

Bioquímica Geral

RFM0004

Lipídeos

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O que são lipídeos?

Qual a importância/função dessas moléculas???

Collaborate!

O que são lipídeos?

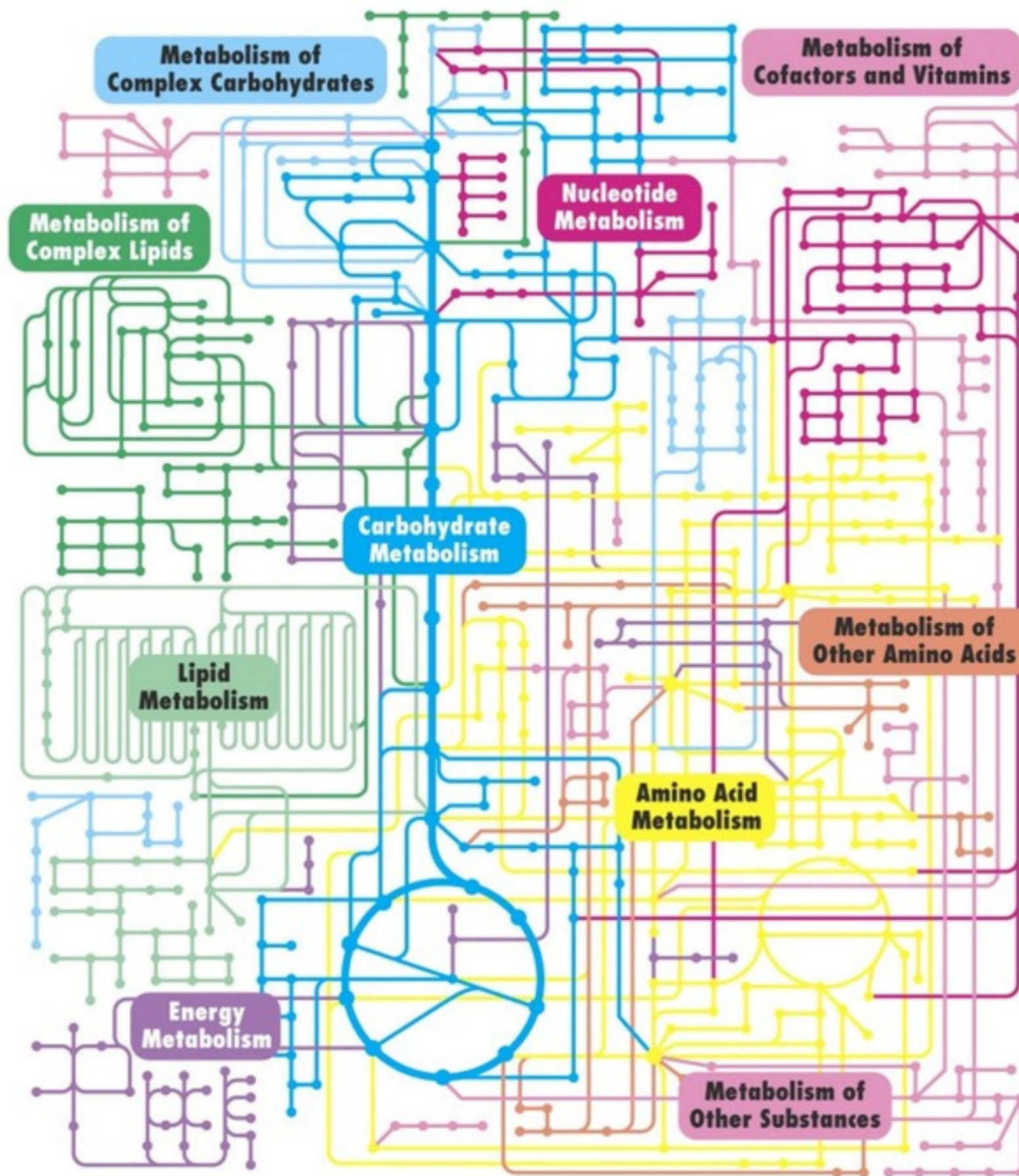


Figure 15-2
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Chapter 22 Opener part 1
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Lipídeos



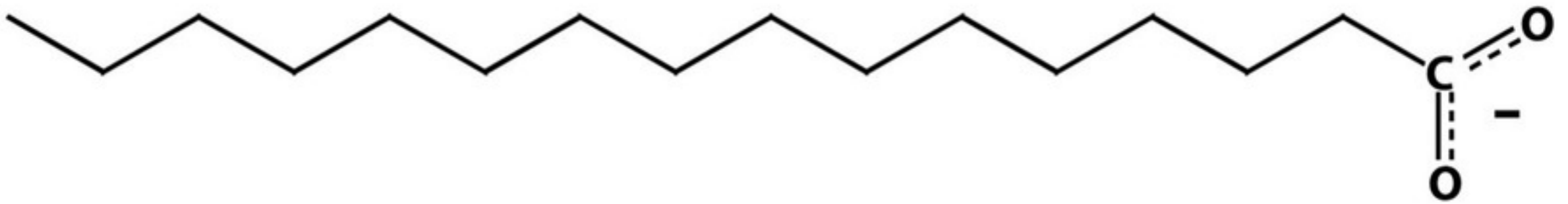
Figure 12-22
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Ácidos Graxos

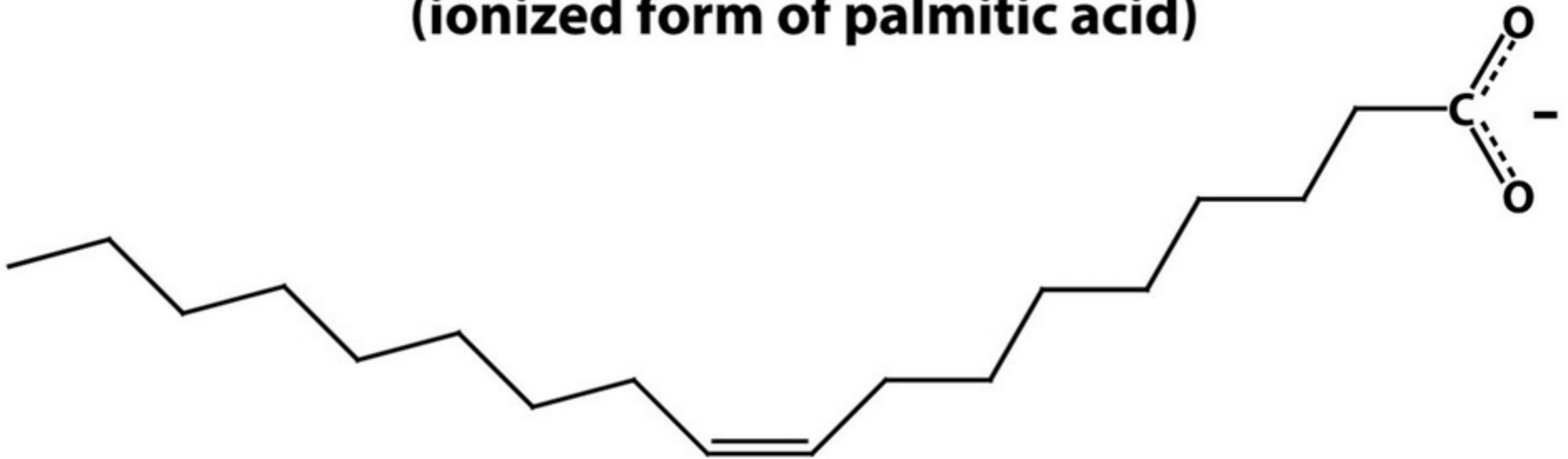
Ácidos carboxílicos
com cadeias hidrocarbônicas
comprimento variando de 4 a 36 carbonos.

Propriedades:

- dependentes do comprimento da cadeia
- dependentes do grau de insaturação



Palmitate
(ionized form of palmitic acid)

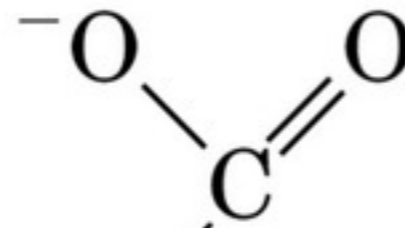


Oleate
(ionized form of oleic acid)

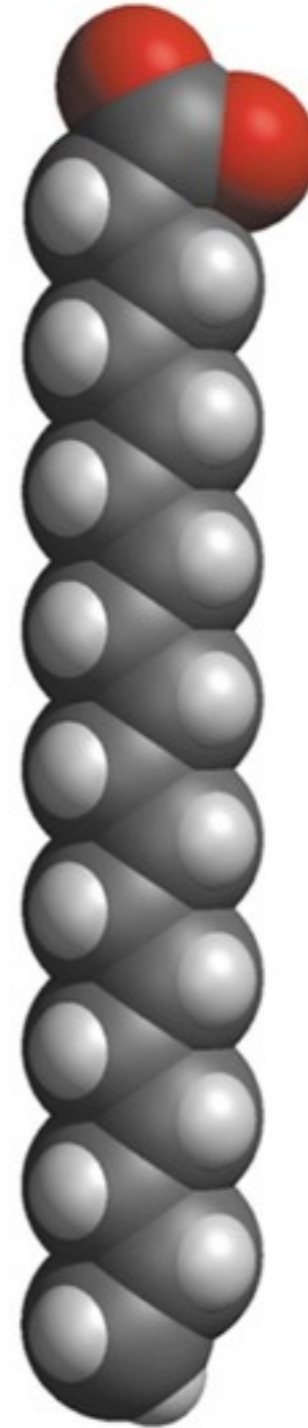
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Um ácido graxo saturado

