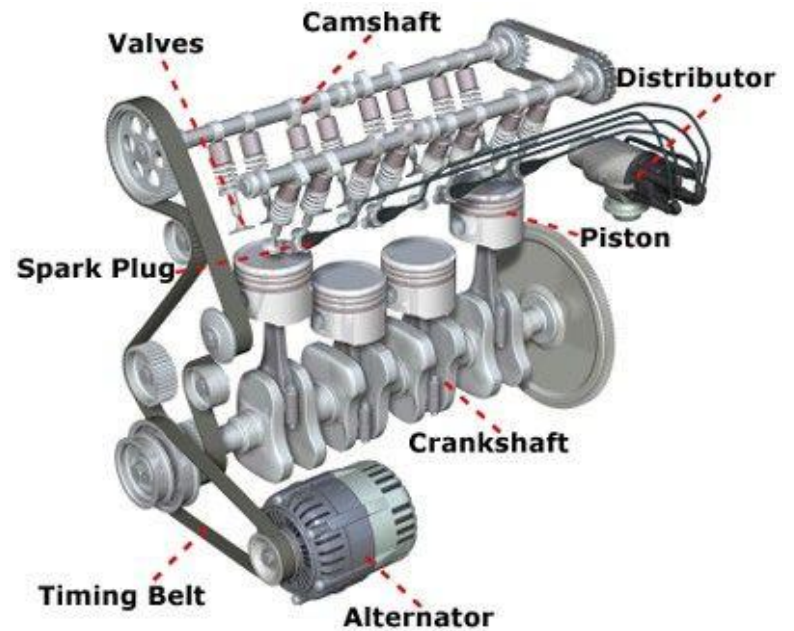


PSI 3561 – Eletrônica Automotiva

Aula 1

INTRODUÇÃO

Motores à Combustão



1. Conceito Fundamental

Motores térmico são máquinas que transformam energia térmica em energia mecânica.

2. Classificação dos motores quanto à combustão:

Motores à combustão interna

Motores à combustão externa

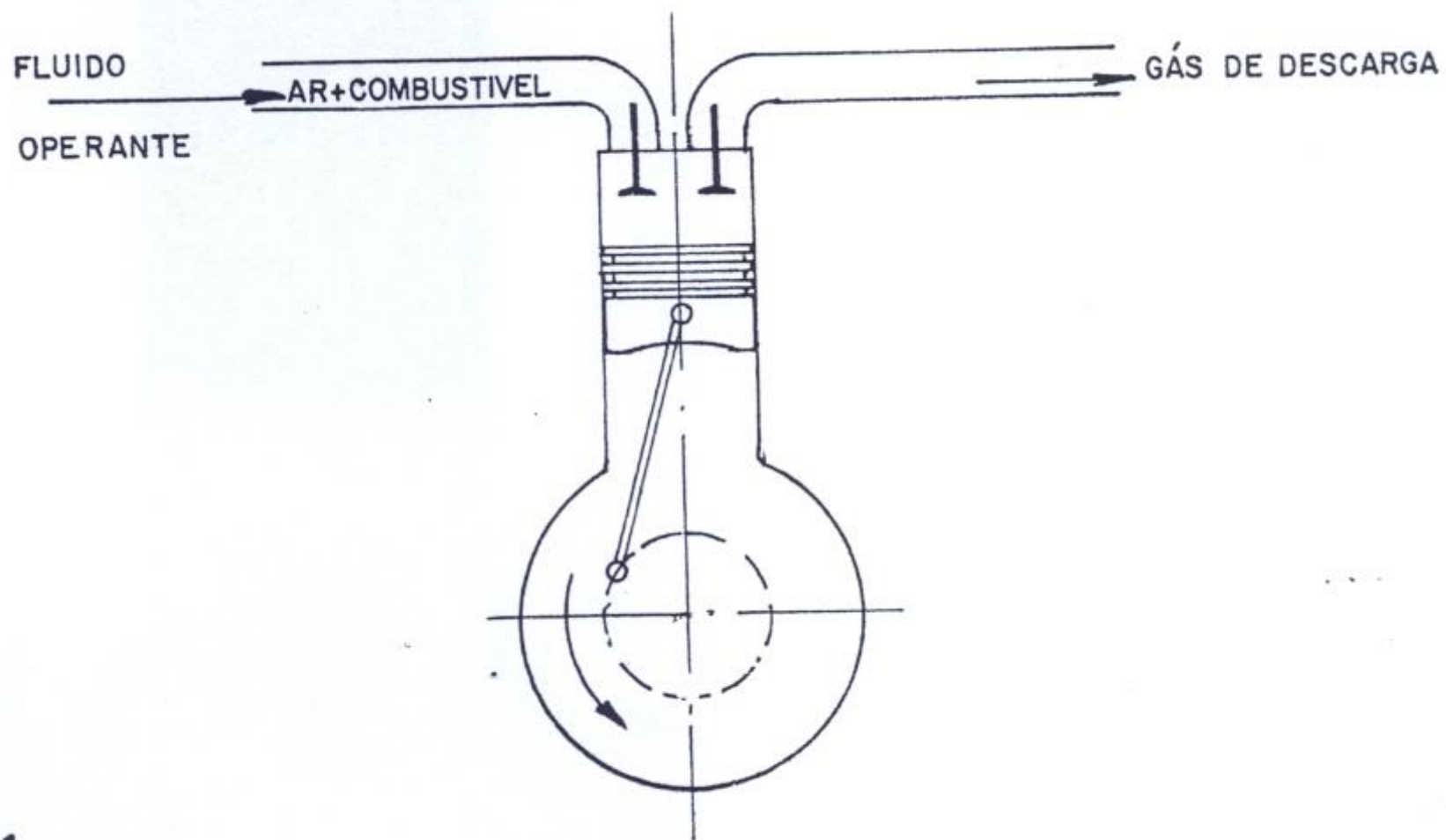


Fig. 1
OS MOTORES

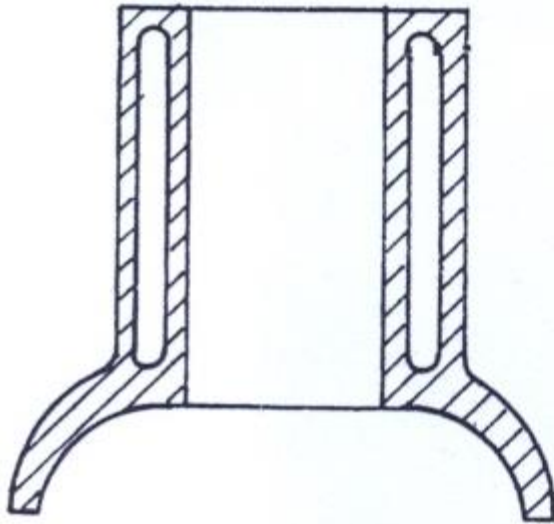
3.PRINCIPAIS COMPONENTES

3.1 Componentes (órgãos) fixos:

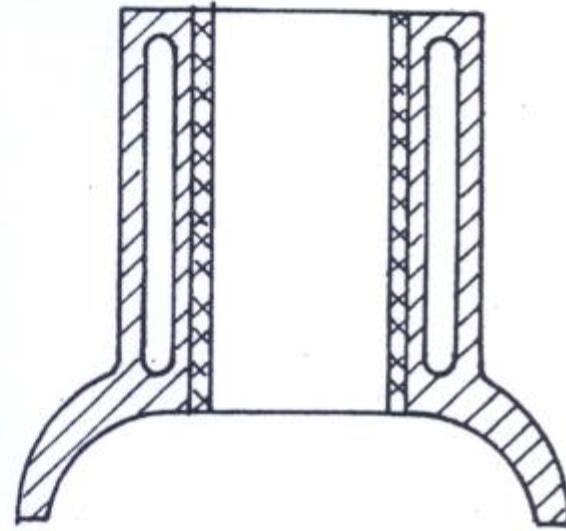
- **Cilindro**
- **Bloco**
- **Cárter**
- **Cabeçote**
- **Câmara de combustão**
- **Sede de válvula**
- **Guia de Válvula**

3.2 Componentes (órgãos) móveis:

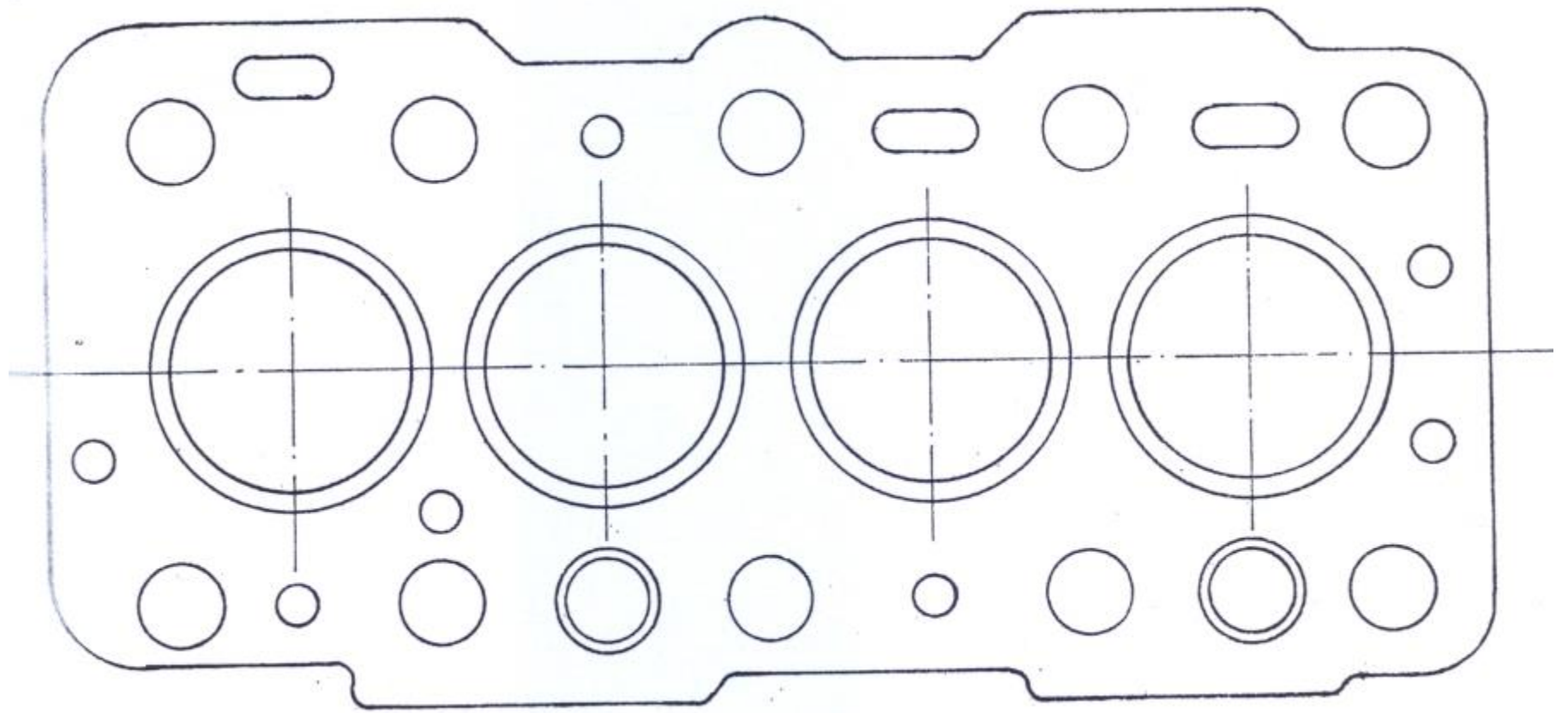
- **Pistão**
- **Pino munhão**
- **Anéis de segmento**
- **Biela**
- **Árvore de manivela**
- **Volante**
- **Casquilho**
- **Válvula**
- **Mola de válvula**
- **Eixo comando de válvula**

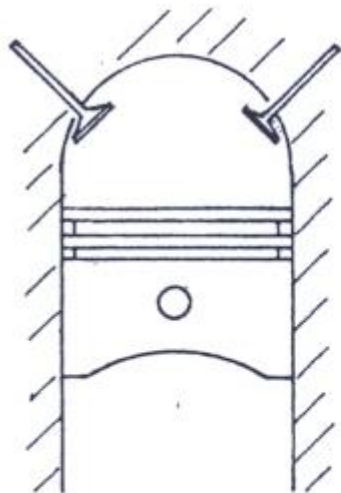


Cilindro fixo

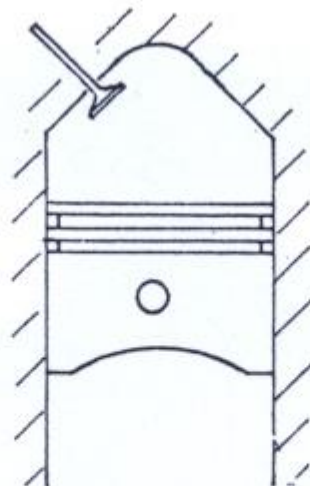


Cilindro com camisa
tipo substituível a seco

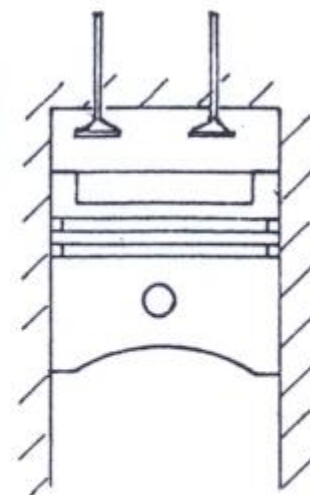




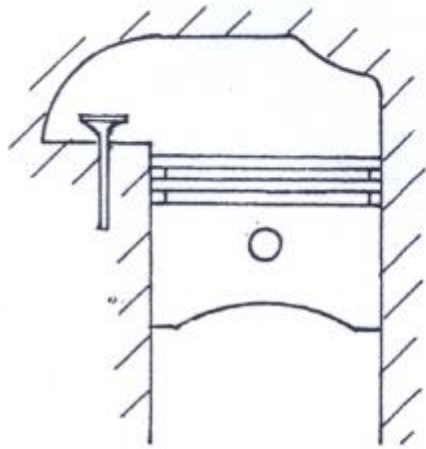
ESFÉRICA



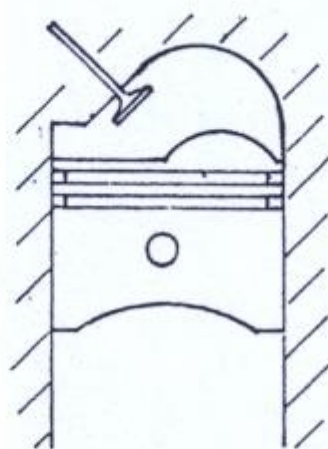
TRIANGULAR



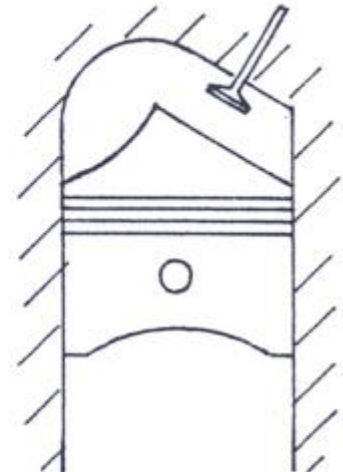
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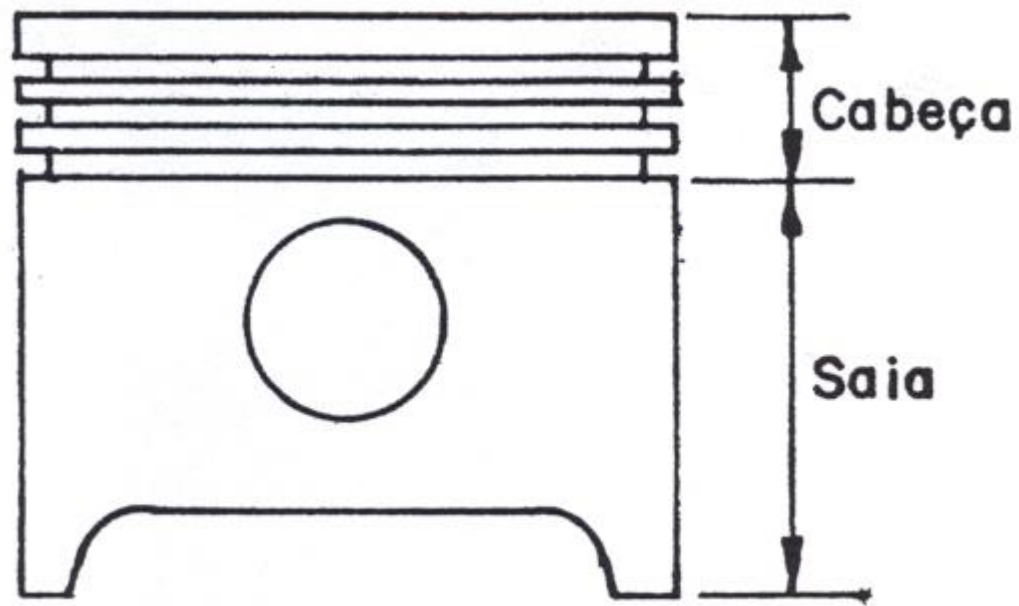
COM VÁLVULA
LATERAL

