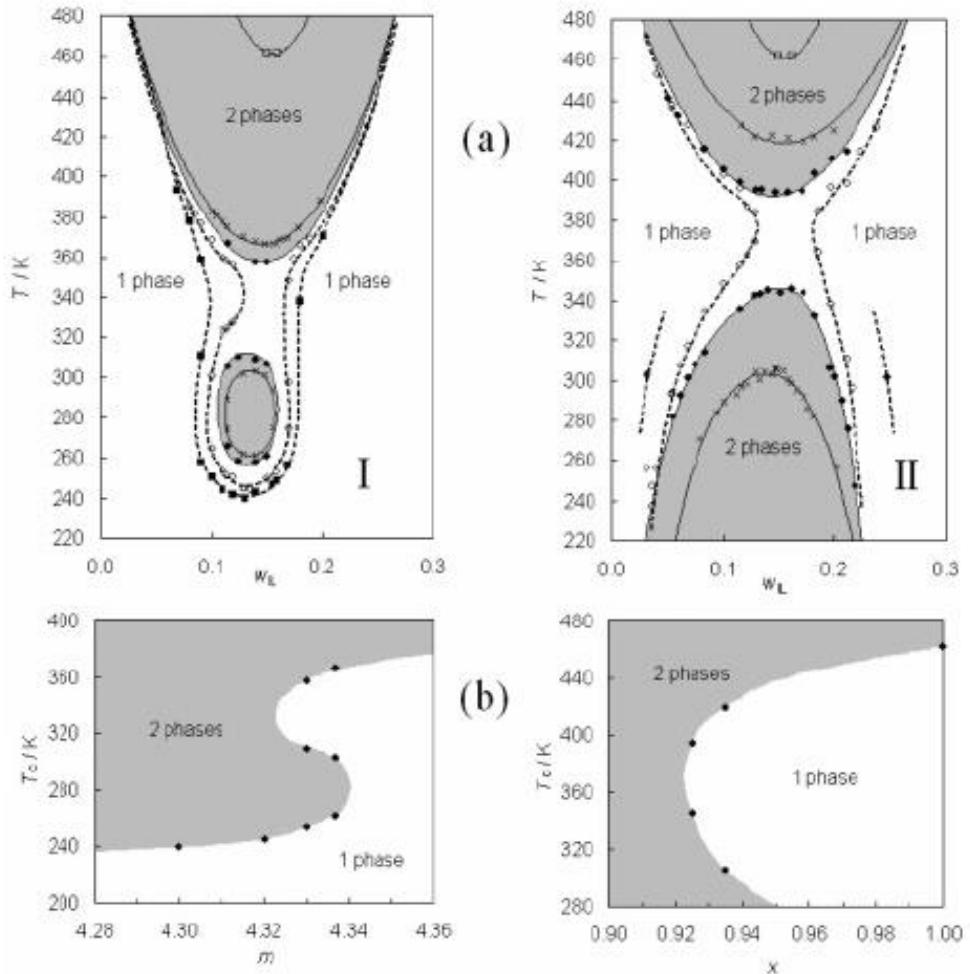
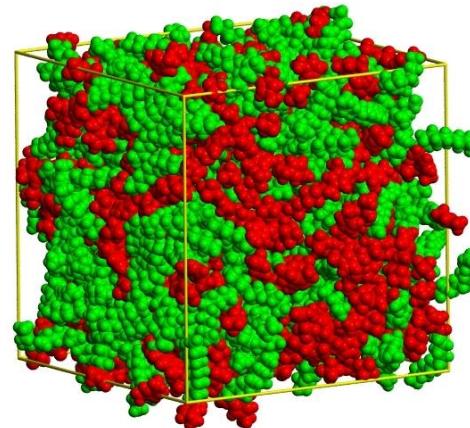


Figure 1. Phase equilibria of  $[C_m\text{mim}][\text{NTf}_2] + \text{CH}_x\text{Cl}_{4-x}$ ;  $w_{\text{IL}}$  = weight



