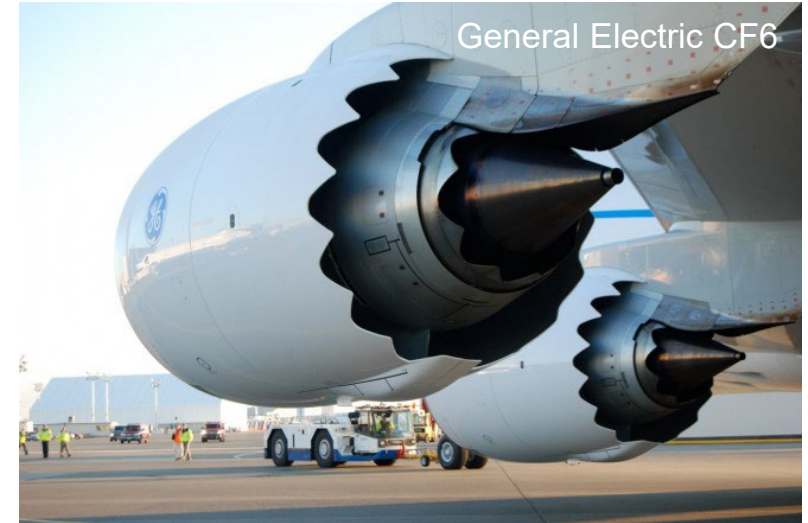


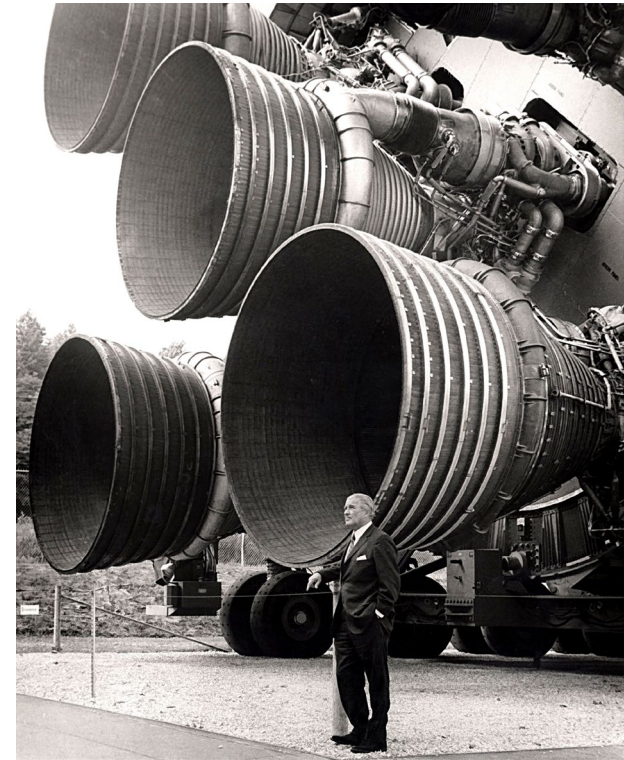
EQUAÇÕES DE BALANÇO DE ENERGIA PARA SISTEMAS ABERTOS

Paulo Seleglim Jr.
Universidade de São Paulo



Aplicação: análise termodinâmica do Boeing 747





150 GW !!!



15 ton/sec !!!





Como um foguete pode acelerar no espaço - a terceira lei de Newton

Prof. P. Seleglim



Minha Equação Malvada Favorita (Tsiolkovsky) - análise do lançamento da missão MARS2020 Perseverance

Prof. P. Seleglim



Entendendo o Estagiamento de um Foguete

Prof. P. Seleglim



Entendendo a pilotagem de um foguete - análise da telemetria do lançamento SpaceX

Prof. P. Seleglim



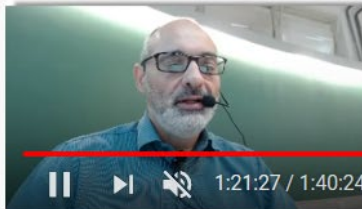
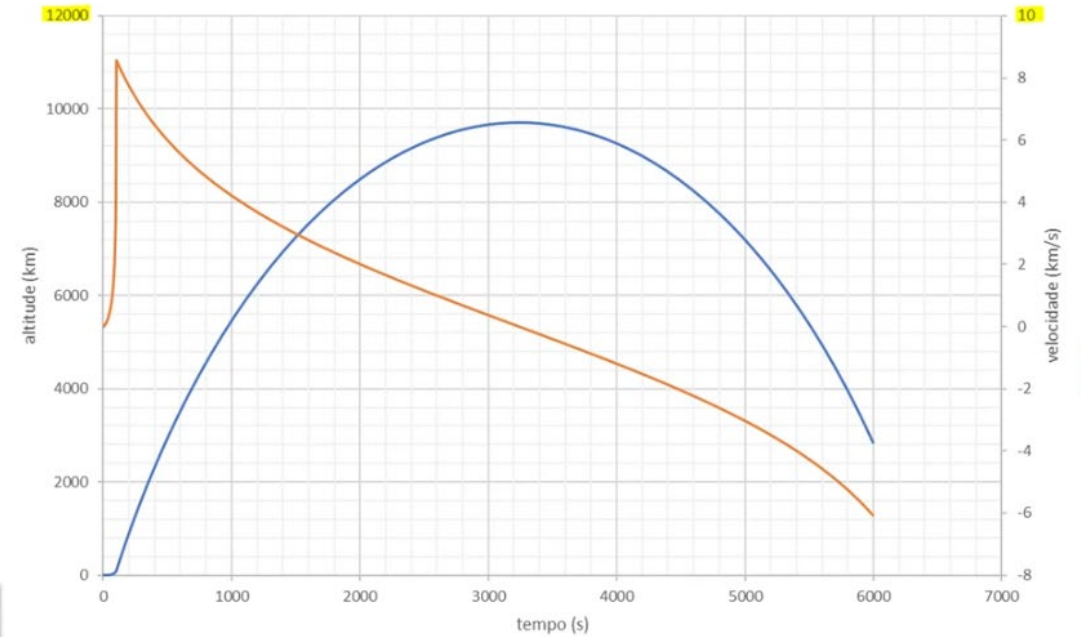
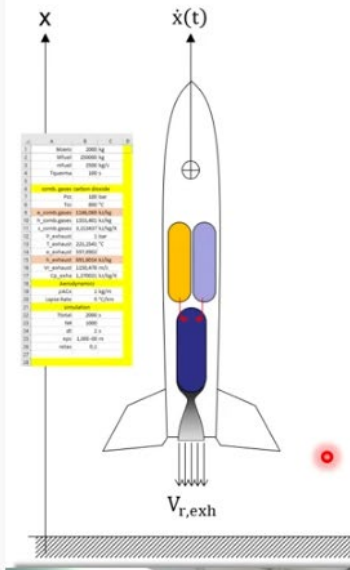
Entendendo o Lançamento de um Foguete / Simulador de Voo

Prof. P. Seleglim



Entendendo os Motores de Foguete 1/2 Configurações

Prof. P. Seleglim



$$[M_0 + M_{fuel} - \dot{m}_{fuel} \cdot t] \cdot \ddot{x}(t) = -\rho A C_x \frac{\dot{x}(t)^2}{2} + V_{r,exh} \cdot \dot{m}_{fuel} - [M_0 + M_{fuel} - \dot{m}_{fuel} \cdot t] \cdot \frac{GM_{terra}}{(R + x(t))^2}$$

Termodinâmica: T5bis Rocket Simulation/ Equação de balanço para volumes de controle 2/2

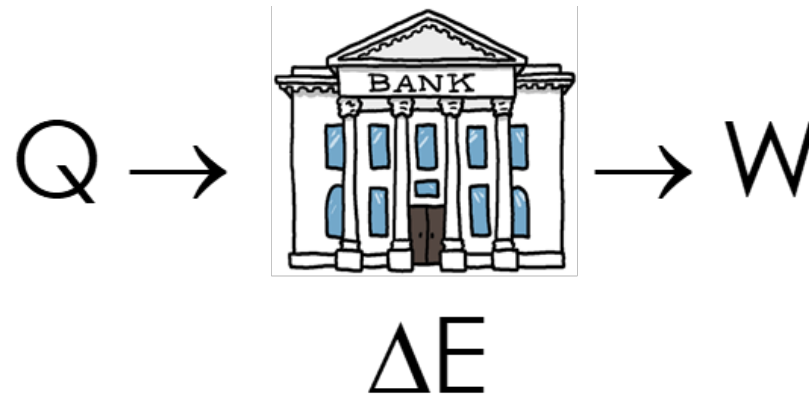
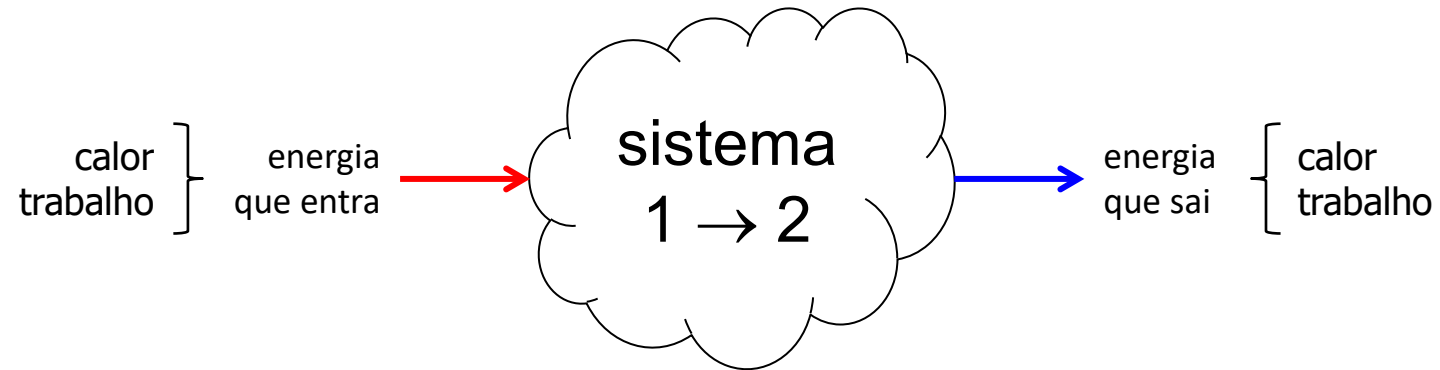


As questões postadas no Chat do YouTube serão respondidas ao final da aula.

Adequando o inventário energético para os sistemas abertos...

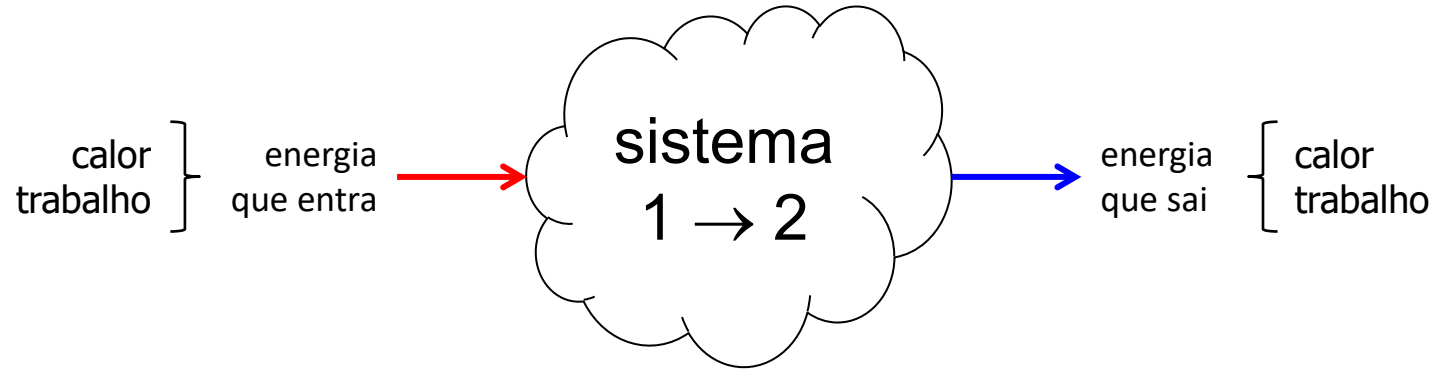


Princípio da conservação da energia: sistema fechado



As questões postadas no Chat do YouTube serão respondidas ao final da aula.

Princípio da conservação da energia: sistema fechado



$$\Delta E = \Delta U + \Delta EC + \Delta EP$$

energia interna
(agitação molecular)

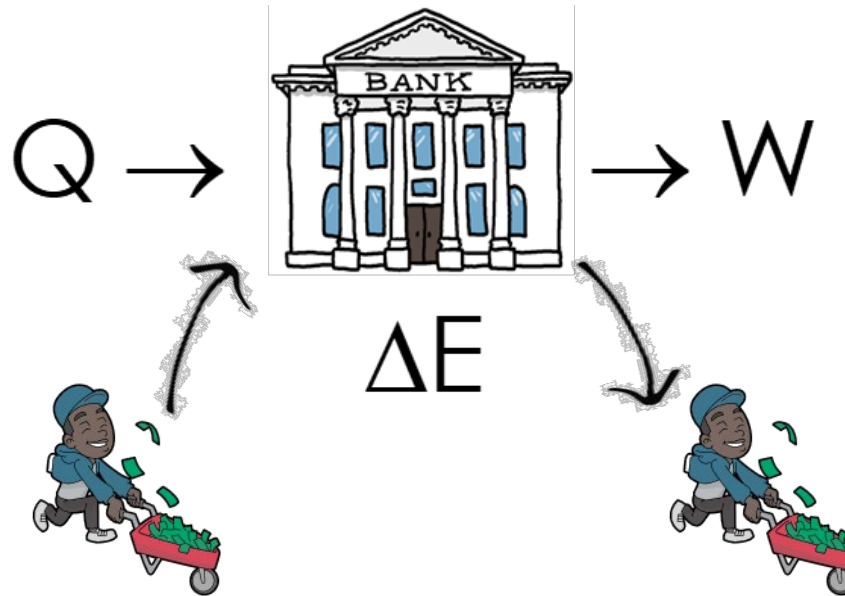
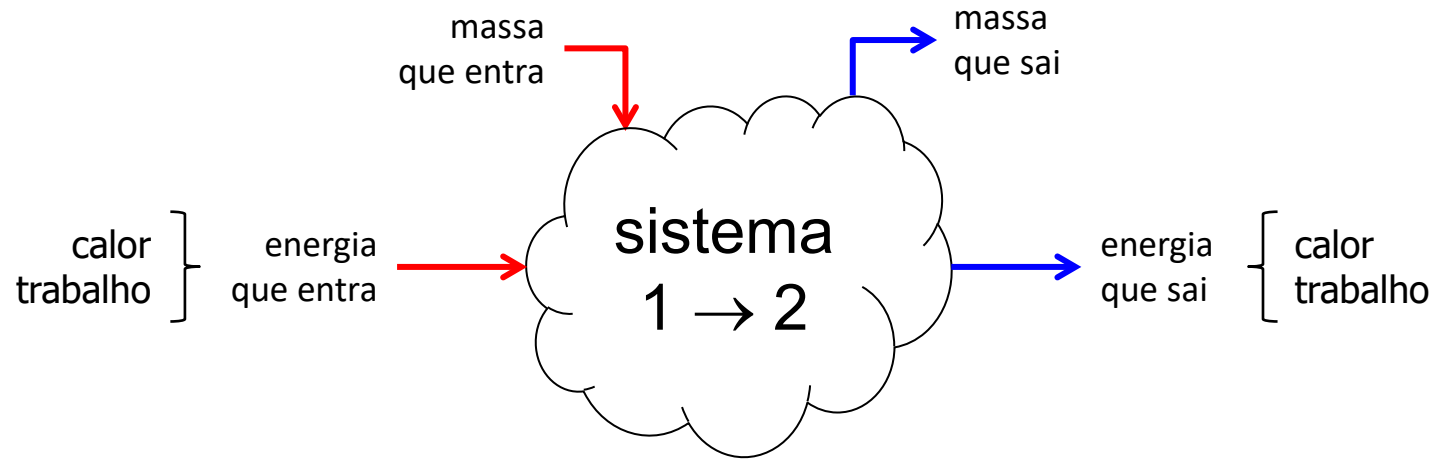
energia potencial
macroscópica