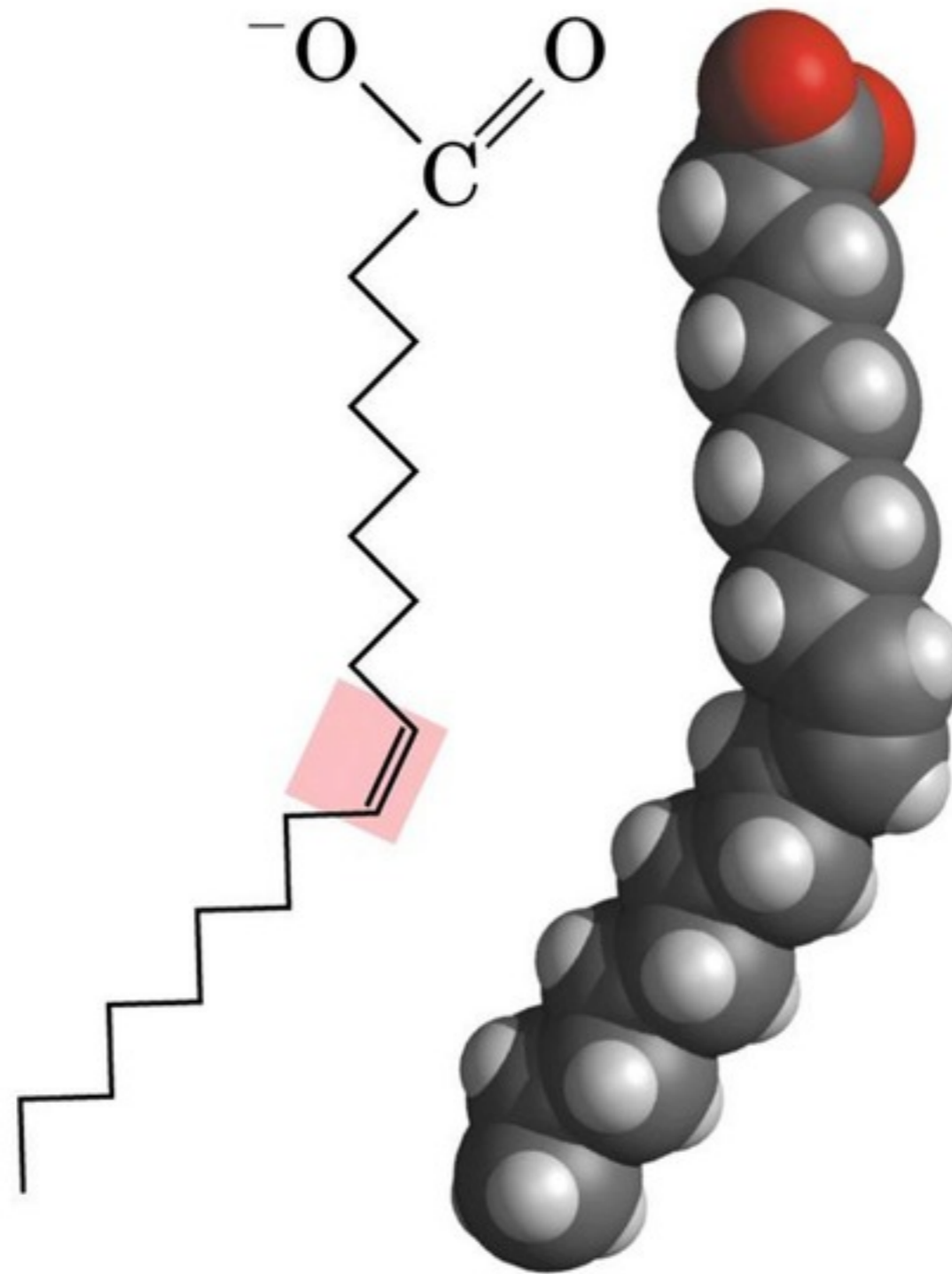
Carboxyl
group

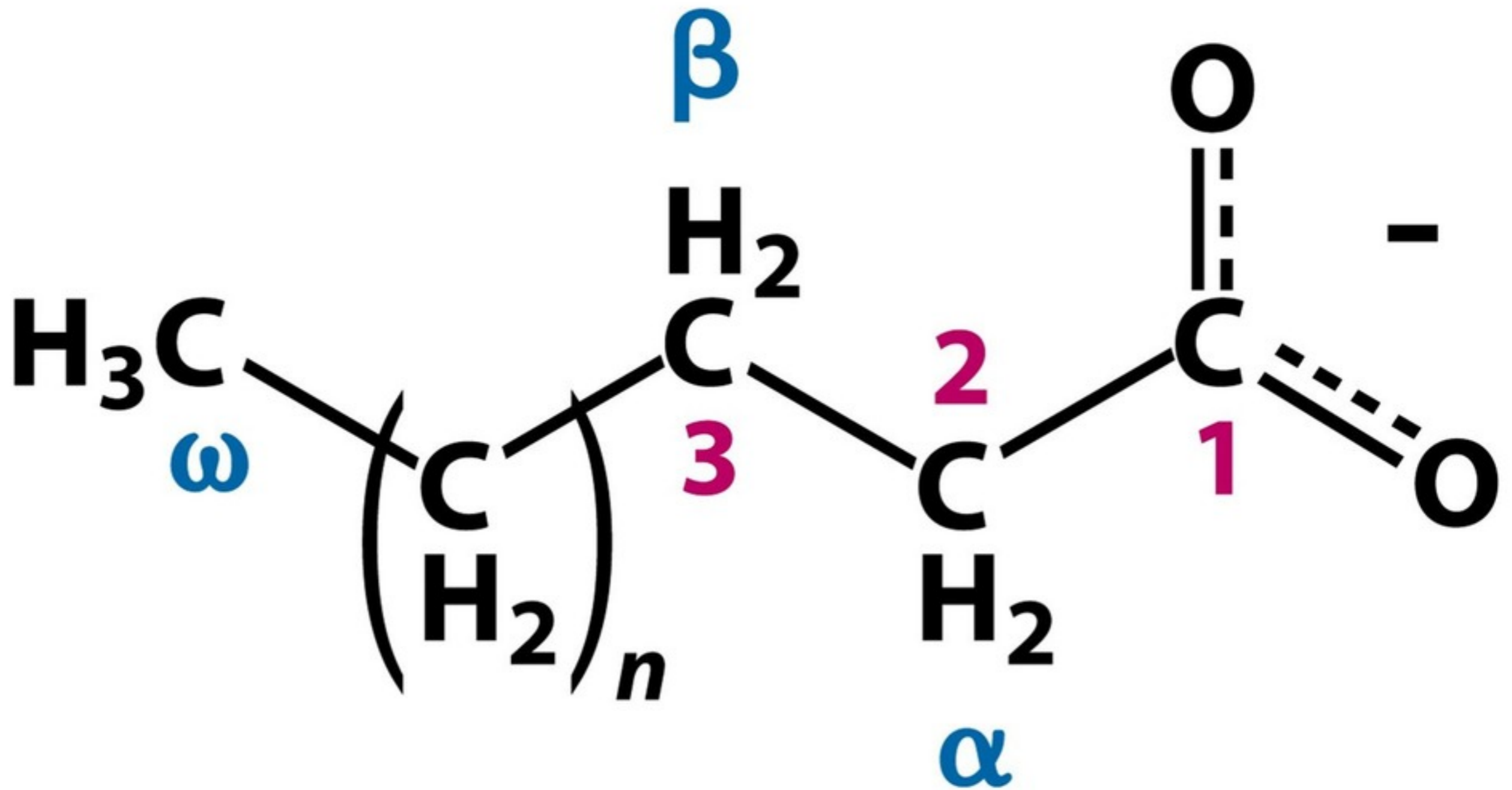


Hydrocarbon
chain



Configuração *cis* de insaturação

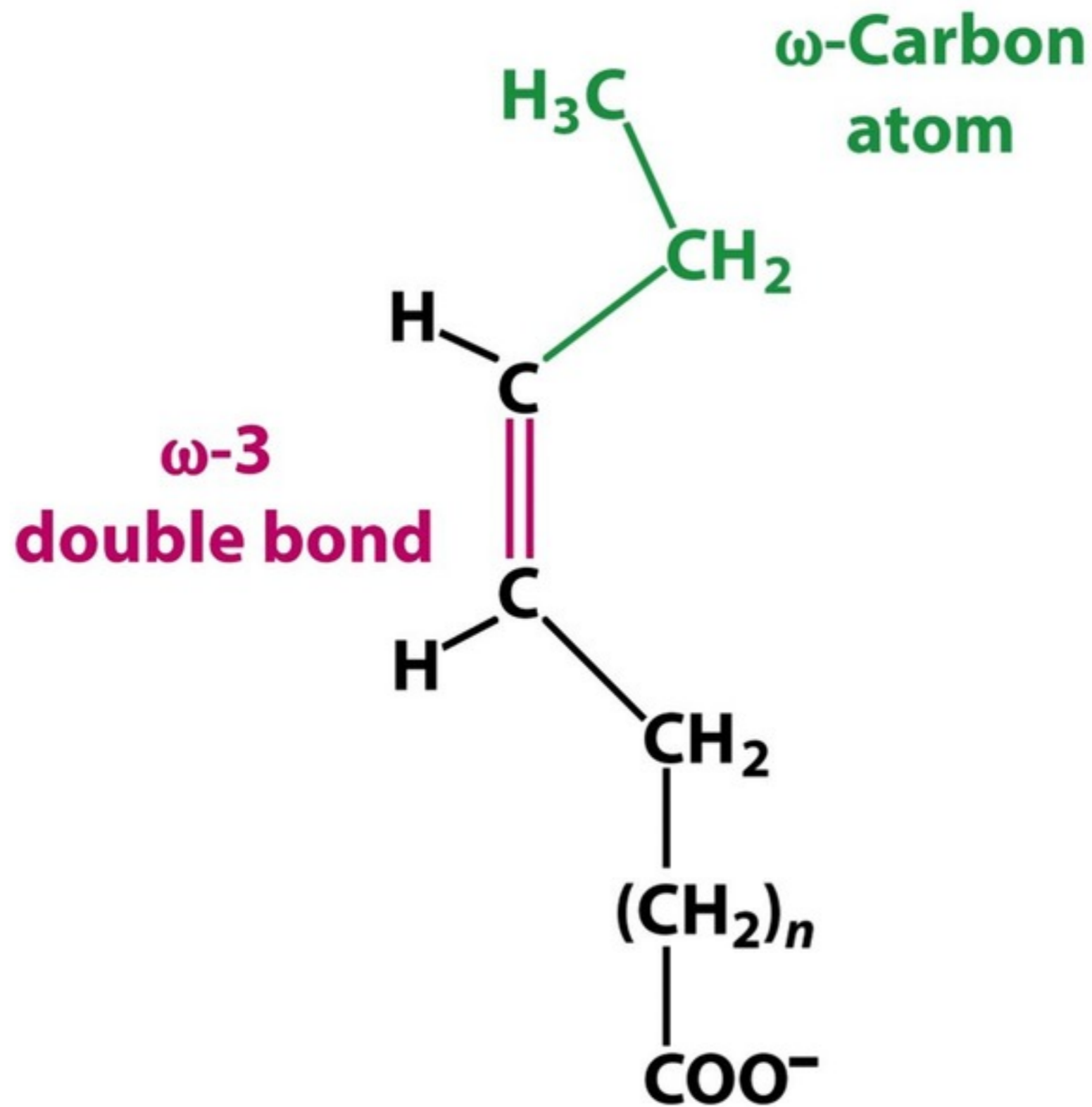




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Some Naturally Occurring Fatty Acids

Carbon skeleton	Structure*	Systematic name†	Common name (derivation)	Melting point (°C)	Solubility at 30 °C (mg/g solvent)	
					Water	Benzene
12:0	$\text{CH}_3(\text{CH}_2)_{10}\text{COOH}$	<i>n</i> -Dodecanoic acid	Lauric acid (Latin <i>laurus</i> , "laurel plant")		0.063	2,600
14:0	$\text{CH}_3(\text{CH}_2)_{12}\text{COOH}$	<i>n</i> -Tetradecanoic acid	Myristic acid (Latin <i>Myristica</i> , nutmeg genus)		0.024	874
16:0	$\text{CH}_3(\text{CH}_2)_{14}\text{COOH}$	<i>n</i> -Hexadecanoic acid	Palmitic acid (Latin <i>palma</i> , "palm tree")		0.0083	348
18:0	$\text{CH}_3(\text{CH}_2)_{16}\text{COOH}$	<i>n</i> -Octadecanoic acid	Stearic acid (Greek <i>stear</i> , "hard fat")		0.0034	124
20:0	$\text{CH}_3(\text{CH}_2)_{18}\text{COOH}$	<i>n</i> -Eicosanoic acid	Arachidic acid (Latin <i>Arachis</i> , legume genus)			
24:0	$\text{CH}_3(\text{CH}_2)_{22}\text{COOH}$	<i>n</i> -Tetracosanoic acid	Lignoceric acid (Latin <i>lignum</i> , "wood" + <i>cera</i> , "wax")			
16:1(Δ^9)	$\text{CH}_3(\text{CH}_2)_5\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$	<i>cis</i> -9-Hexadecenoic acid	Palmitoleic acid			
18:1(Δ^9)	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$	<i>cis</i> -9-Octadecenoic acid	Oleic acid (Latin <i>oleum</i> , "oil")			
18:2($\Delta^{9,12}$)	$\text{CH}_3(\text{CH}_2)_4\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$	<i>cis</i> -, <i>cis</i> -9,12-Octadecadienoic acid	Linoleic acid (Greek <i>linon</i> , "flax")			
18:3($\Delta^{9,12,15}$)	$\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$	<i>cis</i> -, <i>cis</i> -, <i>cis</i> -9,12,15-Octadecatrienoic acid	α -Linolenic acid	-		
20:4($\Delta^{5,8,11,14}$)	$\text{CH}_3(\text{CH}_2)_4\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_3\text{COOH}$	<i>cis</i> -, <i>cis</i> -, <i>cis</i> -, <i>cis</i> -5,8,11,14-Icosatetraenoic acid	Arachidonic acid	-		



An ω -3 fatty acid

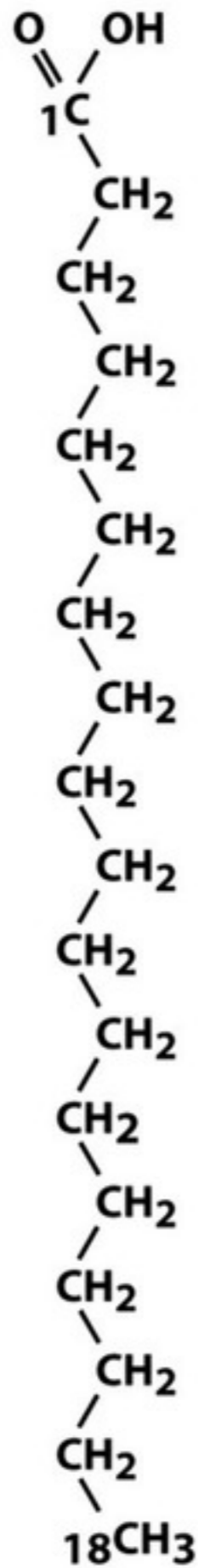
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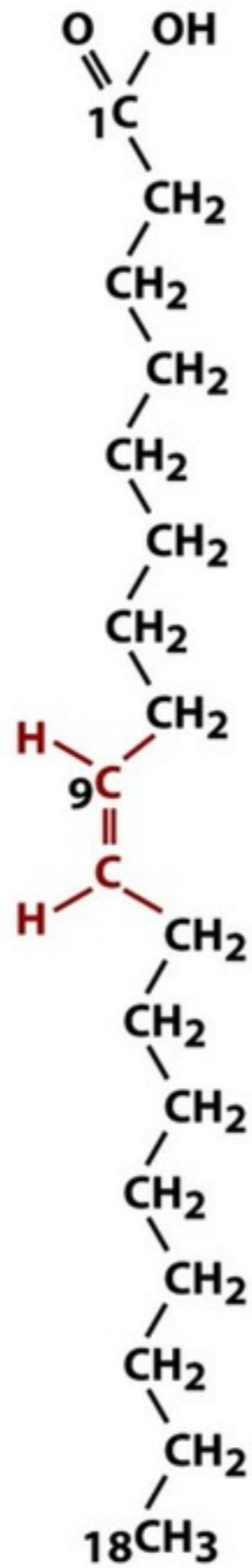
O que são ácidos graxos do tipo ômega-3 e ômega-6? Qual a importância dessas moléculas?

Collaborate!