Two domains

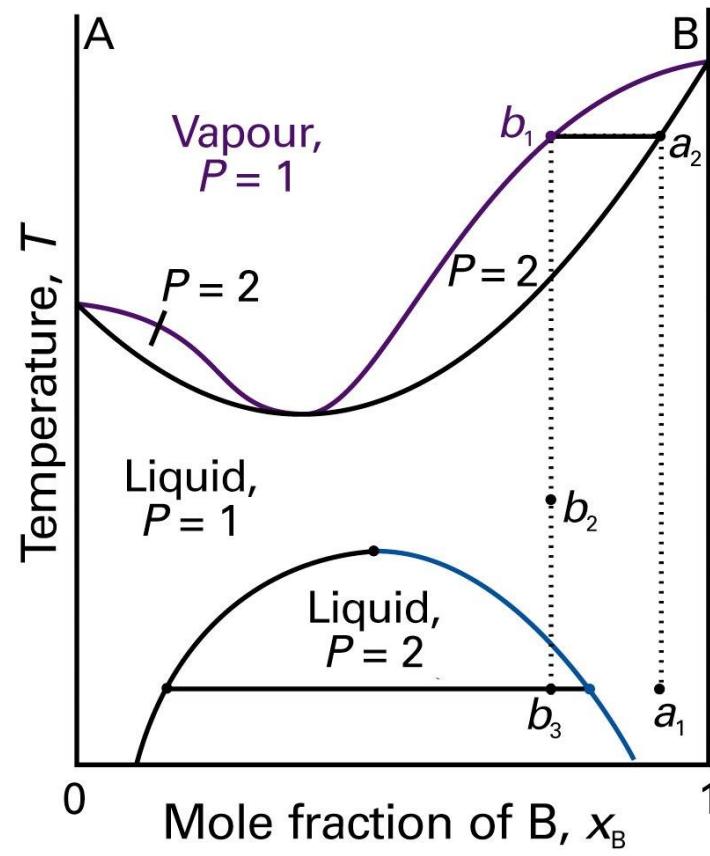
Evidence for Lower Critical Solution Behavior in Ionic Liquid Solutions

Joanna Lachwa, Jerzy Szydłowski, Vesna Nadjadovć-Vlašak, Luis P. N. Rebelo,\*  
Kenneth R. Seddon,\* Manuel Nunes da Ponte, José M. S. S. Esperança, and  
Henrique J. R. Guedes  
Instituto de Tecnologia Química e Biomédica, UNL, Av. República, Apartado 127, 2280-901 Oeiras, Portugal, and  
The QUILL Centre, The Queen's University of Belfast, Stranmillis Road, Belfast BT9 5AG, U.K.

# Liquid Solutions

Low-boiling azeotrope

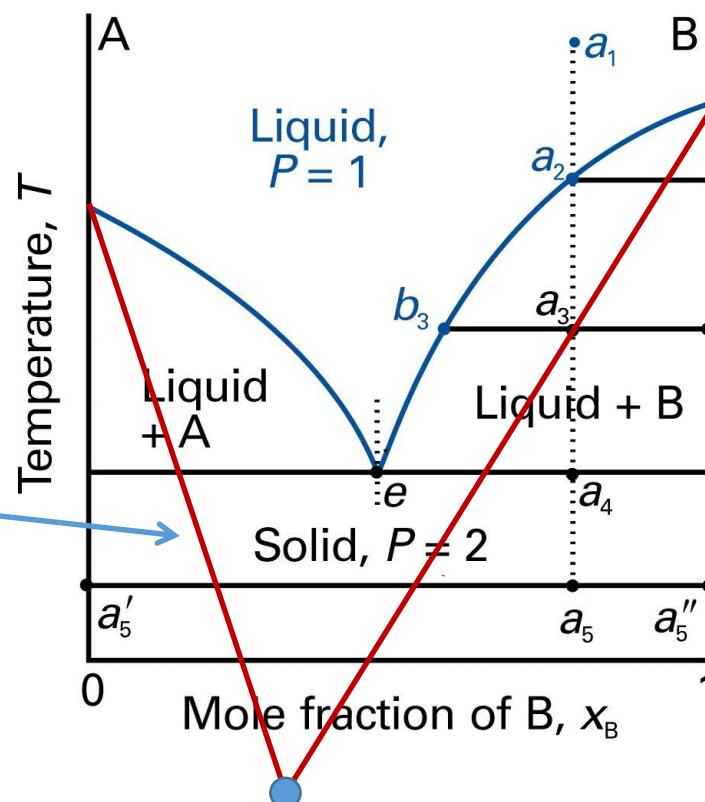
Upper critical  
temperature  
**(thermal motion)**