energia cinética
macroscópica



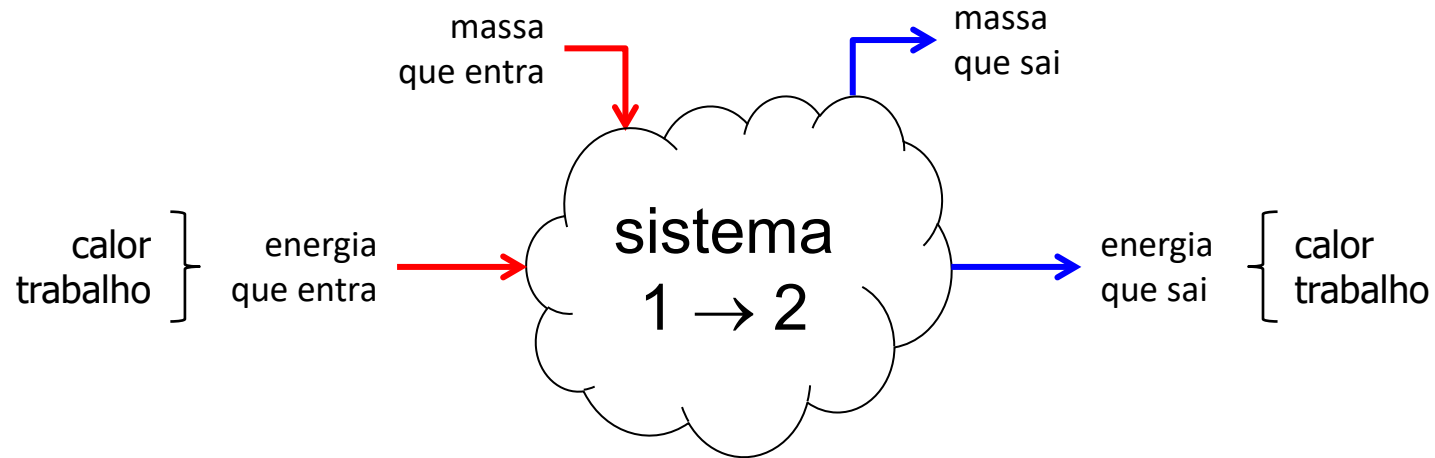
As questões postadas no Chat do YouTube
serão respondidas ao final da aula.

Princípio da conservação da energia: sistema aberto

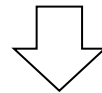


As questões postadas no Chat do YouTube serão respondidas ao final da aula.

Princípio da conservação da energia: sistema aberto



$$\Delta E = (E_{\text{entra}} - E_{\text{sai}})_{Q,W} + (E_{\text{entra}} - E_{\text{sai}})_{\text{massa}}$$



$$\frac{dE}{dt} = \left(\frac{dE_{\text{entra}}}{dt} - \frac{dE_{\text{sai}}}{dt} \right)_{Q,W} + \left(\frac{dE_{\text{entra}}}{dt} - \frac{dE_{\text{sai}}}{dt} \right)_{\text{massa}}$$



Equação generalizada para o balanço de
massa de um sistema aberto...



Depois vamos adaptar para energia...

Equação de balanço de massa: forma generalizada

$$\left(\begin{array}{c} \text{taxa de variação} \\ \text{da massa do sistema} \end{array} \right) - \left(\begin{array}{c} \text{fluxo total de} \\ \text{massa entrando} \end{array} \right) = 0$$

$$\frac{d}{dt} M + \int_{sc} \rho (\vec{V} \cdot \vec{n}) dA = 0$$

$$\frac{d}{dt} \int_{vc} \rho dV + \int_{sc} \rho (\vec{V} \cdot \vec{n}) dA = 0$$



Equação de balanço de massa: forma discreta

$$\left(\begin{array}{c} \text{taxa de variação} \\ \text{da massa do sistema} \end{array} \right) - \left(\begin{array}{c} \text{fluxo total de} \\ \text{massa entrando} \end{array} \right) = 0$$

$$\frac{d}{dt}M + \left(\sum_{\text{saídas}} m_{\text{sai},k} - \sum_{\text{entradas}} m_{\text{ent},k} \right) = 0$$

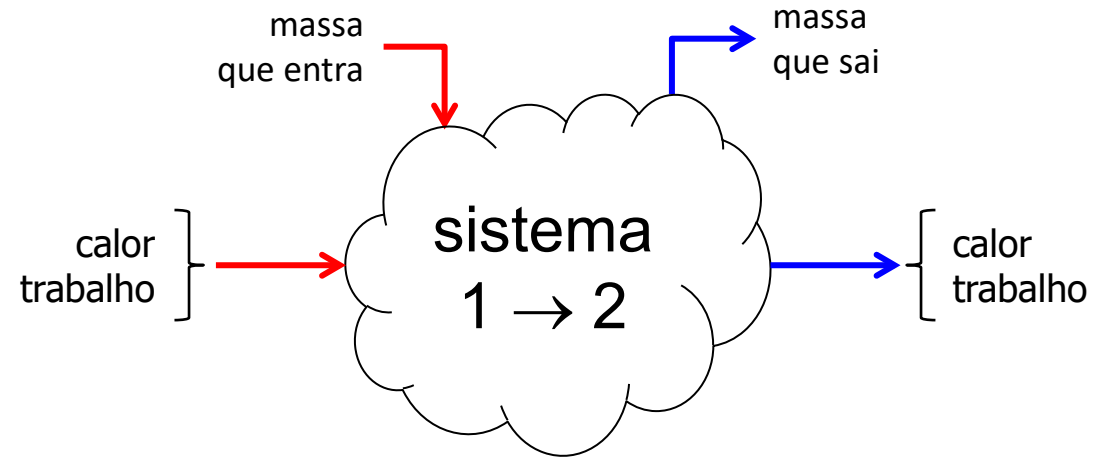
$$\frac{d}{dt}M + \left(\sum_{\text{saídas}} \rho_k v_{n,k} A_k - \sum_{\text{entradas}} \rho_k v_{n,k} A_k \right) = 0$$



Calculando a energia associada
aos fluxos de massa...

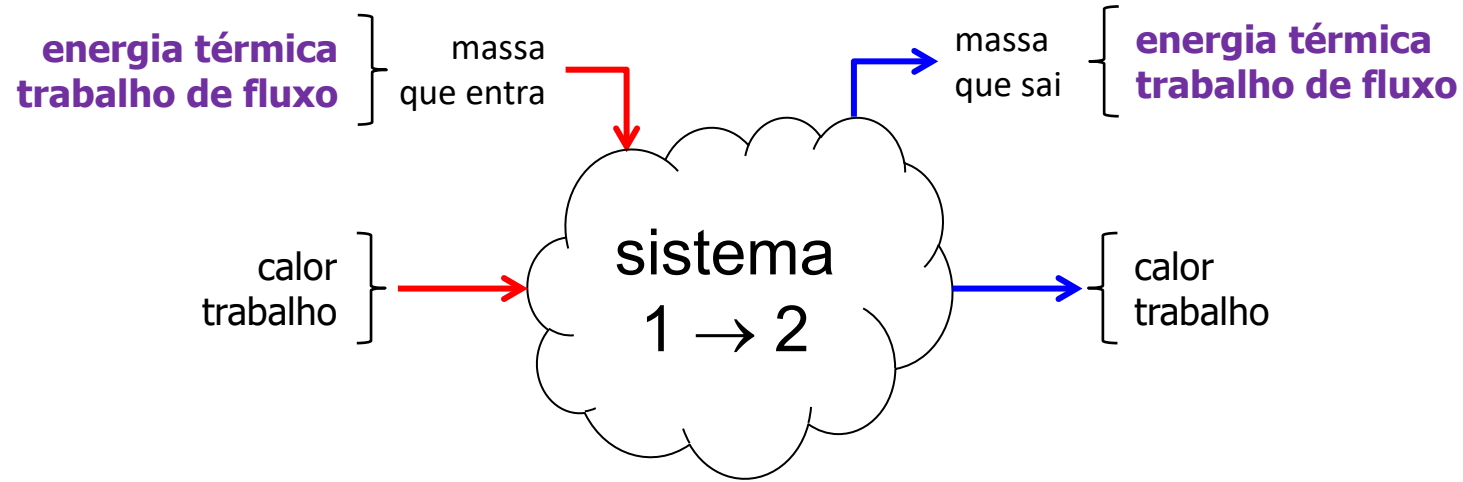


Princípio da conservação da energia: sistema aberto



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Princípio da conservação da energia: sistema aberto



transporte de energia térmica



Energia térmica, ou interna ($u + ep + ec$), associada ao fluxo de massa entrando/saindo do VC e que deve ser contabilizado

trabalho de fluxo

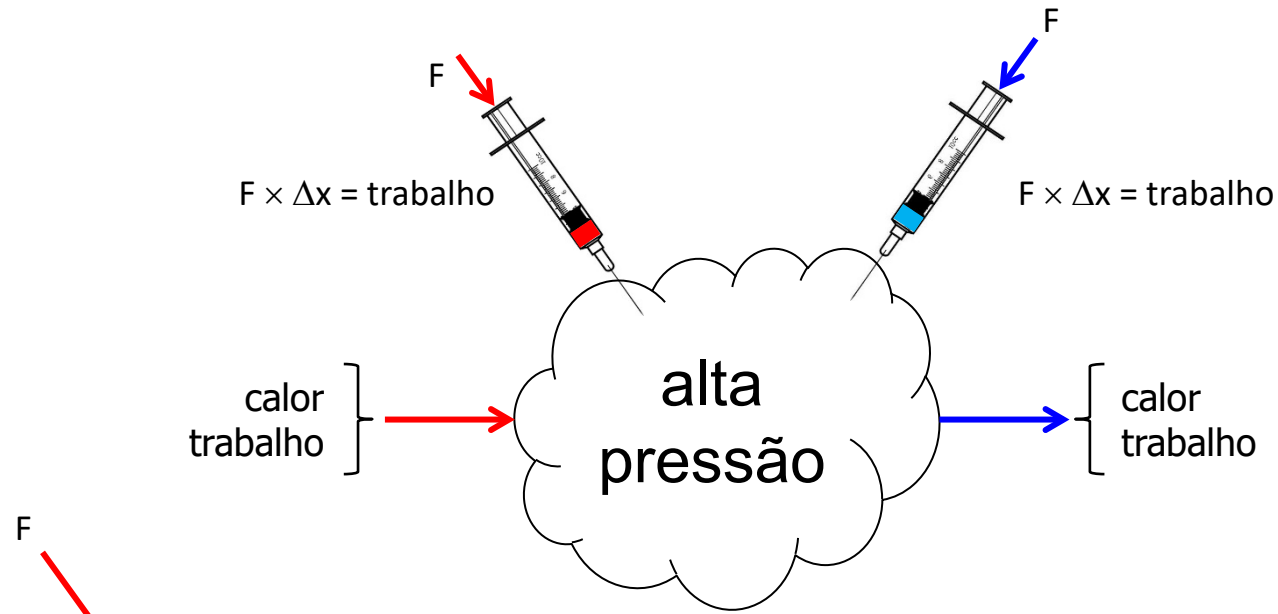


Trabalho mecânico (força \times deslocamento) necessário para que um elemento de massa entre/saia do VC...



As questões postadas no Chat do YouTube serão respondidas ao final da aula.

Trabalho necessário para entrar no sistema: trabalho de fluxo...



$$W_{\text{fluxo,e}} = F \times \Delta x = PA \cdot \Delta x$$

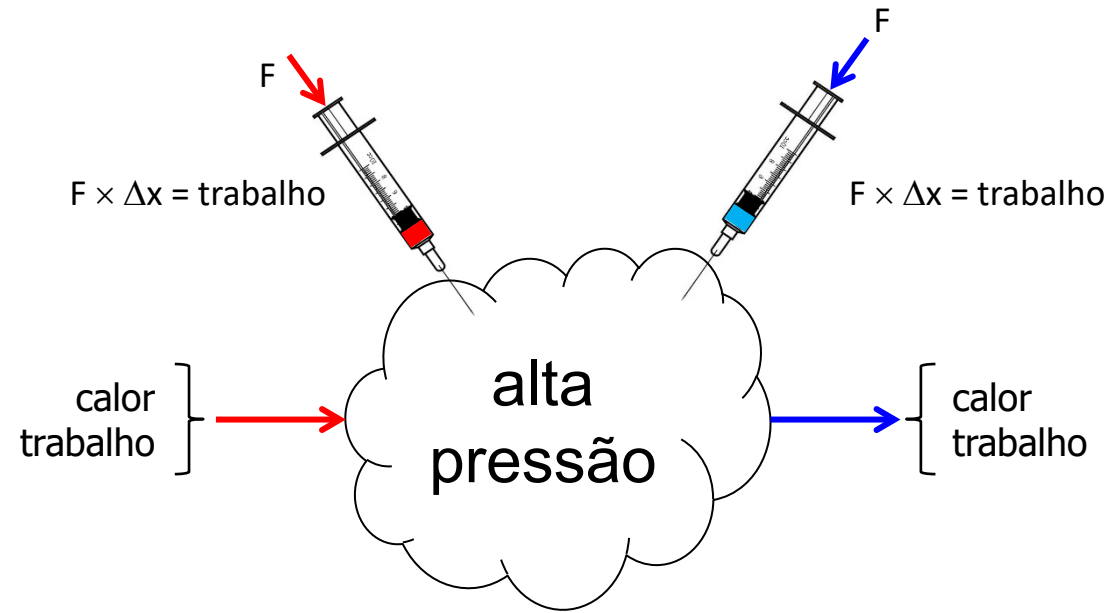
$$W_{\text{fluxo,e}} = F \times \Delta x = P \cdot (A \cdot \Delta x)$$

$$W_{\text{fluxo,e}} = P \cdot (A \cdot \Delta x) = P \cdot V$$



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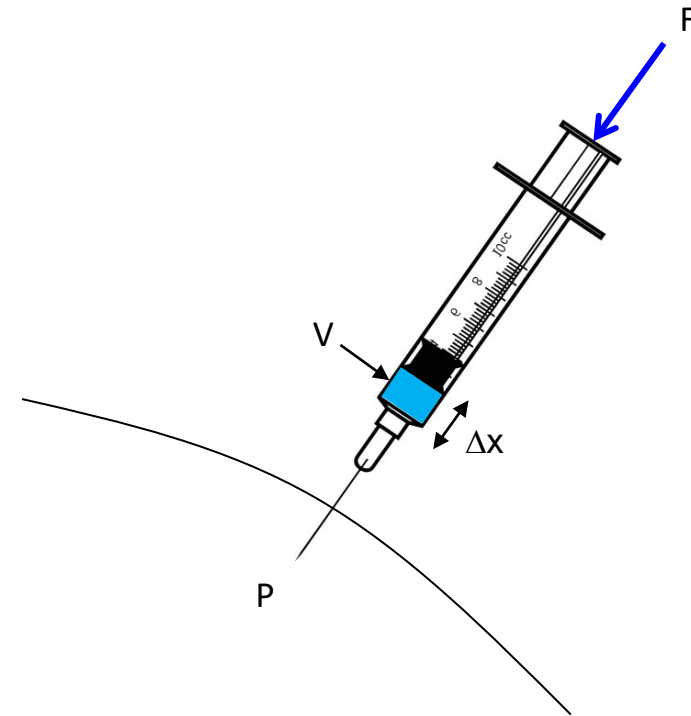
Trabalho necessário para sair no sistema: trabalho de fluxo...



