

APPENDIX E

FIGURES

ISOTHERMAL TRANSFORMATION FOR AISI 1050 STEEL

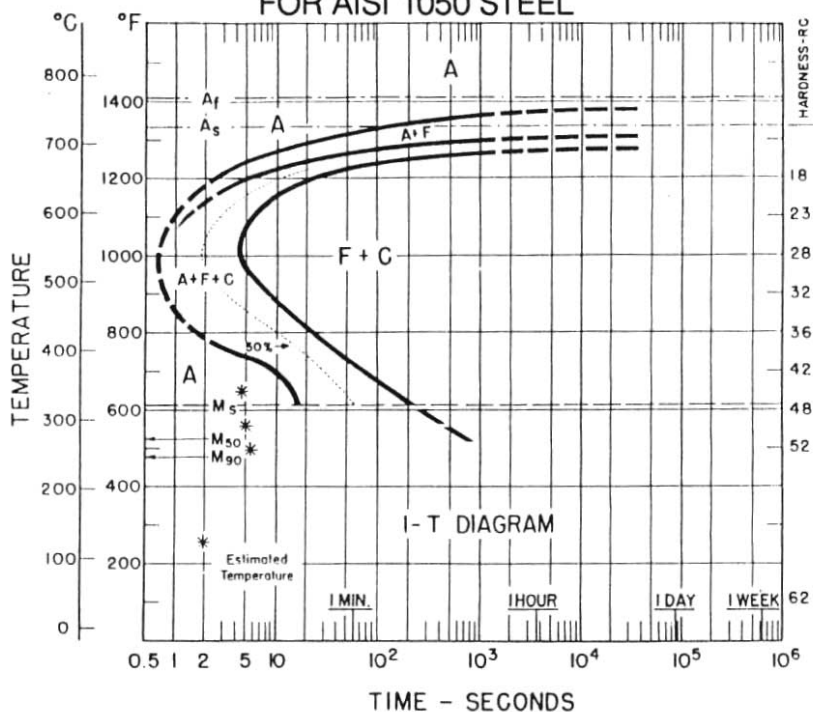


Figure E.1 Isothermal Transformation (I-T) diagram for AISI 1050 Steel. (From the Atlas of Isothermal Transformation and Cooling Transformation Diagrams, 1977, p. 15. Reprinted by permission of ASM International, Metals Park, Ohio.)