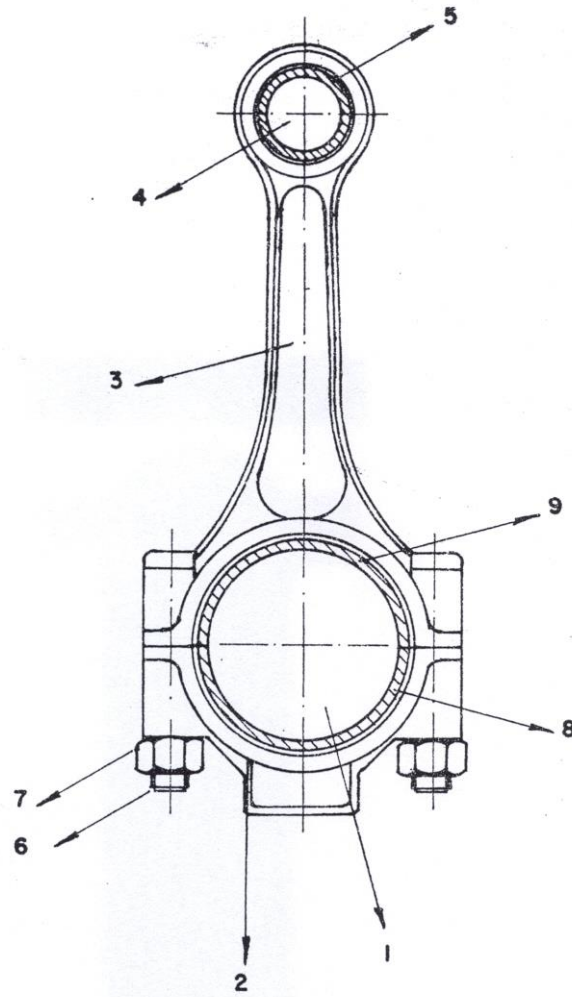


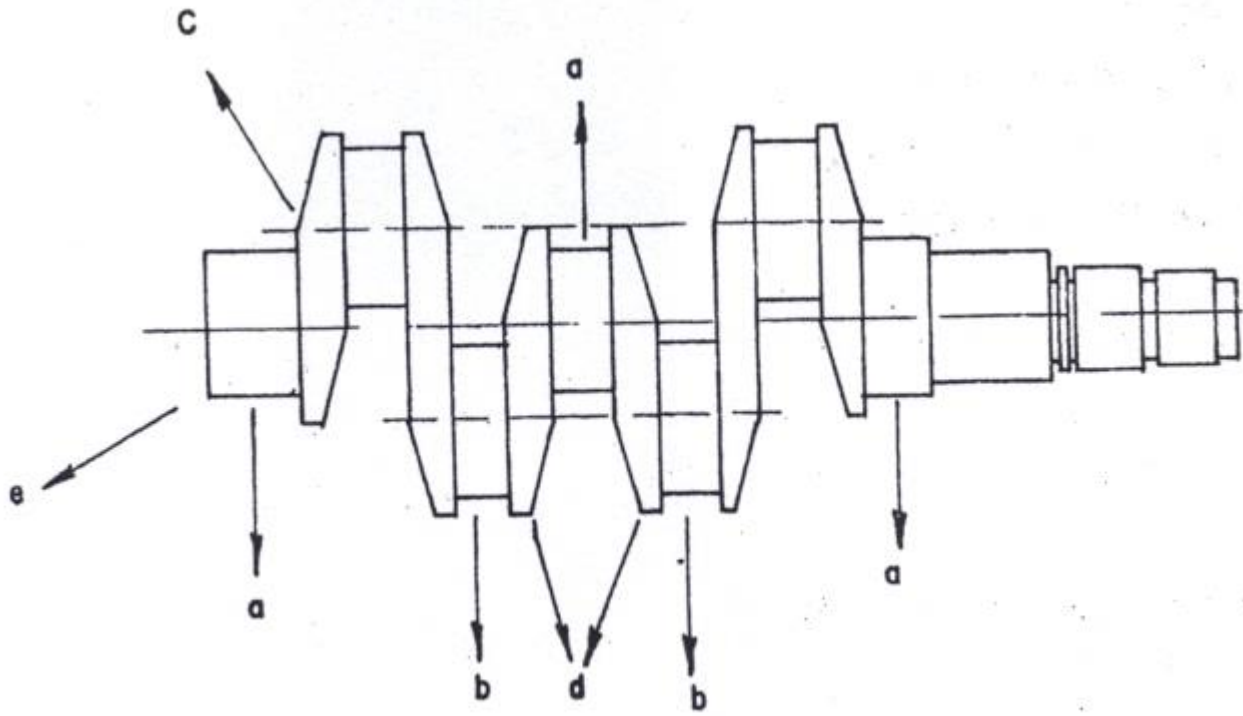
A BANHEIRA

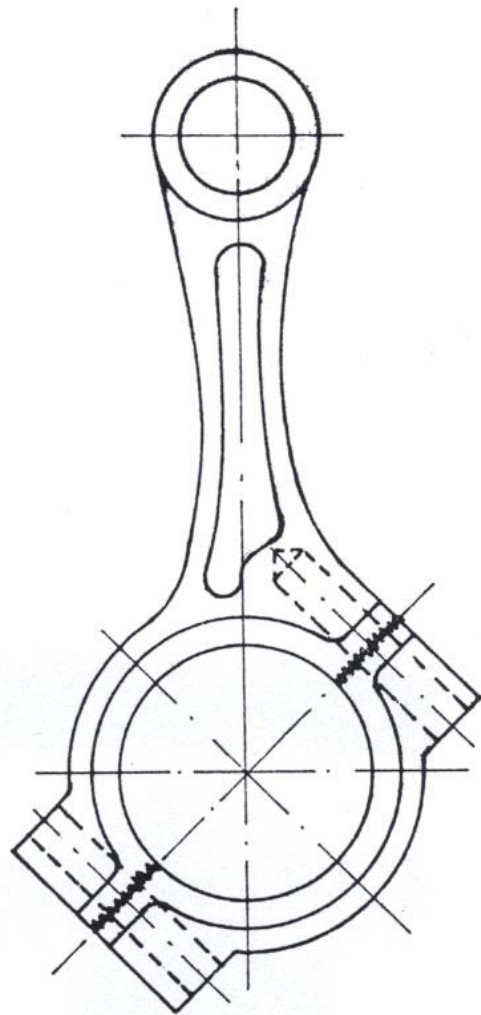


TRAPESOIDAL

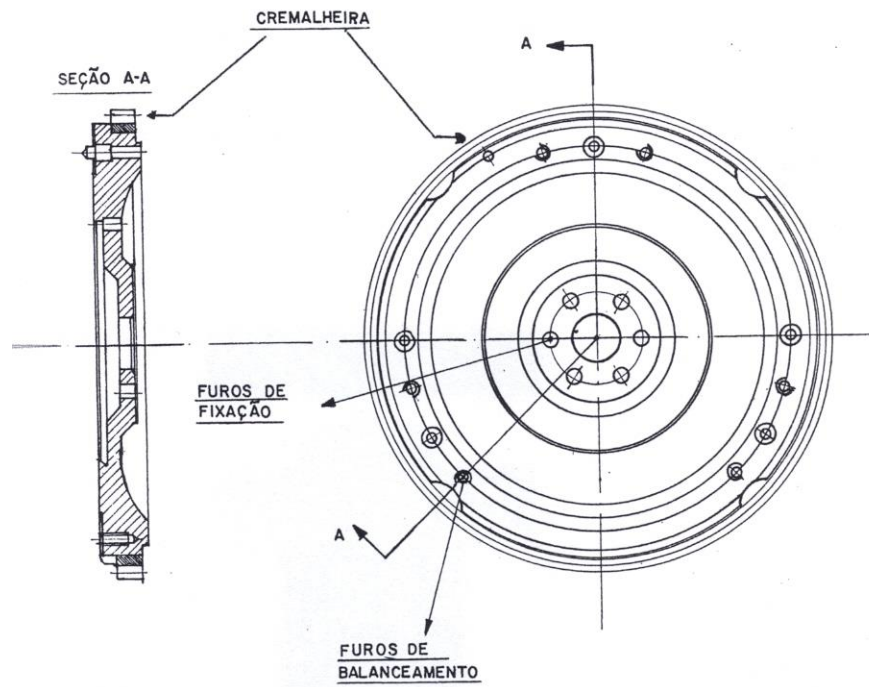


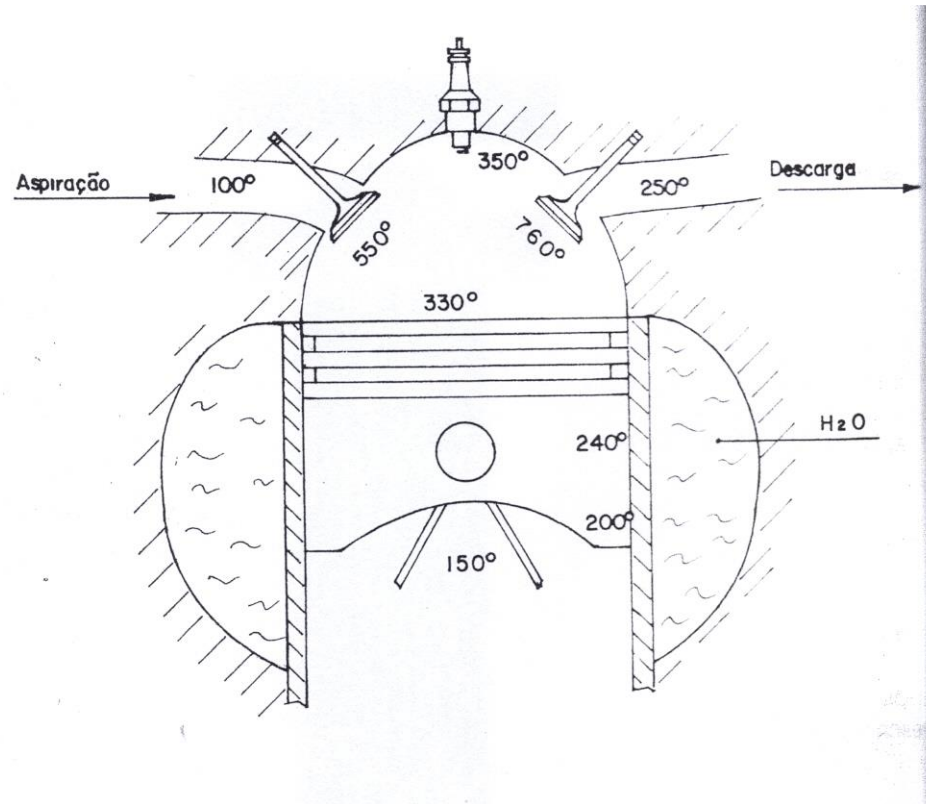


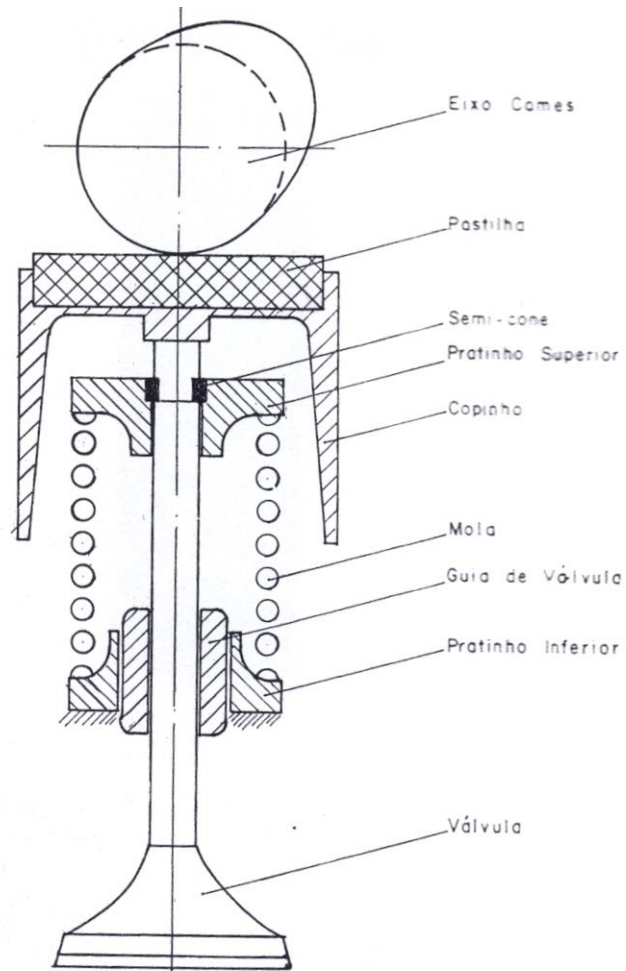


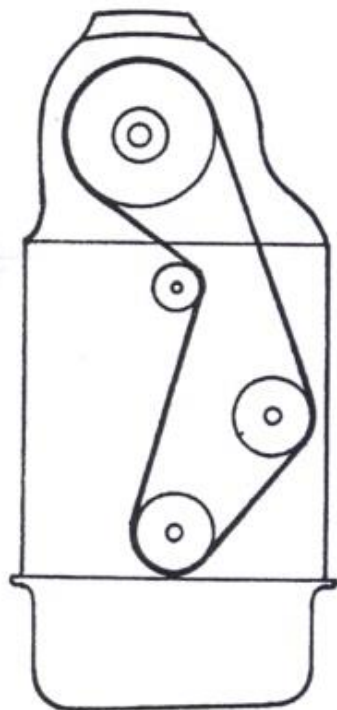


Árvore de manivela



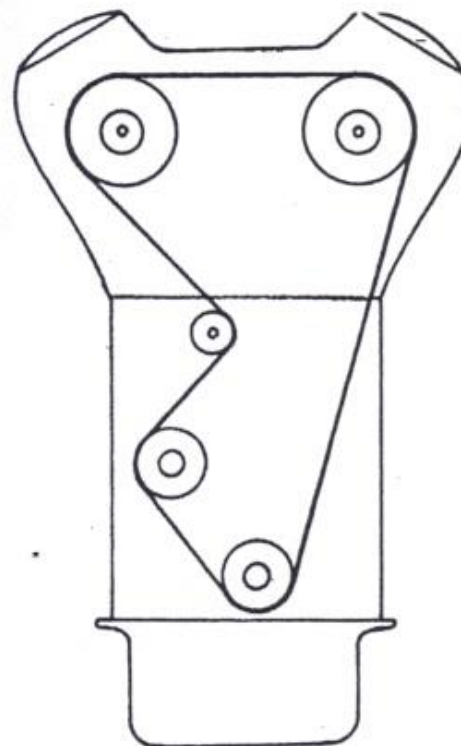






COMANDO SIMPLES

ITORES



COMANDO DUPLO

4. MOVIMENTO DO CILINDRO

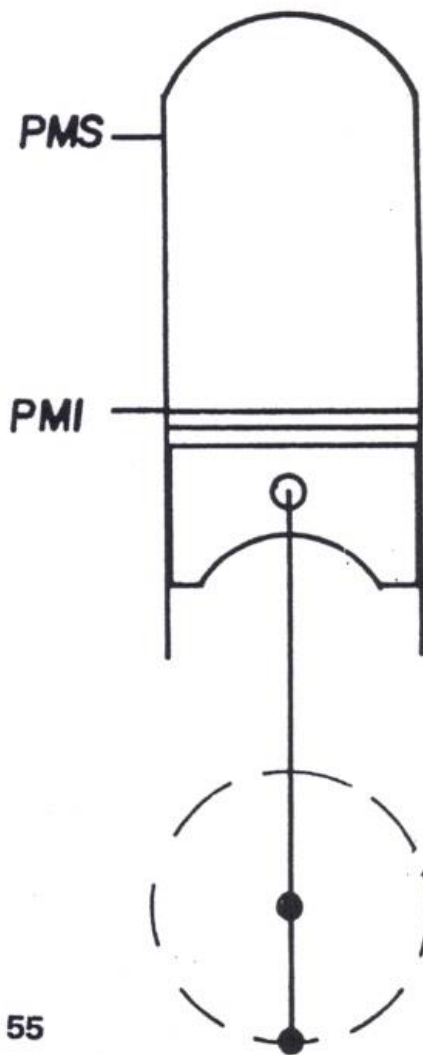
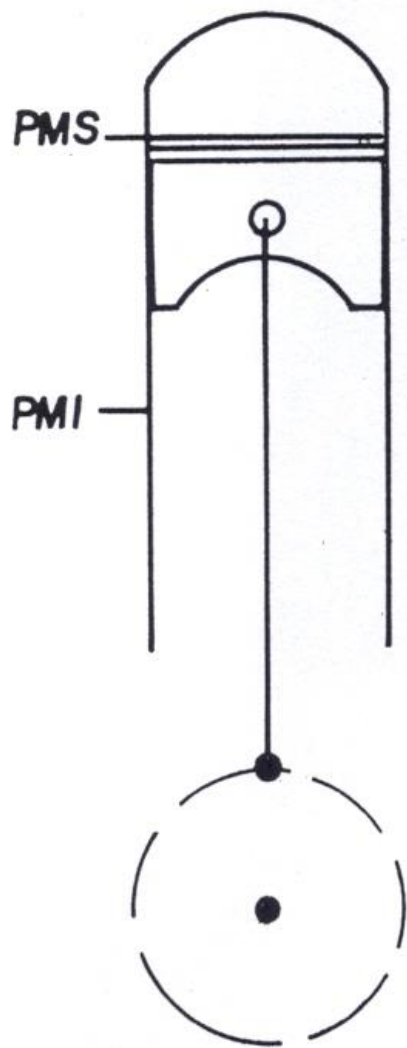