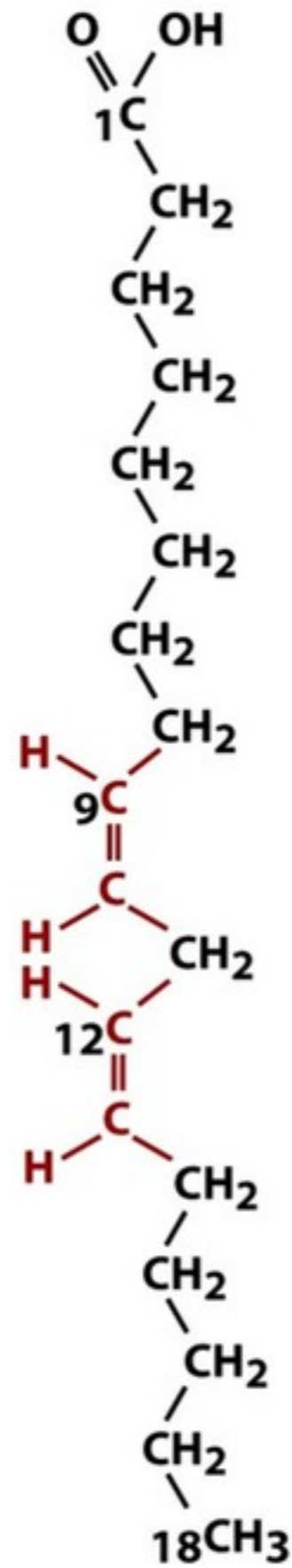
ômega 3 e 6



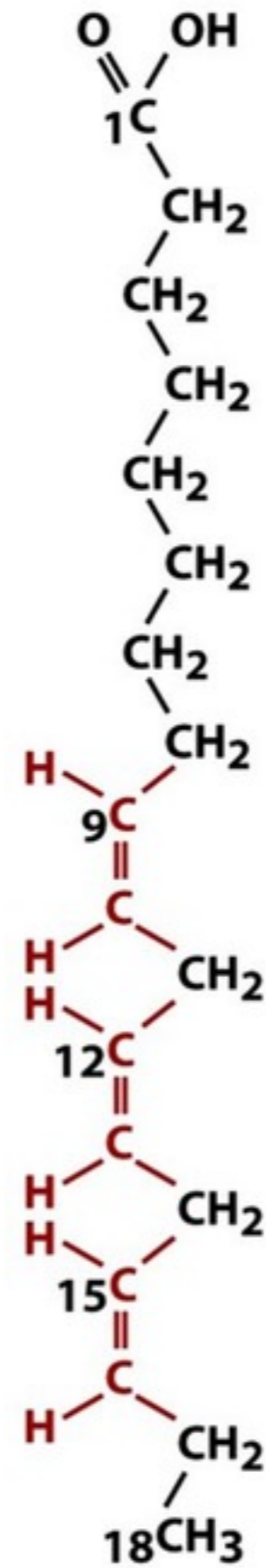
Stearic acid



Oleic acid



Linoleic acid

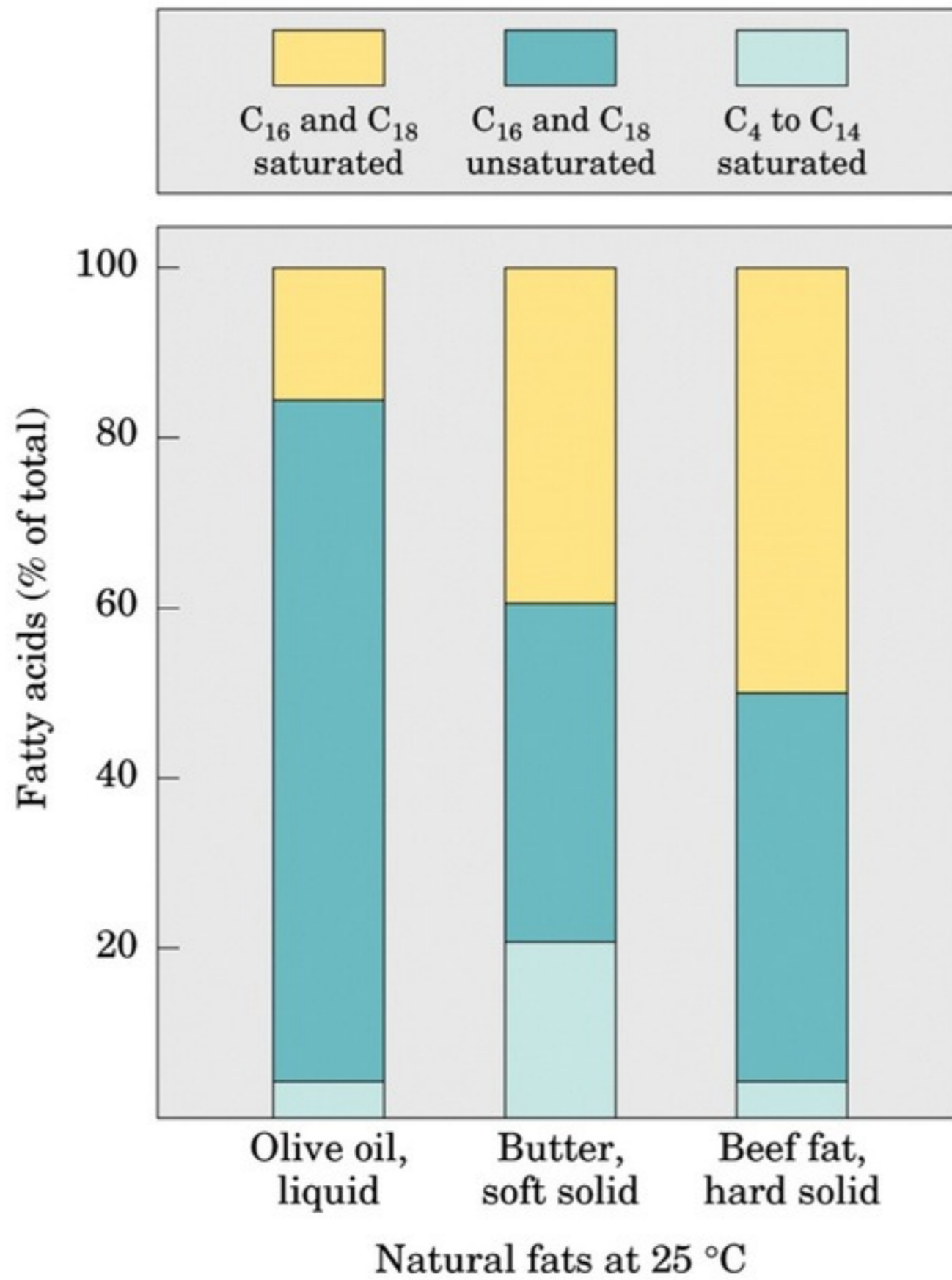


α -Linolenic acid

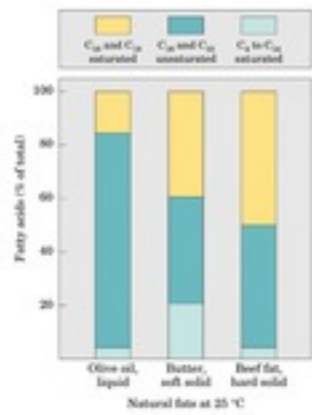
Nomenclatura

18:1(Δ^9)	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$	<i>cis</i> -9-Octadecenoic acid	Oleic acid (Latin <i>oleum</i> , "oil")
18:2($\Delta^{9,12}$)	$\text{CH}_3(\text{CH}_2)_4\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$	<i>cis</i> -, <i>cis</i> -9,12-Octadecadienoic acid	Linoleic acid (Greek <i>linon</i> , "flax")
18:3($\Delta^{9,12,15}$)	$\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CHCH}_2\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$	<i>cis</i> -, <i>cis</i> -, <i>cis</i> -9,12,15-Octadecatrienoic acid	α -Linolenic acid

18:1 $n-9$	Oleic acid	9-Octadecenoic acid	$\text{CH}_3(\text{CH}_2)_7\text{CH}=\text{CH}(\text{CH}_2)_7\text{COOH}$
18:2 $n-6$	Linoleic acid	9,12-Octadecadienoic acid	$\text{CH}_3(\text{CH}_2)_4(\text{CH}=\text{CHCH}_2)_2(\text{CH}_2)_6\text{COOH}$
18:3 $n-3$	α -Linolenic acid	9,12,15-Octadecatrienoic acid	$\text{CH}_3\text{CH}_2(\text{CH}=\text{CHCH}_2)_3(\text{CH}_2)_6\text{COOH}$



Open Ended Question



A composição de três alimentos compostos de ácidos graxos está mostrado no gráfico. Qual a relação entre a proporção de ácidos graxos saturados ou insaturados e a temperatura de fusão desses alimentos?

Some Naturally Occurring Fatty Acids

Carbon skeleton	Structure*	Systematic name†	Common name (derivation)	Melting point (°C)	Solubility at 30 °C (mg/g solvent)	
					Water	Benzene
12:0	CH ₃ (CH ₂) ₁₀ COOH	<i>n</i> -Dodecanoic acid	Lauric acid (Latin <i>laurus</i> , "laurel plant")	44.2	0.063	2,600
14:0	CH ₃ (CH ₂) ₁₂ COOH	<i>n</i> -Tetradecanoic acid	Myristic acid (Latin <i>Myristica</i> , nutmeg genus)	53.9	0.024	874
16:0	CH ₃ (CH ₂) ₁₄ COOH	<i>n</i> -Hexadecanoic acid	Palmitic acid (Latin <i>palma</i> , "palm tree")	63.1	0.0083	348
18:0	CH ₃ (CH ₂) ₁₆ COOH	<i>n</i> -Octadecanoic acid	Stearic acid (Greek <i>stear</i> , "hard fat")	69.6	0.0034	124
20:0	CH ₃ (CH ₂) ₁₈ COOH	<i>n</i> -Eicosanoic acid	Arachidic acid (Latin <i>Arachis</i> , legume genus)	76.5		
24:0	CH ₃ (CH ₂) ₂₂ COOH	<i>n</i> -Tetracosanoic acid	Lignoceric acid (Latin <i>lignum</i> , "wood" + <i>cera</i> , "wax")	86.0		
16:1(Δ ⁹)	CH ₃ (CH ₂) ₅ CH=CH(CH ₂) ₇ COOH	<i>cis</i> -9-Hexadecenoic acid	Palmitoleic acid	-0.5		
18:1(Δ ⁹)	CH ₃ (CH ₂) ₇ CH=CH(CH ₂) ₇ COOH	<i>cis</i> -9-Octadecenoic acid	Oleic acid (Latin <i>oleum</i> , "oil")	13.4		
18:2(Δ ^{9,12})	CH ₃ (CH ₂) ₄ CH=CHCH ₂ CH=CH(CH ₂) ₇ COOH	<i>cis</i> -, <i>cis</i> -9,12-Octadecadienoic acid	Linoleic acid (Greek <i>linon</i> , "flax")	-5		
18:3(Δ ^{9,12,15})	CH ₃ CH ₂ CH=CHCH ₂ CH=CHCH ₂ CH=CH(CH ₂) ₇ COOH	<i>cis</i> -, <i>cis</i> -, <i>cis</i> -9,12,15-Octadecatrienoic acid	α-Linolenic acid	-11		
20:4(Δ ^{5,8,11,14})	CH ₃ (CH ₂) ₄ CH=CHCH ₂ CH=CHCH ₂ CH=CHCH ₂ CH=CH(CH ₂) ₃ COOH	<i>cis</i> -, <i>cis</i> -, <i>cis</i> -, <i>cis</i> -5,8,11,14-Icosatetraenoic acid	Arachidonic acid	-49.5		

Ácidos Graxos

Ponto de fusão dos ácidos graxos:

- dependentes do comprimento da cadeia hidrocarbônica
- dependentes do grau de insaturação
 - monoinsaturados
 - poliinsaturados

Por quê um ácido graxo de mesmo número de carbonos, em uma dada temperatura, assume um aspecto de cera se for saturado, ou de óleo se for insaturado ?

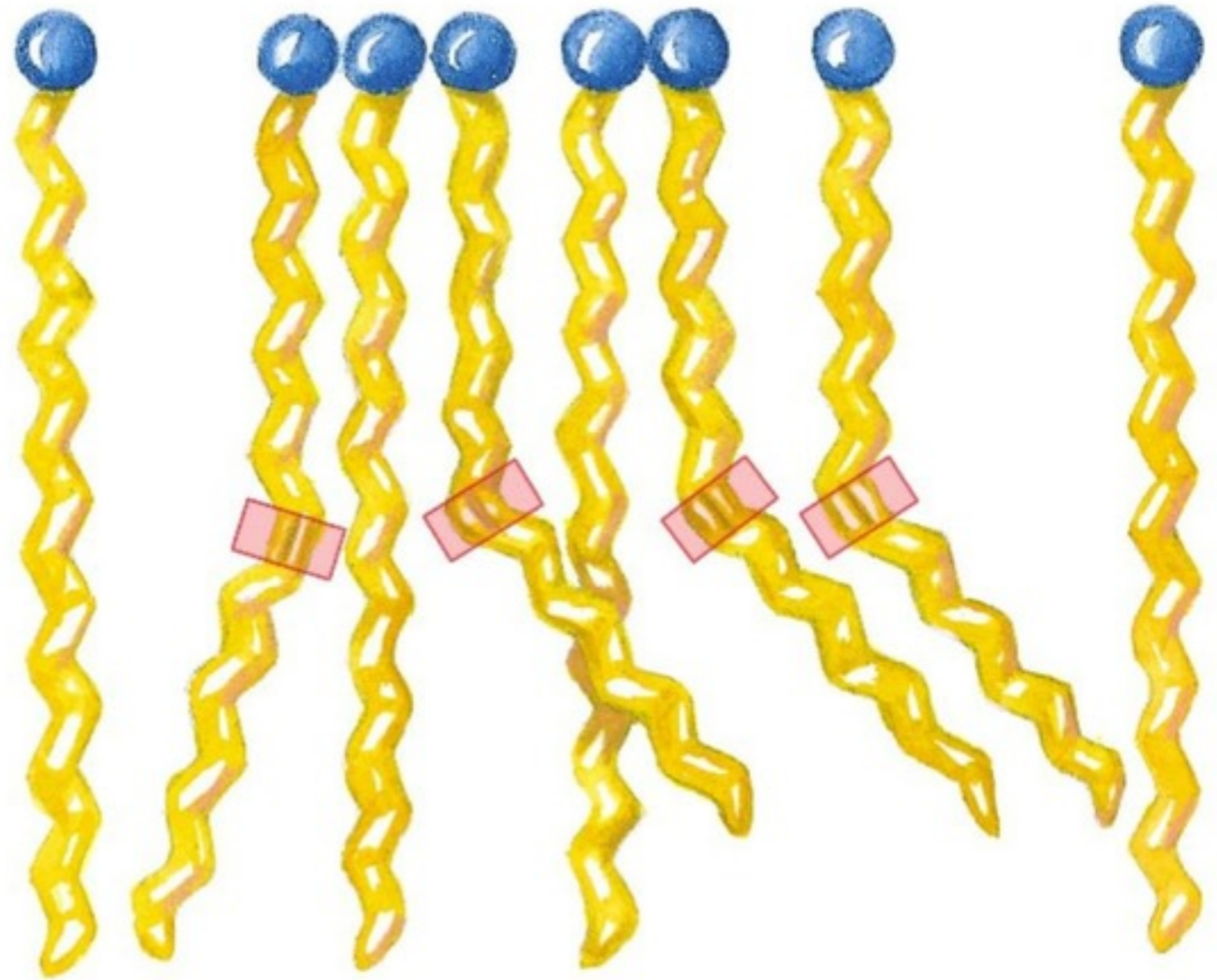
Open Ended Question

Por que um ácido graxo de mesmo número de carbonos, em uma dada temperatura, assume um aspecto de cera se for saturado, ou de óleo se for insaturado ?