ΔG^0 FOR METAL OXIDES AS A FUNCTION OF TEMPERATURE

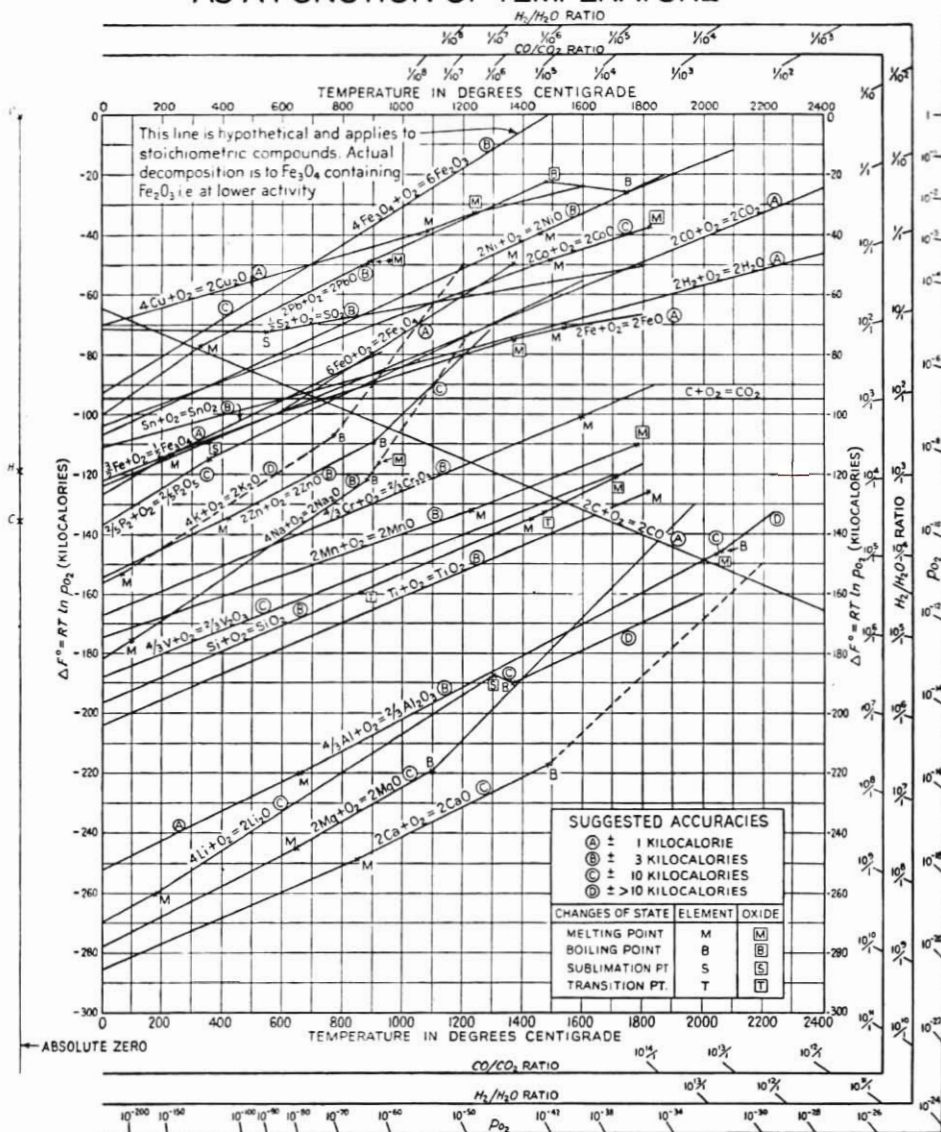


Figure E.2 The standard free energy of formation of metal oxides as a function of temperature. (From L.S. Darken and R.W. Gurry, 1953, Physical Chemistry of Metals, Fig. 14.4. Reprinted by permission of McGraw-Hill, Inc., New York.)

ΔG^0 (ΔF_T^0) FOR METAL SULFIDES AS A FUNCTION OF TEMPERATURE

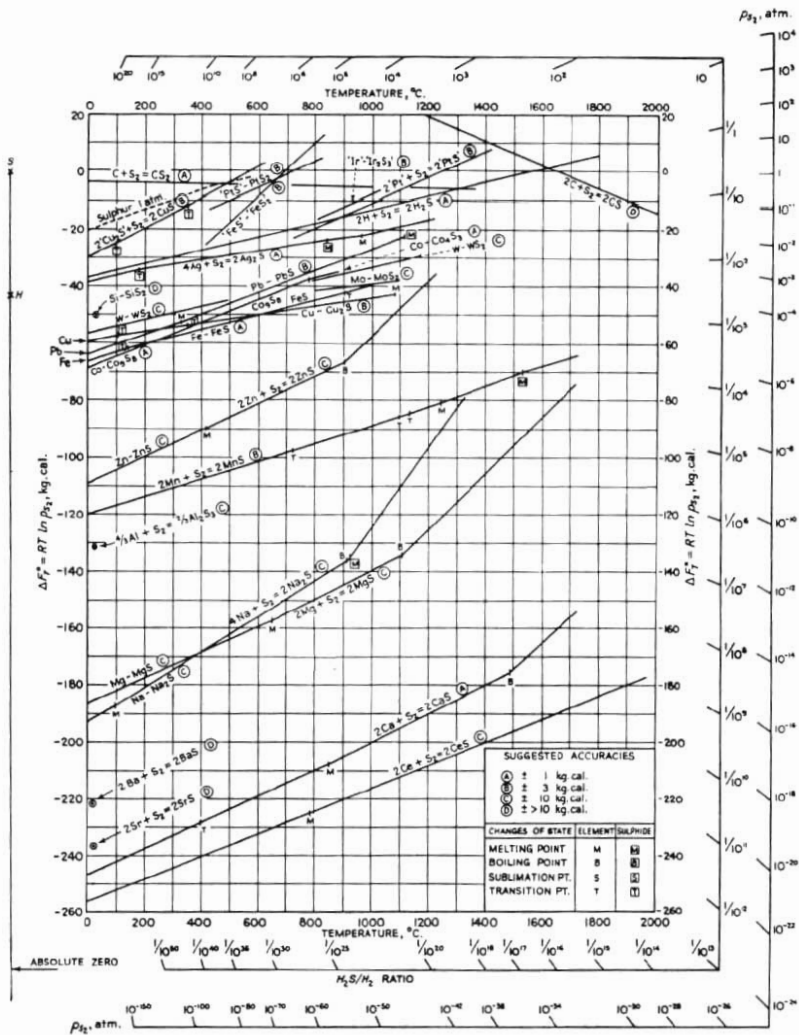


Figure E.3 The standard free energy of formation of metal sulfides as a function of temperature. (From L.S. Darken and R.W. Gurry, 1953, *Physical Chemistry of Metals*, Fig. 14-9. Reprinted by permission of McGraw-Hill, Inc., New York.)