# Liquid ..eutetic

Deep eutetic

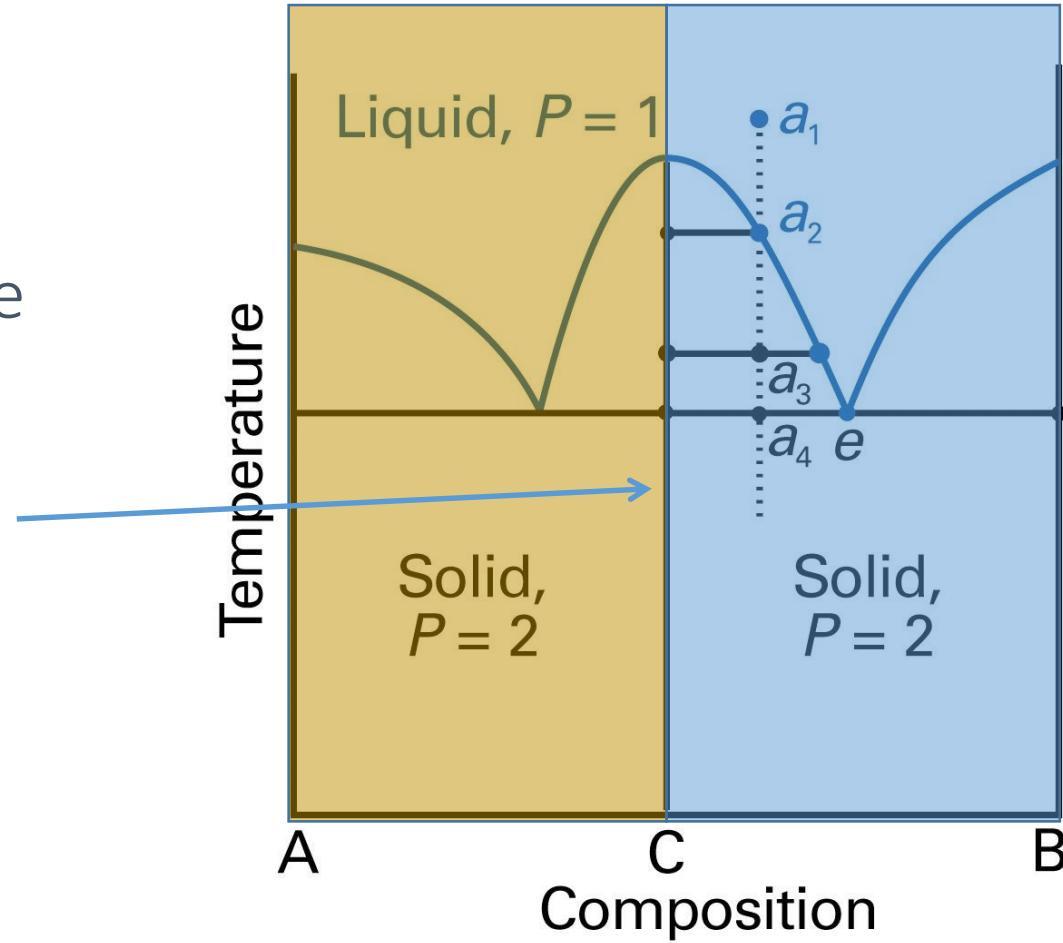
Strong Liquid  
solution stabilization



# Liquid Solutions

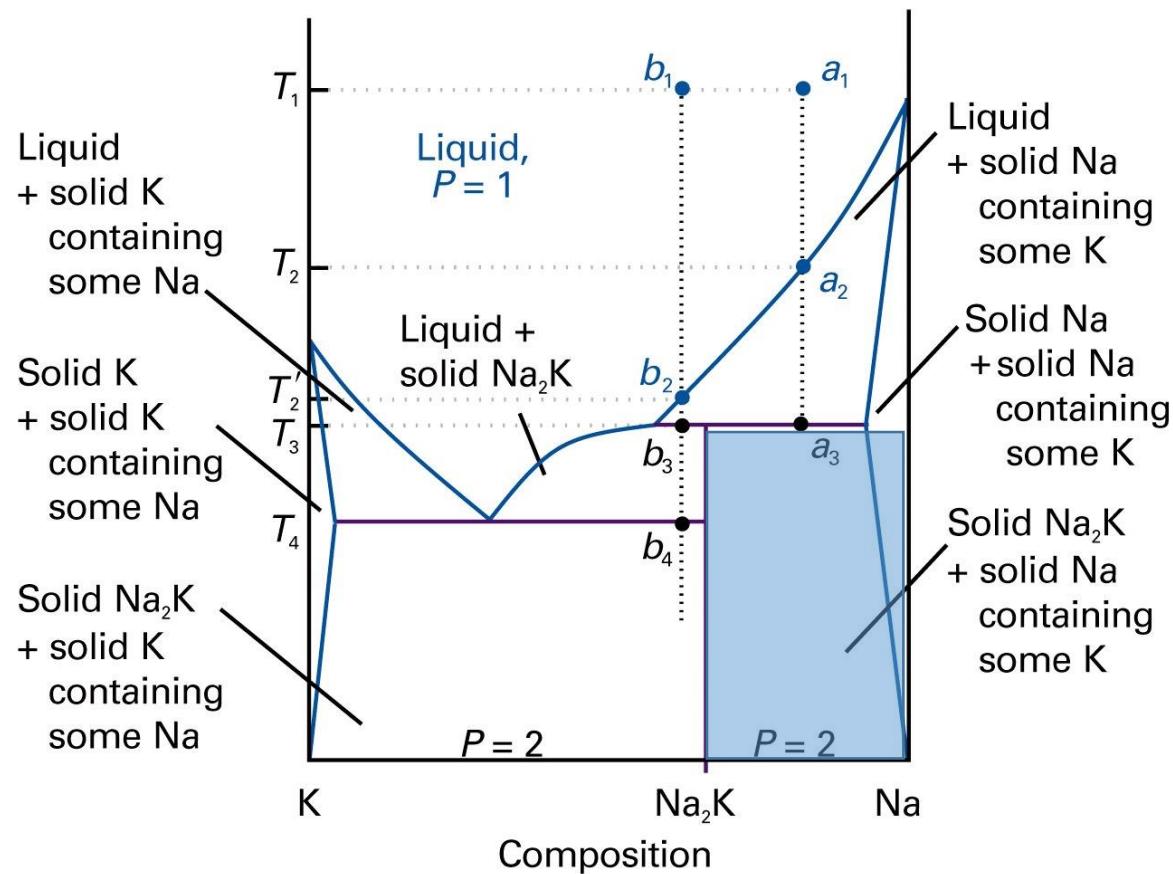
Congruente  
melting