Relação entre insaturação e grau de empacotamento



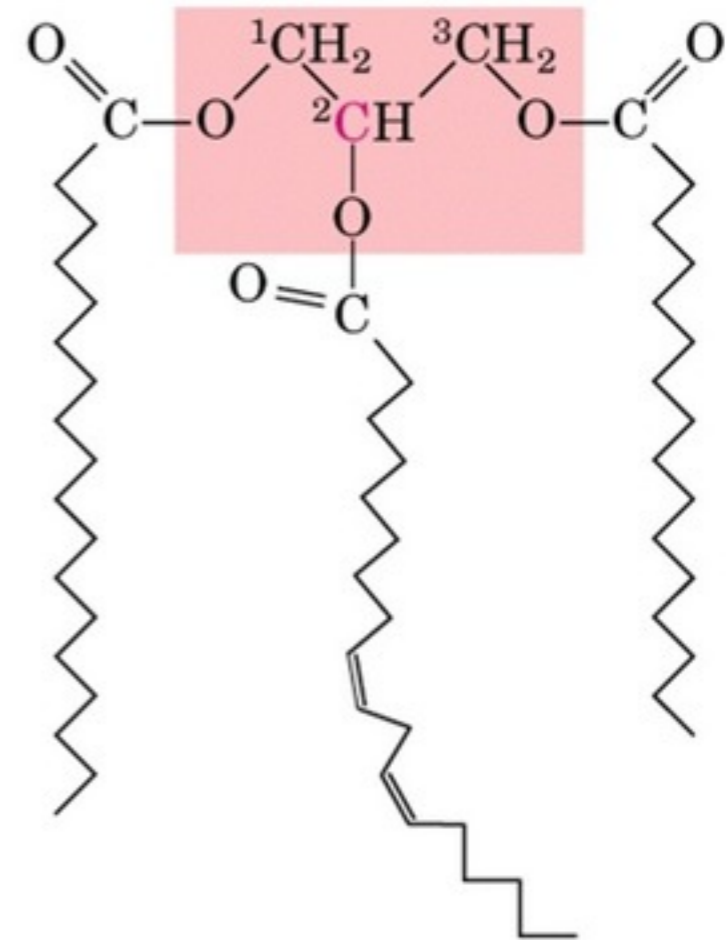
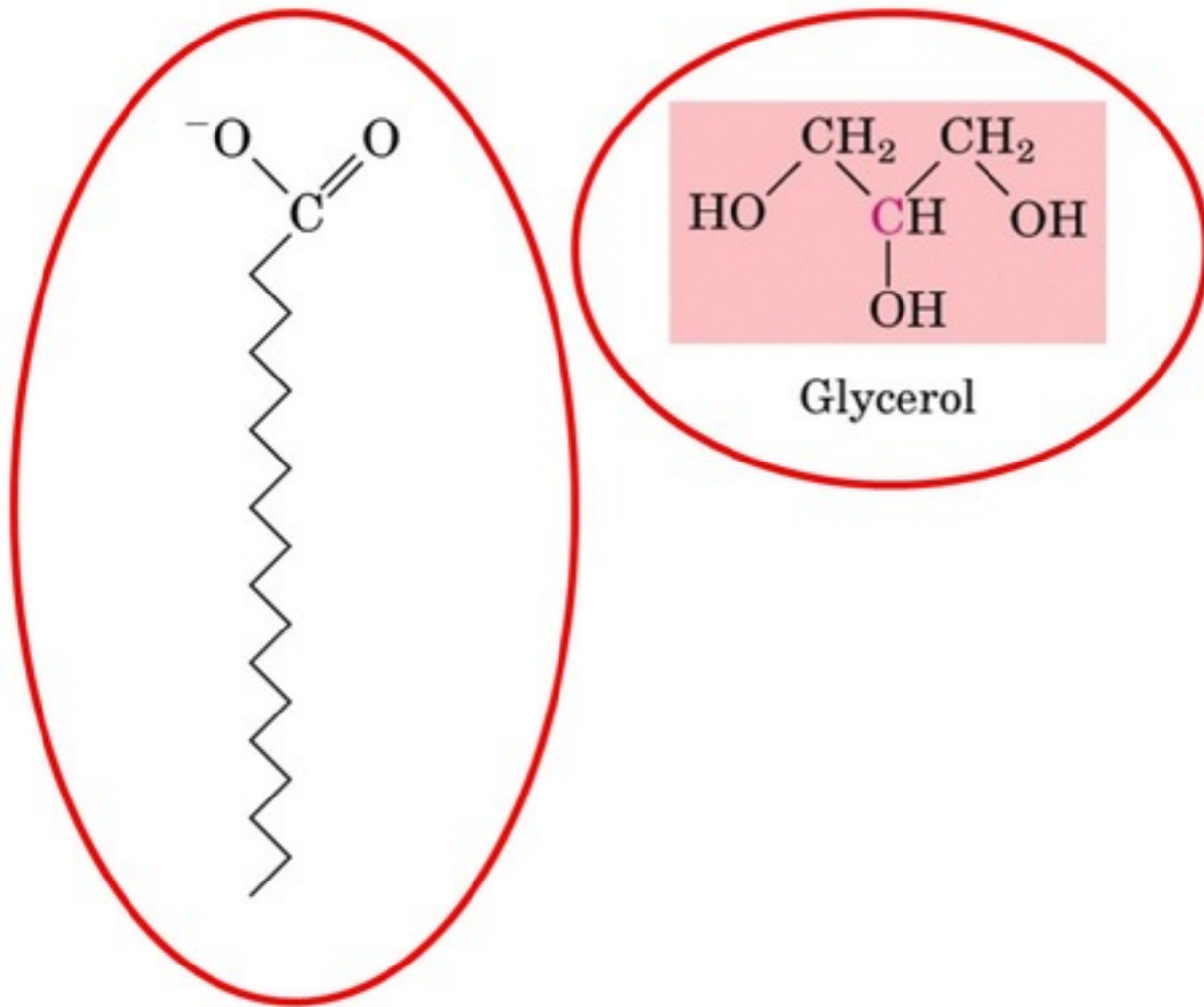
Saturated fatty acids



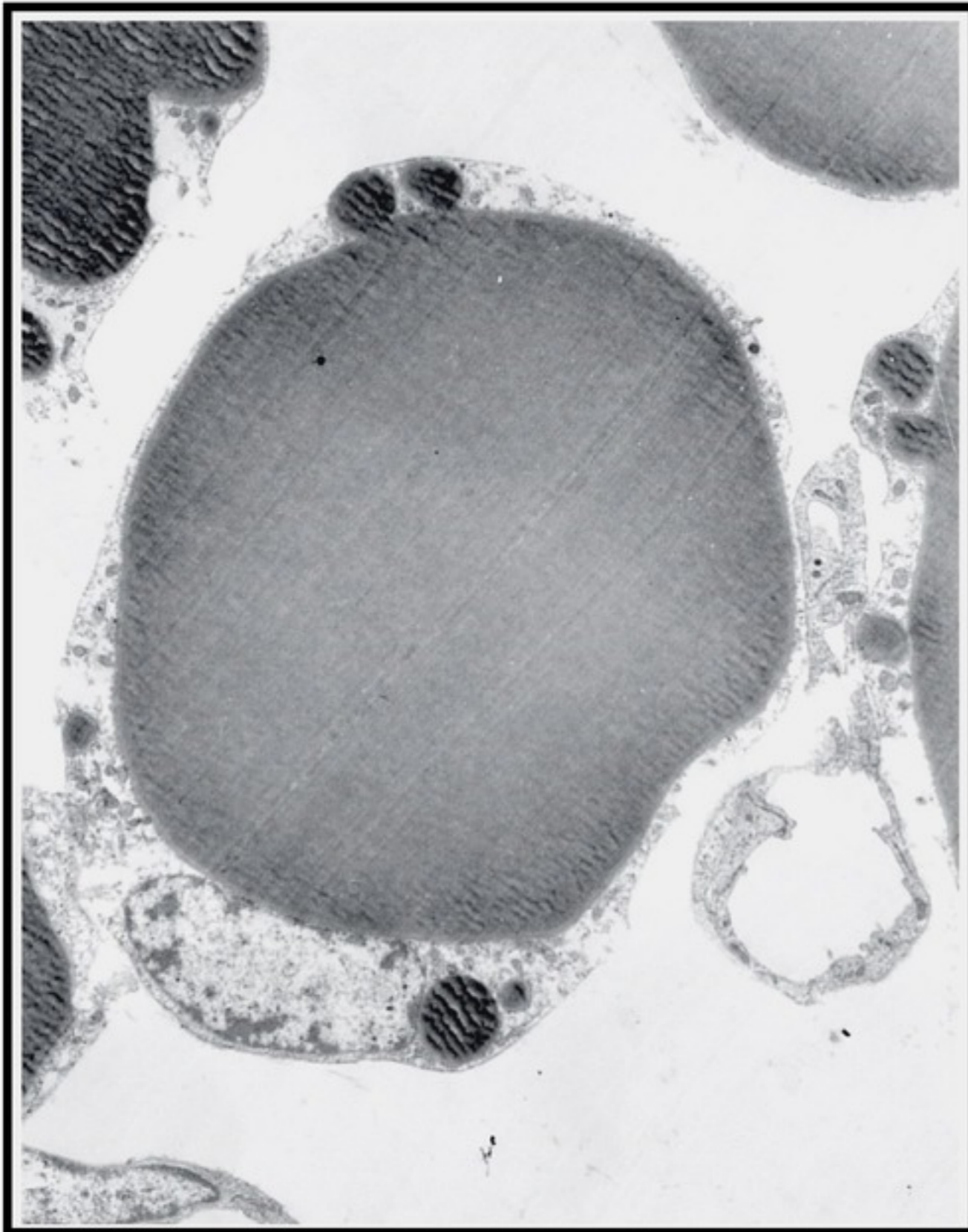
Mixture of saturated and unsaturated fatty acids

Triacilgliceróis

Também chamados de triglicerídios ou gorduras neutras, os triacilgliceróis podem ser simples ou mistos.



Os triacilgliceróis são reservas altamente concentradas de energia metabólica porque são reduzidos e anidros.