LOG $P_{N_2}^{1/2}$ VS. $10^4/T$
FOR SELECT METAL-NITRIDE SYSTEMS

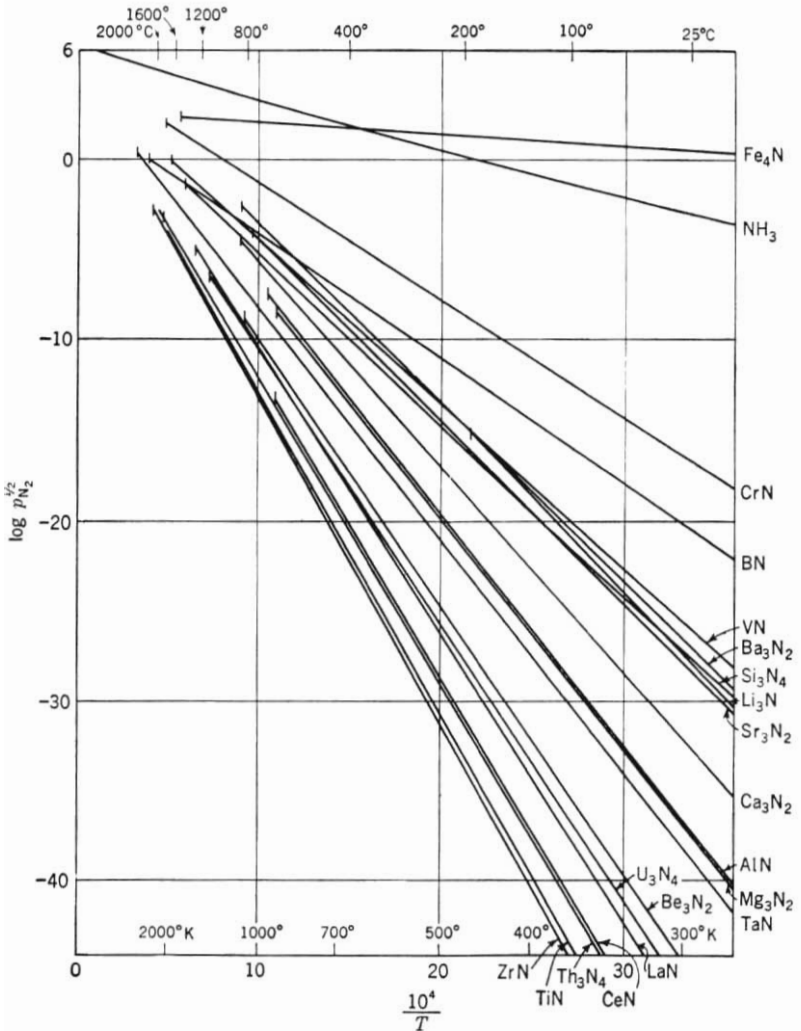


Figure E.4 Log $P_{N_2}^{1/2}$ vs. $10^4/T$ for select metal-nitride systems. (From L.S. Darken and R.W. Gurry, 1953, Physical Chemistry of Metals, Fig. 14-11. Reprinted by permission of McGraw-Hill, Inc., New York.)