Fig. 55

5. DEFINIÇÕES

6. CICLOS OPERATIVOS

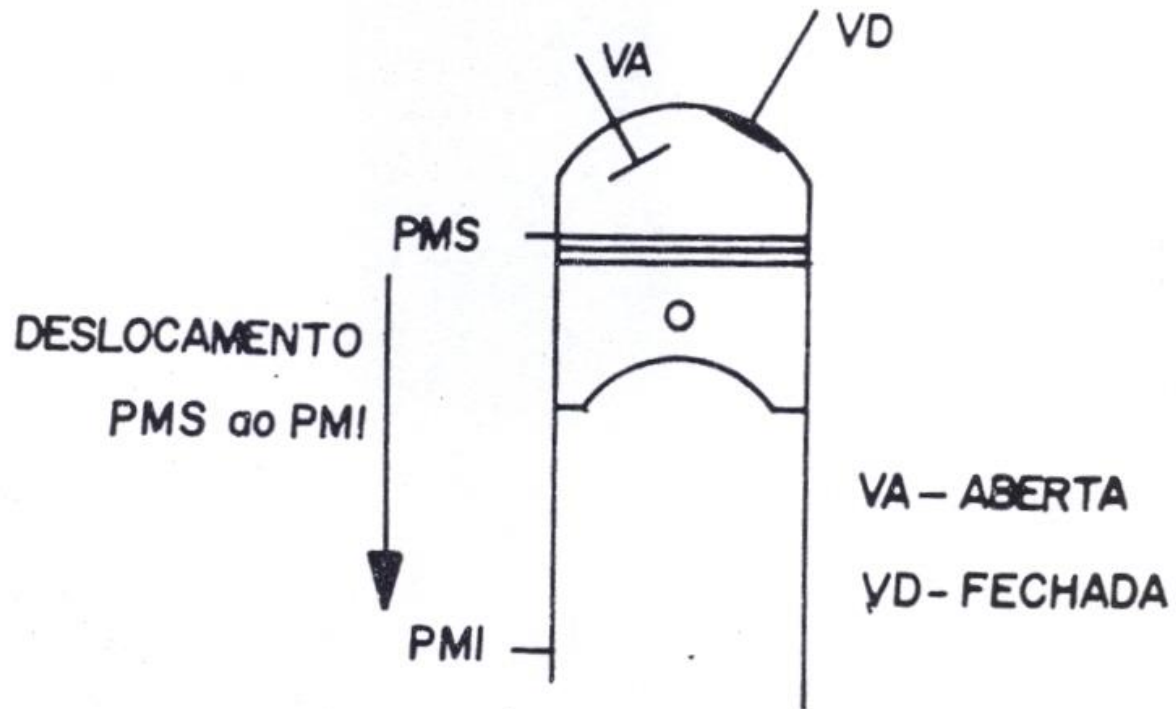


Fig. 59

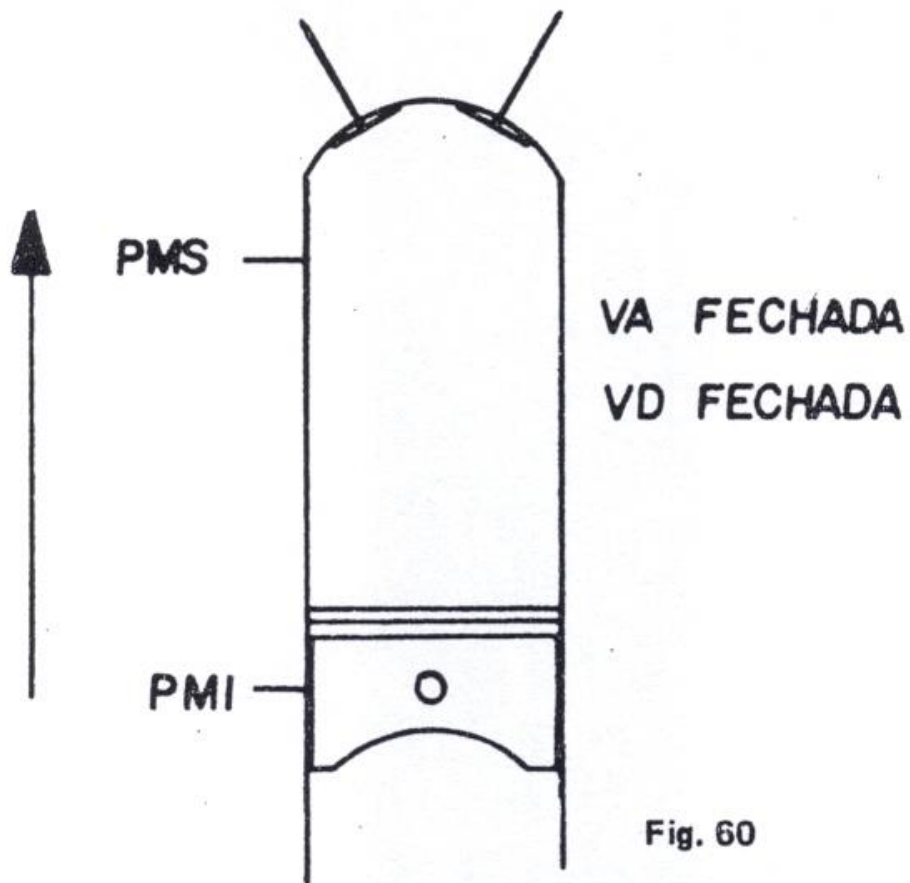


Fig. 60

3º TEMPO: COMBUSTÃO (EXPANSÃO)

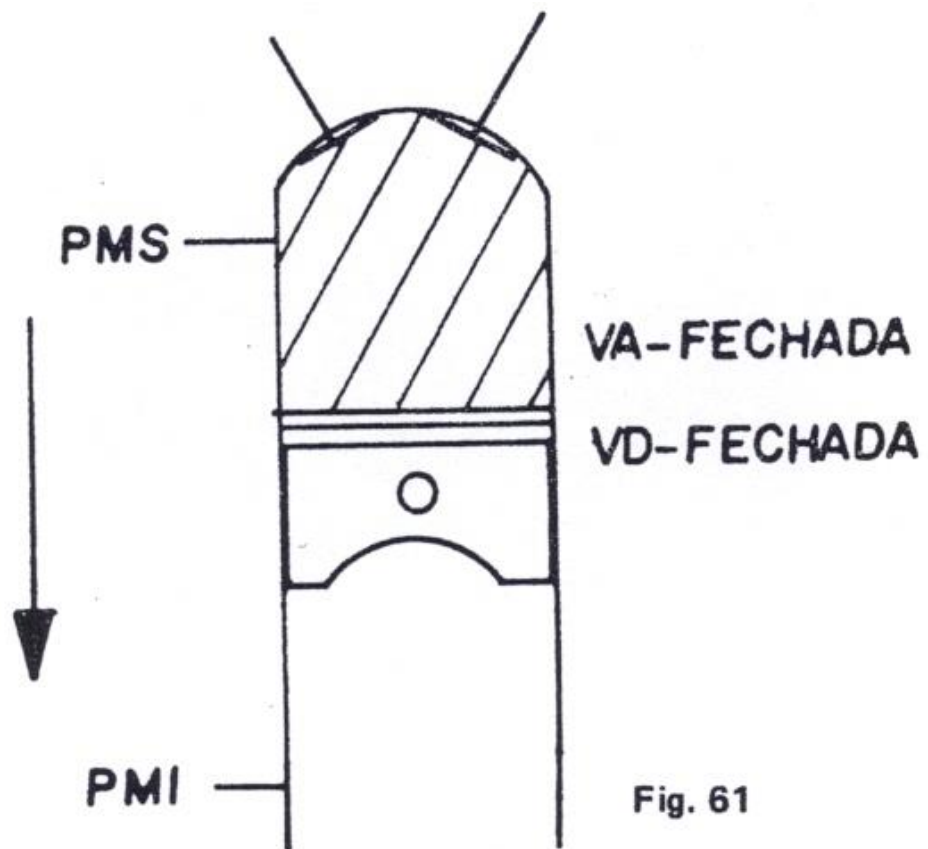


Fig. 61

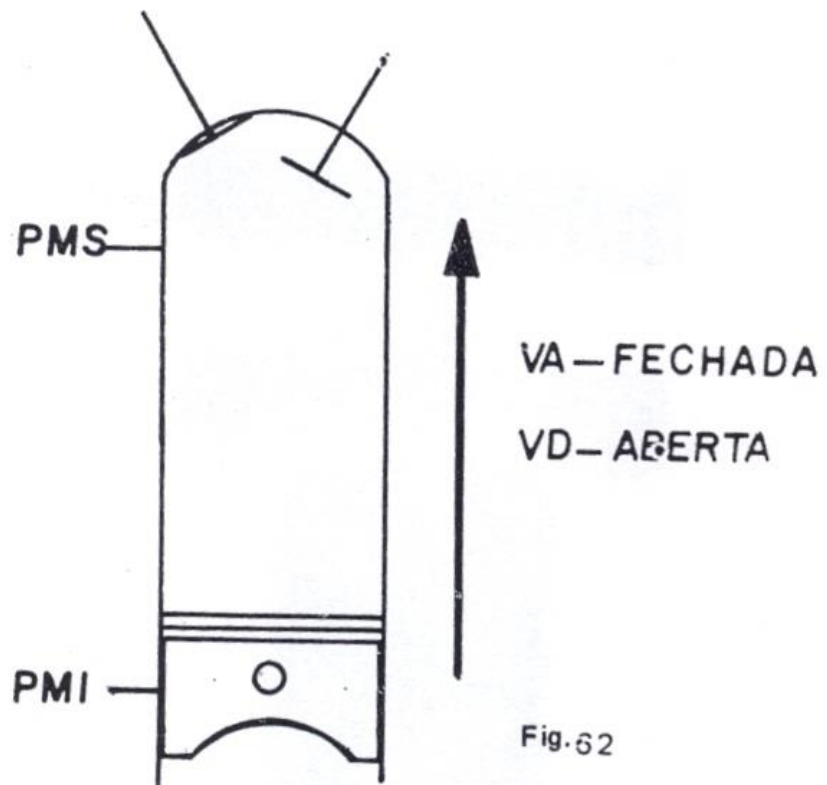
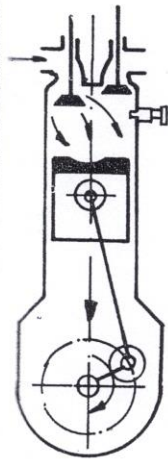
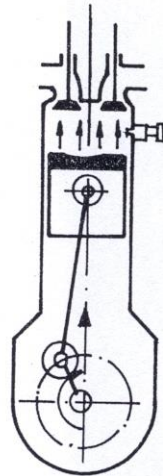


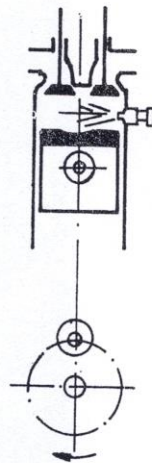
Fig. 62



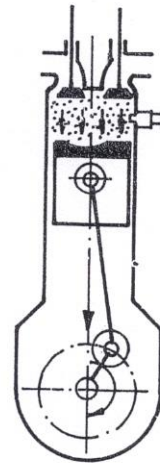
1º TEMPO
Aspiração



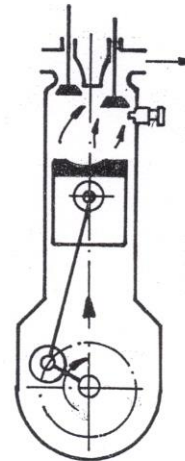
2º TEMPO
Compressão



Injeção do
Combustível

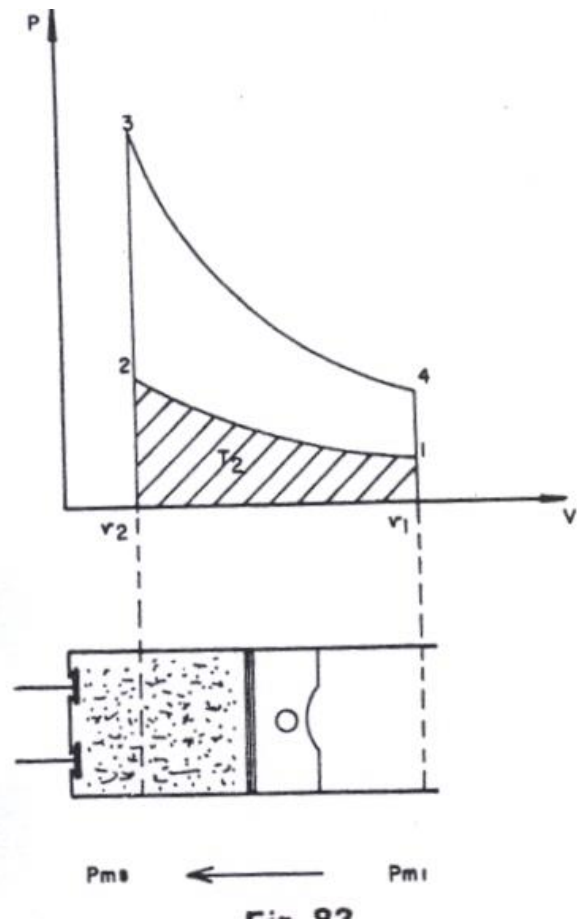
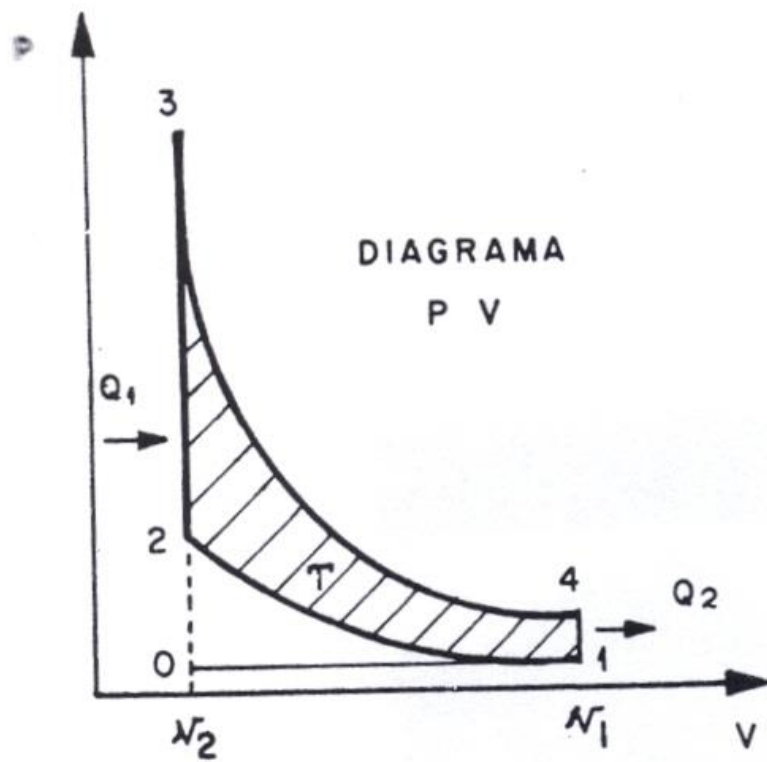