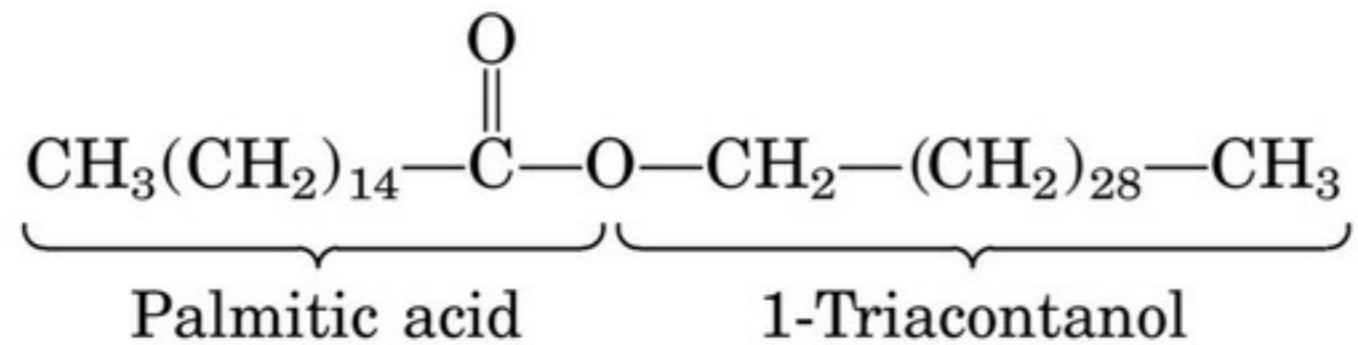


Micrografia eletrônica de uma adipócito

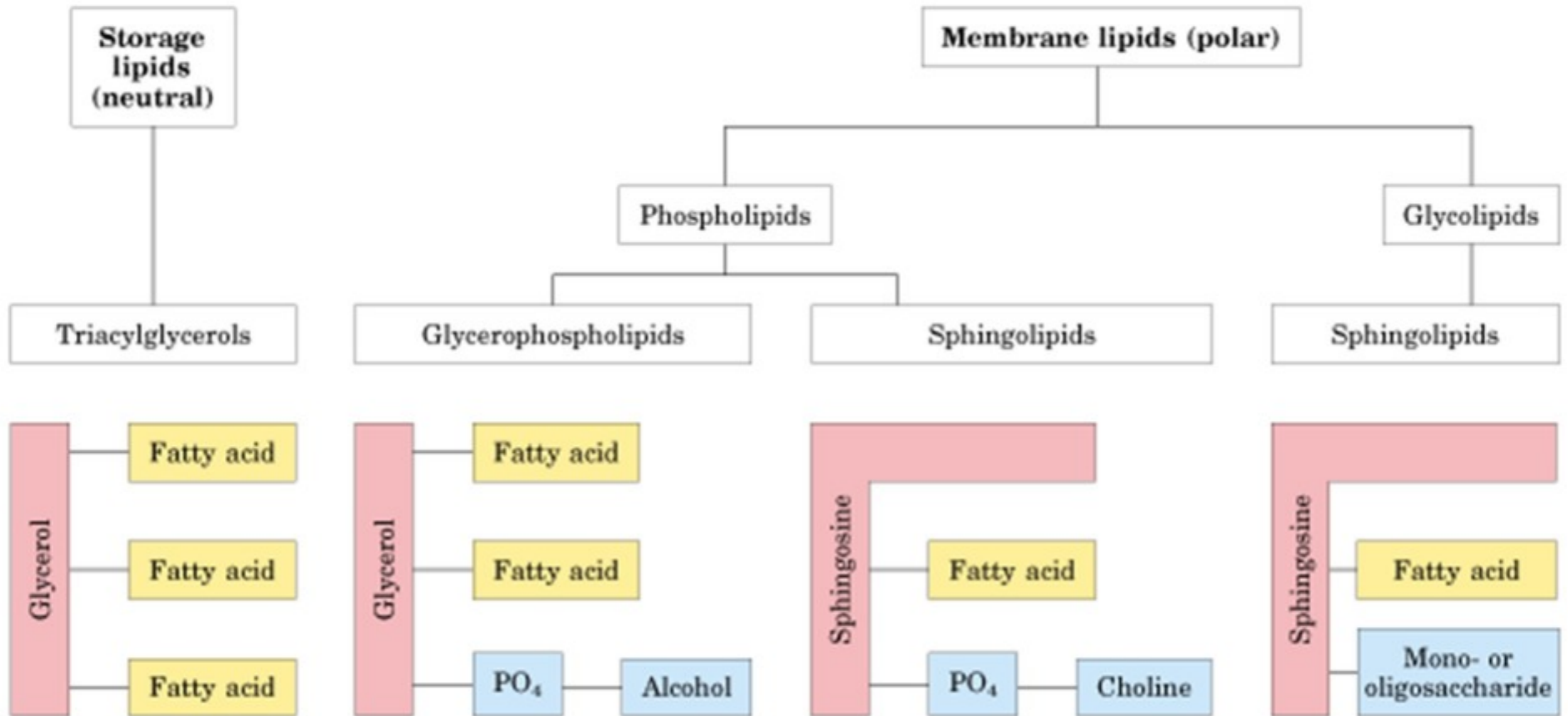
- **As células adiposas são especializadas para síntese e armazenamento de triacilgliceróis.**
- **Os triacilgliceróis são apolares e por isto são armazenados em forma quase anidra (sem água).**
- **As reservas de triacilgliceróis podem sustentar energeticamente as funções biológicas por várias semanas.**

Ceras

São ésteres de ácidos graxos saturados e insaturados de cadeia longa, com álcoois de cadeia longa. Além de armazenar energia, também servem como repelentes de água.



Lipídeos





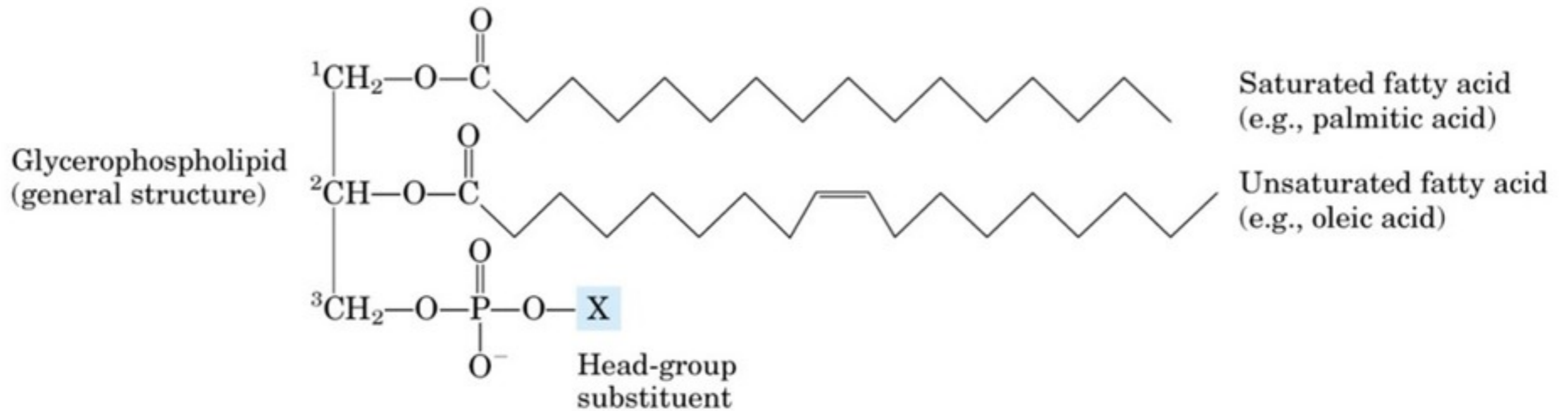
Lipídeos de armazenamento e estruturais

Qual é a diferença entre um lipídeo de armazenamento e um lipídeo estrutural?

Collaborate!

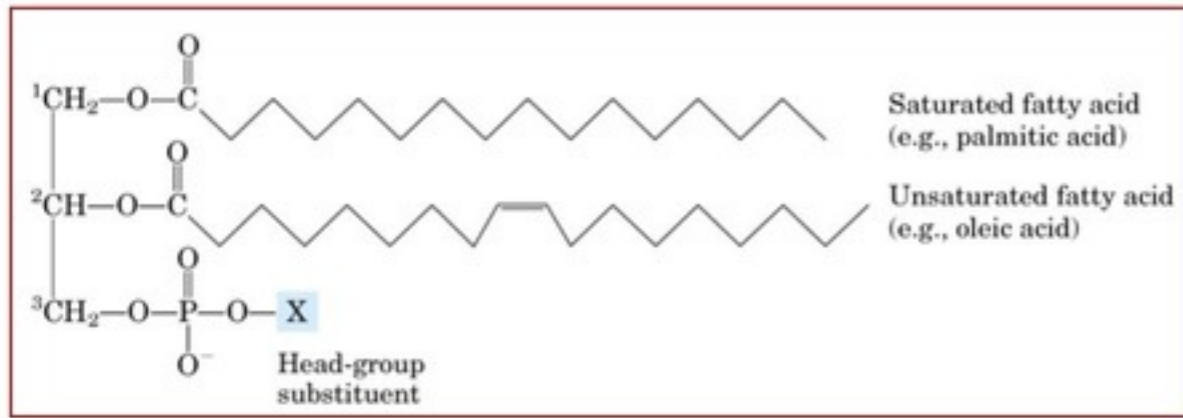
Lipídeos de armazenamento e estruturais

Glicerofosfolipídios



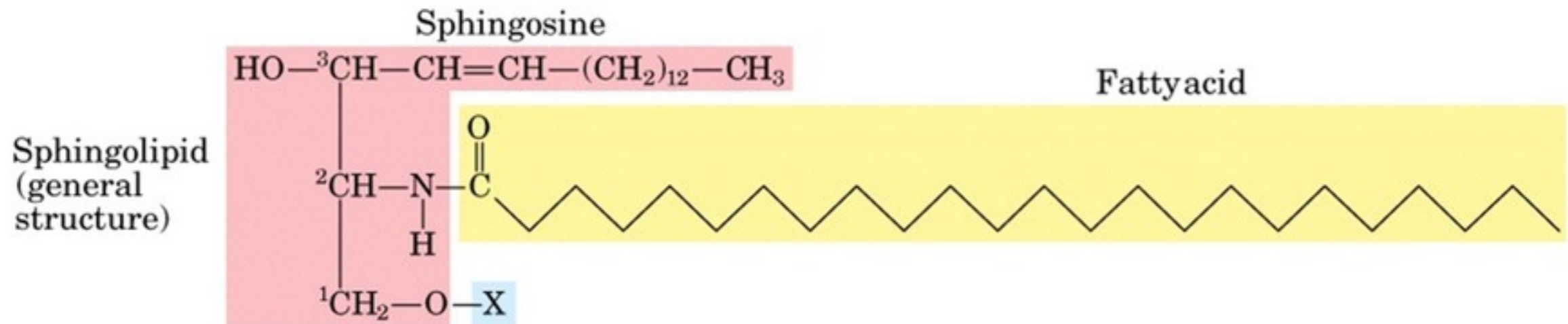
Existe uma enorme variedade de ácidos graxos que podem compor cada um dos glicerofosfolipídios

Glicerofosfolípídios



Name of glycerophospholipid	Name of X	Formula of X
Phosphatidic acid	—	— H
Phosphatidylethanolamine	Ethanolamine	— CH ₂ -CH ₂ -NH ₃ ⁺
Phosphatidylcholine	Choline	— CH ₂ -CH ₂ -N ⁺ (CH ₃) ₃
Phosphatidylserine	Serine	— CH ₂ -CH(NH ₃ ⁺)COO ⁻
Phosphatidylglycerol	Glycerol	— CH ₂ -CH(OH)-CH ₂ -OH
Phosphatidylinositol 4,5-bisphosphate	<i>myo</i> -Inositol 4,5-bisphosphate	

Esfingolipídios



É a segunda maior classe de lipídios de membrana.

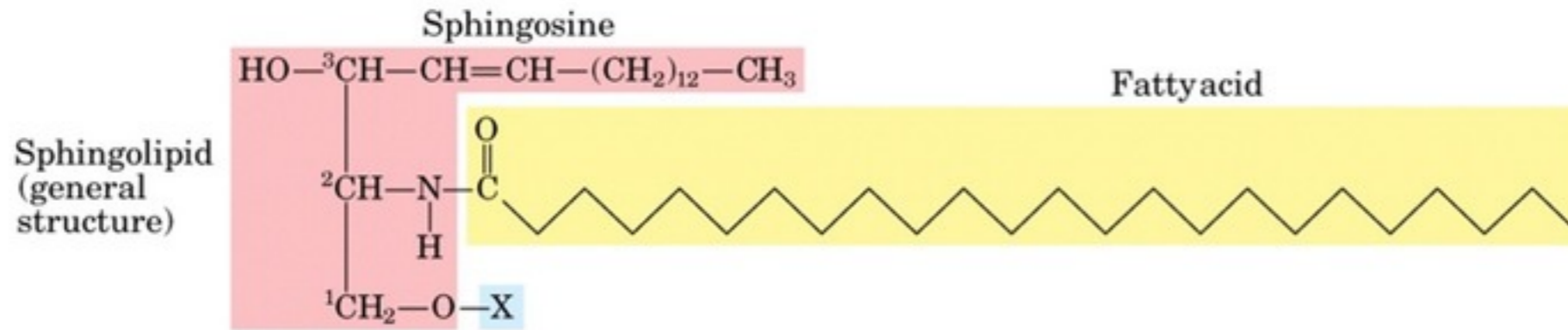
Também possuem um grupo cabeça polar e duas caudas apolares.

Ao invés do glicerol → um aminoálcool chamado esfingosina.

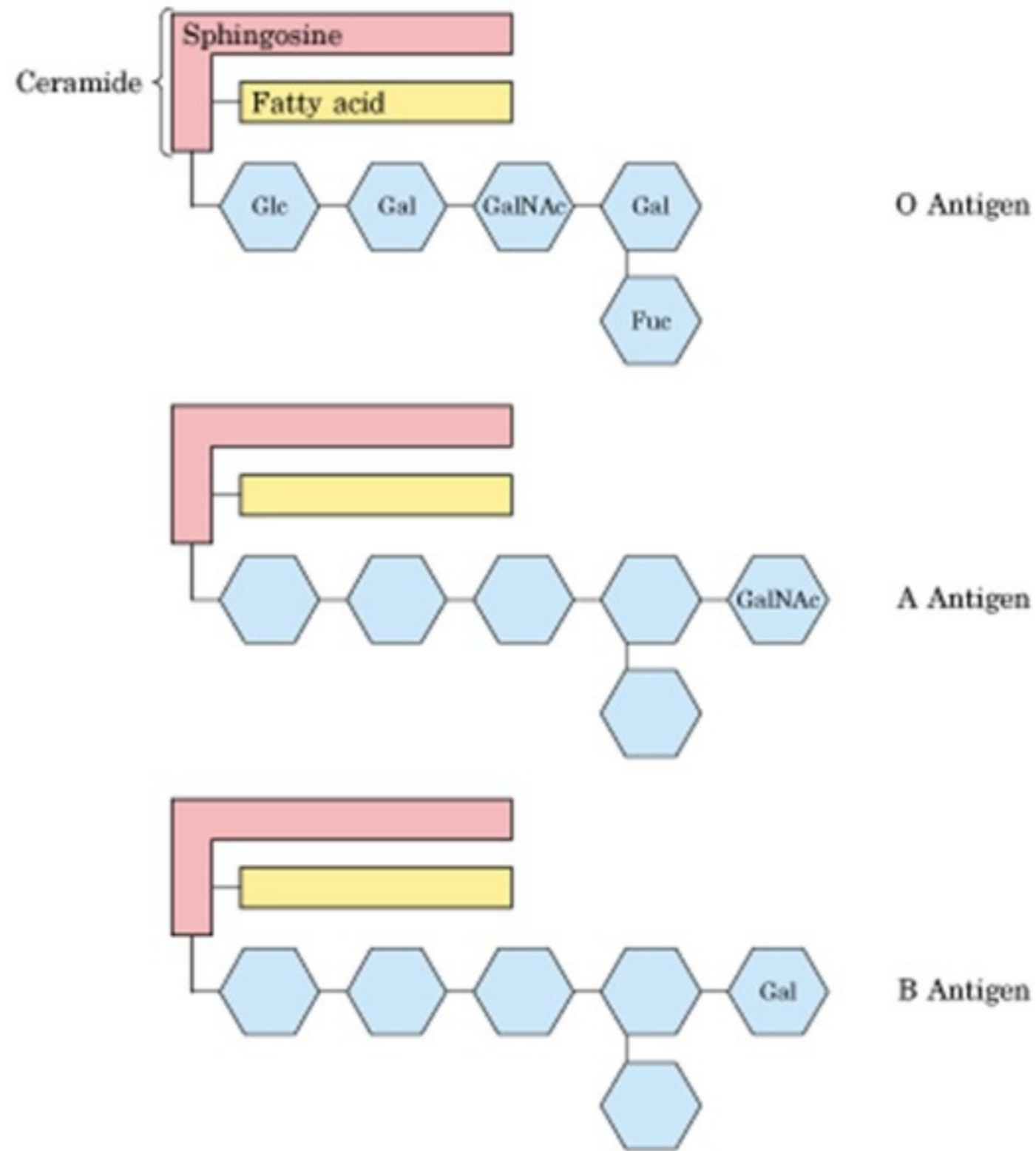
O grupo cabeça polar pode estar ligado através de uma ligação glicosídica ou de uma ligação fosfodiéster.