E^0 AND $-\Delta F^0$ ($-\Delta G^0$) FOR SELECT CHLORIDES AS A FUNCTION OF TEMPERATURE

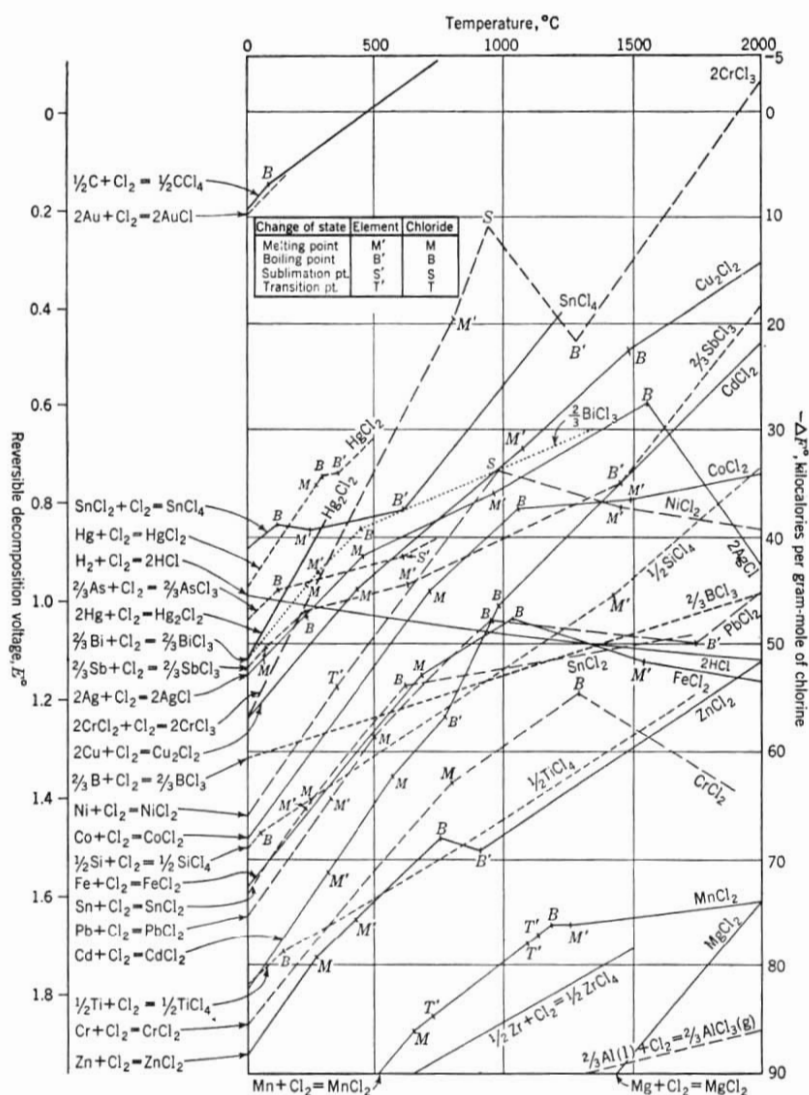


Figure E.5 The standard free energy of formation and reversible decomposition voltage of select chlorides as a function of temperature. (From L.S. Darken and R.W. Gurry, 1953, *Physical Chemistry of Metals*, Fig. 14-12. Reprinted by permission of McGraw-Hill, Inc., New York.)