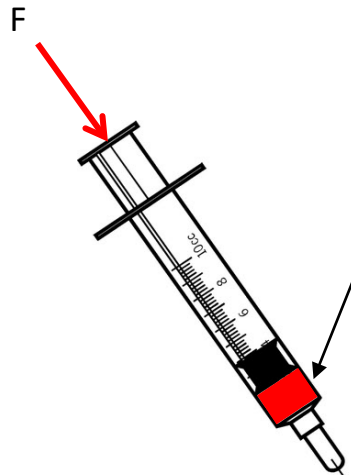
$$W_{\text{fluxo},s} = F \times \Delta x = PA \cdot \Delta x$$

$$W_{\text{fluxo},s} = F \times \Delta x = P \cdot (A \cdot \Delta x)$$

$$W_{\text{fluxo},s} = P \cdot (A \cdot \Delta x) = P \cdot V$$



Energia total associada ao fluxo de massa entrando/saindo...



$$\begin{cases} E = m \cdot (u + gz + V^2 / 2) \\ E = \rho V \cdot (u + gz + V^2 / 2) \end{cases}$$

energia interna
+
trabalho de fluxo

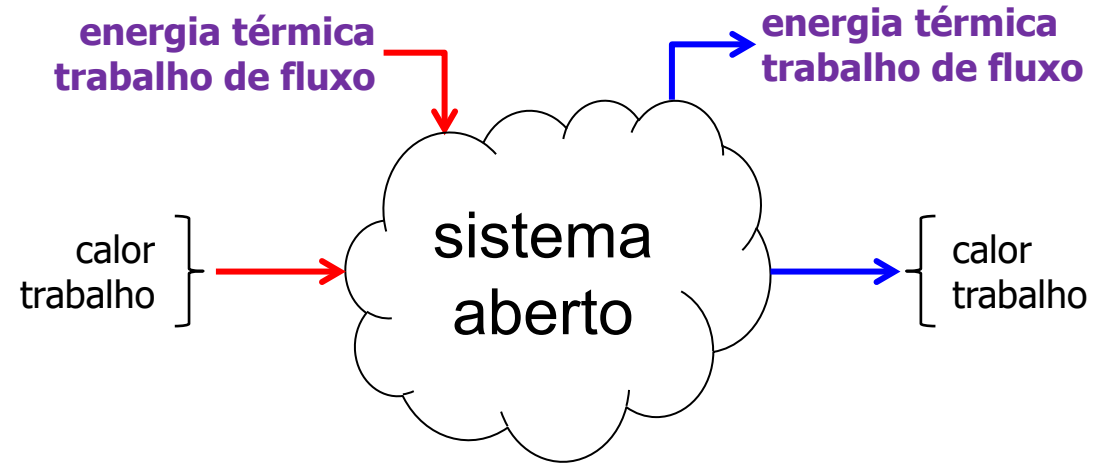
$$\begin{cases} \theta = E + W_{\text{fluxo}} \\ \theta = m \cdot (u + gz + V^2 / 2) + (F \cdot \Delta x) \\ \theta = m \cdot (u + gz + V^2 / 2) + m \cdot (Pv) \end{cases}$$

$$\theta = m \cdot (h + gz + V^2 / 2)$$

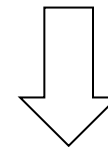
Formulando o inventário de energia
para um sistema aberto...



Inventário de energia no VC... (regime permanente)



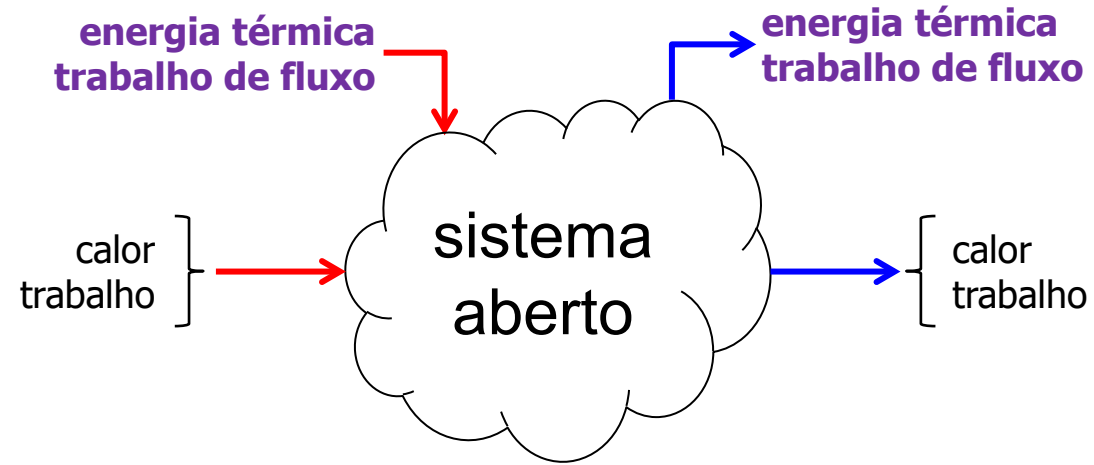
$$\left[\begin{array}{l} \text{energia total (líquida)} \\ \text{entrando no VC} \end{array} \right] = \left[\begin{array}{l} \text{taxa de variação da} \\ \text{energia interna do VC} \end{array} \right]$$



regime permanente

$$\left[\begin{array}{l} \text{energia total (líquida)} \\ \text{entrando no VC} \end{array} \right] = \left[\begin{array}{c} \mathbf{0} \end{array} \right]$$

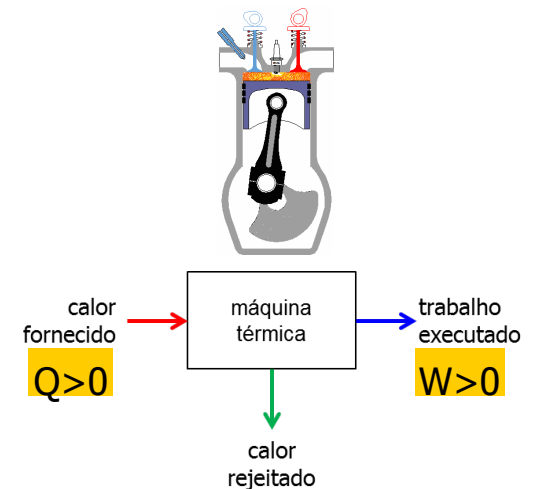
Inventário de energia no VC... (regime permanente)



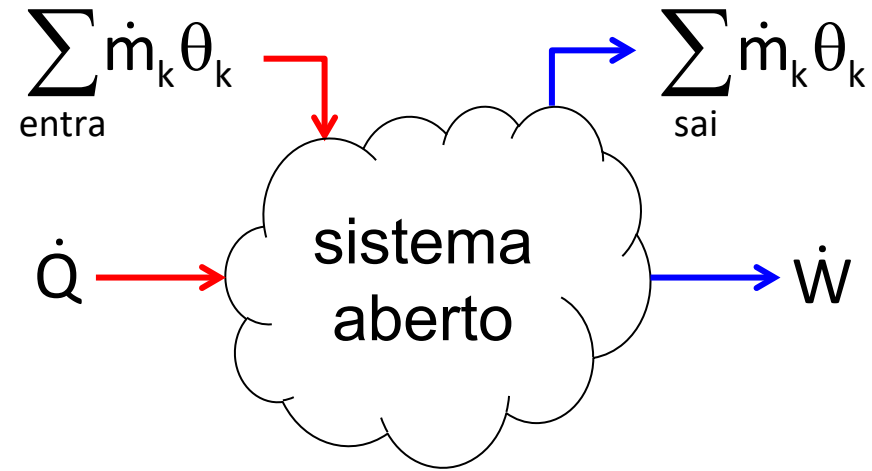
$$\dot{Q}_e + \dot{W}_e + \sum_{\text{entra}} \dot{m}_k \theta_k = \dot{Q}_s + \dot{W}_s + \sum_{\text{sai}} \dot{m}_k \theta_k$$

$$(\dot{Q}_e - \dot{Q}_s) - (\dot{W}_s - \dot{W}_e) = \sum_{\text{sai}} \dot{m}_k \theta_k - \sum_{\text{entra}} \dot{m}_k \theta_k$$

$$\dot{Q} - \dot{W} = \sum_{\text{sai}} \dot{m}_k \theta_k - \sum_{\text{entra}} \dot{m}_k \theta_k$$



Inventário de energia no VC... (regime permanente)



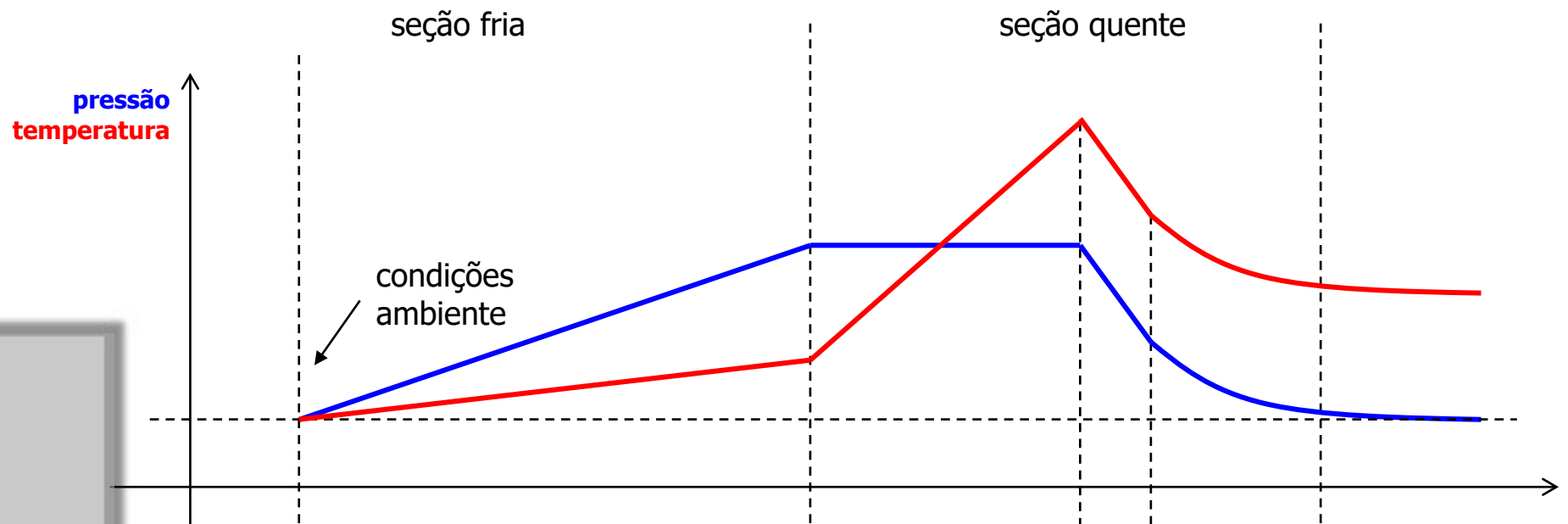
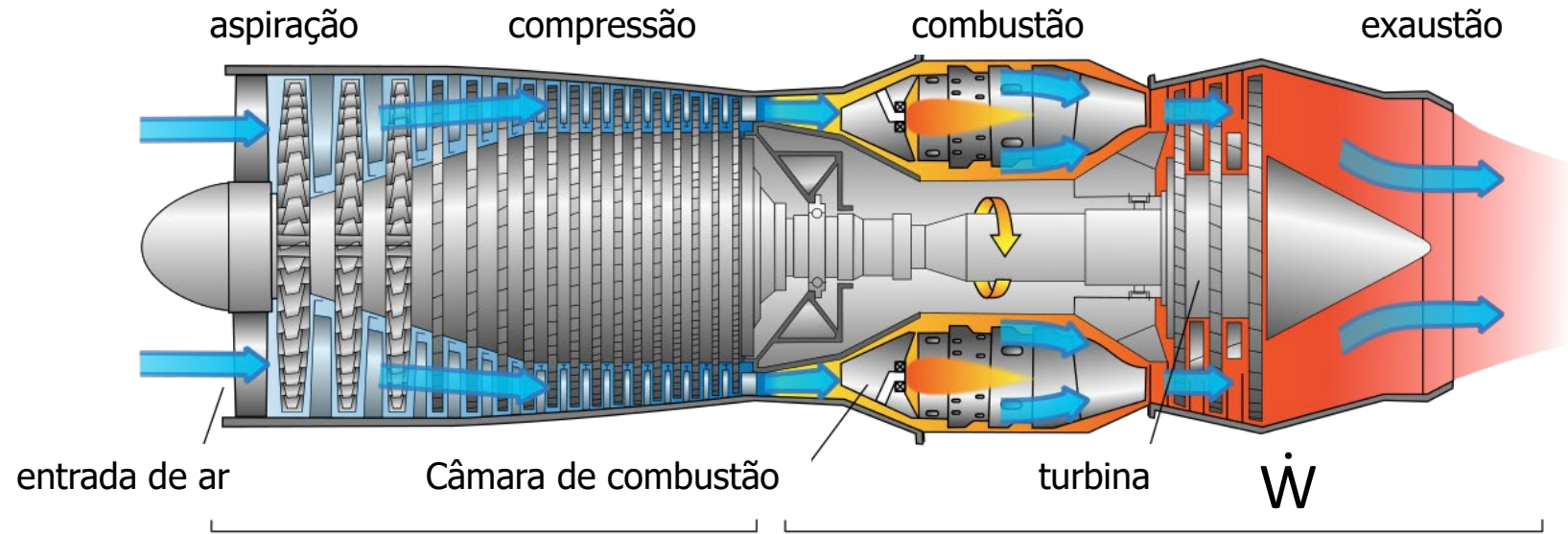
$$\dot{Q} - \dot{W} = \sum_{\text{sai}} \dot{m}_k \theta_k - \sum_{\text{entra}} \dot{m}_k \theta_k$$

$$\dot{Q} - \dot{W} = \sum_{\text{sai}} \dot{m}_k \cdot \left(h_k + gz_k + \frac{V_k^2}{2} \right) - \sum_{\text{entra}} \dot{m}_k \cdot \left(h_k + gz_k + \frac{V_k^2}{2} \right)$$