São importantes sítios de reconhecimento celular.

Esfingolípídios

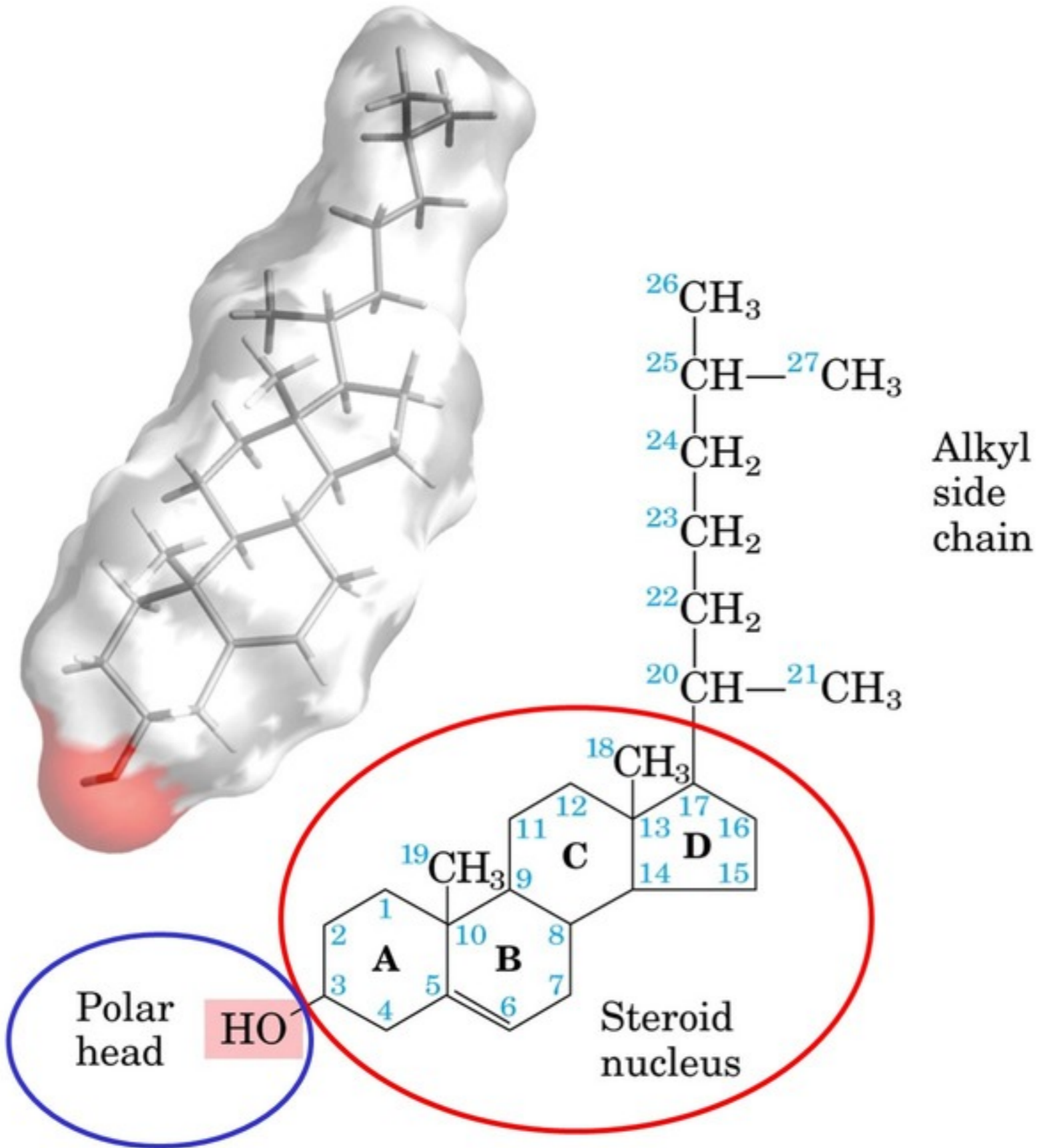


Name of sphingolipid	Name of X	Formula of X
Ceramide	—	—H
Sphingomyelin	Phosphocholine	$\begin{array}{c} \text{O} \\ \parallel \\ \text{—P—O—CH}_2\text{—CH}_2\text{—N}^+(\text{CH}_3)_3 \\ \\ \text{O}^- \end{array}$
Neutral glycolipids Glucosylcerebroside	Glucose	
Lactosylceramide (a globoside)	Di-, tri-, or tetrasaccharide	
Ganglioside GM2	Complex oligosaccharide	

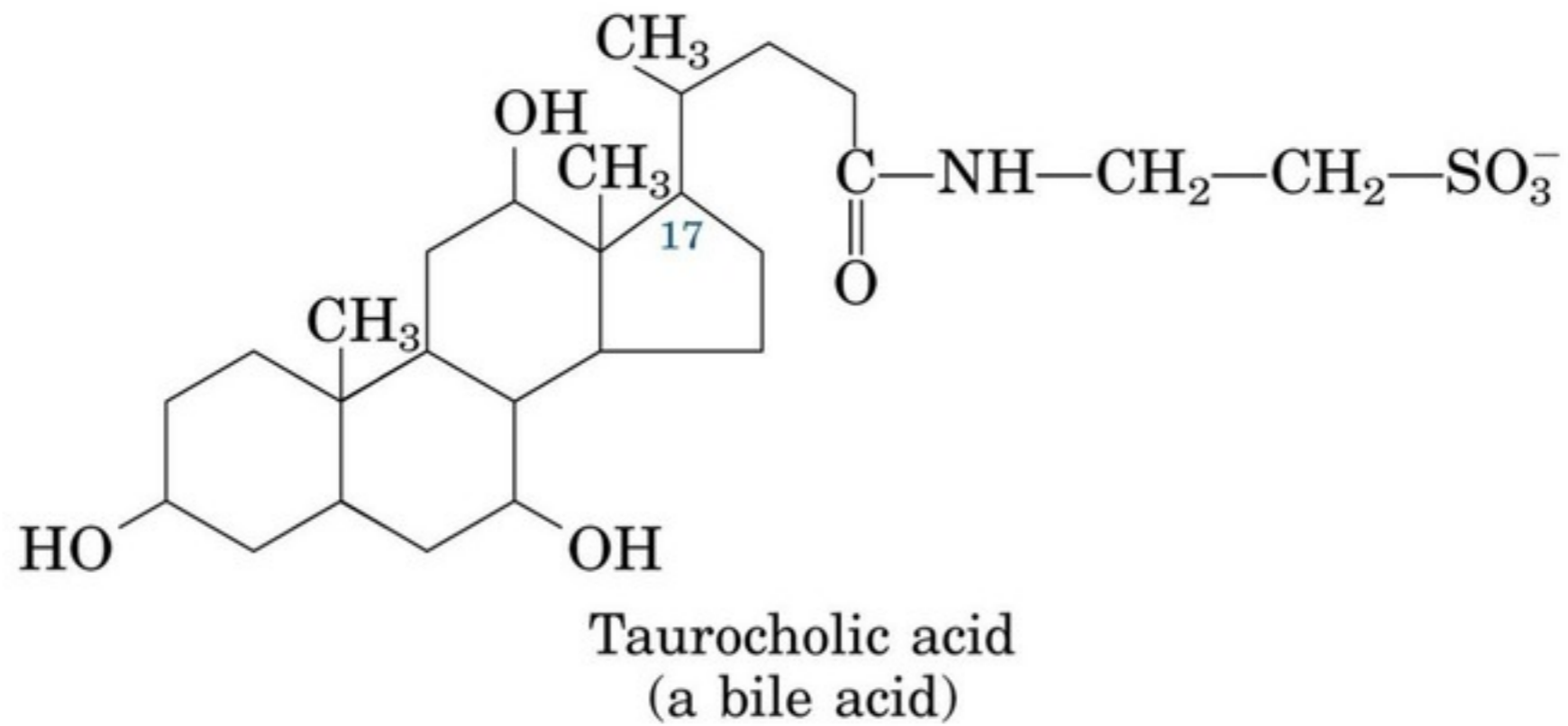


Glicoesfingolipideos : grupo sanguineo

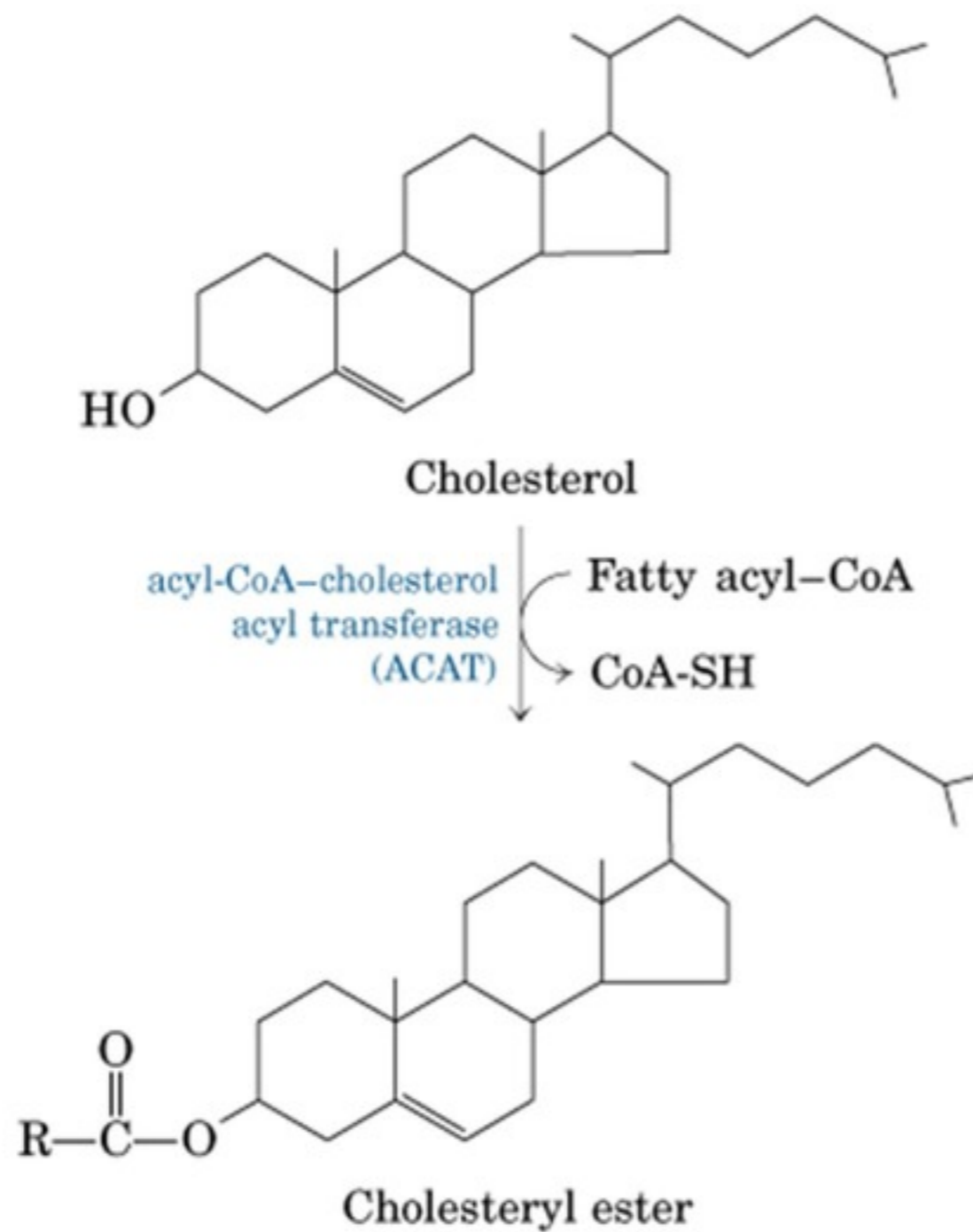
Esteróis



O colesterol também é precursor dos ácidos biliares, que emulsificam as gorduras para facilitar ação das lipases.