ALUMINUM-SILICON EUTECTIC PHASE DIAGRAM

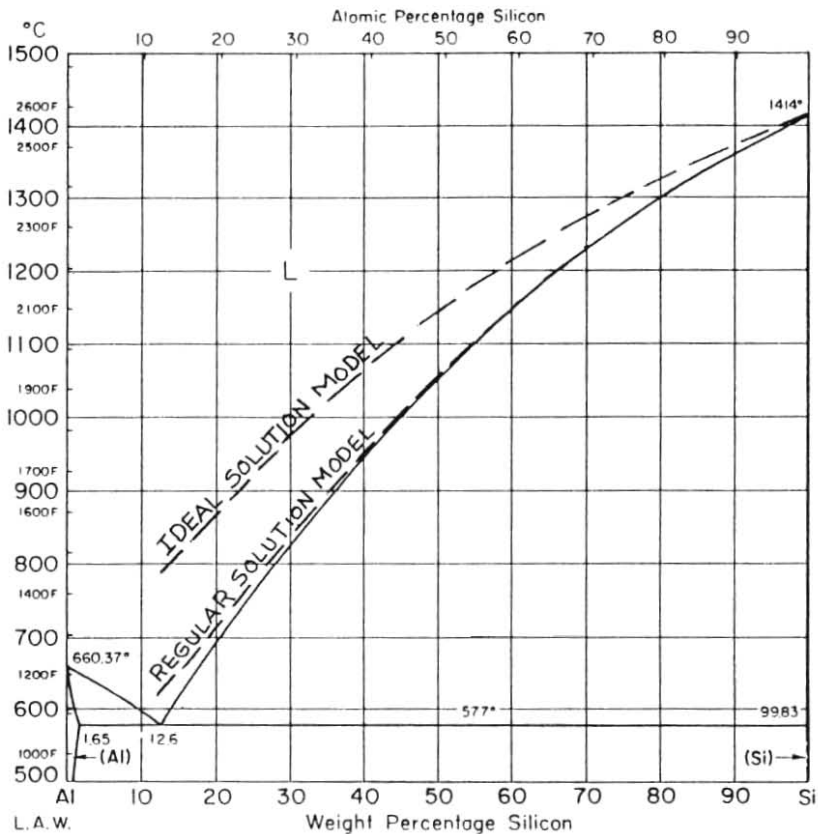


Figure E.6 Al-Si Eutectic Phase Diagram: No solid solubility. (From the Metals Handbook, 8th edition, v. 8, Metallurgy, Structures, and Phase Diagrams, 1973, p. 263. Reprinted by permission of ASM International, Metals Park, Ohio.)

SILVER-COPPER EUTECTIC PHASE DIAGRAM

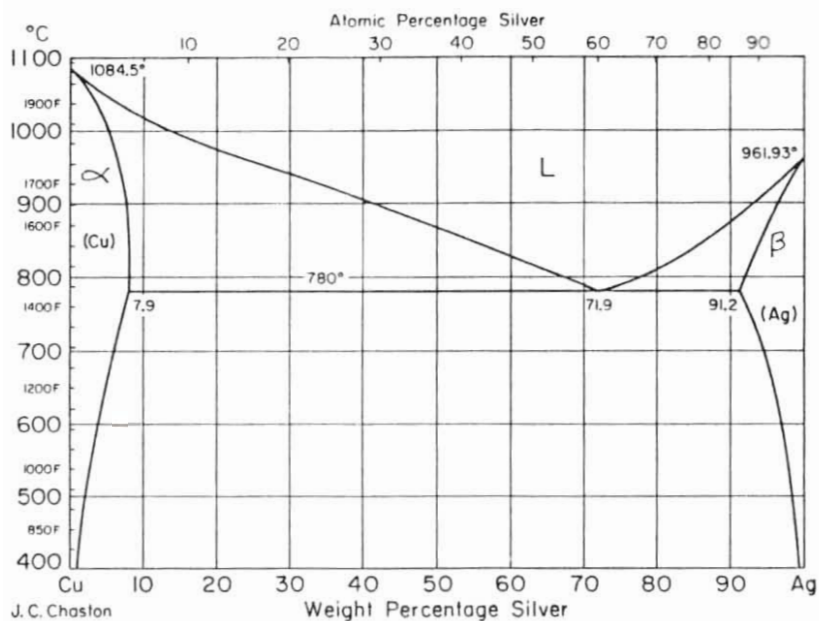


Figure E.7 Ag-Cu Eutectic Phase Diagram: Terminal Solid Solubility. (From the Metals Handbook, 8th edition, v. 8, Metallography, Structures, and Phase Diagrams, 1973, p. 253. Reprinted by permission of ASM International, Metals Park, Ohio.)

BISMUTH-LEAD PHASE DIAGRAM

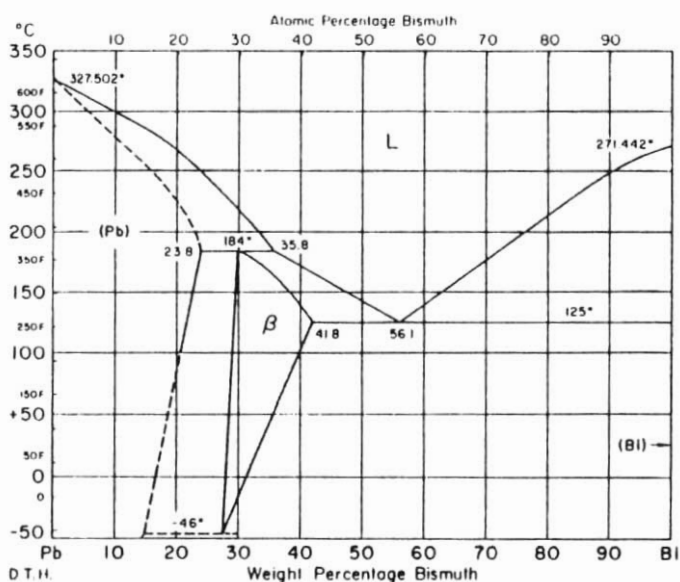


Figure E.8 Bi-Pb Phase Diagram. (From the Metals Handbook, 8th edition, v. 8, Metallurgy, Structures, and Phase Diagrams, 1973, p. 273. Reprinted by permission of ASM International, Metals Park, Ohio.)