3º TEMPO
Expansão



4º TEMPO
Descarga

7. ASPECTOS DE TERMODINÂMICA



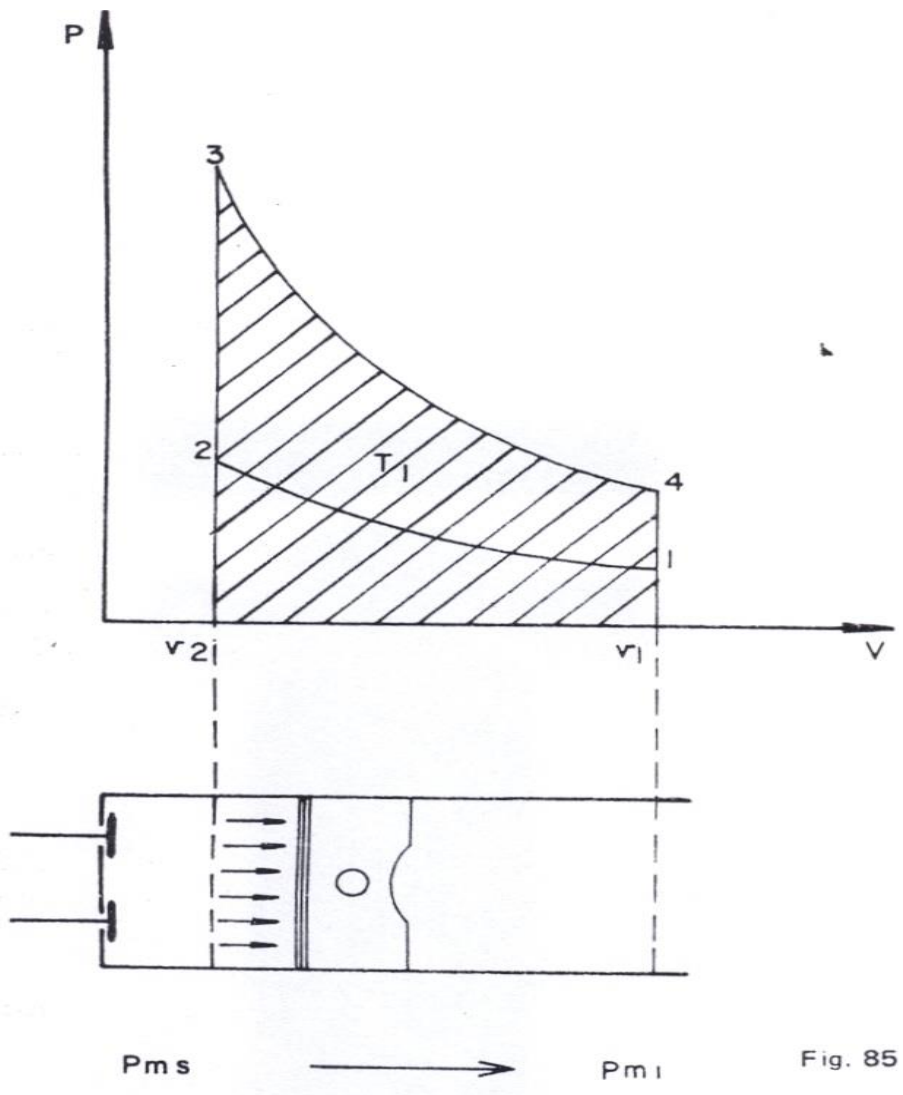


Fig. 85

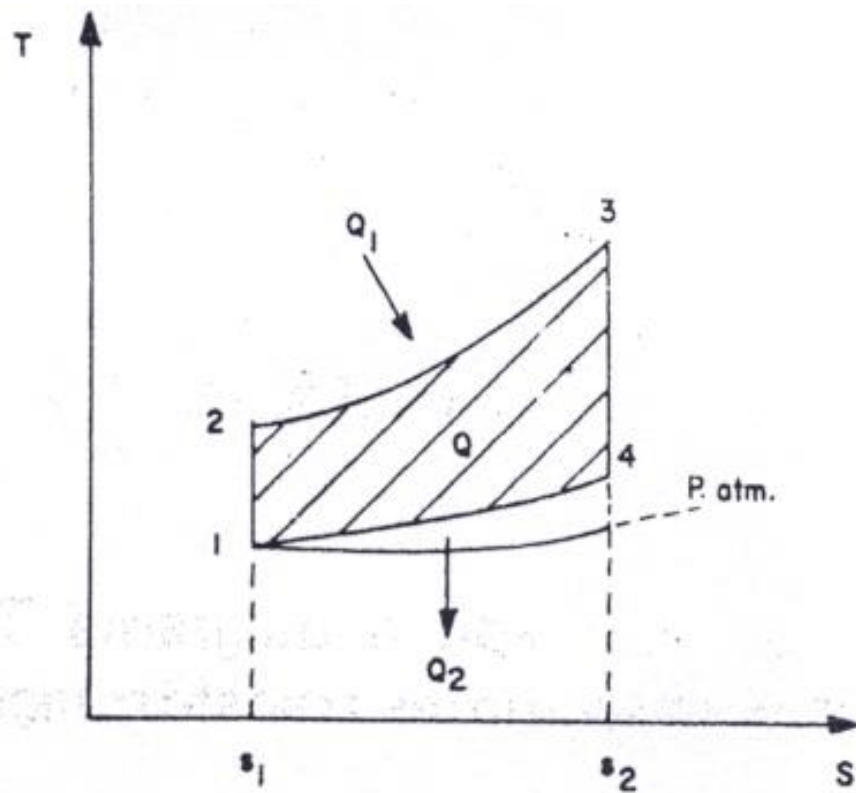


Fig. 86

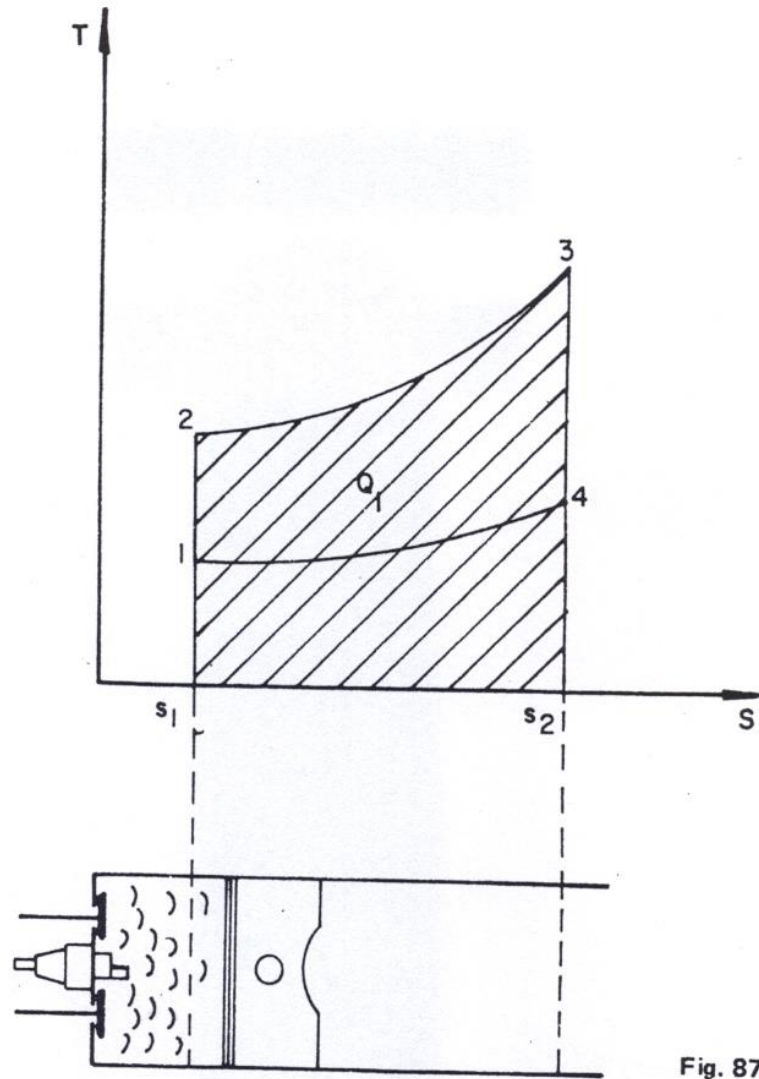


Fig. 87 a

8. COMBUSTIÙEL

COMBUSTÍVEL	TÍTULO ESTEQUIOMÉTRICO	COMBUSTÍVEL	TÍTULO ESTEQUIOMÉTRICO
ciclopentano	14,7	hidrogênio	29,6
ciclohexano	14,7	monóxido carbono	29,6
benzina	13,2	metano	9,5
tolueno	13,4	etano	5,7
xileno	13,6	propano	4
propeno	14,7	butano	3,1
buteno	14,7	benzol	2,7
penteno	14,7		
hexeno	14,7		
hepteno	14,7		
hexadecano	14,7		
metanol	6,4		
etanol	9,0		
propanol	10,5		
butanol	11,1		

Fig. 126

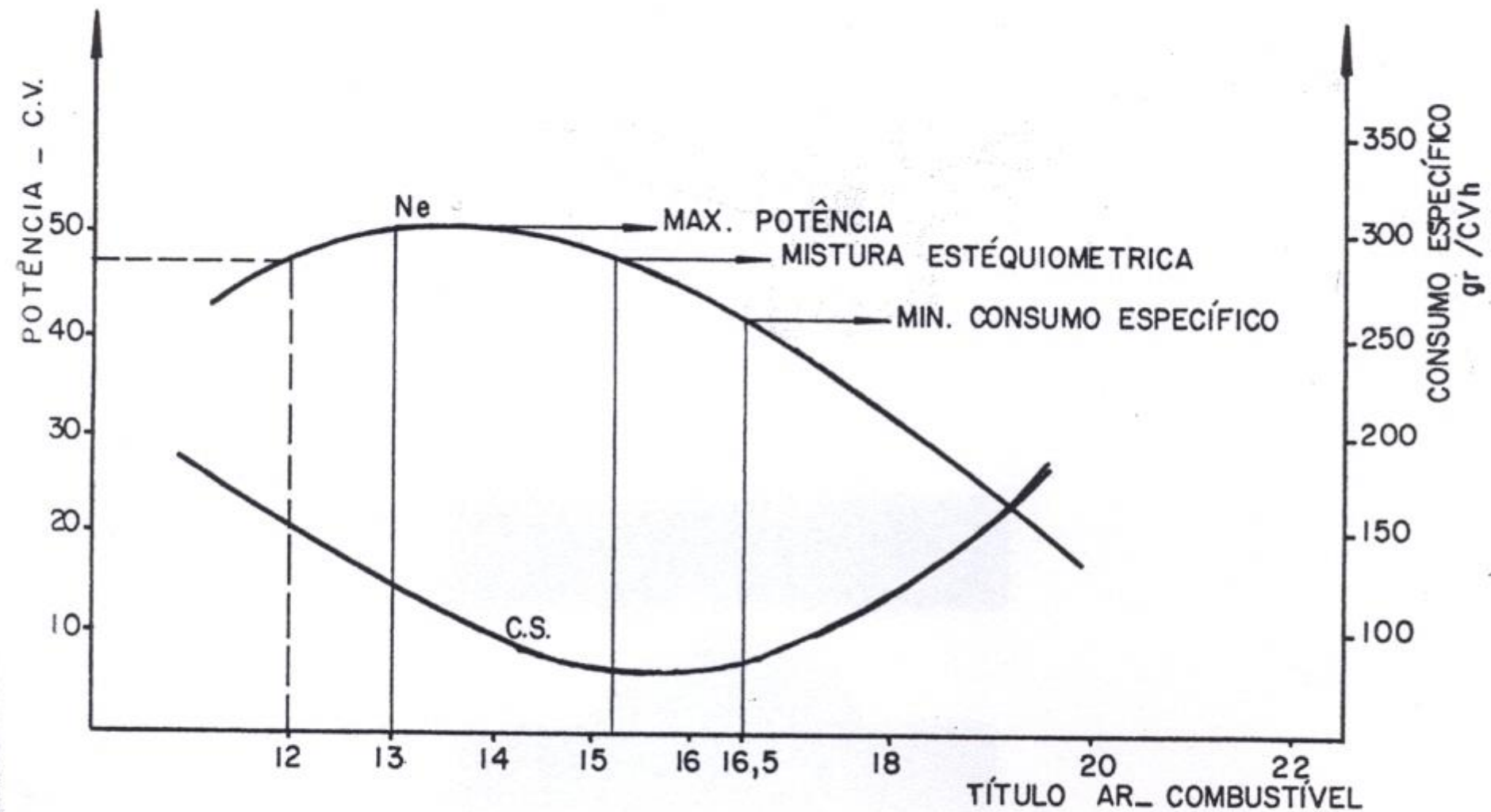
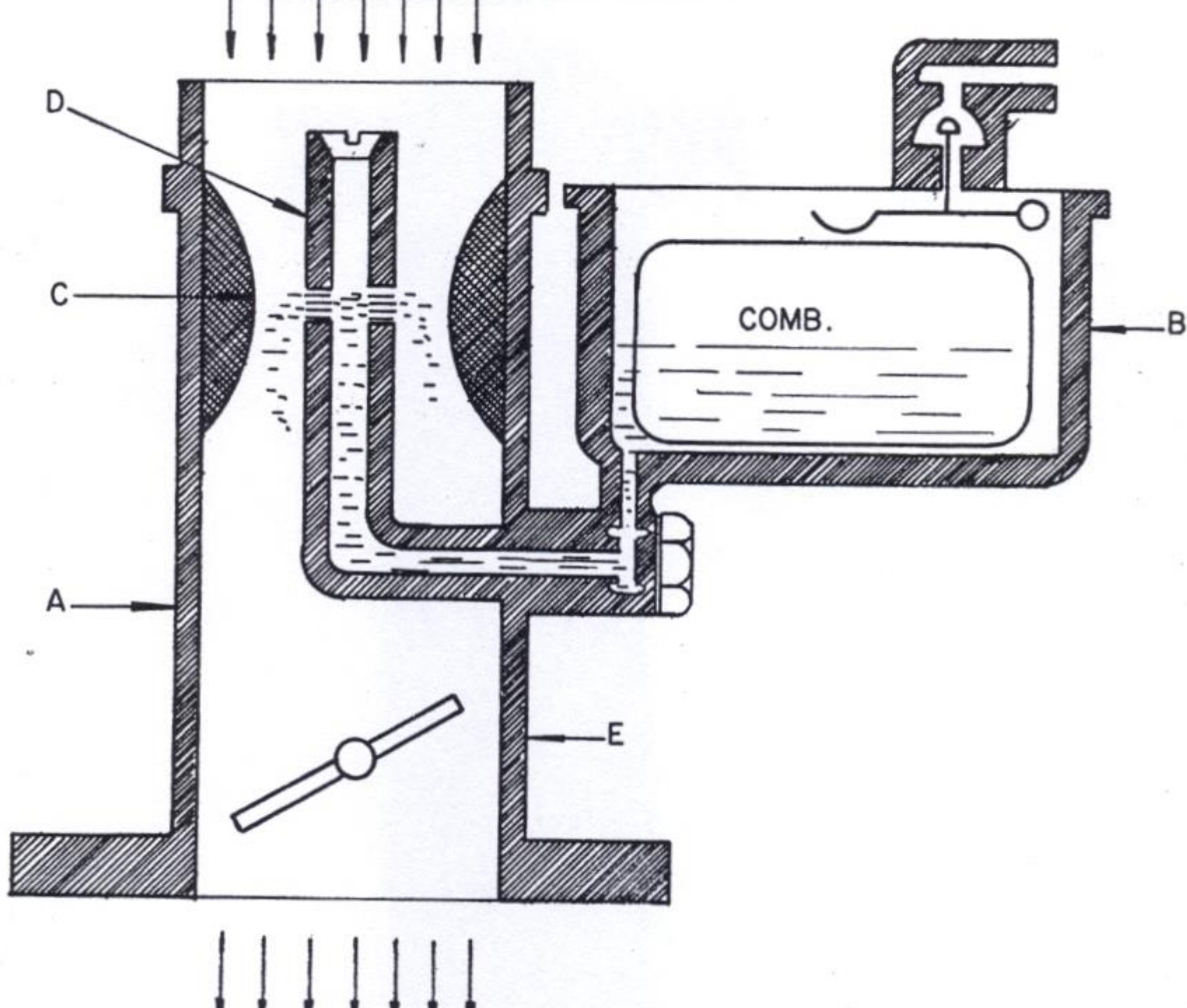


Fig. 127

8. CARBURADOR



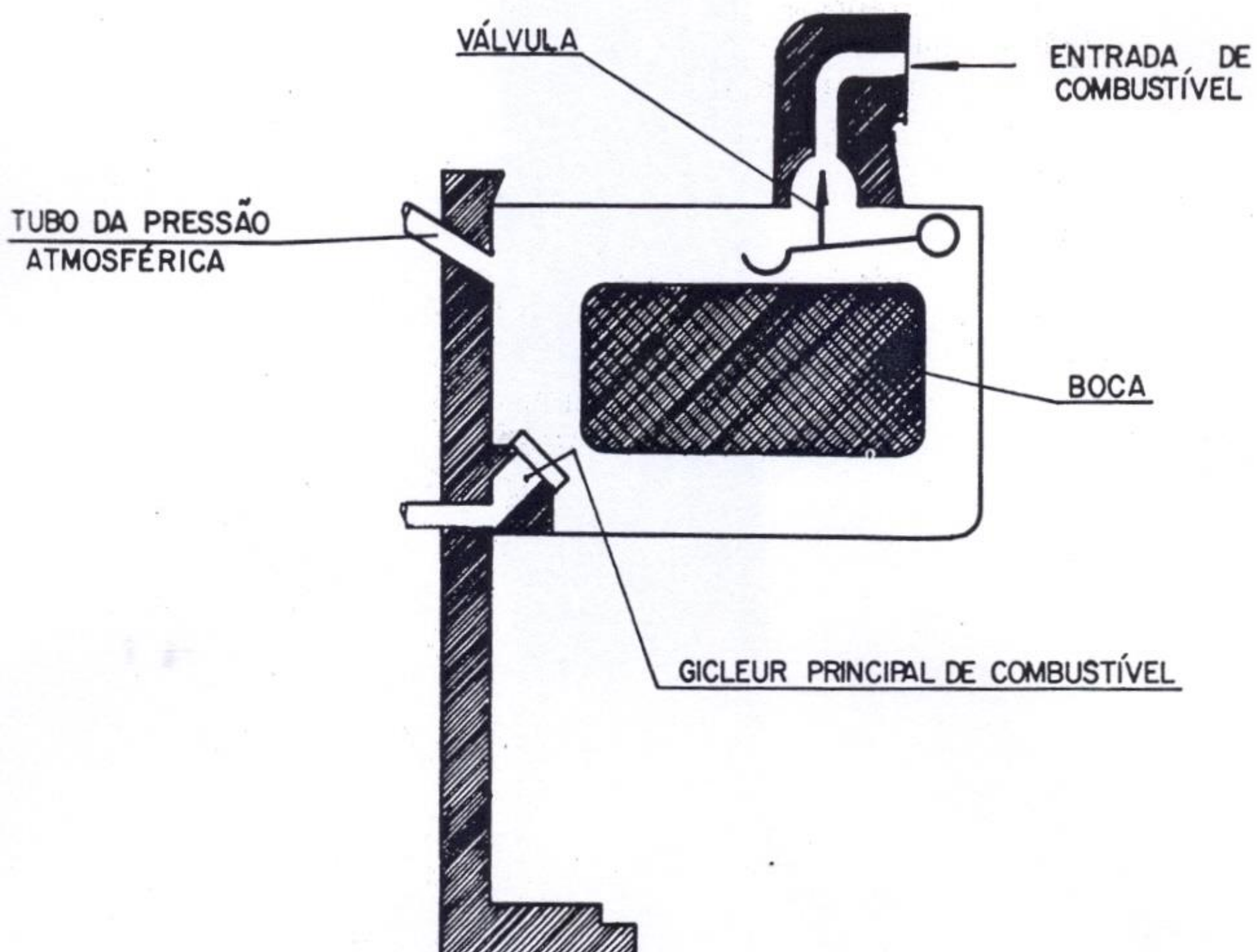
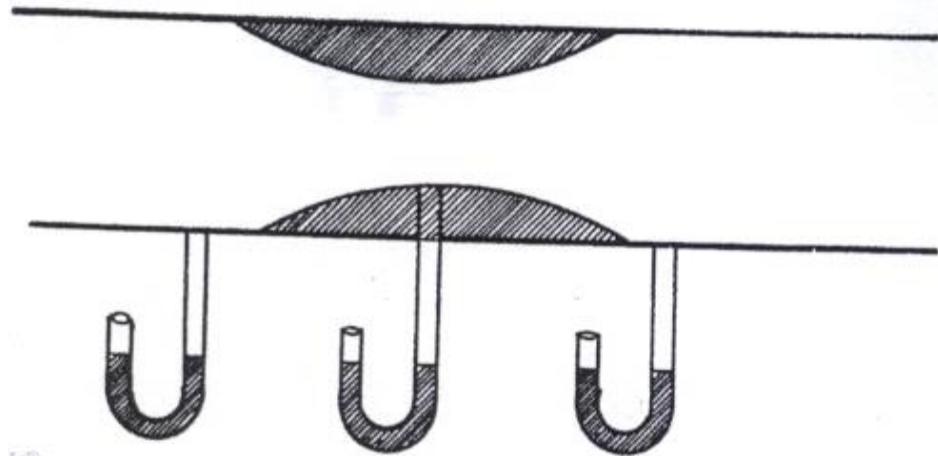