Destino do Colesterol



Destino do Colesterol

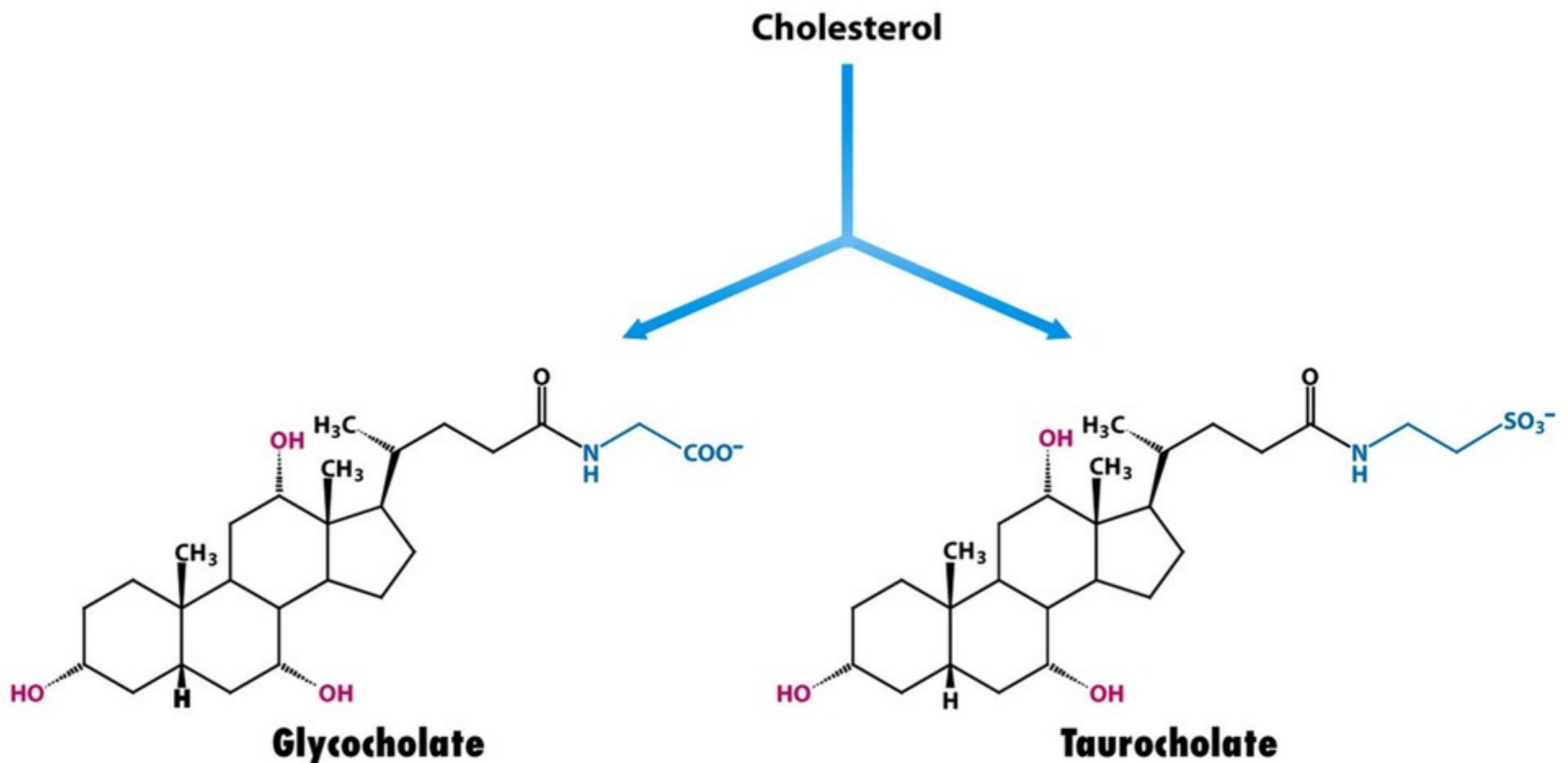


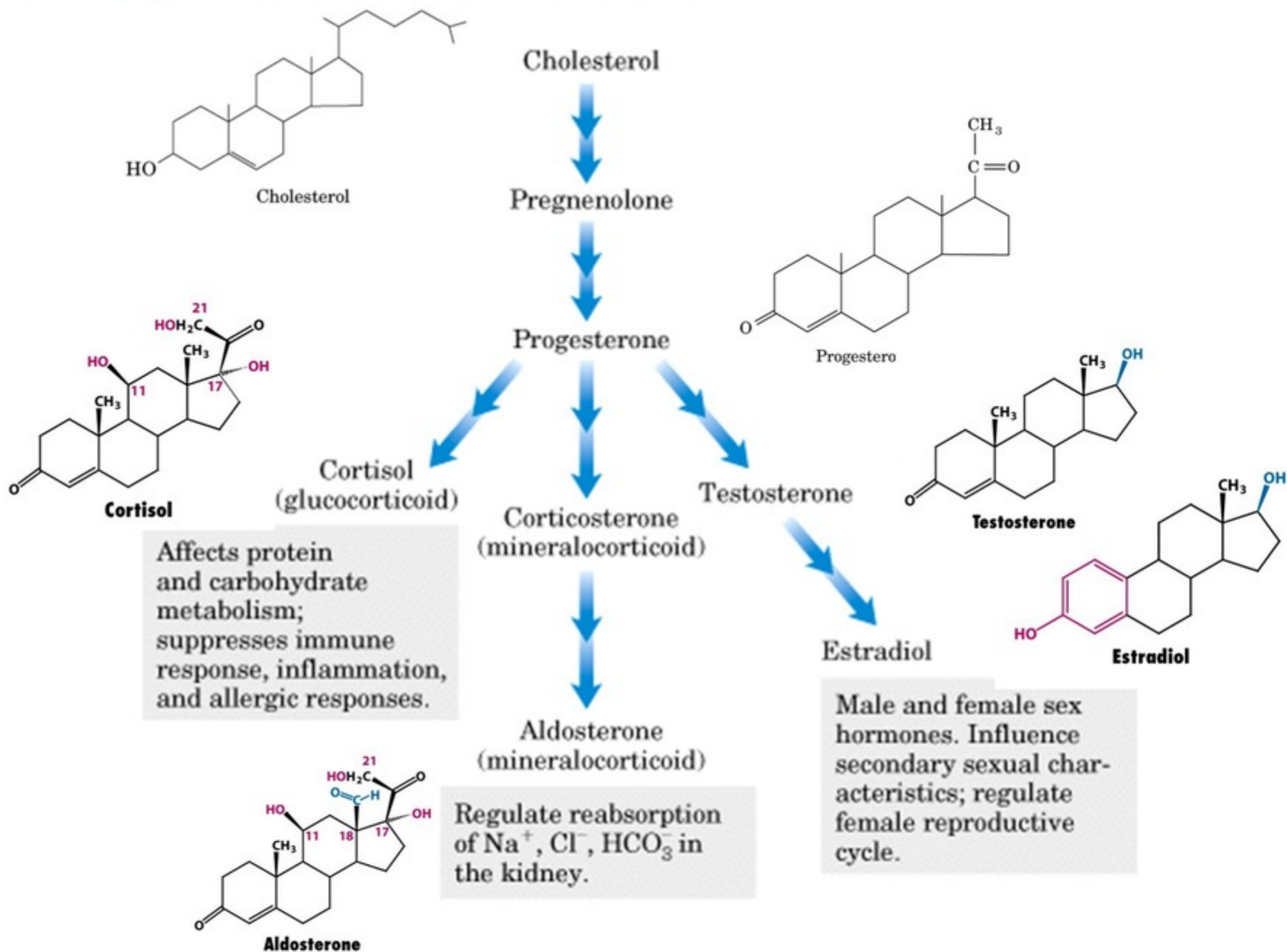
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Sais biliares

Aumento da excreção de colesterol para bile = cálculos biliares

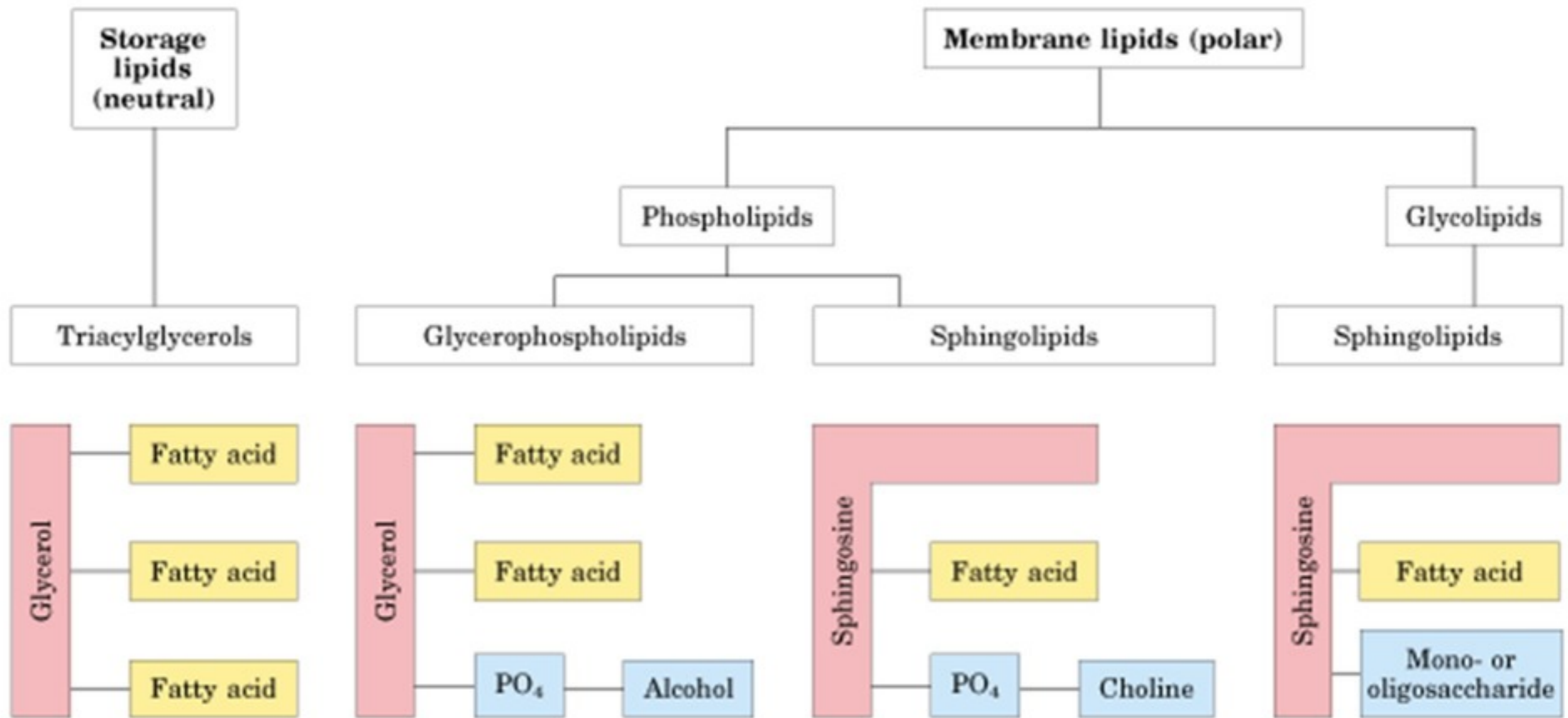
Emulsificação e solubilização lipídeos

Destino do Colesterol



Lipídeos:

Sinalizadores, cofatores e pigmentos



80% massa de um adipócito (sem contar água)

5-10% massa da célula (sem contar água)

Fosfatitilinositóis

Phosphatidylinositol

phosphorylation
in plasma
membrane

2ATP → 2ADP

Phosphatidylinositol 4,5-bisphosphate

hormone-sensitive
phospholipase C
in plasma
membrane

H₂O

Inositol 1,4,5-trisphosphate

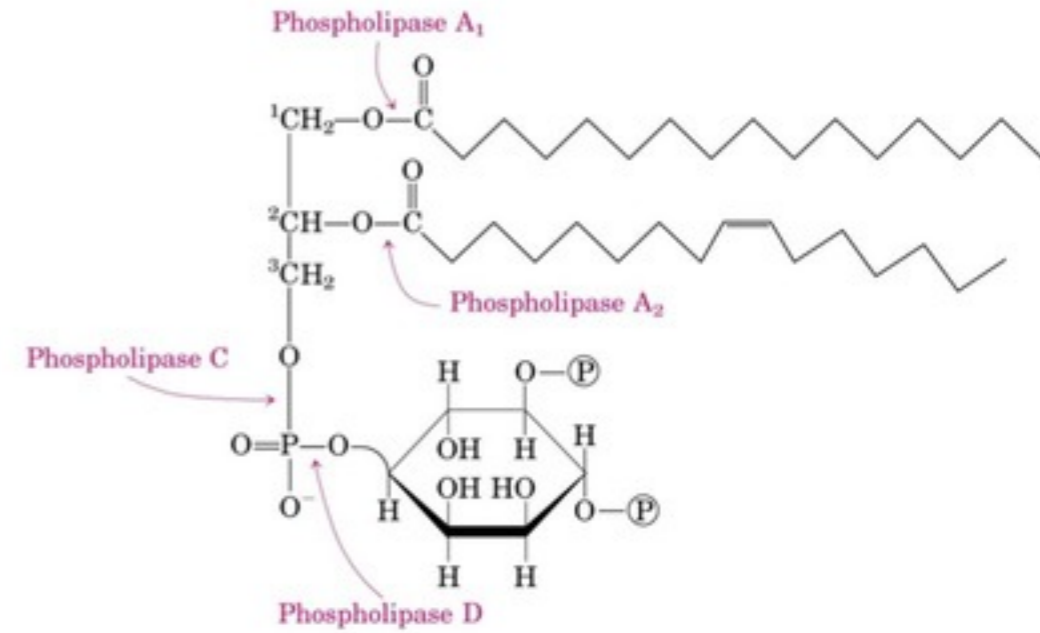
Diacylglycerol

Release of intracellular Ca²⁺

Activation of
protein kinase C

Regulation of other enzymes
(by Ca²⁺)