LEAD-ANTIMONY PHASE DIAGRAM

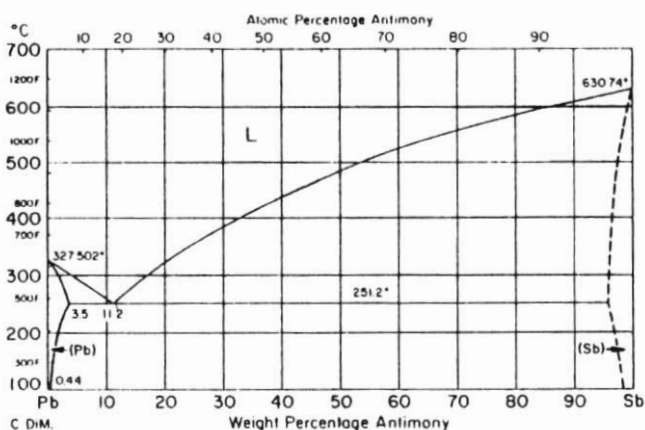


Figure E.9 Pb-Sb Phase Diagram. (From the Metals Handbook, 8th edition, v. 8, Metallurgy, Structures, and Phase Diagrams, 1973, p. 329. Reprinted by permission of ASM International, Metals Park, Ohio.)

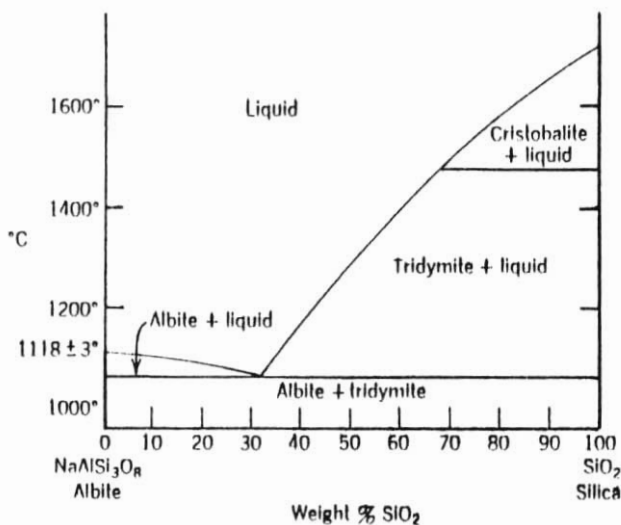
NaAlSi₃O₈-SiO₂ EUTECTIC PHASE DIAGRAM

Figure E.10 NaAlSi₃O₈-SiO₂ eutectic phase diagram: No solid solubility. (From C. Klein and C.S. Hurlbut, Jr., 1985, *Manual of Mineralogy*, Fig. 12.5. Reprinted by permission of John Wiley and Sons, Inc., Copyright © 1985.)

DIAMOND-GRAPHITE UNIVARIANT PHASE DIAGRAM

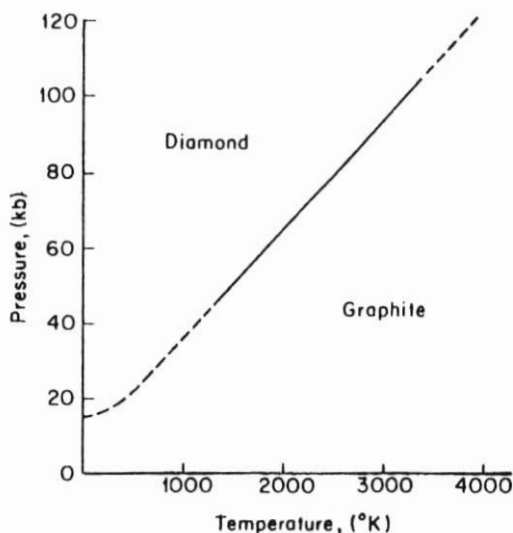


Figure E.11 Diamond-graphite univariant P-T phase diagram. (From E.G. Ehlers, 1972, *The Interpretation of Geological Phase Diagrams*, Fig. 93. Reprinted by permission of W.H. Freeman and Company.)