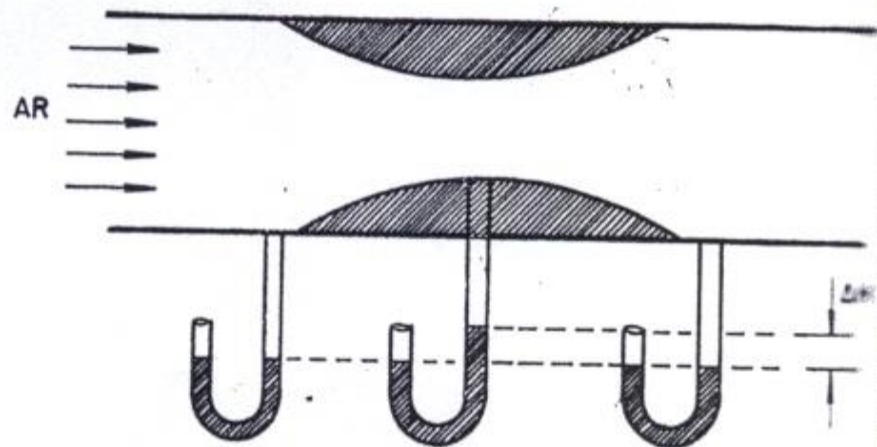


Fig. 134



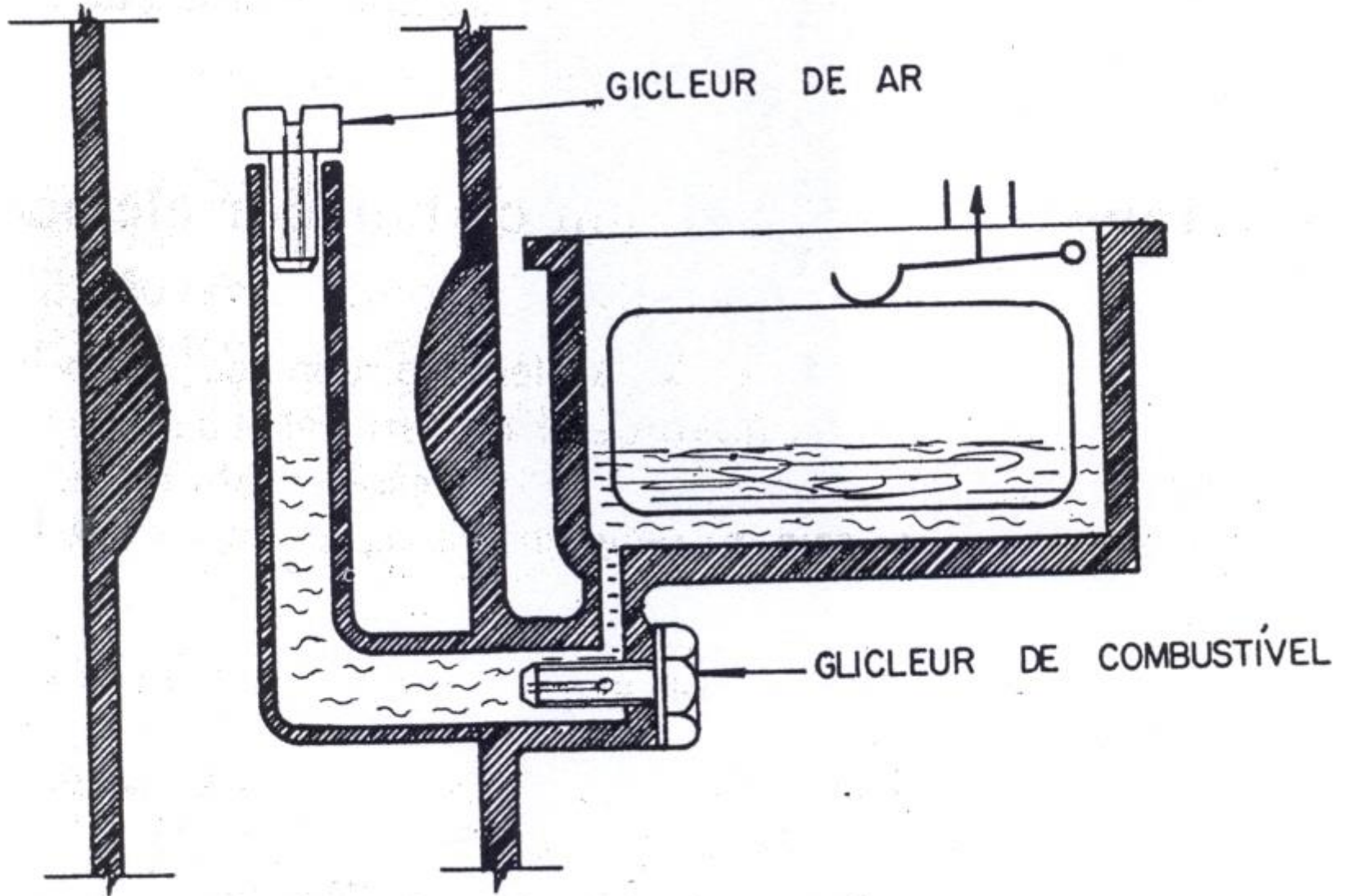
SEM FLUXO DE AR

Fig. 135

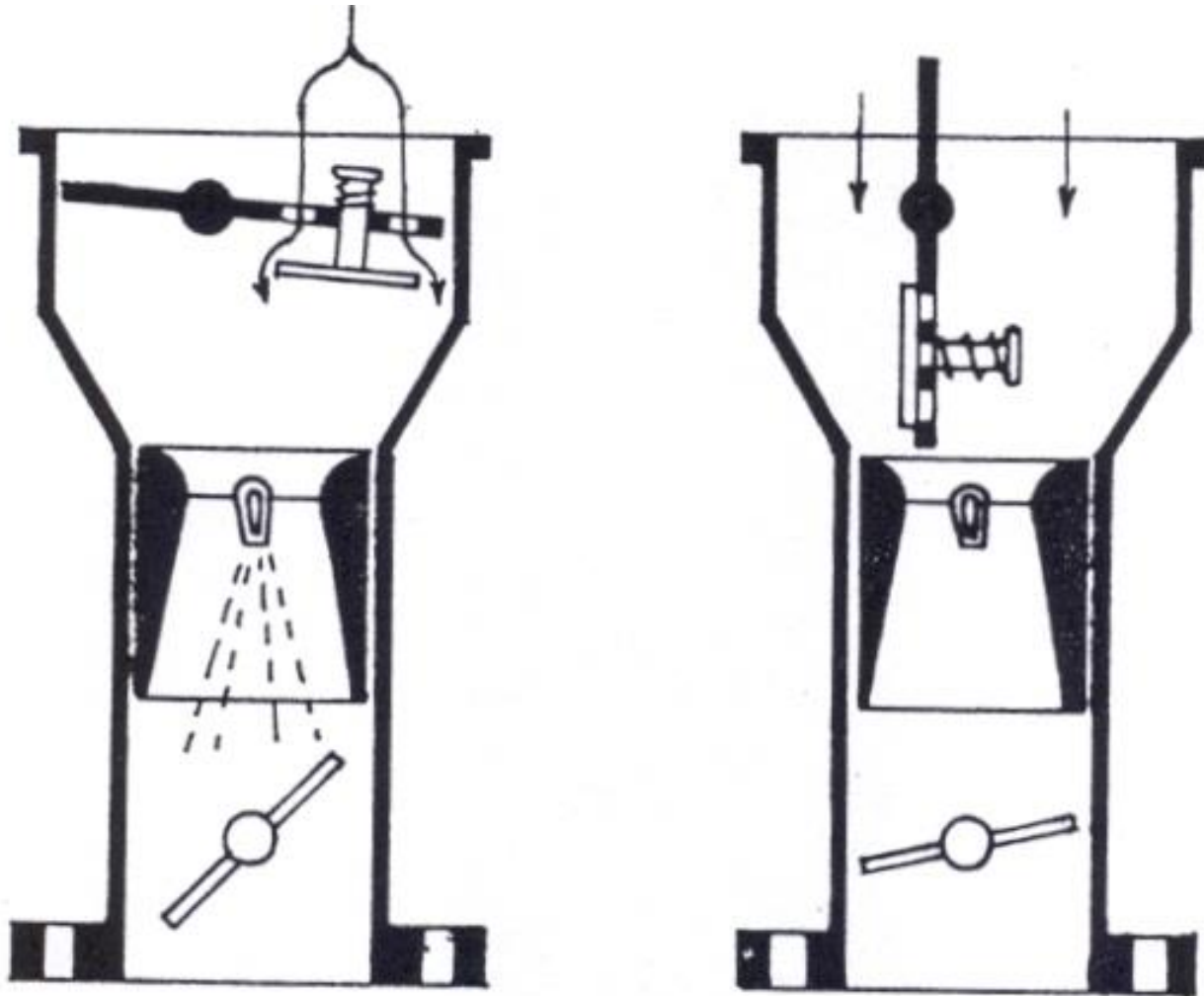


COM FLUXO DE AR

Fig. 136



PARTIDA



ACELERAÇÃO

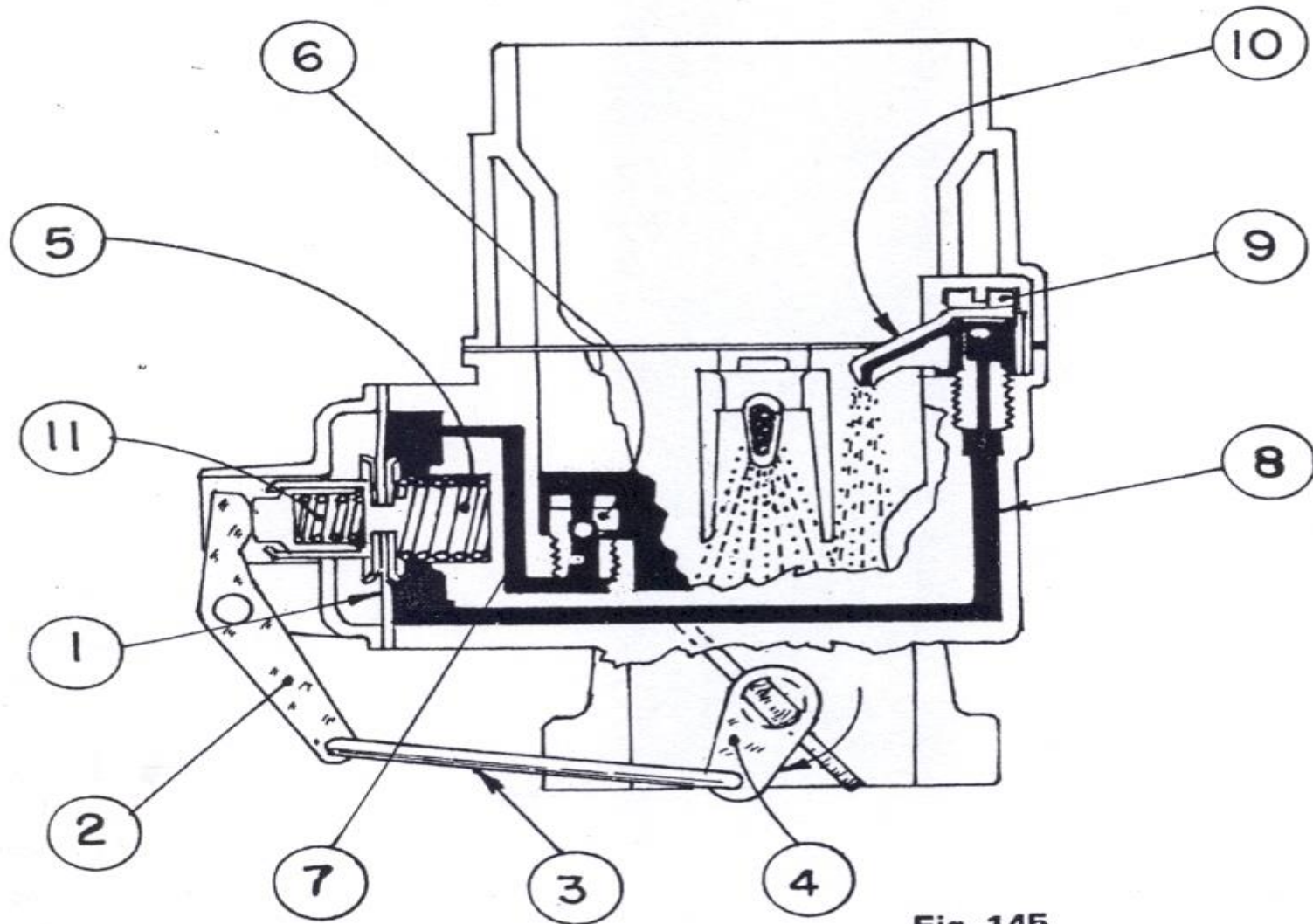
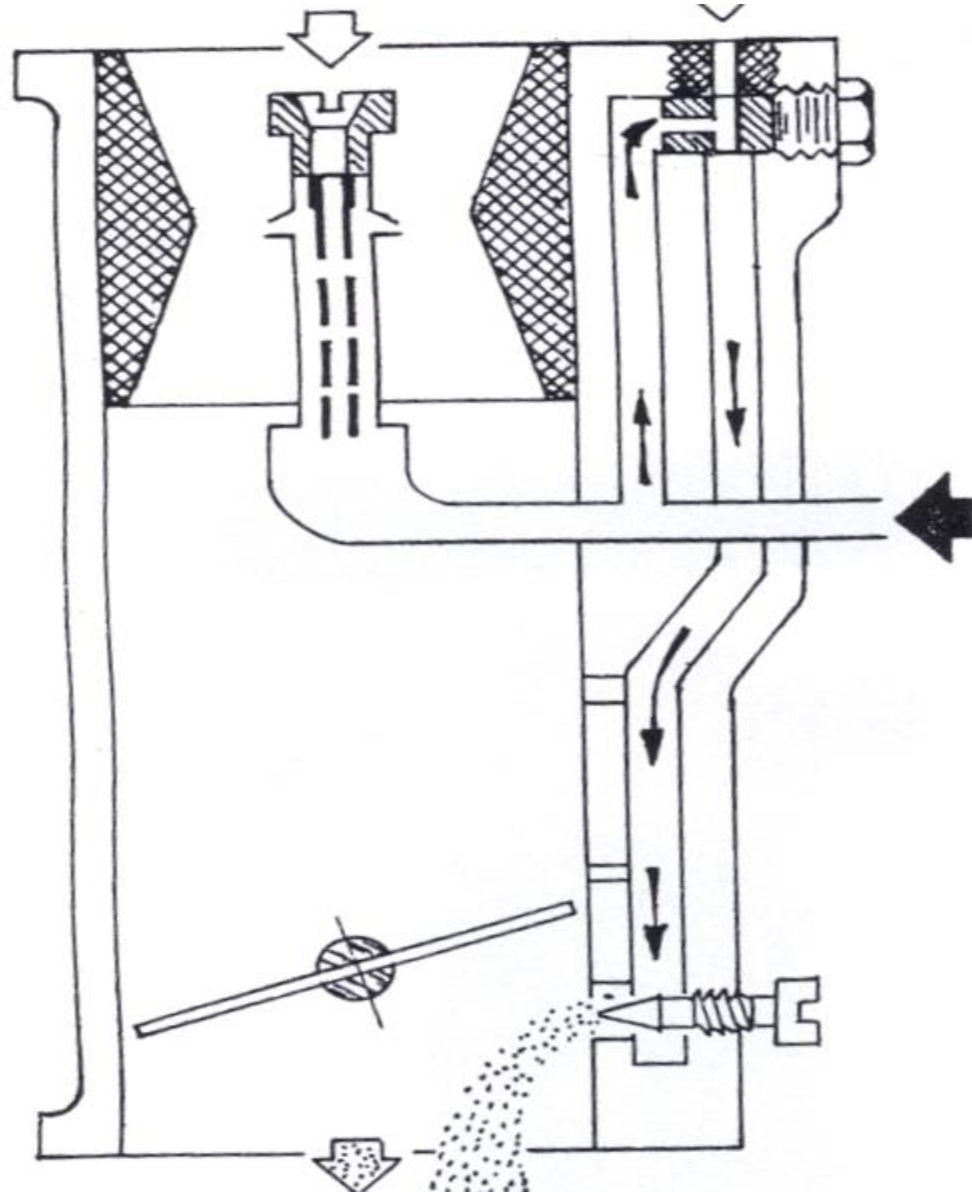