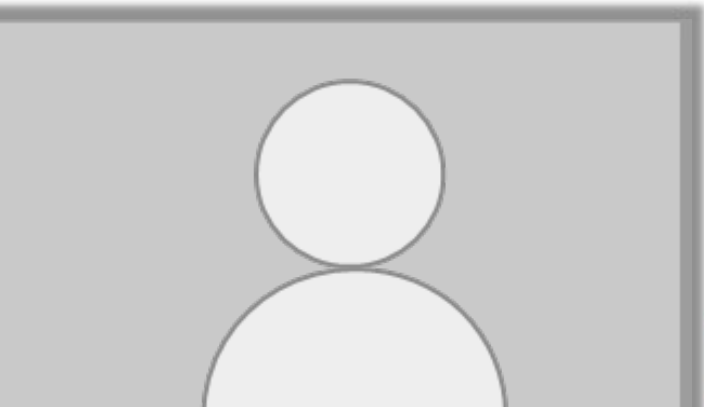
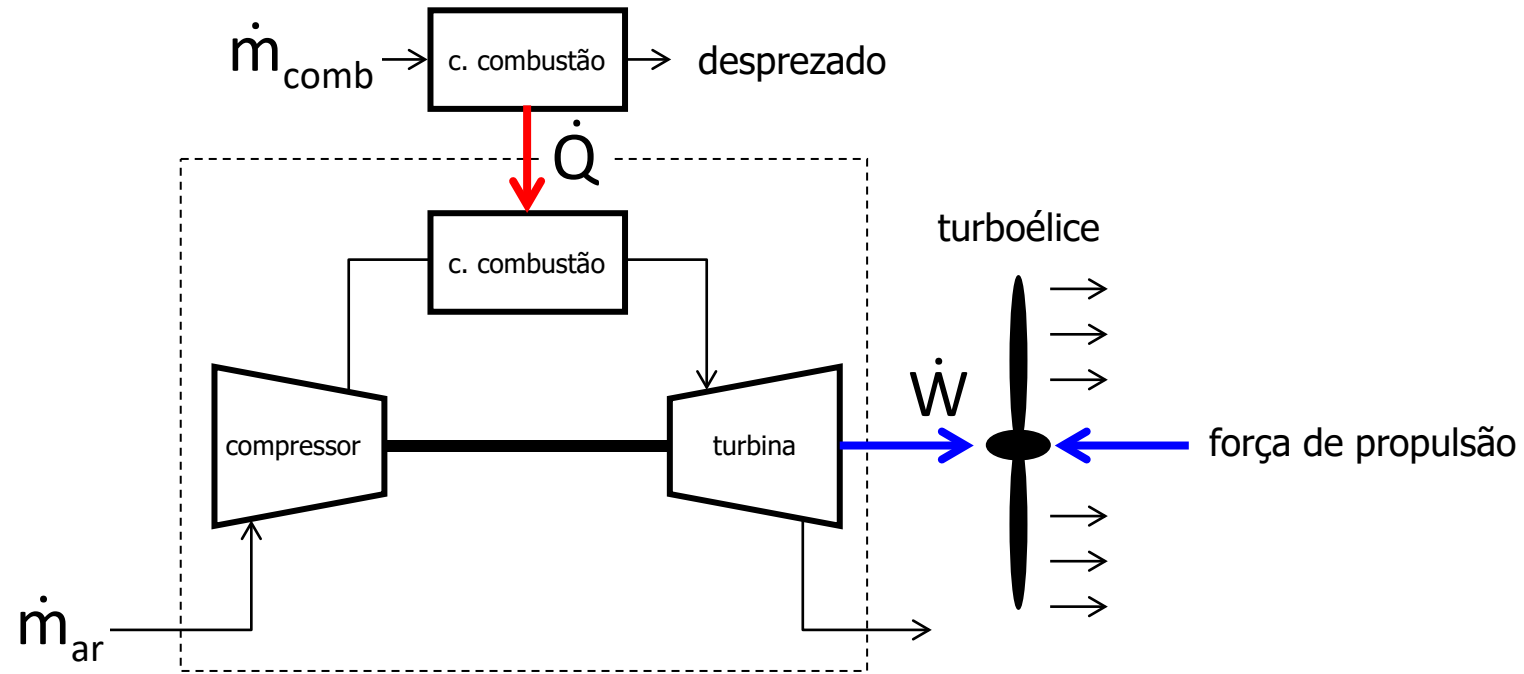
Aplicação: análise de motores
de propulsão a jato e turboélice



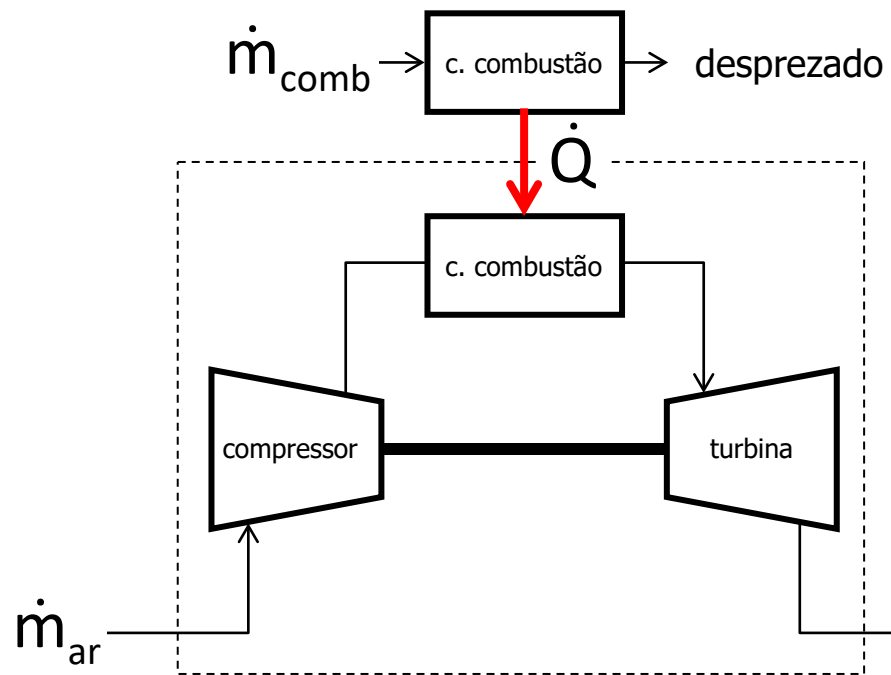


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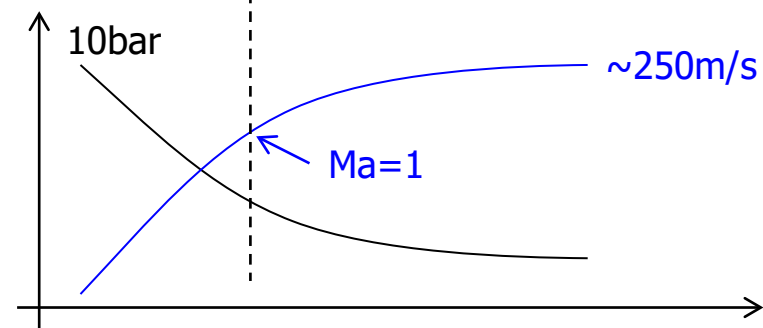
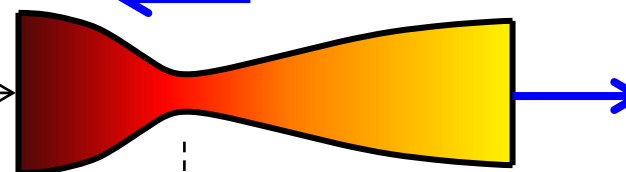
Modelo simplificado: turboélice



Modelo simplificado: motor a jato

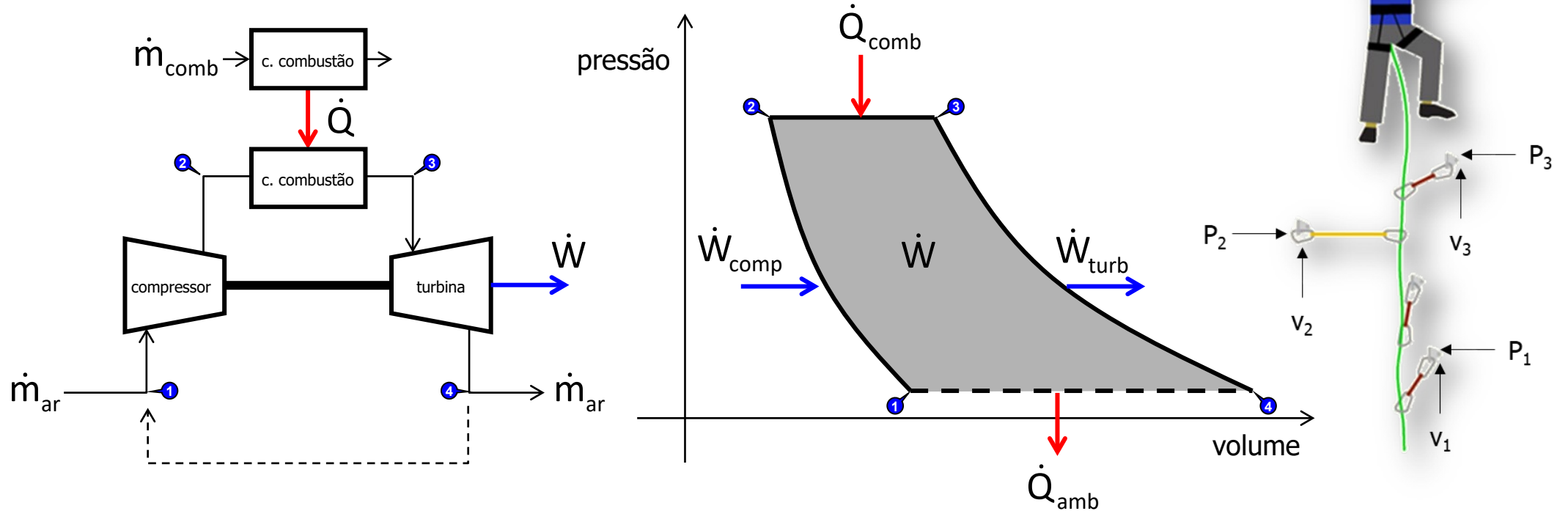


força de propulsão



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Análise termodinâmica – ciclo de Brayton



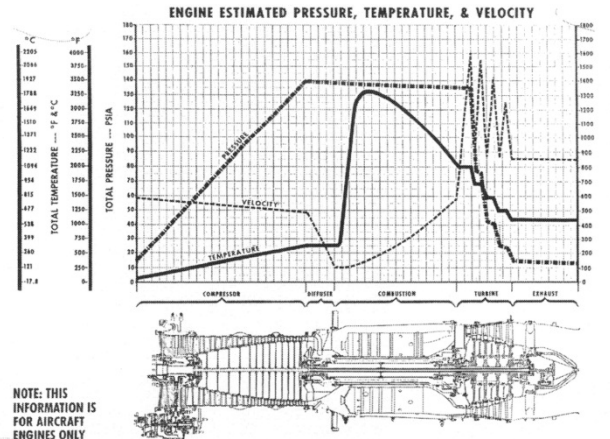
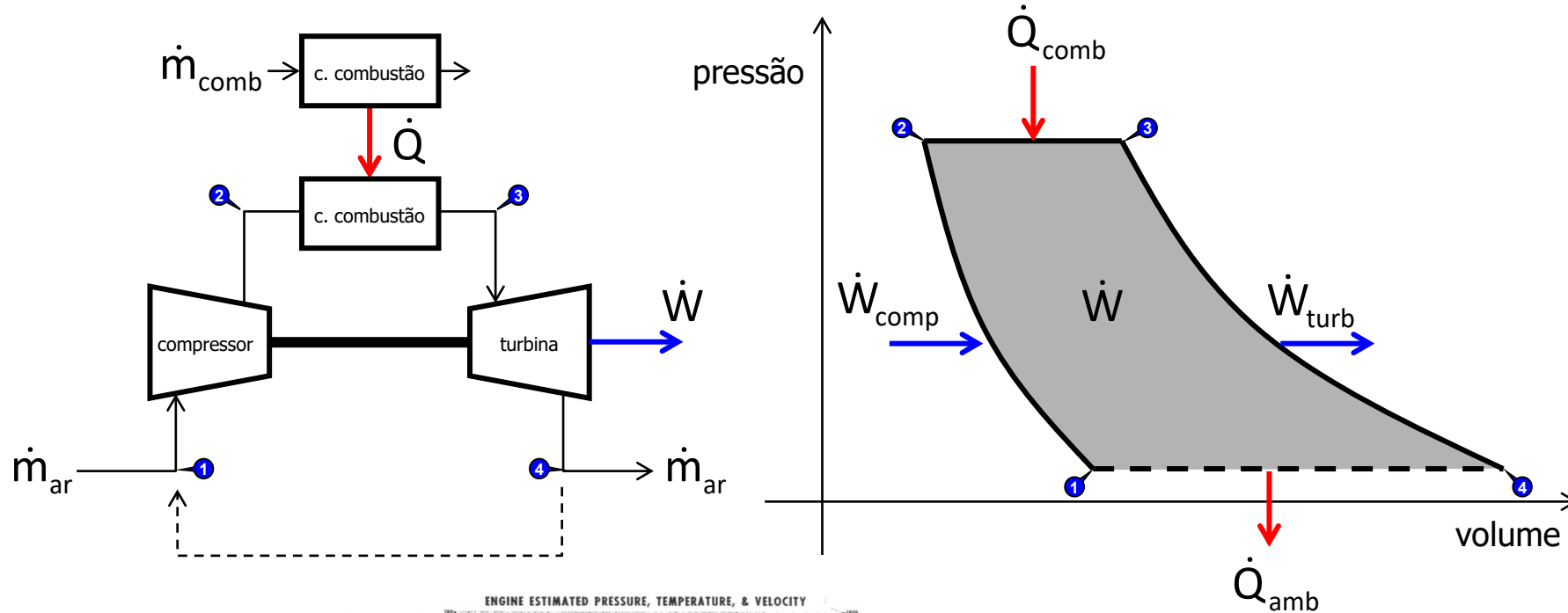
Análise termodinâmica

⇒ "PASSO ZERO": Determinação das propriedades de estado



As questões postadas no Chat do YouTube serão respondidas ao final da aula.

Análise termodinâmica – ciclo de Brayton



estimativas práticas

P2=10bar

T2=?????