Solid C  
stabilization



# Liquid Solutions

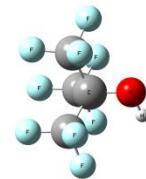
## Phase diagrams



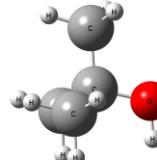
# Liquid Solutions

## Phase diagrams

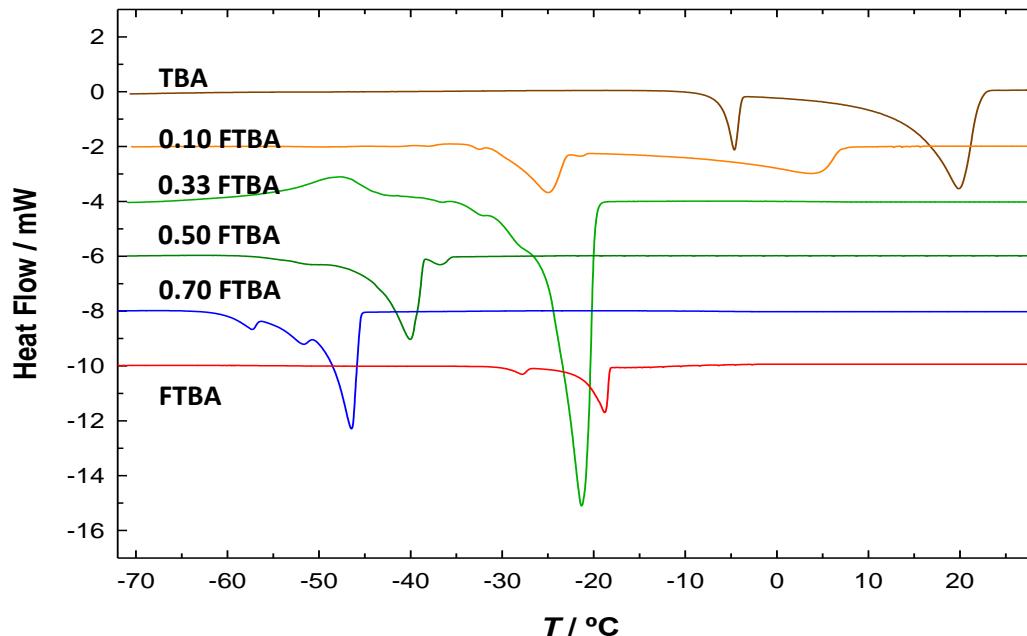
DSC....



FTBA



TBA



R-OH ..acidity

H-bond interaction

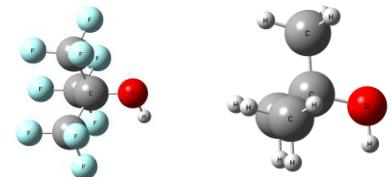
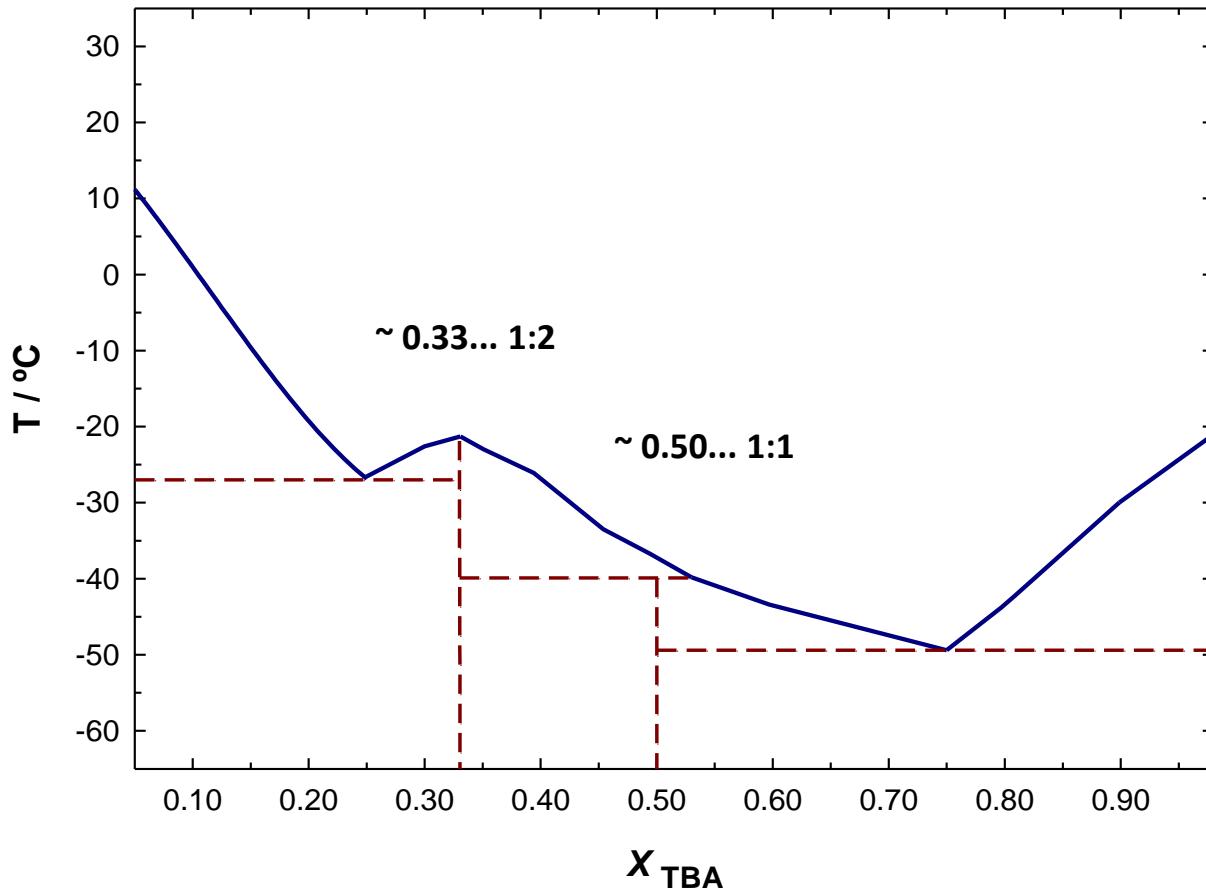
Steric hindrance