Fig. 145

MARCHA LENTA



PROGRESSÃO DE ACELERAÇÃO

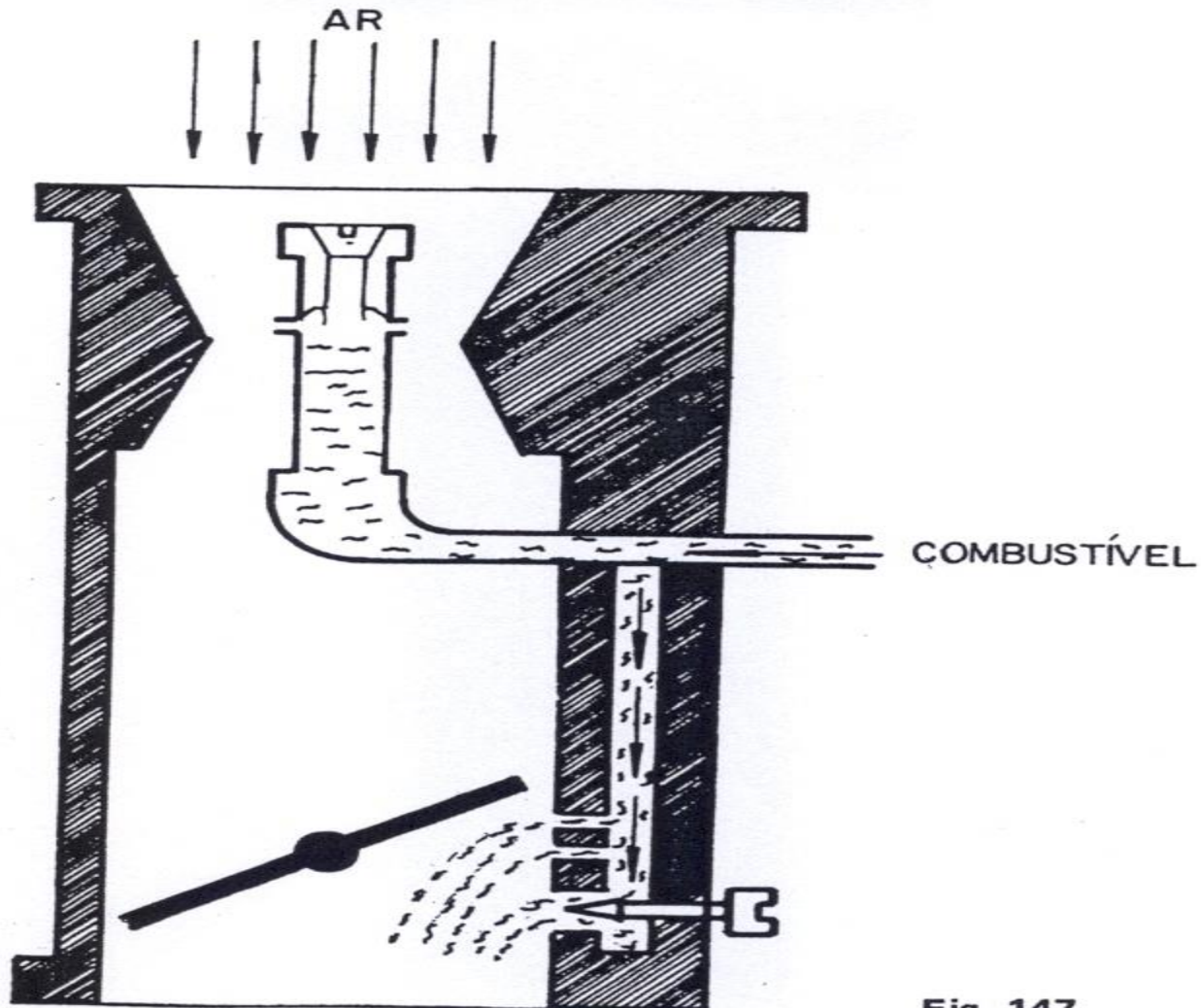


Fig. 147

CIRCUITO SUPLEMENTAR DE POTÊNCIA

a mistura.

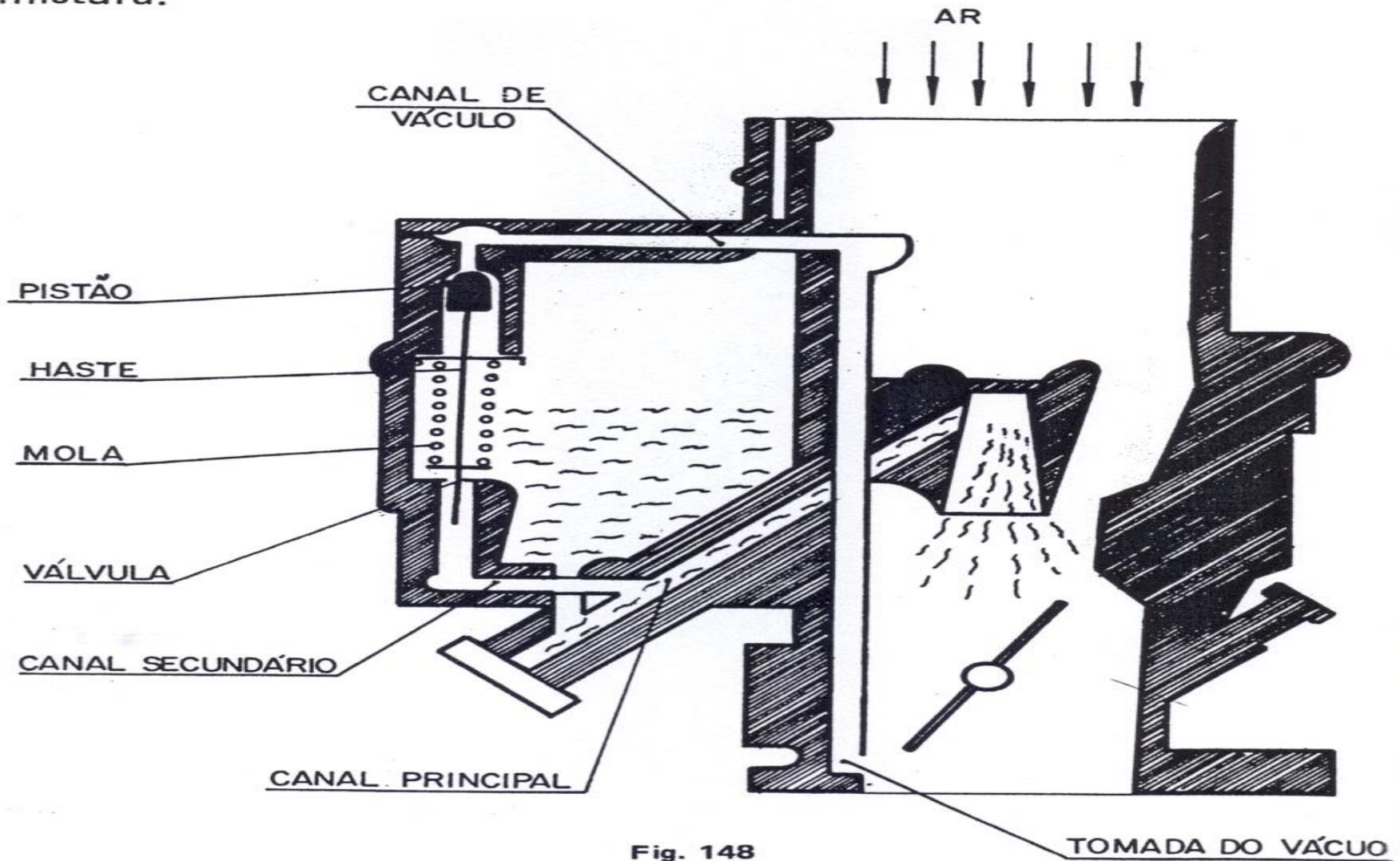
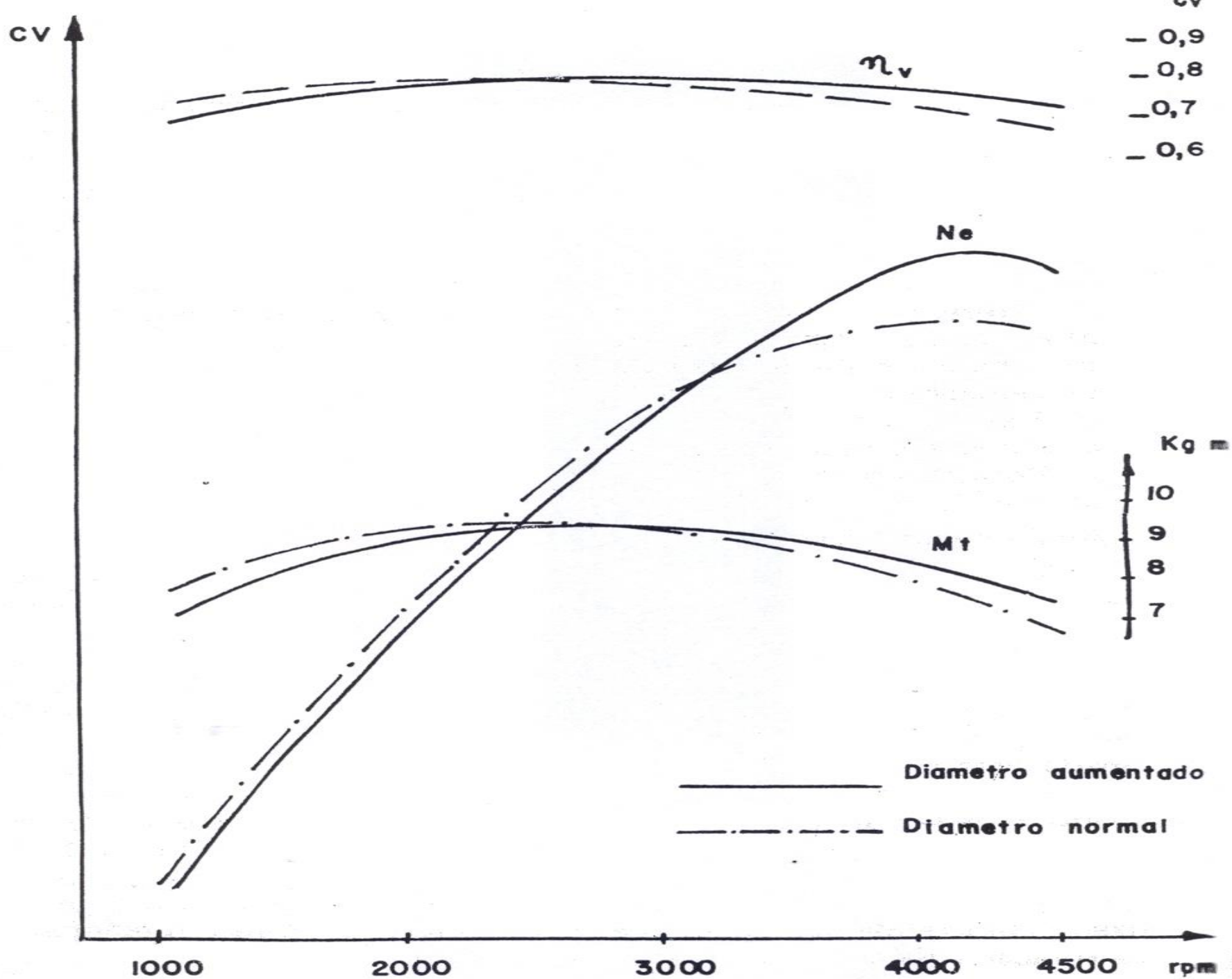


Fig. 148



ΤΕΡΜΟΔΥΝΑΜΙΚΑ

2. THERMODYNAMIC ENGINE CYCLES

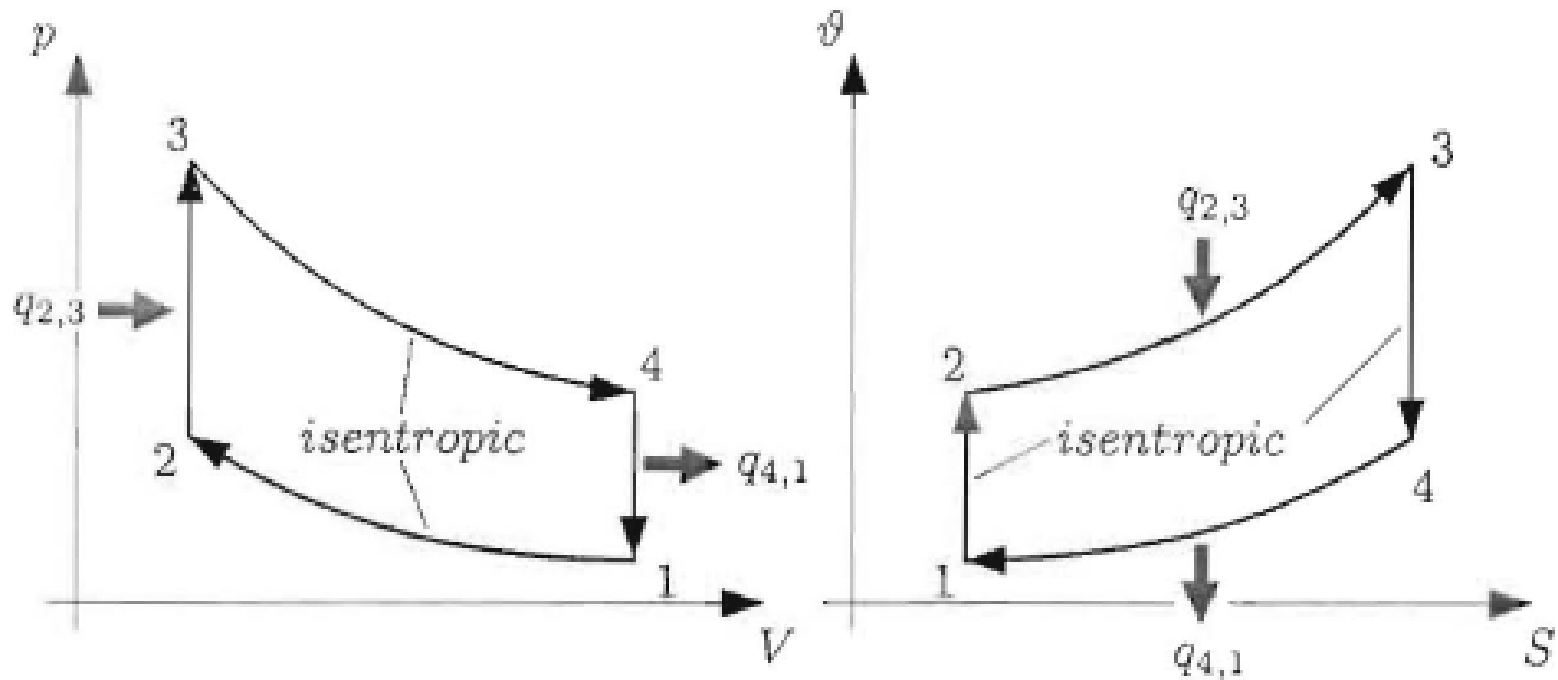


Figure 2.1 pV -diagram (left) and θS -diagram (right) of the SI engine process

$$\epsilon = \frac{V_1}{V_2}$$

1 \rightarrow 2 : Isentropic compression, $dq = 0$:

$$dq = du + dw = 0$$

$$q_{1,2} = 0$$

$$dw = -du = -m c_v d\vartheta$$

$$w_{1,2} = - \int_1^2 m c_v d\vartheta = -m c_v (\vartheta_2 - \vartheta_1)$$

The work $w_{1,2}$ is used to compress the gas and therefore, it is negative.

2 \rightarrow 3 : Isochoric input of thermal energy, $dV = 0$:

$$dw = p dV = 0$$

$$w_{2,3} = \int_2^3 p dV = 0$$

$$dq = du = m c_v d\vartheta$$

$$q_{2,3} = m c_v \int_2^3 d\vartheta = m c_v (\vartheta_3 - \vartheta_2)$$

The increased thermal energy $q_{2,3}$ is caused by combustion of the gas.

3 \rightarrow 4 : Isentropic expansion, $dq = 0$:

$$q_{3,4} = 0$$

$$dw = -du = -m c_v d\vartheta$$

$$w_{3,4} = - \int_3^4 m c_v d\vartheta = -m c_v (\vartheta_4 - \vartheta_3)$$

This state change describes the power stroke of the engine where $w_{3,4}$ is the output of kinetic energy from the gas, which is positive ($\vartheta_4 < \vartheta_3$).

4 → 1 : Isochoric heat loss, $dV = 0$:

$$dw = p dV = 0$$

$$w_{4,1} = \int_4^1 p dV = 0$$

$$dq = du + dw = m c_v d\vartheta$$

$$q_{4,1} = m c_v \int_4^1 d\vartheta = m c_v (\vartheta_1 - \vartheta_4)$$

The loss of thermal energy $q_{4,1}$ is due to the gas exchange: The burnt hot gas is pumped into the exhaust and the combustion chamber is filled with a cold mixture of unburnt fuel vapour and air ($q_{4,1}$ is negative because of $\vartheta_1 < \vartheta_4$).

$$\begin{aligned}
 \eta_{th} &= \frac{w_{1,2} + w_{2,3} + w_{3,4} + w_{4,1}}{q_{2,3}} \\
 &= \frac{m c_v (-\vartheta_2 + \vartheta_1 - \vartheta_4 + \vartheta_3)}{m c_v (\vartheta_3 - \vartheta_2)} \\
 &= 1 - \frac{\vartheta_4 - \vartheta_1}{\vartheta_3 - \vartheta_2} \\
 &= 1 - \frac{\vartheta_1}{\vartheta_2} \frac{\vartheta_4/\vartheta_1 - 1}{\vartheta_3/\vartheta_2 - 1}
 \end{aligned}$$

- $K = C_p/C_v$

$$\frac{v_4}{v_3} = \left(\frac{V_3}{V_4} \right)^{\kappa-1} = \frac{1}{\epsilon^{\kappa-1}} = \frac{v_1}{v_2}$$

$$\eta_{th} = 1 - \frac{1}{\epsilon^{\kappa-1}}$$

Please note that the thermal efficiency η_{th} does not depend on the absolute temperature values. It mainly depends on the compression ratio ϵ . Example: For a compression ratio of $\epsilon = 11$ and an adiabatic coefficient of $\kappa = 1.4$ the theoretical thermal efficiency η_{th} is:

$$\eta_{th} = 0.617$$

$$w_i = \frac{1}{V_d} \sum_{j=1}^{CYL} \oint (p_j(V_j) - p_0) dV_j \quad , \quad ($$

where:

$V_d = CYL \cdot (V_1 - V_2)$ is the displacement volume of all cylinders
 CYL is the number of cylinders
 w_i is the (normalised) indicated specific work.