P1=1bar

T1=25°C

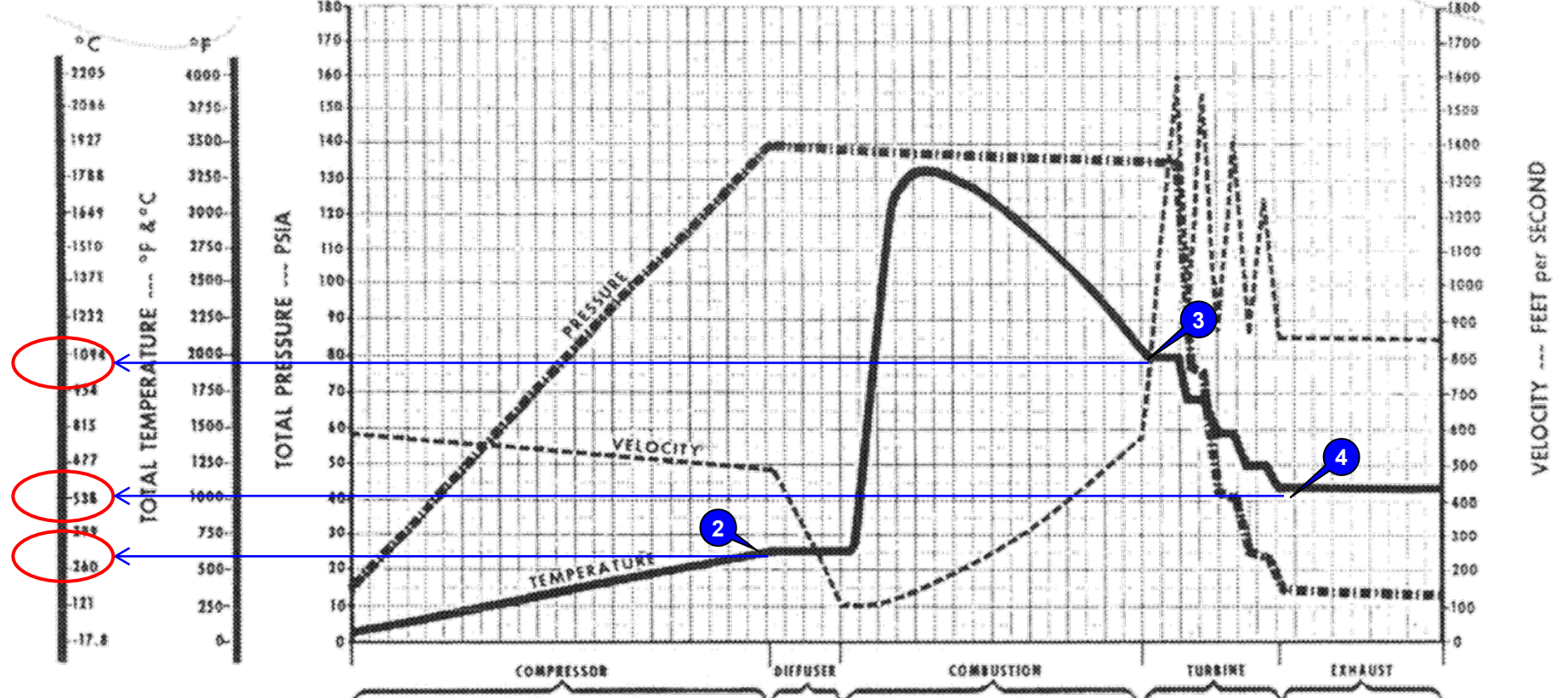
P3=10bar

T3=?????

P4=1bar

T4=?????

ENGINE ESTIMATED PRESSURE, TEMPERATURE, & VELOCITY



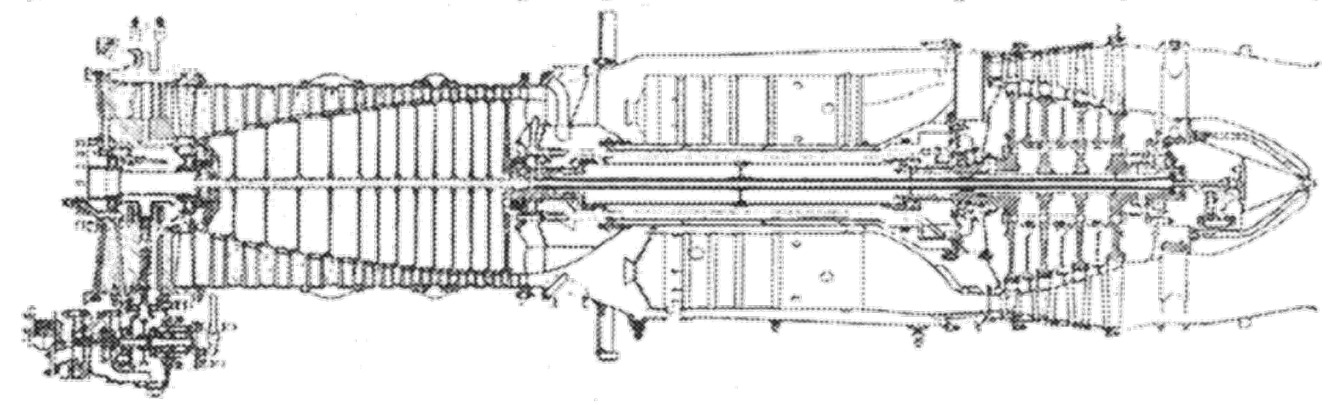
1094
538
240

2

3

4

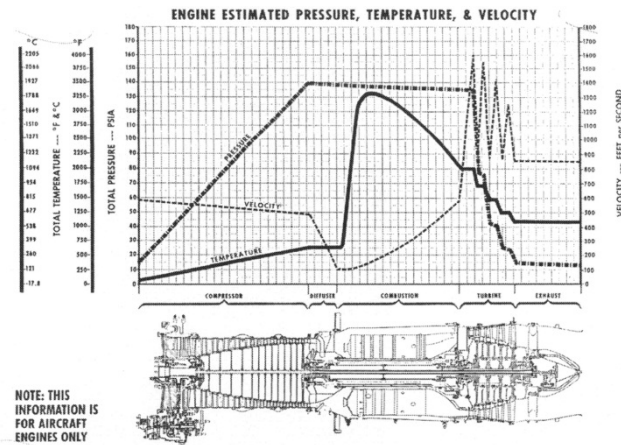
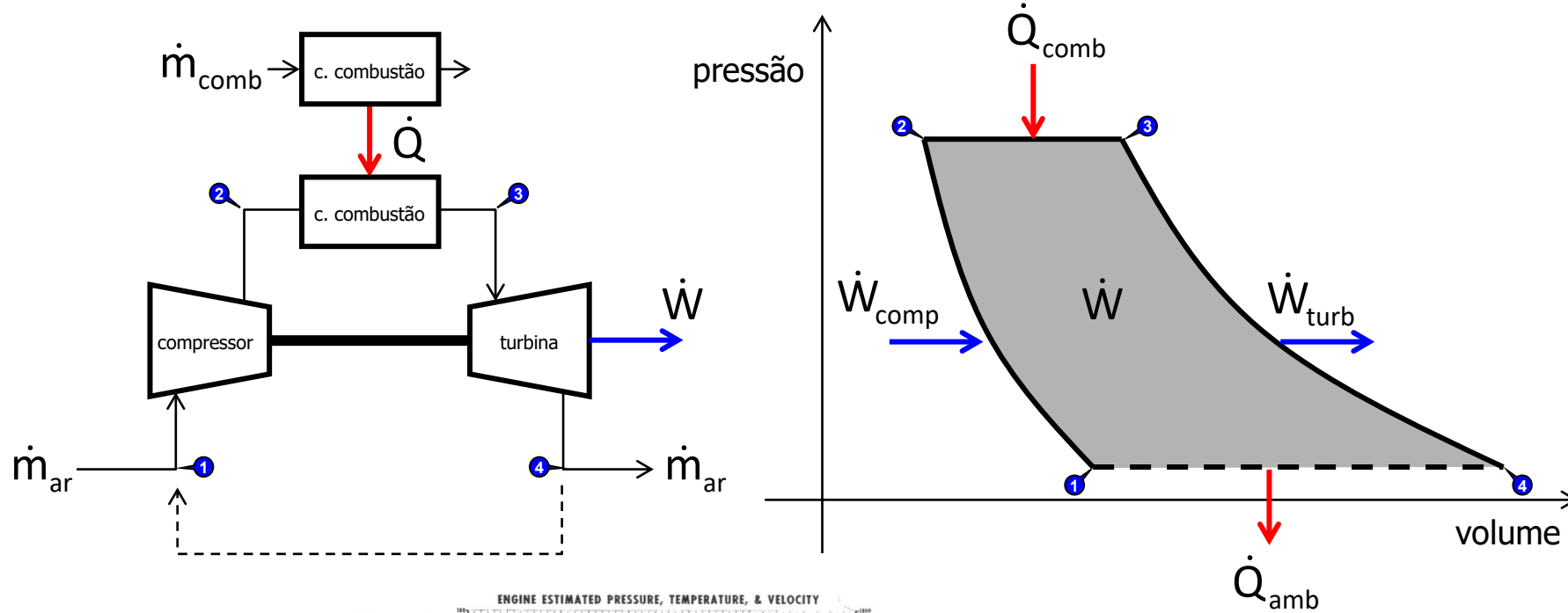
NOTE: THIS INFORMATION IS FOR AIRCRAFT ENGINES ONLY



Estimativa das variáveis termodinâmicas...



Análise termodinâmica – ciclo de Brayton



estimativas práticas

$$P_2 = 10 \text{ bar}$$

$$T_2 \approx 290^\circ\text{C}$$

$$P_1 = 1 \text{ bar}$$

$$T_1 = 25^\circ\text{C}$$

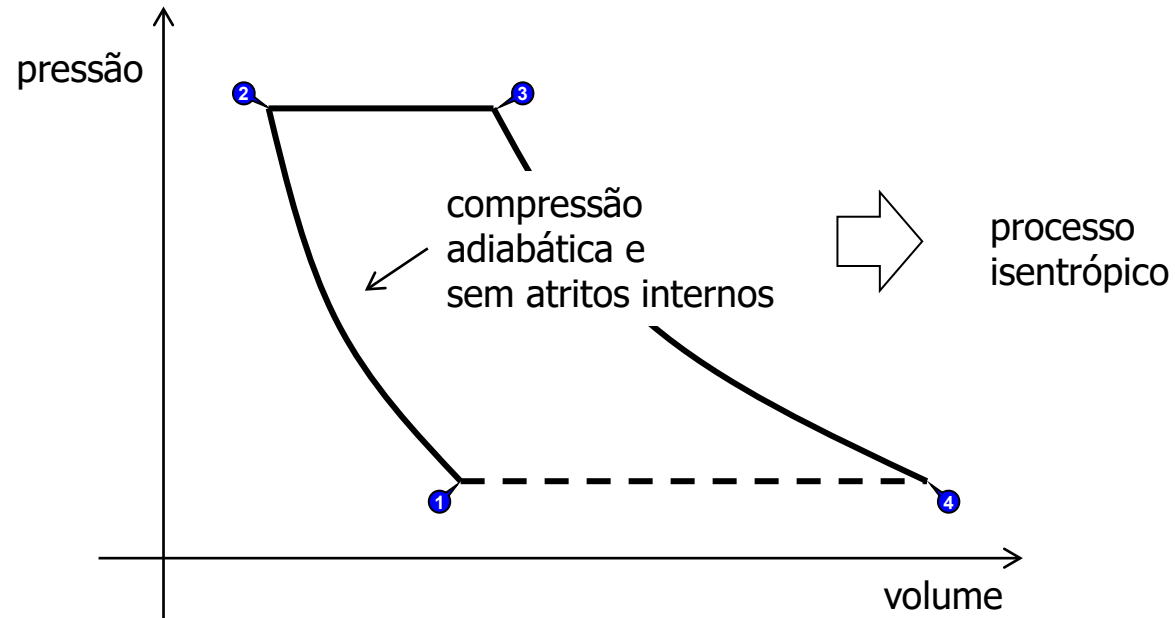
$$P_3 = 10 \text{ bar}$$

$$T_3 \approx 1094^\circ\text{C}$$

$$P_4 = 1 \text{ bar}$$

$$T_4 \approx 570^\circ\text{C}$$

Determinação das propriedades em (2)



$P_1 = 1 \text{ bar}$
 $T_1 = 25^\circ\text{C}$
 $s_1 = 3,88 \text{ kJ/kg/K}$

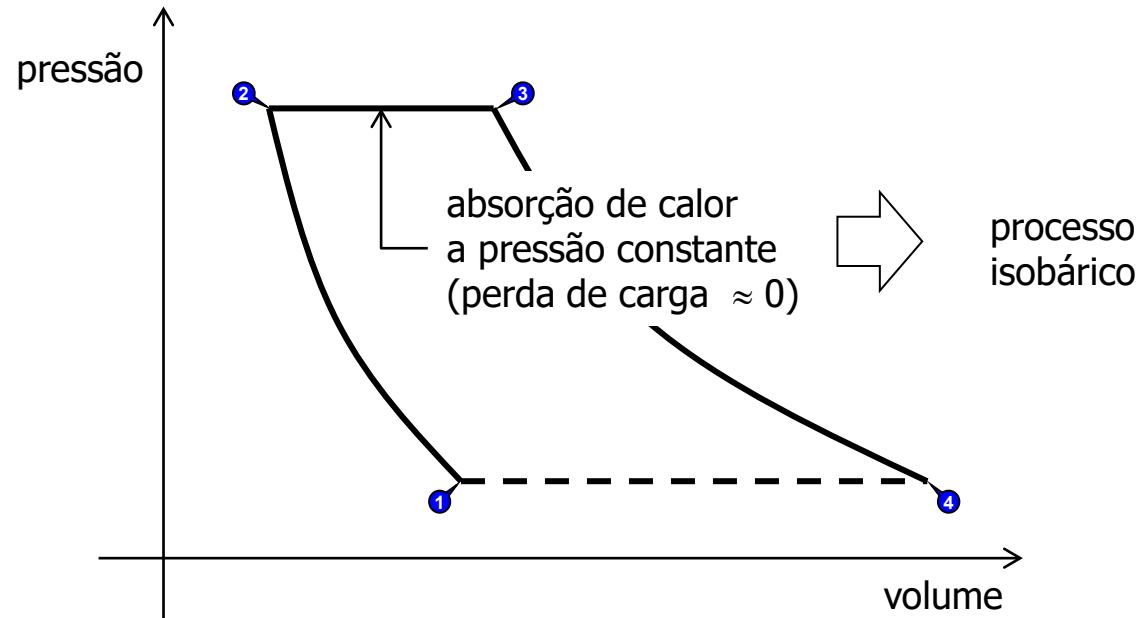
$P_2 = 10 \text{ bar}$
 $s_2 = s_1 = 3,88 \text{ kJ/kg/K}$
 $T_2 = 297,8^\circ\text{C}$
 $h_2 = 703,05 \text{ kJ/kg}$

→ estimado: $T_2 \approx 290^\circ\text{C}$

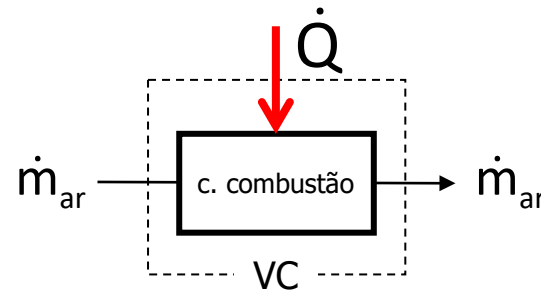


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Determinação das propriedades em (3)



$$\dot{Q} - \dot{W} = \sum_{\text{sai}} \dot{m}_k \theta_k - \sum_{\text{entra}} \dot{m}_k \theta_k$$



$$\dot{Q} = \dot{m}_{ar} \cdot \left[\left(h + \cancel{gz} + \cancel{\frac{v^2}{2}} \right)_3 - \left(h + \cancel{gz} + \cancel{\frac{v^2}{2}} \right)_2 \right]$$