.....

# Liquid Solutions

## Phase diagrams

DSC.... Solid – Liquid Equilibrium



FTBA      TBA

R-OH .. acidity  
H-bond interaction  
Steric hindrance

.....

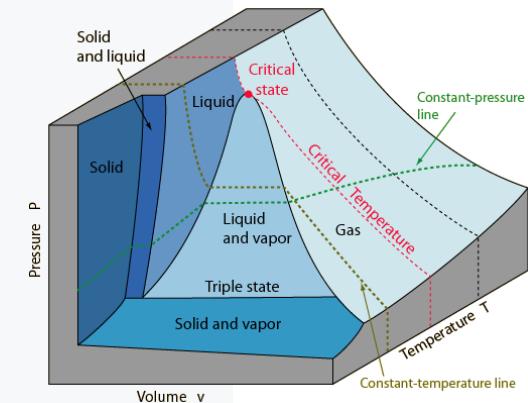
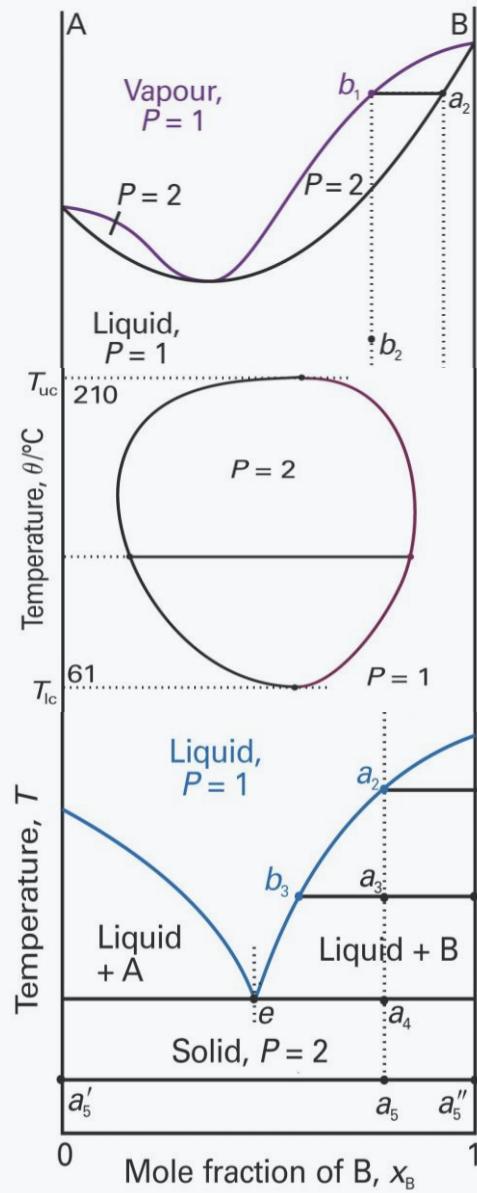
Liquid ..vapor



Liquid ..Liquid



Solid ..Liquid



All together