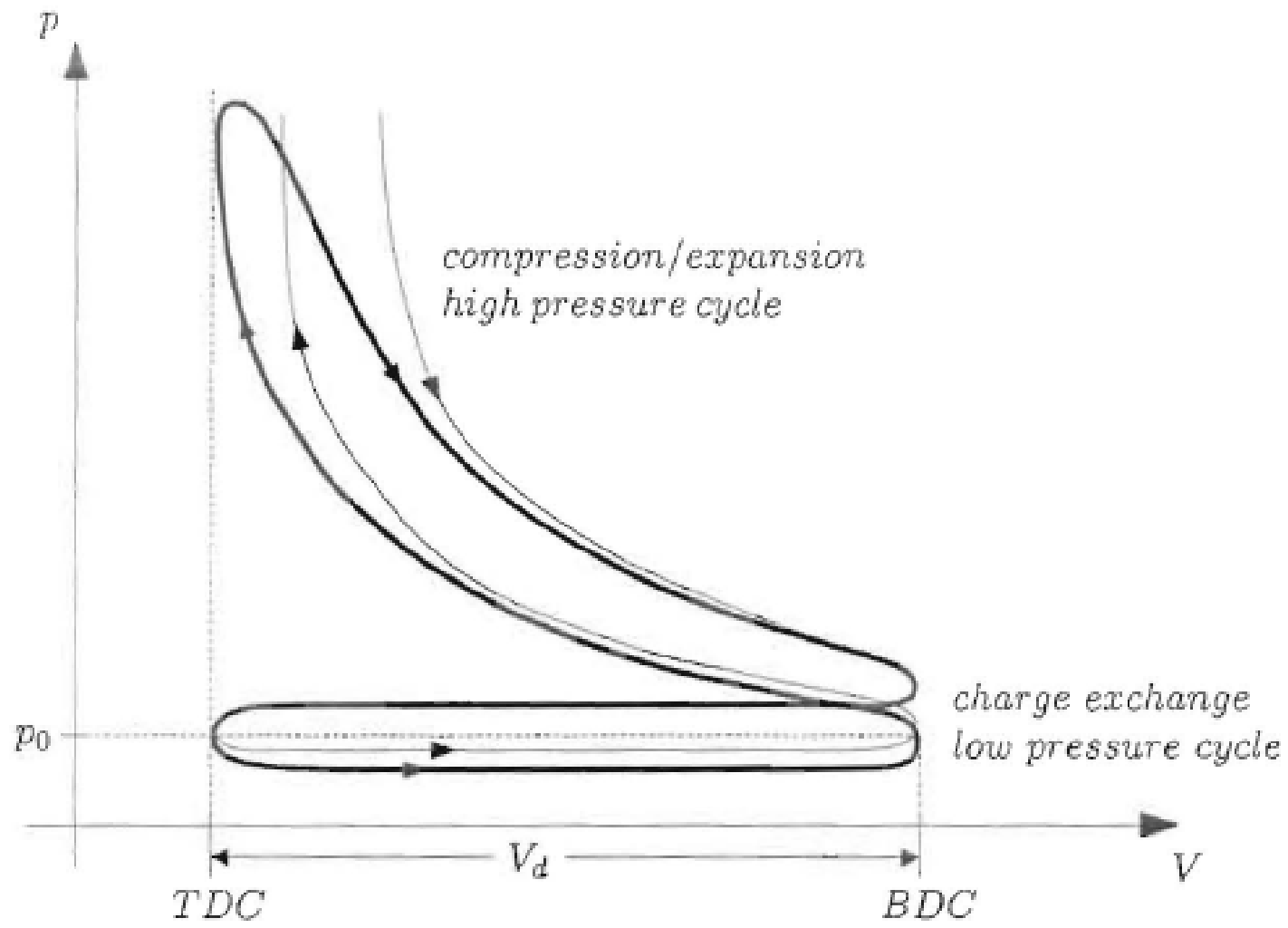


Figure 3.1 pV-diagram of four-stroke combustion engine

The piston stroke from Top Dead Center (TDC) is

$$s(\alpha_{CS}) = l(1 - \cos \beta) + r(1 - \cos \alpha_{CS}) \quad .$$

From Figure 3.2 we get

$$\begin{aligned} l \sin \beta &= r \sin \alpha_{CS} \quad , \\ \cos \beta &= \sqrt{1 - \frac{r^2}{l^2} \sin^2 \alpha_{CS}} \quad , \end{aligned} \quad (3.2)$$

which yields the piston stroke as

$$s(\alpha_{CS}) = r \left(1 - \cos \alpha_{CS} + \frac{l}{r} \left(1 - \sqrt{1 - \frac{r^2}{l^2} \sin^2 \alpha_{CS}} \right) \right) \quad . \quad (3.3)$$

At Top Dead Center, we have $\alpha_{CS} = 0$, $s(\alpha_{CS}) = 0$, and at Bottom Dead Center

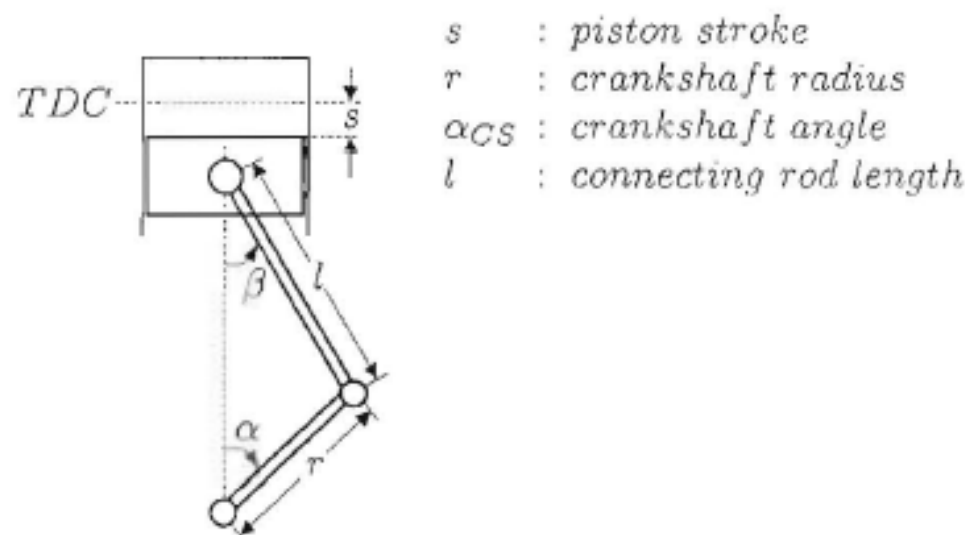


Figure 3.2 Piston and crankshaft motion

$\alpha_{CS} = \pi$, $s(\alpha_{CS}) = 2r$ respectively. The derivatives of the piston stroke are

$$\frac{ds}{d\alpha_{CS}} = r \left(\sin \alpha_{CS} + \frac{r}{l} \cdot \frac{\sin \alpha_{CS} \cos \alpha_{CS}}{\sqrt{1 - \frac{r^2}{l^2} \sin^2 \alpha_{CS}}} \right)$$

and

$$\frac{d^2s}{d\alpha_{CS}^2} = r \left(\cos \alpha_{CS} + \frac{\frac{r}{l} (\cos^2 \alpha_{CS} - \sin^2 \alpha_{CS}) + \frac{r^2}{l^2} \sin^4 \alpha_{CS}}{\left(\sqrt{1 - \frac{r^2}{l^2} \sin^2 \alpha_{CS}} \right)^3} \right) \quad (3.4)$$

These derivatives over crankshaft angle can be related to the derivatives over time as follows:

$$\begin{aligned}
 \dot{s} &= \frac{ds}{dt} = \frac{ds}{d\alpha_{CS}} \cdot \frac{d\alpha_{CS}}{dt} = \frac{ds}{d\alpha_{CS}} \cdot \dot{\alpha}_{CS} \\
 \ddot{s} &= \frac{d^2s}{dt^2} = \frac{d}{dt} \left(\frac{ds}{d\alpha_{CS}} \cdot \frac{d\alpha_{CS}}{dt} \right) = \frac{d}{dt} \left(\frac{ds}{d\alpha_{CS}} \right) \cdot \frac{d\alpha_{CS}}{dt} + \frac{ds}{d\alpha_{CS}} \cdot \frac{d^2\alpha_{CS}}{dt^2} \\
 &= \frac{d^2s}{d\alpha_{CS}^2} \cdot \dot{\alpha}_{CS}^2 + \frac{ds}{d\alpha_{CS}} \cdot \ddot{\alpha}_{CS}
 \end{aligned} \tag{3.5}$$

The indicated specific work can be written as

$$\begin{aligned}
 w_i &= \frac{1}{V_d} \oint \sum_{j=1}^{CYL} (p_j(\alpha_{CS}) - p_0) A_p \frac{ds_j(\alpha_{CS})}{d\alpha_{CS}} d\alpha_{CS} \\
 &= \frac{1}{V_d} \oint T_{comb}(\alpha_{CS}) d\alpha_{CS} \quad .
 \end{aligned} \tag{3.6}$$

The combustion torque at the crankshaft is thus defined as

$$T_{comb}(\alpha_{CS}) = \sum_{j=1}^{CYL} (p_j(\alpha_{CS}) - p_0) A_p \frac{ds_j}{d\alpha_{CS}} \quad . \tag{3.7}$$

The piston strokes in different cylinders are shifted by phase.

$$s_j(\alpha_{CS}) = s \left(\alpha_{CS} - (j - 1) \cdot \frac{4\pi}{CYL} \right) \quad , \quad j = 1, \dots, CYL \quad (3.8)$$

The average combustion torque is

$$\begin{aligned} \bar{T}_{comb} &= \frac{1}{4\pi} \oint T_{comb}(\alpha_{CS}) d\alpha_{CS} \\ &= \frac{P_i}{\dot{\alpha}_{CS}} \quad , \end{aligned} \quad (3.9)$$

where P_i is the mean indicated power. The total indicated work $w_i V_d$ can now be written at stationary engine operation as

$$w_i V_d = 4\pi \bar{T}_{comb} = 4\pi \frac{P_i}{\dot{\alpha}_{CS}} = \frac{4\pi P_i}{2\pi n} = \frac{2P_i}{n} \quad ,$$

and the normalised work

$$w_i = \frac{2P_i}{V_d n} \quad , \quad (3.10)$$

$$\eta_c = \frac{P_e}{m_f H_f} = \frac{w_e V_d n}{2m_f n H_f} \cdot \frac{2}{CYL} = \frac{w_e}{m_f H_f} \cdot \frac{V_d}{CYL} \quad .$$

- P_e is the effective power in W
 w_e is the effective specific work per cycle in J/m^3
 m_f is the mass of fuel measured per cylinder in kg
 \dot{m}_f is the fuel flow in kg/s
 H_f is the specific energy of the fuel released in the combustion J/kg
 V_d is the total displacement volume in m^3
 (V_d/CYL displacement volume per cylinder)

The indicated thermodynamic efficiency (friction not considered) is:

$$\eta_i = \frac{w_i}{2m_f H_f} \cdot \frac{V_d}{CYL} \quad (3.12)$$

Table 3.1 Indicated specific work w_i , theoretical heat loss $q_{hl,th}$, and realistic heat loss $q_{hl,r}$ for different engine types, related to fuel combustion heat.

Engine Type	SI	Diesel	Big Diesel
w_i	33-35 %	40-43 %	45-48 %
$q_{hl,th}$	23-28 %	22-25 %	12-14 %
$q_{hl,r}$	37-44 %	35-40 %	26-33 %

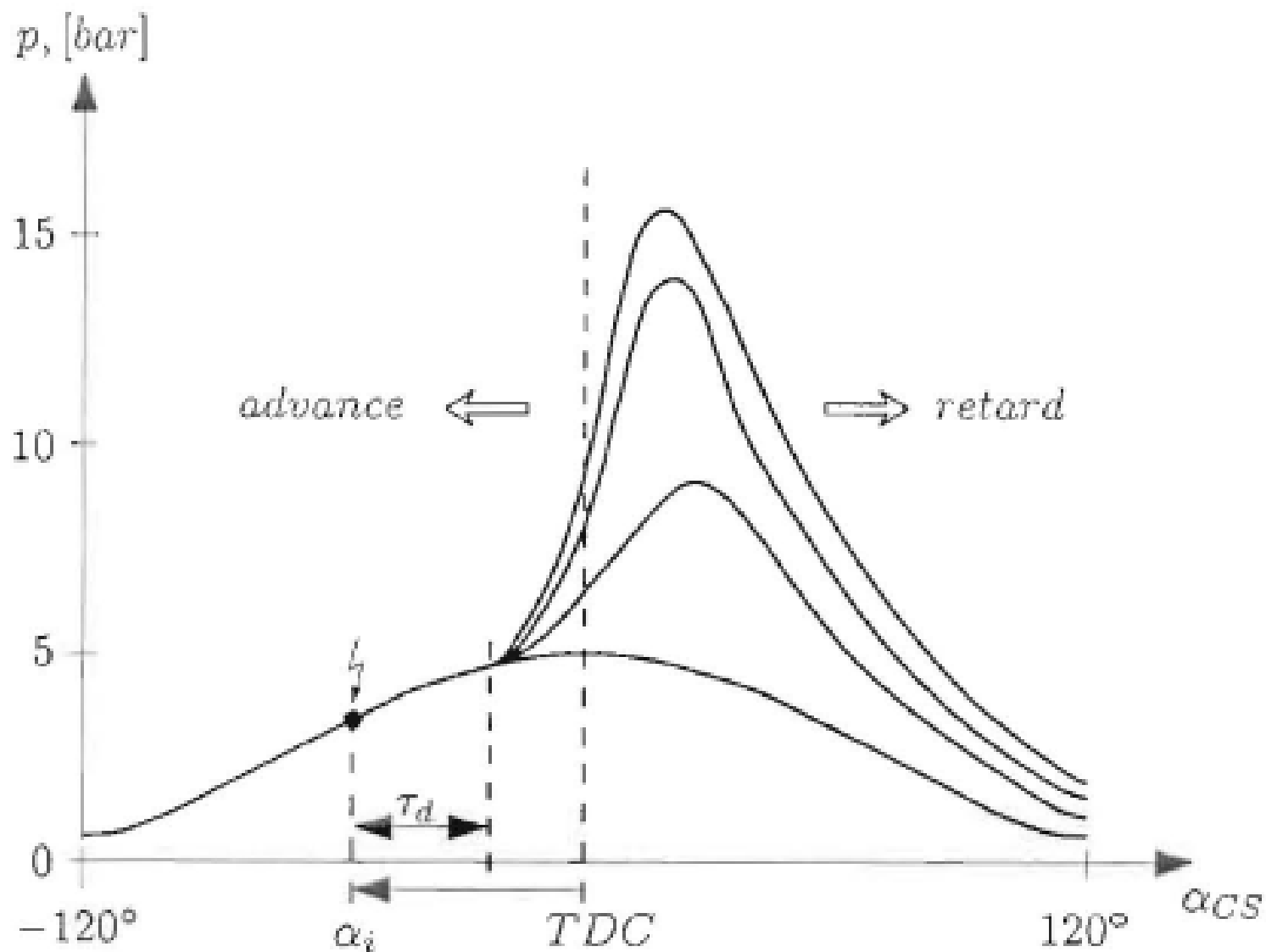


Figure 3.9 In-cylinder pressure p over crankshaft angle α_{CS} .

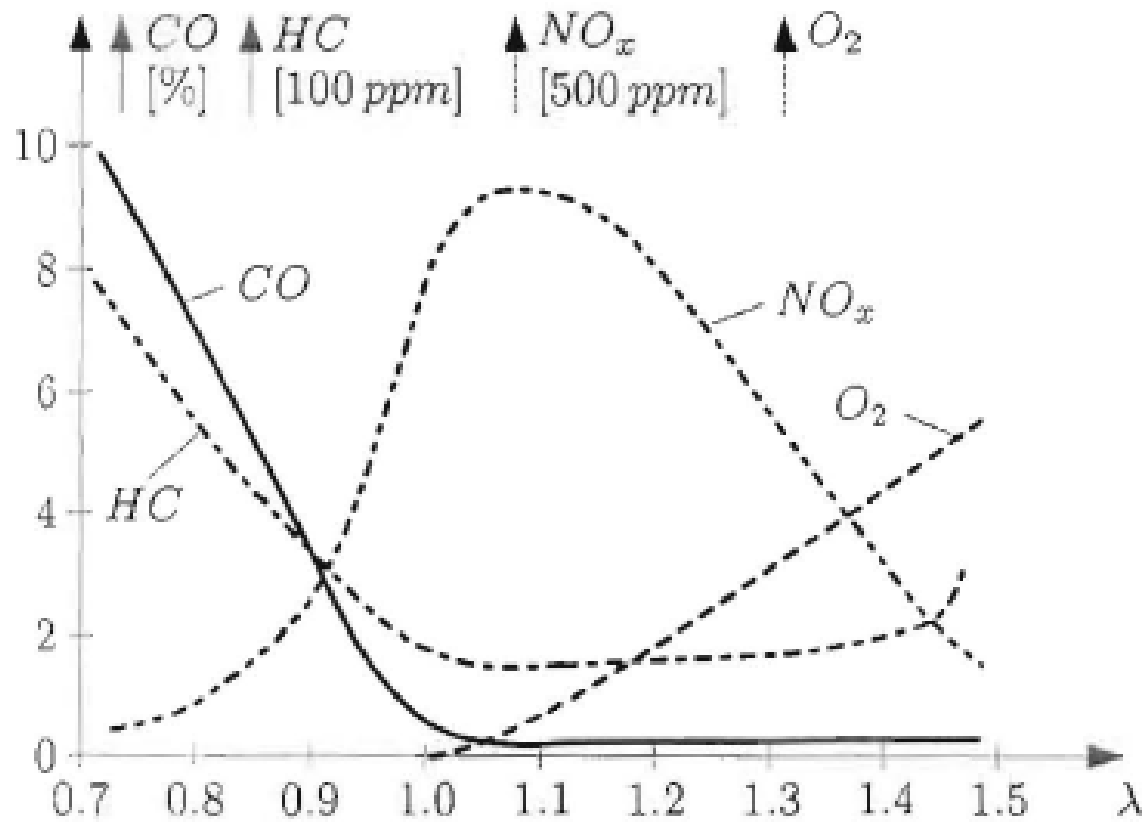


Figure 3.14 Raw emissions of CO , HC , NO_x and O_2 over air-fuel ratio λ for SI engines.