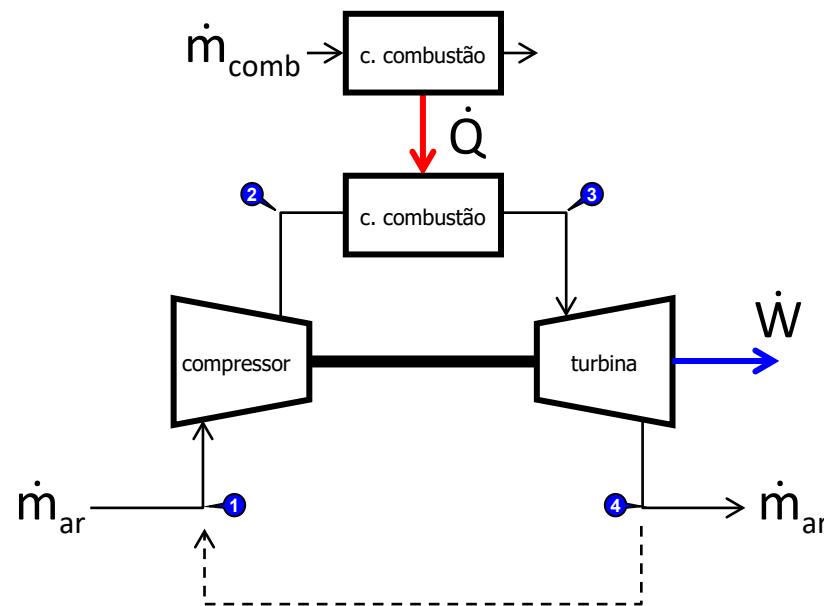
$$\dot{Q} = \dot{m}_{ar} \cdot (h_3 - h_2) \quad \Rightarrow \quad \dot{Q} / \dot{m}_{ar} = h_3 - h_2$$



Determinação das propriedades em (3)

$$\dot{Q} = \dot{m}_{\text{comb}} \cdot (\eta_{\text{comb}} \text{PCI}_{\text{comb}}) = \dot{m}_{\text{ar}} \cdot (h_3 - h_2)$$

$$h_3 = h_2 + \dot{m}_{\text{comb}} / \dot{m}_{\text{ar}} \cdot \eta_{\text{comb}} \cdot \text{PCI}_{\text{comb}}$$



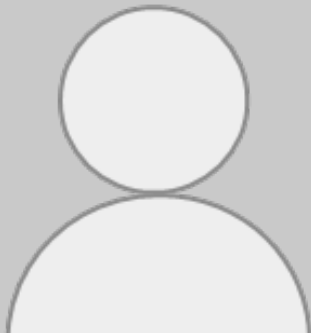
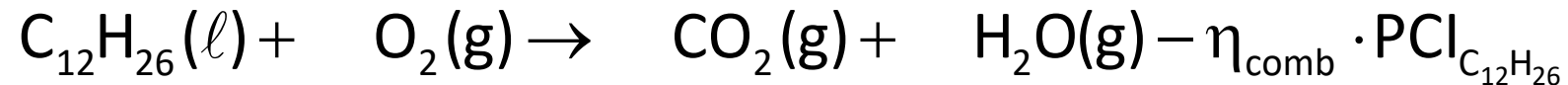
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Determinação das propriedades em (3)

$$\dot{Q} = \dot{m}_{\text{comb}} \cdot (\eta_{\text{comb}} \text{PCI}_{\text{comb}}) = \dot{m}_{\text{ar}} \cdot (h_3 - h_2)$$

$$h_3 = h_2 + \dot{m}_{\text{comb}} / \dot{m}_{\text{ar}} \cdot \eta_{\text{comb}} \cdot \text{PCI}_{\text{comb}}$$

Equação estequiométrica de combustão



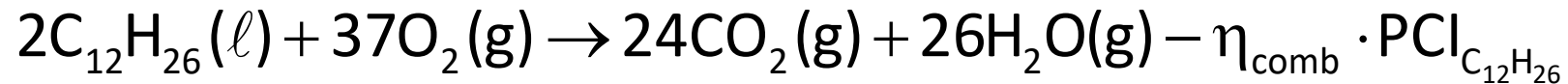
As questões postadas no Chat do YouTube serão respondidas ao final da aula.

Determinação das propriedades em (3)

$$\dot{Q} = \dot{m}_{\text{comb}} \cdot (\eta_{\text{comb}} \text{PCI}_{\text{comb}}) = \dot{m}_{\text{ar}} \cdot (h_3 - h_2)$$

$$h_3 = h_2 + \dot{m}_{\text{comb}} / \dot{m}_{\text{ar}} \cdot \eta_{\text{comb}} \cdot \text{PCI}_{\text{comb}}$$

Equação estequiométrica de combustão



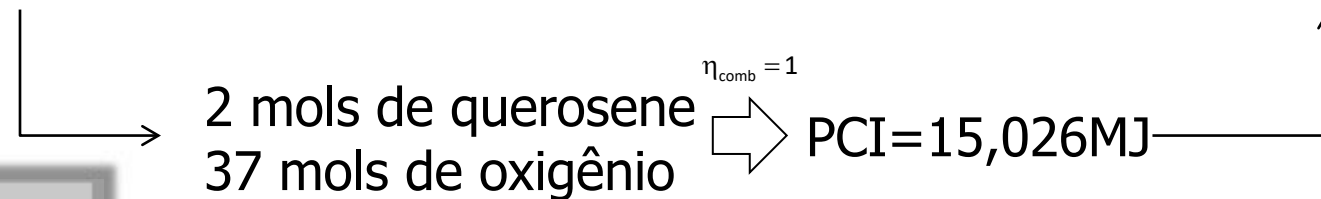
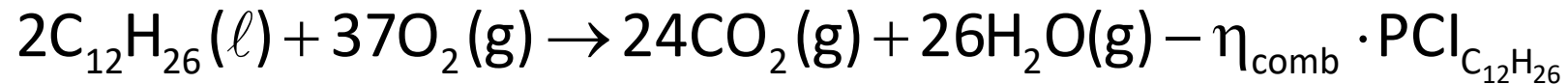
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Determinação das propriedades em (3)

$$\dot{Q} = \dot{m}_{\text{comb}} \cdot (\eta_{\text{comb}} \text{PCI}_{\text{comb}}) = \dot{m}_{\text{ar}} \cdot (h_3 - h_2)$$

$$h_3 = h_2 + \dot{m}_{\text{comb}} / \dot{m}_{\text{ar}} \cdot \eta_{\text{comb}} \cdot \text{PCI}_{\text{comb}}$$

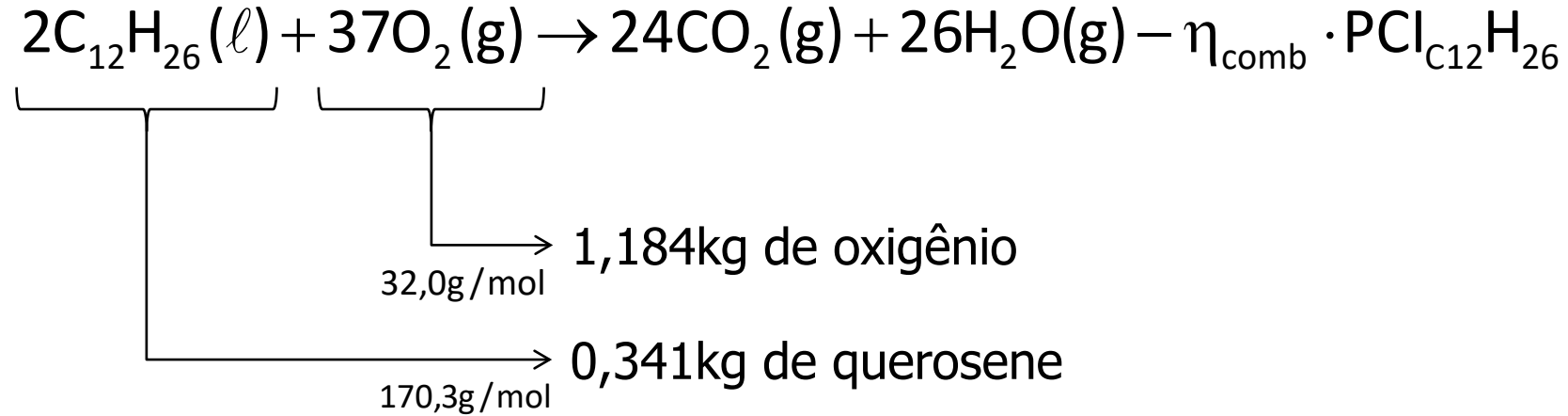
Equação estequiométrica de combustão



$$M_{\text{C}_{12}\text{H}_{26}} = 170,3 \text{ g/mol} \Rightarrow \text{PCI} = \frac{15,026 \text{ MJ}}{2 \cdot 170,3 \cdot 10^{-3} \text{ kg}} = 44,19 \text{ MJ/kg}$$



Determinação das propriedades em (3)



Cálculo da massa de ar para combustão estequiométrica

Major constituents of dry air, by volume^[6]

Gas		Volume ^(A)	
Name	Formula	in ppmv ^(B)	in %
Nitrogen	N ₂	780,840	78.084
Oxygen	O ₂	209,460	20.946
Argon	Ar	9,340	0.9340
Carbon dioxide	CO ₂	397	0.0397
Neon	Ne	18.18	0.001818
Helium	He	5.24	0.000524
Methane	CH ₄	1.79	0.000179
Not included in above dry atmosphere:			
Water vapor ^(C)	H ₂ O	10–50,000 ^(D)	0.001%–5% ^(D)

notes:
 (A) volume fraction is equal to mole fraction for ideal gas only, also see [volume \(thermodynamics\)](#)
 (B) ppmv: parts per million by volume
 (C) Water vapor is about 0.25% by mass over full atmosphere
 (D) Water vapor strongly varies locally^[4]

$$0,78084 \cdot N_2 + 0,20946 \cdot O_2 + 0,00934 \cdot Ar = 1\text{mol}$$

$\xrightarrow{28\text{g/mol}}$ 21,86g
 $\xrightarrow{32\text{g/mol}}$ 6,70g
 $\xrightarrow{40\text{g/mol}}$ 0,37g

$$0,7555\text{kg} + 0,2316\text{kg} + 0,0129\text{kg} = 1\text{kg}$$

Determinação das propriedades em (3)

Cálculo da massa de ar que contém 1,181kg de oxigênio

1 kg de ar \longrightarrow 0,232kg de oxigênio
x kg de ar \longrightarrow 1,184kg de oxigênio \Rightarrow 5,112kg de ar

Cálculo da razão ar/combustível

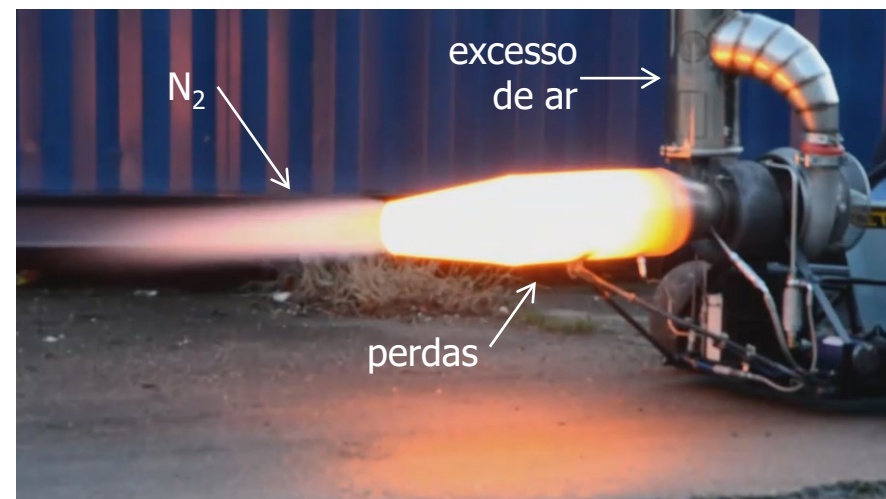
$$\frac{\dot{m}_{\text{comb}}}{\dot{m}_{\text{ar}}} = \frac{0,341\text{kg}}{5,112\text{kg}} = 0,066705$$

Rendimento da combustão

$$\eta_{\text{comb}} = \text{PCI} / \dot{Q}_{\text{útil}}$$

$$\eta_{\text{comb}}^{\text{arb}} = 0,333$$

A ser elaborado em
uma aula
específica...



Determinação das propriedades em (3)

Cálculo do salto entálpico

$$h_3 - h_2 = \dot{m}_{\text{comb}} / \dot{m}_{\text{ar}} \cdot \eta_{\text{comb}} \cdot \text{PCI}_{\text{comb}}$$

$$h_3 - 703,05 \text{ kJ/kg} = 0,066705 \cdot 0,333 \cdot 44,19 \cdot 10^3 \text{ kJ/kg}$$

$$h_3 = 1684,644 \text{ kJ/kg}$$

P3=10bar

h3=1684,644kJ/kg



T3=1161,6°C

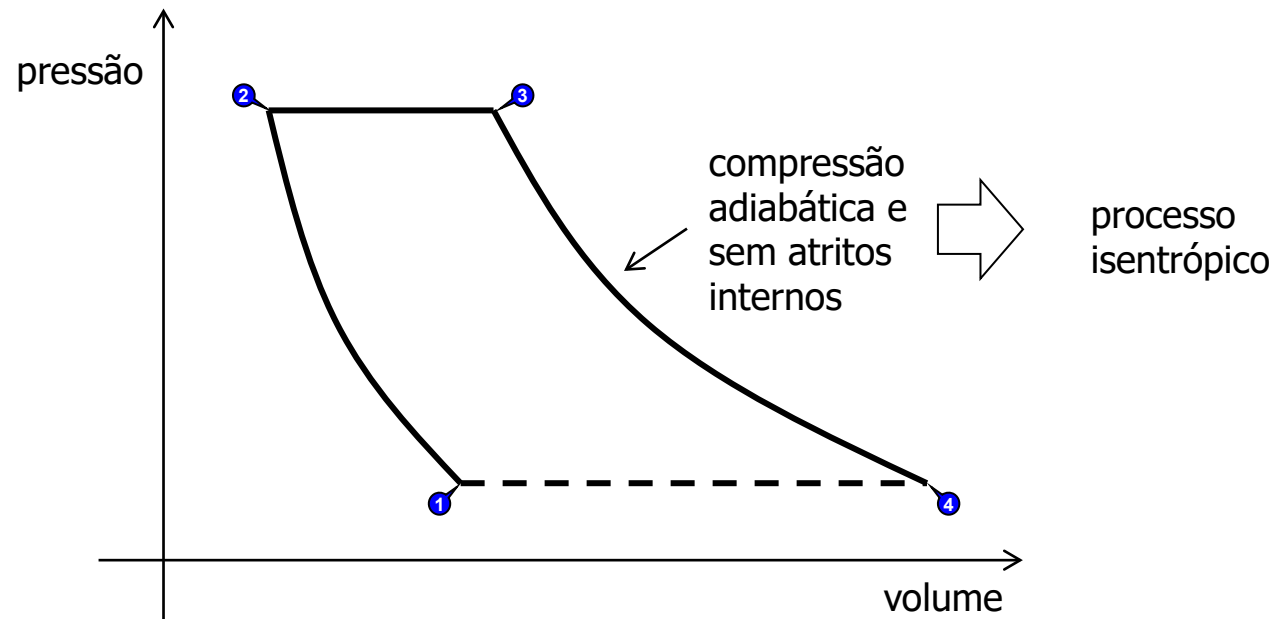
→ **estimado: T3≈1094°C**

s3=4,9208kJ/kg/K



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Determinação das propriedades em (4)



$P_3 = 10\text{bar}$

$h_3 = 1684,644\text{kJ/kg}$



$T_3 = 1161,6^\circ\text{C}$

$s_3 = 4,9208\text{kJ/kg/K}$

$P_4 = 1\text{bar}$

$s_4 = 4,9208\text{kJ/kg/K}$



$T_4 = 535,9^\circ\text{C}$

$h_4 = 958,6\text{kJ/kg}$

→ **estimado: $T_3 \approx 570^\circ\text{C}$**

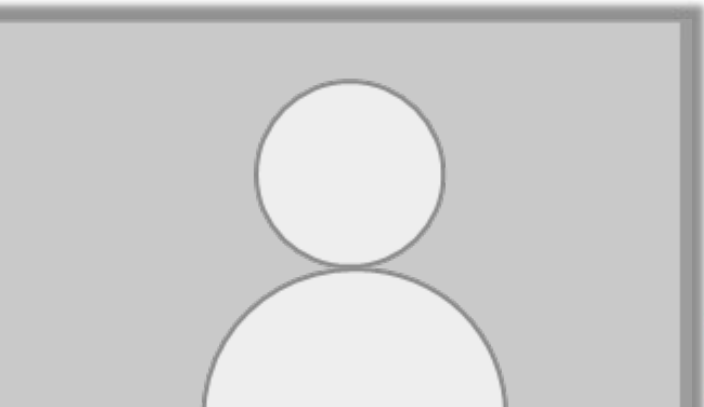
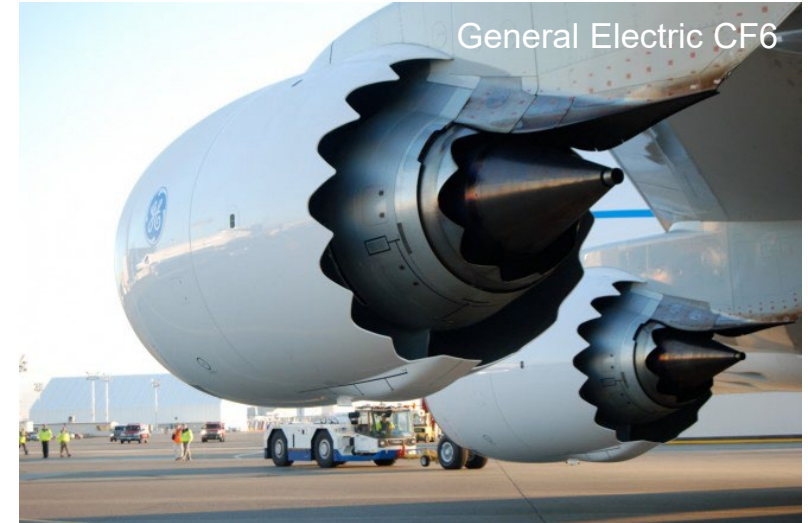


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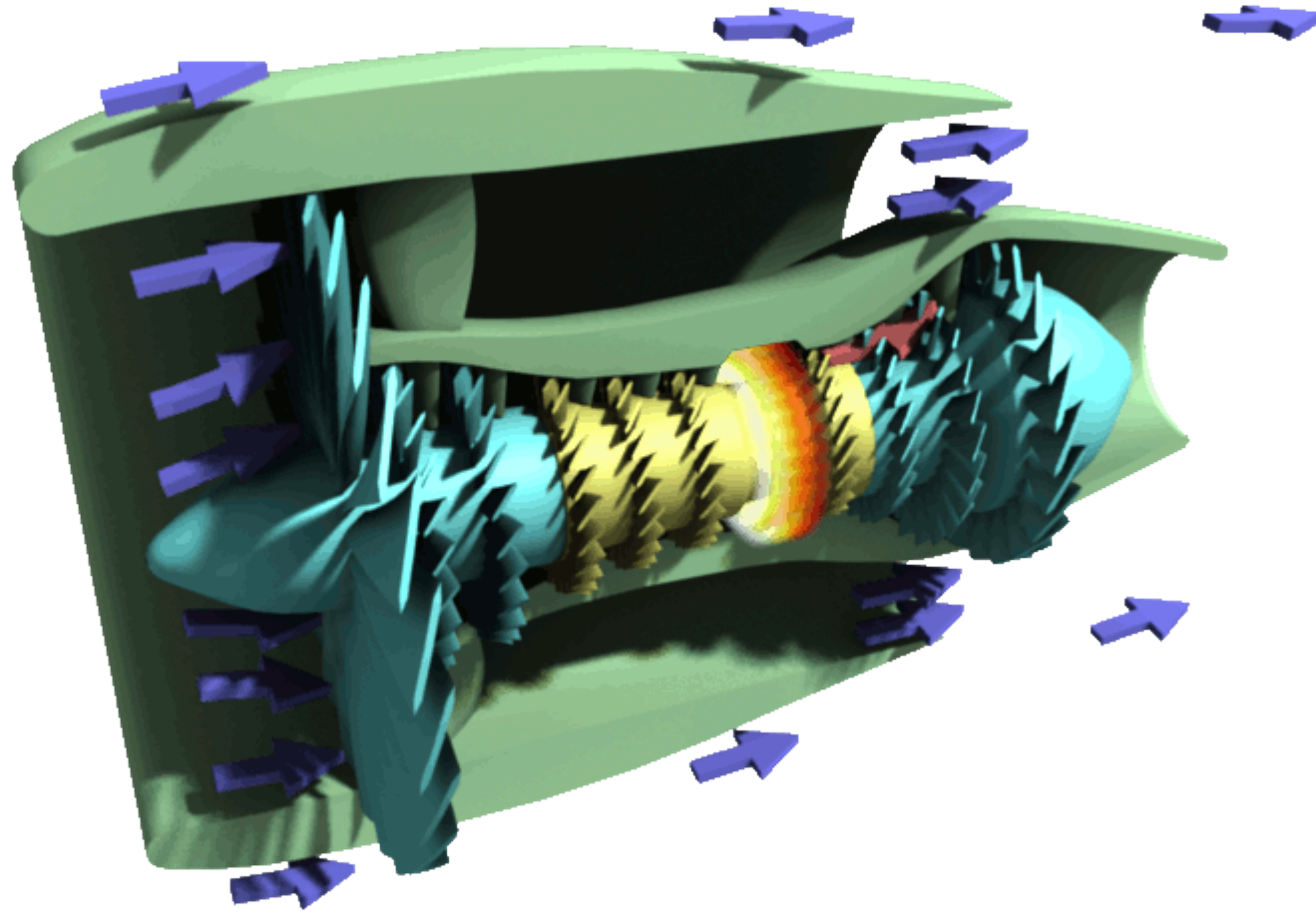
Comparando com o Boeing 747 ...



Configurações e motores que equipam o Boeing 747

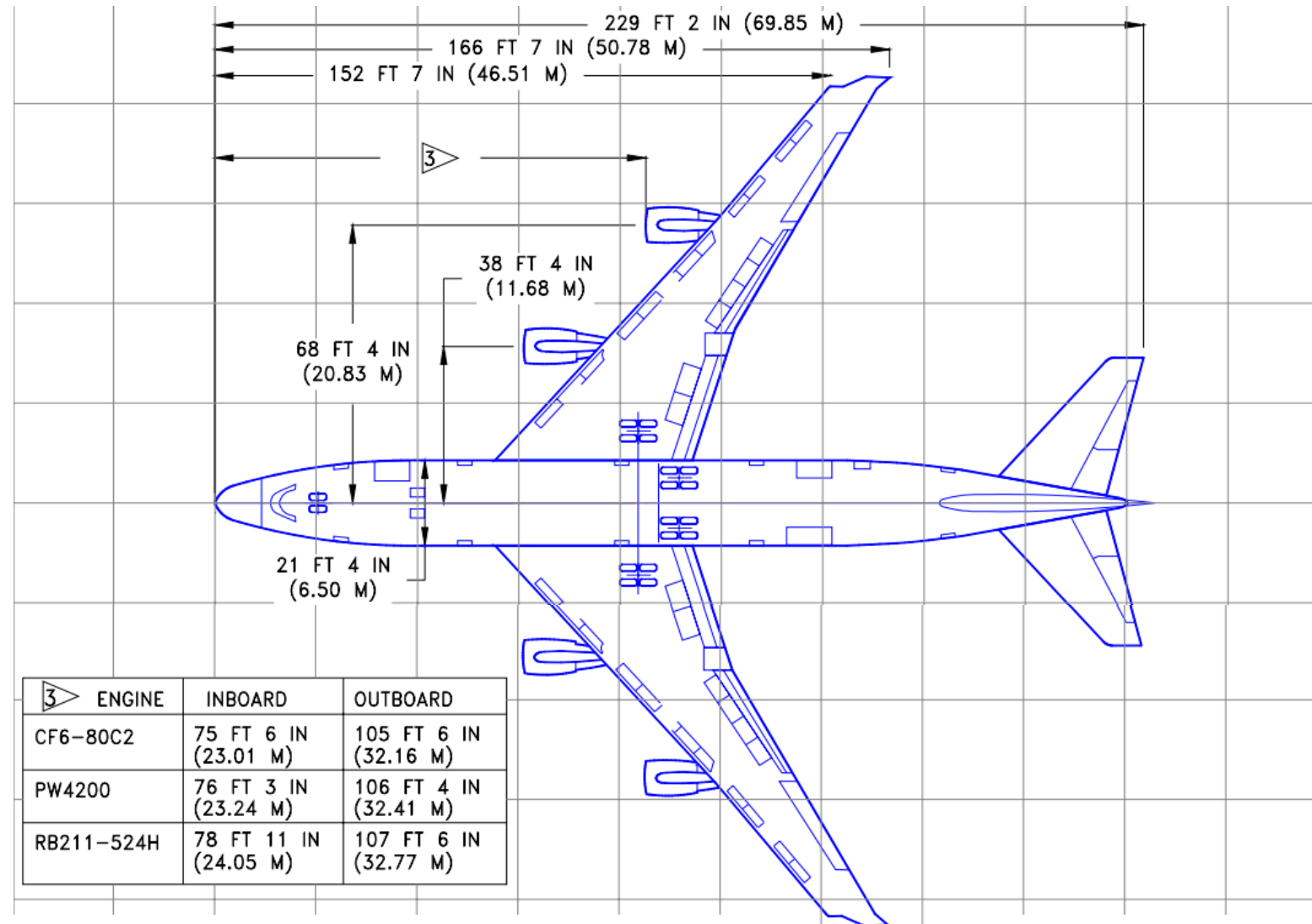


Configurações e motores que equipam o Boeing 747



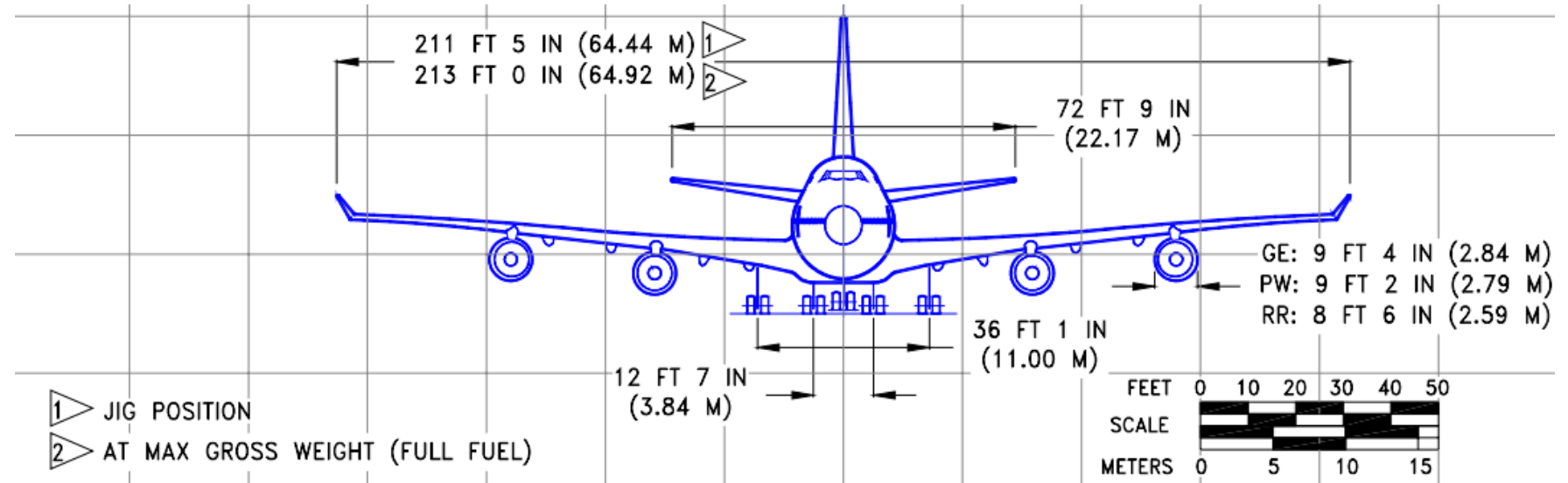
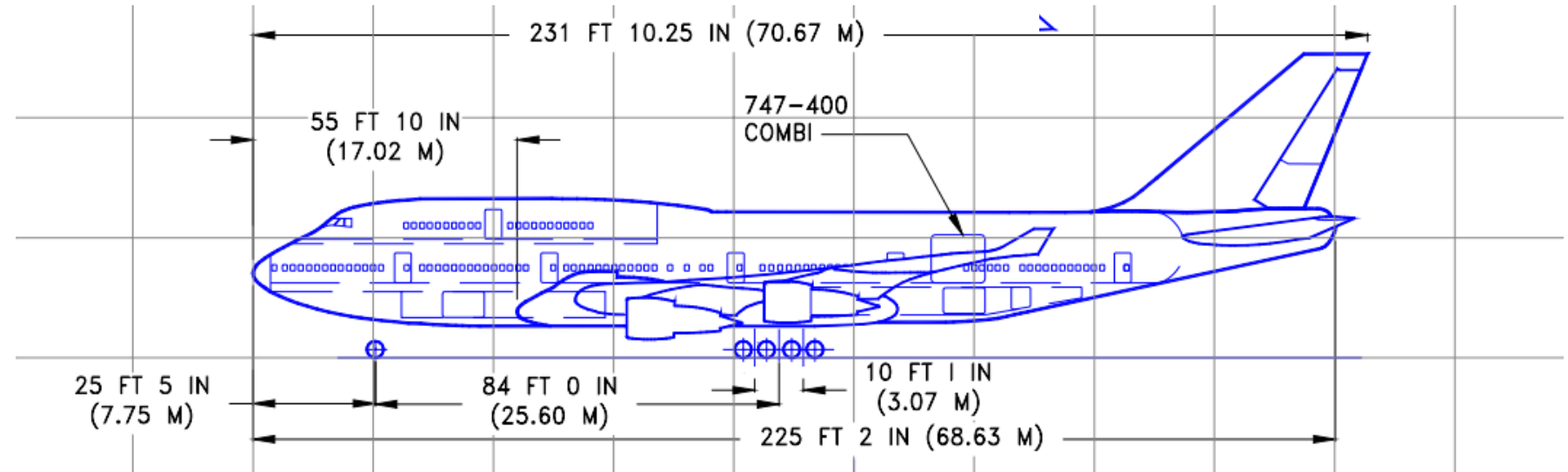
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Configurações e motores que equipam o Boeing 747