3.3. IGNITION CONTROL IN SI ENGINES

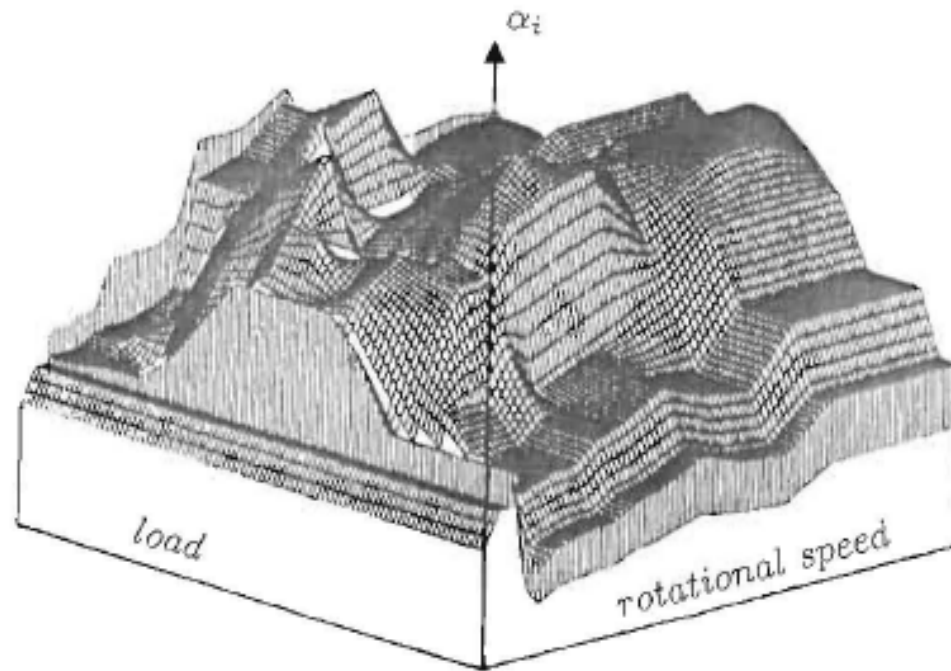


Figure 3.21 Ignition angle map

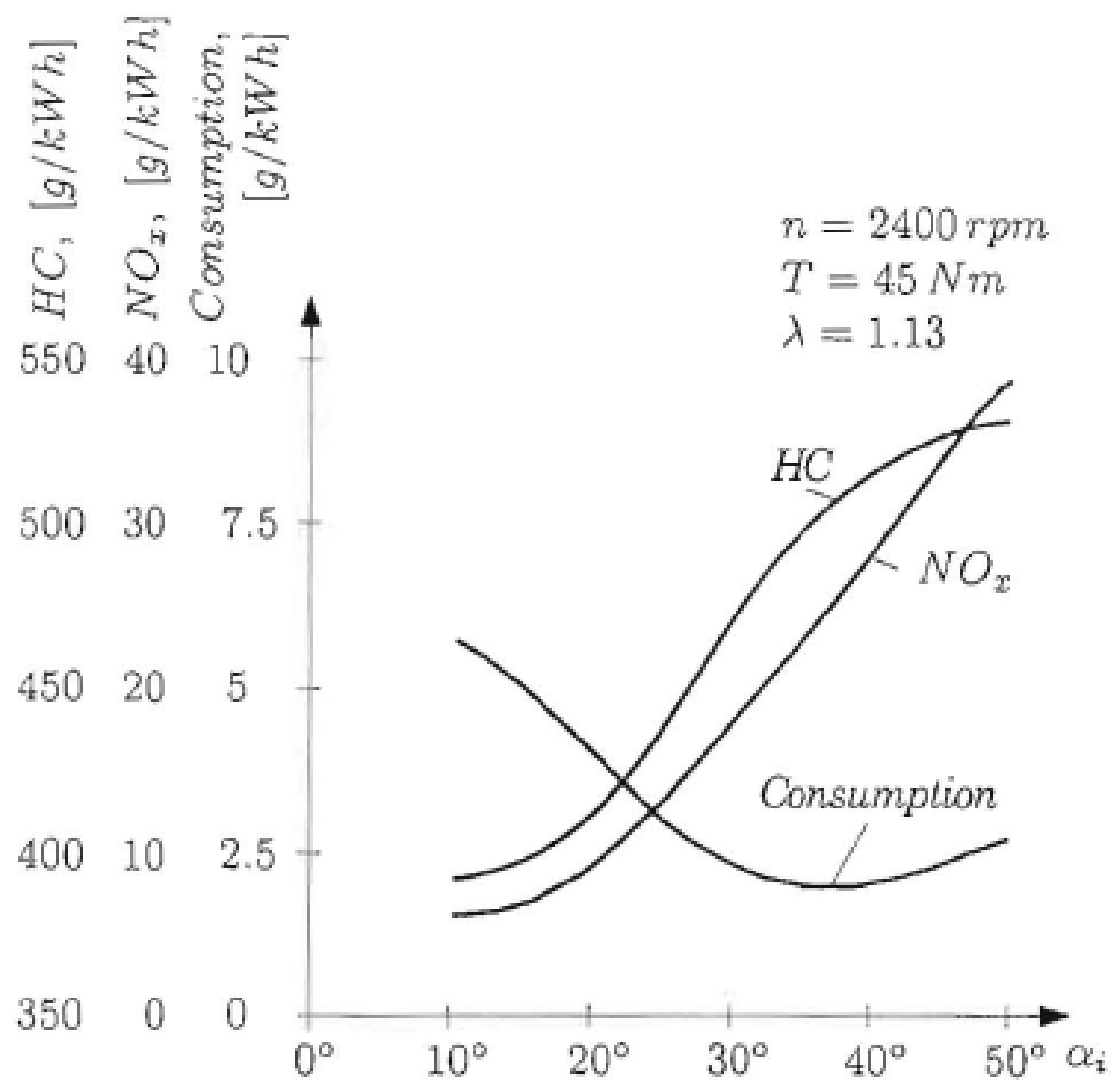


Figure 3.22 Fuel consumption and emission levels over ignition angle α_i .

GERENCIAMENTO DO MOTOR

Technology

Powertrain Control

Driver Information Systems

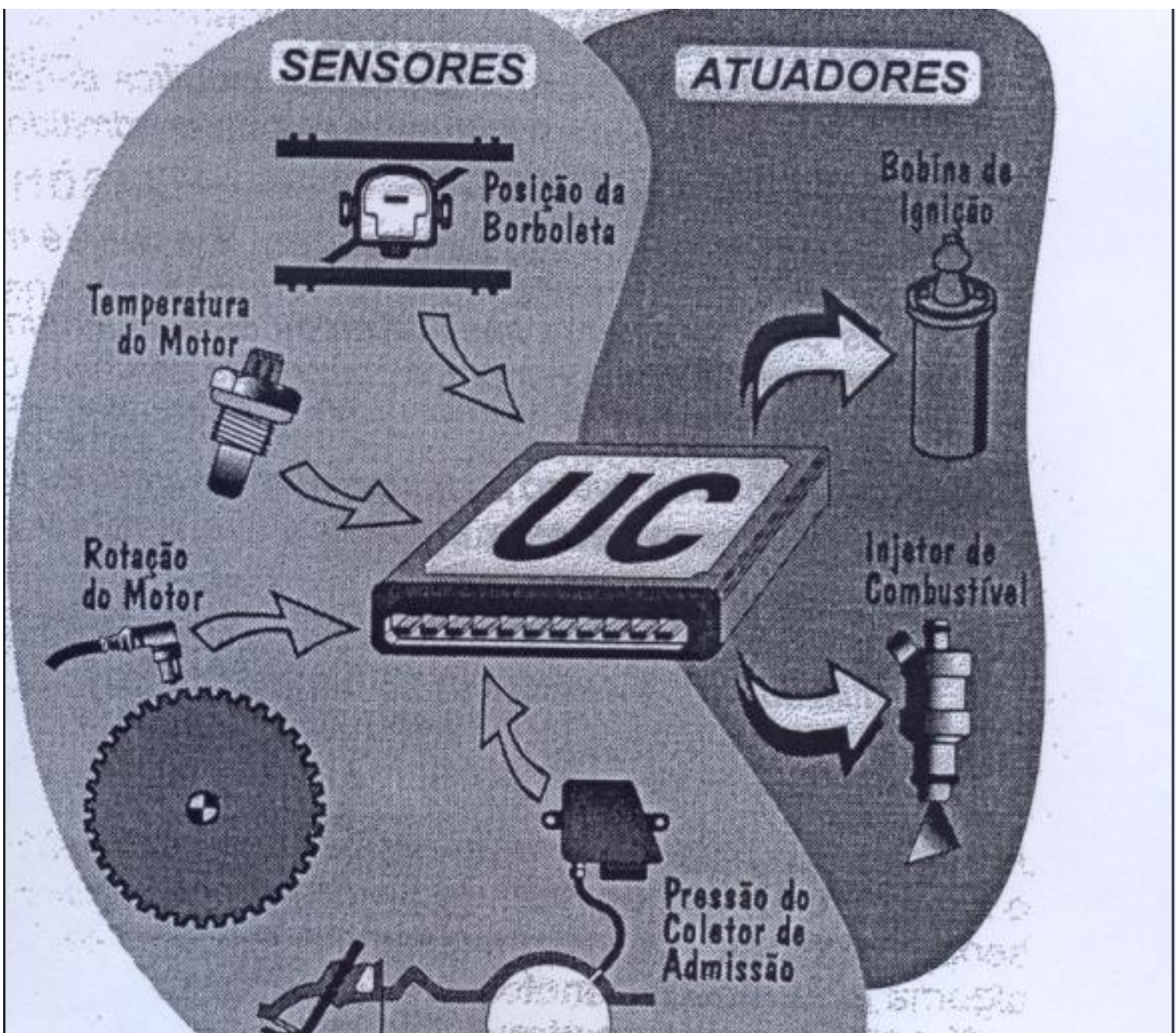
Vehicle Network

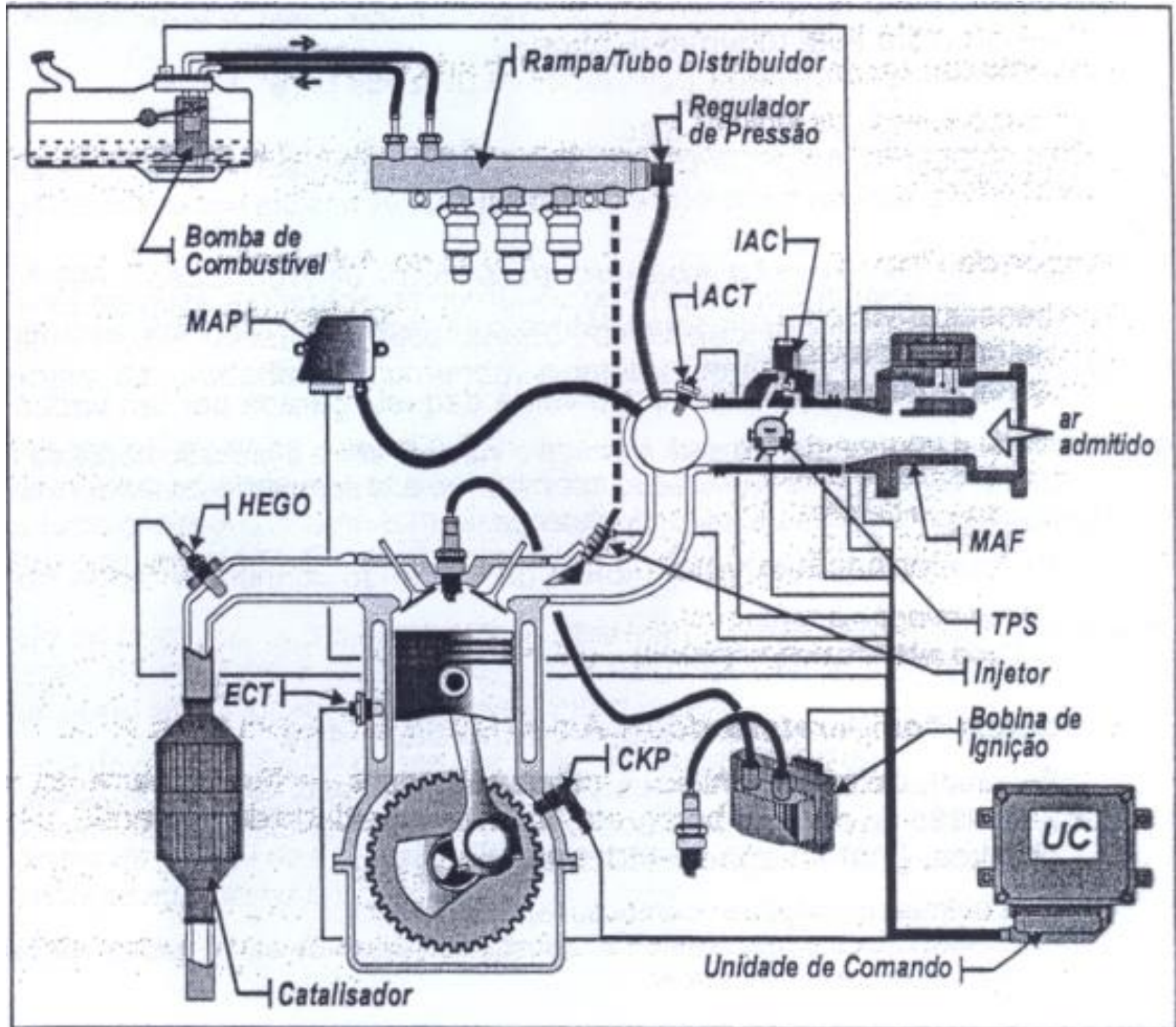
- ▶ Engine Control and Management
- ▶ Hybrid and Electric Auxiliaries

POWERTRAIN CONTROL

SENSORES

ATUADORES





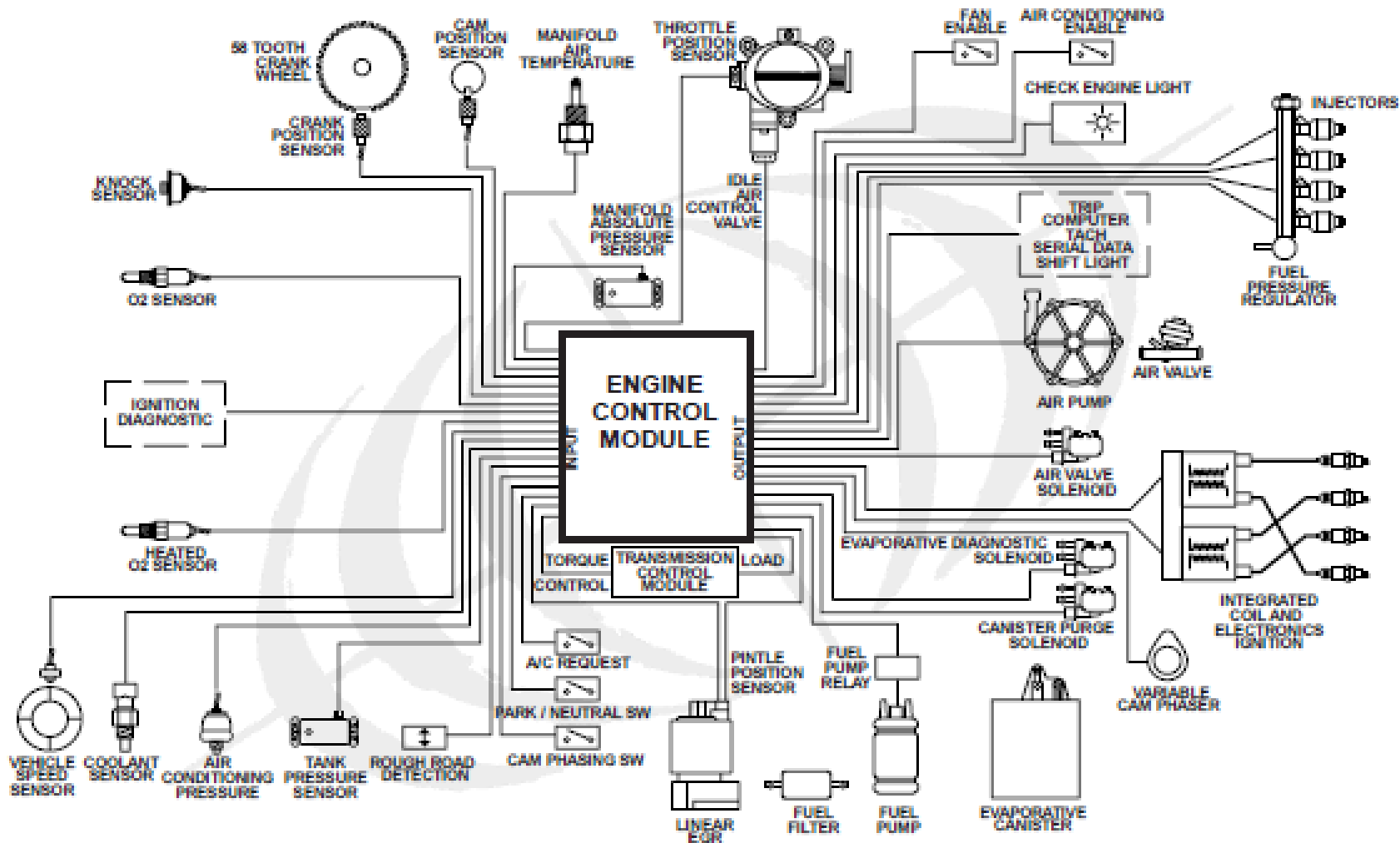


Figure 4-1. Ignition System Interfaces.