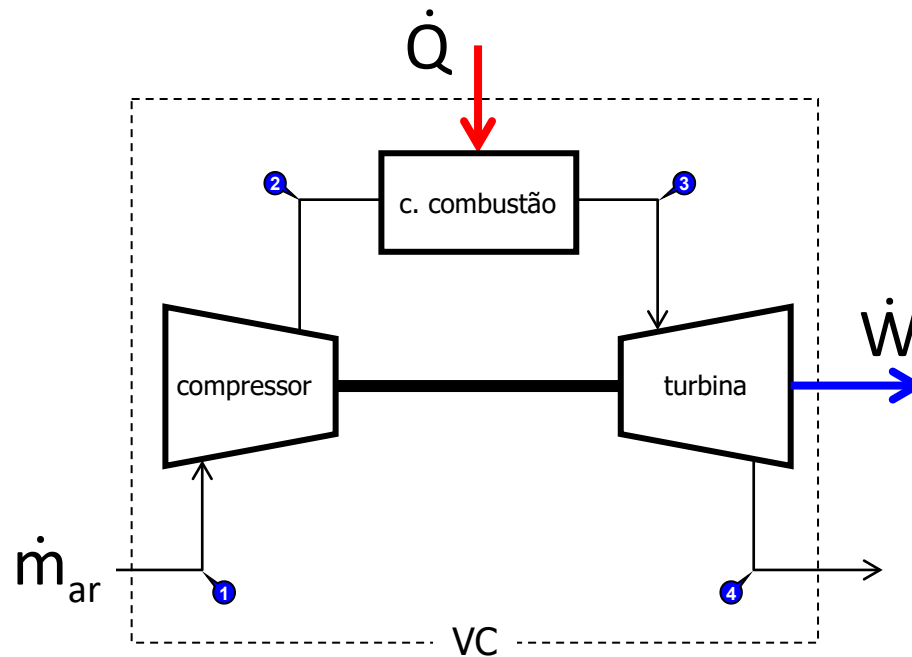
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Configurações e motores que equipam o Boeing 747



Configurações e motores que equipam o Boeing 747

Measurement	747-100B	747-200B	747-300	747-400 747-400ER	747-8I
Maximum cargo capacity	6,190 ft ³ (175.3 m ³) = 30xLD1			6,025 ft ³ (170.5 m ³) ER: 5,599 ft ³ (158.6 m ³)	5,705 ft ³ (161.5 m ³)
Operating empty weight	358,000 lb (162,400 kg)	383,000 lb (174,000 kg)	392,800 lb (178,100 kg)	393,263 lb (178,756 kg) ER: 406,900 lb (184,600 kg)	472,900 lb (214,503 kg)
Maximum takeoff weight	735,000 lb (333,390 kg)	833,000 lb (377,842 kg)		875,000 lb (396,890 kg) ER: 910,000 lb (412,775 kg)	987,000 lb (447,696 kg)
Cruising speed (at 35,000 ft (11,000 m) altitude)	Mach 0.84 (555 mph, 893 km/h, 481 knots)			Mach 0.85 (567 mph, 913 km/h, 493 kn) ER: Mach 0.855 (570 mph, 918 km/h, 495 kn)	Mach 0.855 (570 mph, 918 km/h, 495 kn)
Maximum speed	Mach 0.89 (594 mph, 955 km/h, 516 kn)			Mach 0.92 (614 mph, 988 km/h, 533 kn)	
Maximum fuel capacity	48,445 U.S. gal (40,339 imp gal/183,380 L)	52,410 U.S. gal (43,640 imp gal/199,158 L)		57,285 U.S. gal (47,700 imp gal/216,840 L) ER: 63,705 U.S. gal (53,045 imp gal/241,140 L)	64,225 U.S. gal (53,478 imp gal/243,120 L)
Engine models (x 4)	PW JT9D-7A/-7F/-7J RR RB211-524B2 GE CF6-45A2 (747SR)	PW JT9D-7R4G2 GE CF6-50E2 RR RB211-524D4	PW JT9D-7R4G2 GE CF6-80C2B1 RR RB211-524D4	PW 4062 GE CF6-80C2B5F RR RB211-524G/H ER: GE CF6-80C2B5F	GEEnx-2B67
Engine thrust (per engine) estático	PW 46,500 lbf (207 kN) RR 50,100 lbf (223 kN)	PW 54,750 lbf (244 kN) GE 52,500 lbf (234 kN) RR 53,000 lbf (236 kN)	PW 54,750 lbf (244 kN) GE 55,640 lbf (247 kN) RR 53,000 lbf (236 kN)	PW 63,300 lbf (282 kN) GE 62,100 lbf (276 kN) RR 59,500/60,600 lbf (265/270 kN) ER: GE 62,100 lbf (276 kN)	66,500 lbf (296 kN)



Consumo de um motor: 0,8kg/s de querosene

$$\dot{Q} = \dot{m}_{\text{comb}} \cdot (\eta_{\text{comb}} \text{PCI}_{\text{comb}})$$

$$\dot{Q} = 0,8 \text{kg/s} \cdot (0,333 \cdot 44,19 \text{MJ/kg})$$

$$\dot{Q} = 11,7722 \text{MW}$$

Inventário de energia no volume de controle

$$\dot{Q} - \dot{W} = \dot{m}_{\text{ar}} \cdot \left[\left(h + \cancel{gz} + \frac{\cancel{v^2}}{2} \right)_s - \left(h + \cancel{gz} + \frac{\cancel{v^2}}{2} \right)_e \right]$$

$$\dot{Q} - \dot{W} = \dot{m}_{\text{ar}} \cdot (h_4 - h_1) \quad m_{\text{C}_{12}\text{H}_{26}} = 0,066705 \cdot m_{\text{ar}}$$

$$11,7722 \text{MW} - \dot{W} = \frac{0,8 \text{kg/s}}{0,066705} \cdot (958,6 - 424,5) \cdot 10^{-3} \text{MJ/kg}$$

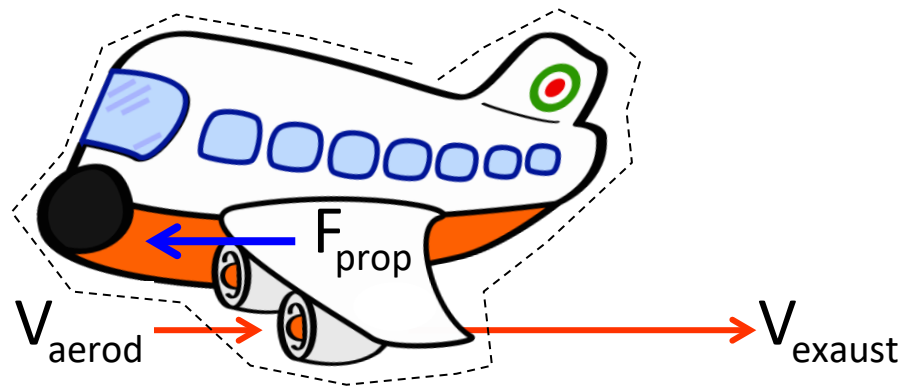
$$\dot{W} = 5,367 \text{MW}$$

$$\eta = \dot{W} / \dot{Q}$$

$$\eta = 45,6\%$$



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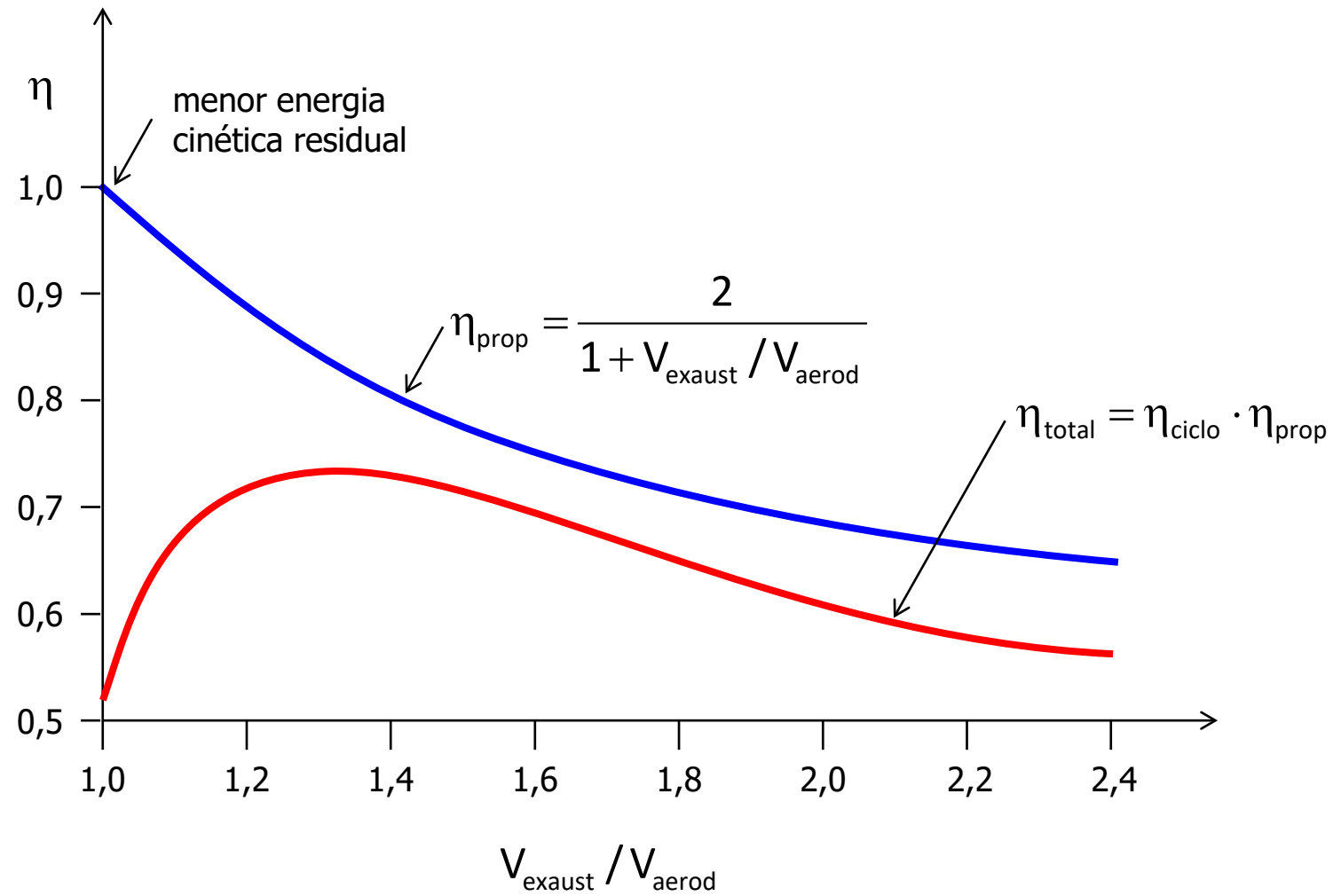
$$\eta_{\text{prop}} = \frac{F_{\text{prop}} \cdot V_{\text{aerod}}}{\frac{\dot{m}_{\text{exhaust}} V_{\text{exhaust}}^2}{2} - \frac{\dot{m}_{\text{aspirac}} V_{\text{aerod}}^2}{2}}$$

$$F_{\text{prop}} \cong \dot{m}_{\text{ar}} (V_{\text{exhaust}} - V_{\text{aerod}}) \quad \dot{m}_{\text{aspirac}} \cong \dot{m}_{\text{ar}} \cong \dot{m}_{\text{exhaust}}$$

$$\eta_{\text{prop}} = \frac{2}{1 + \frac{V_{\text{exhaust}}}{V_{\text{aerod}}}}$$

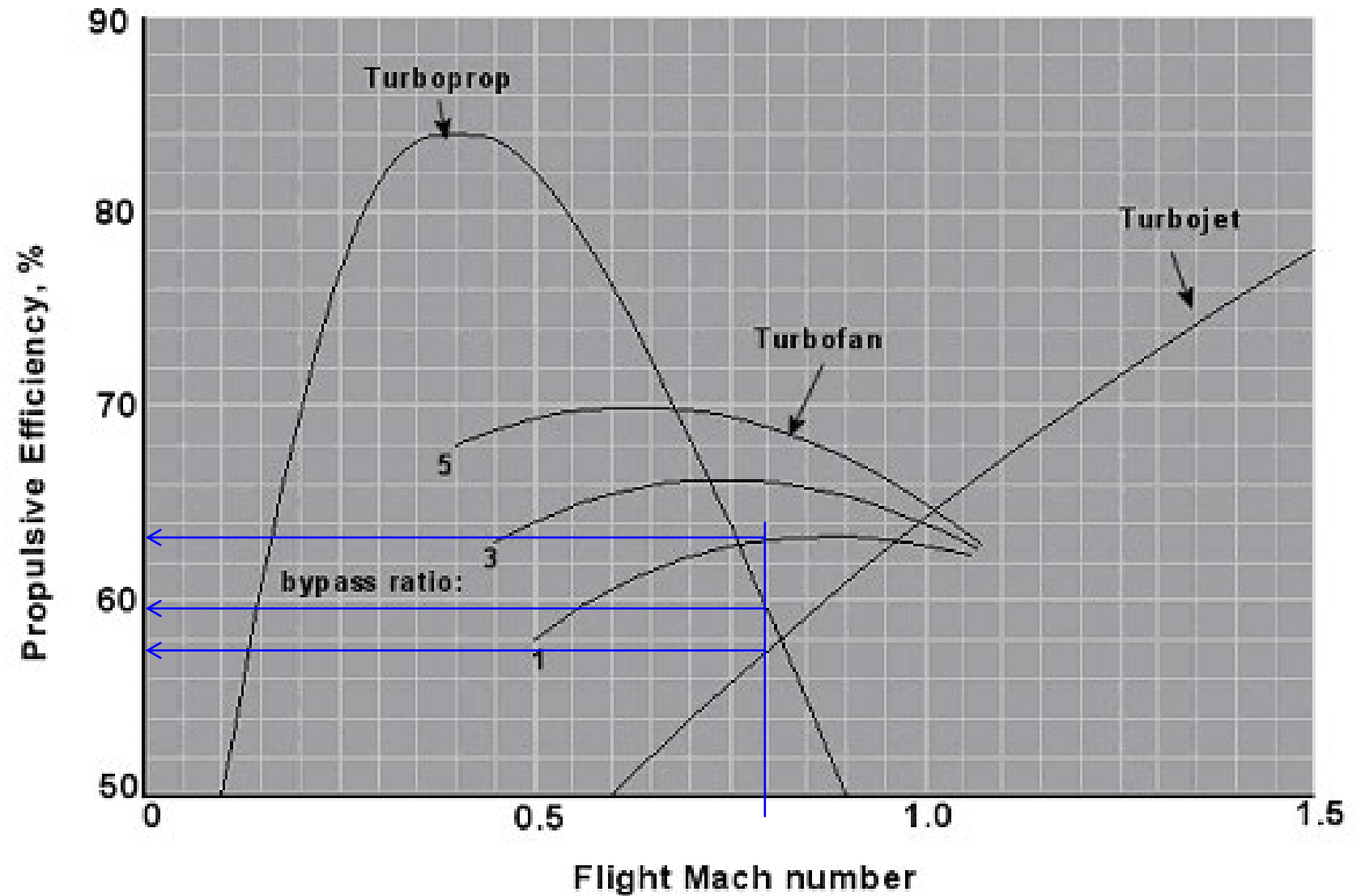


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Balanco de Energia em VC em regime
transiente....

Lançamento do Saturno V / Apollo



<https://youtu.be/f3NhpIu10T8>