Regulation of other enzymes
(by protein phosphorylation)

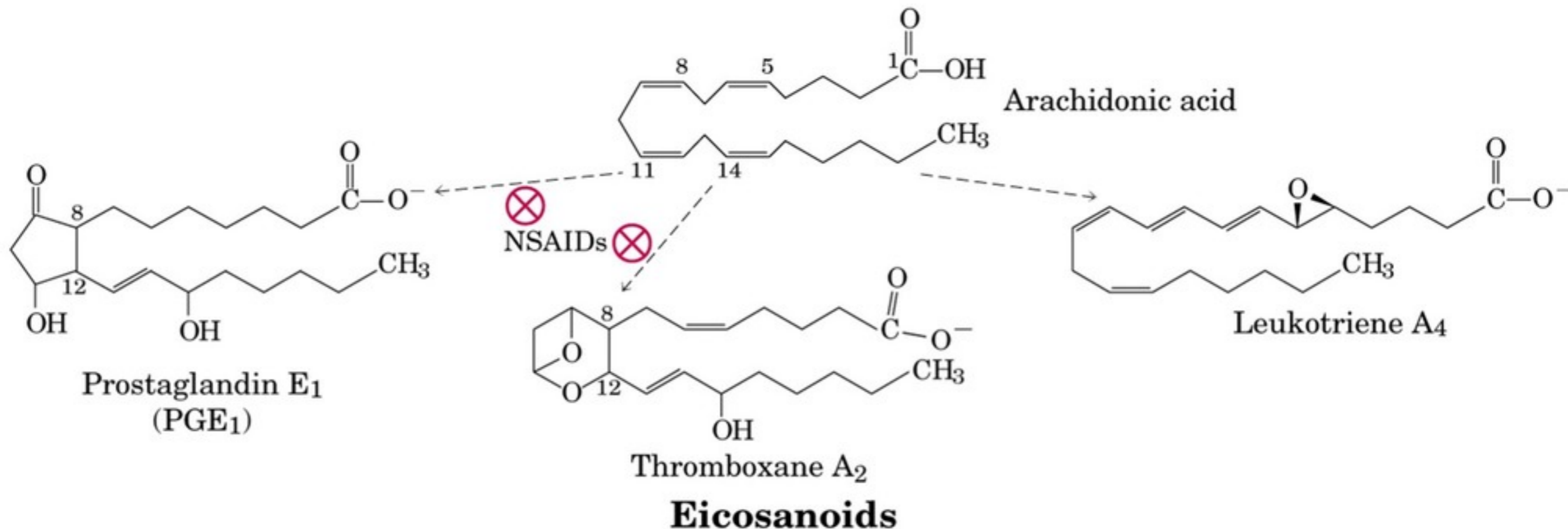
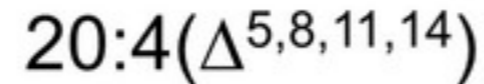


?

Eicosanóides

São derivados de ácidos graxos que agem como hormônios parácrinos. Envolvidos em inflamação, dor, funções reprodutivas, etc.

Derivados do ácido graxo de 20 carbonos, ácido aracdônico.

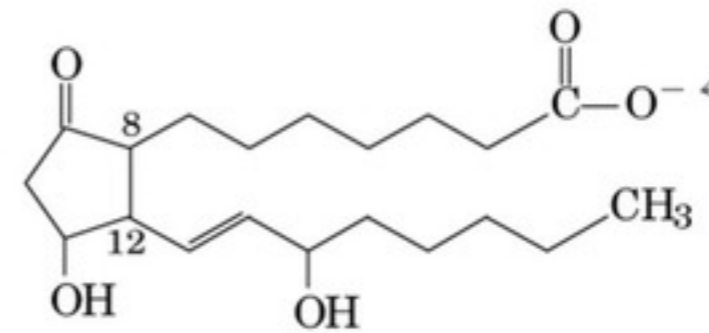


Eicosanóides

Podem ser divididos em:

- prostaglandinas

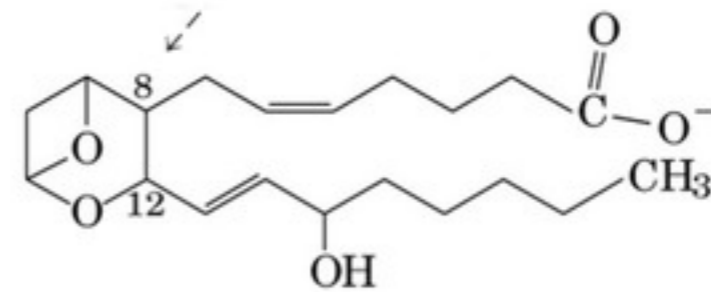
Ex. Ação em vários tecidos, regulando a síntese de cAMP. Contração uterina, dor, inflamação.



Prostaglandin E₁
(PGE₁)

- tromboxanos

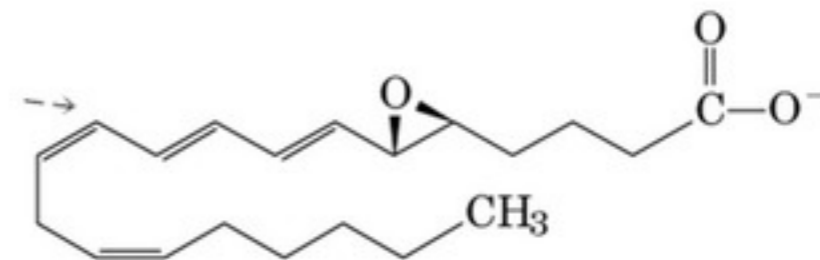
Ação na formação de coágulos, redução do fluxo sanguíneo.



Thromboxane A₂

- leucotrienos

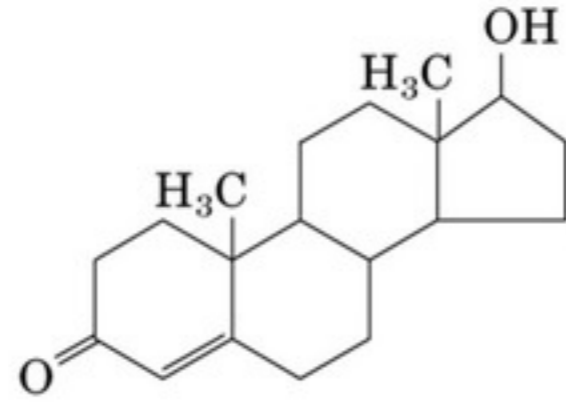
Ex. Contração de músculos relacionados a vias aéreas/pulmão



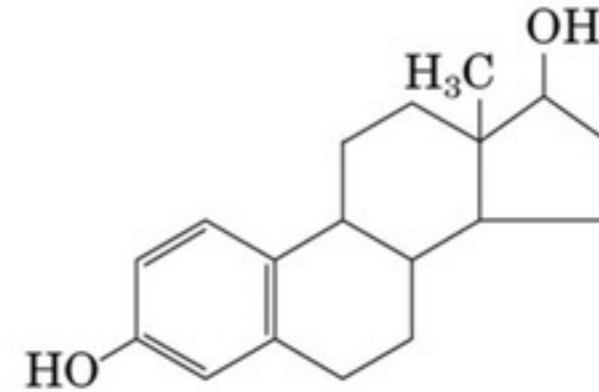
Leukotriene A₄

prednisona

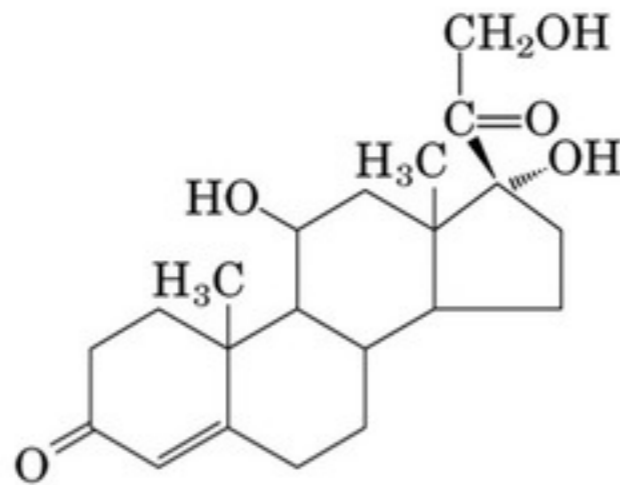
Hormônios esteróides



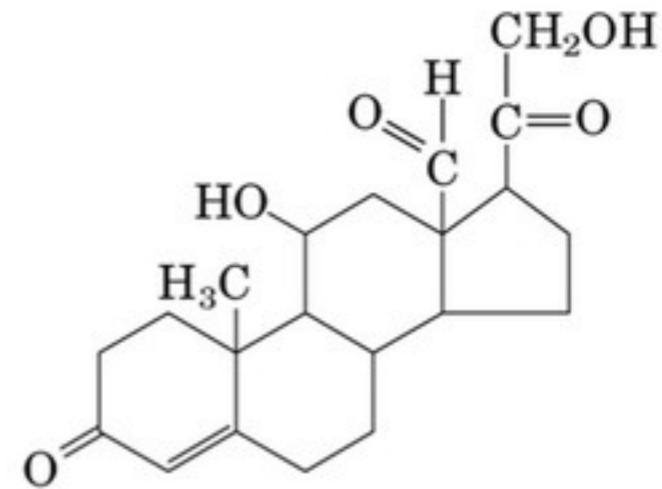
Testosterone



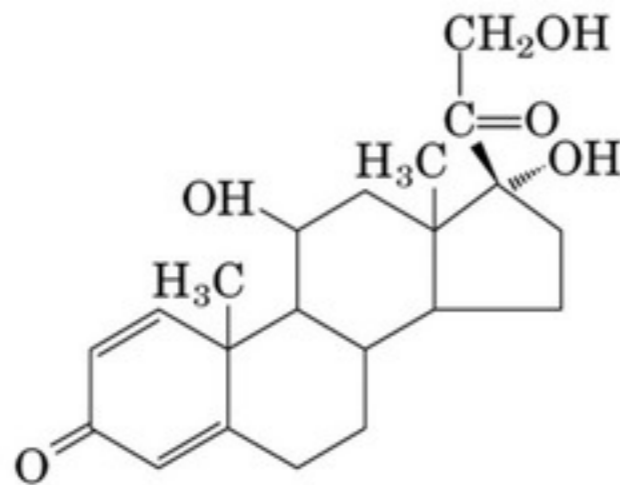
Estradiol



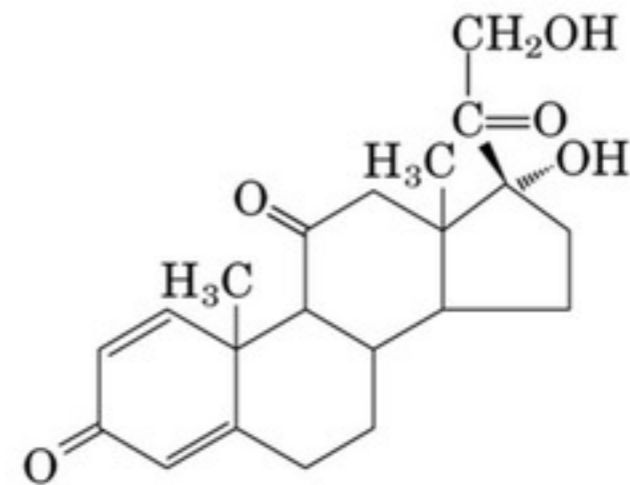
Cortisol



Aldosterone



Prednisolone



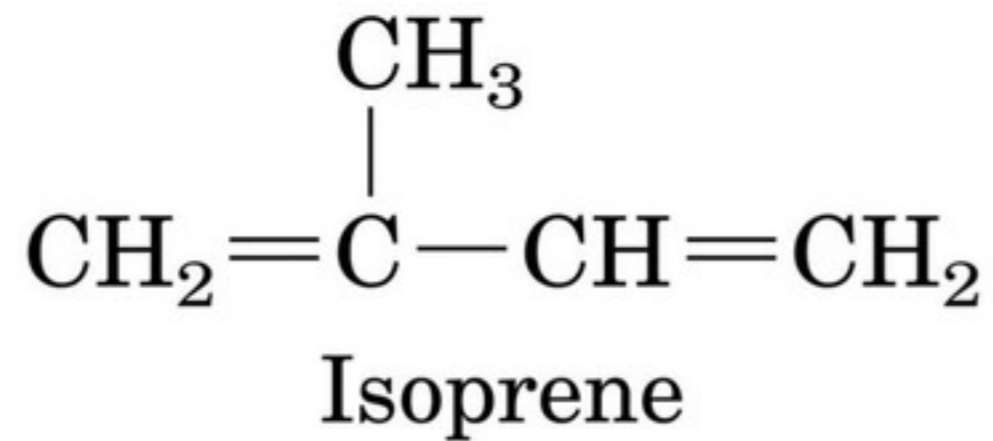
Prednisone

Vitaminas

As vitaminas foram divididas inicialmente entre solúveis e insolúveis.

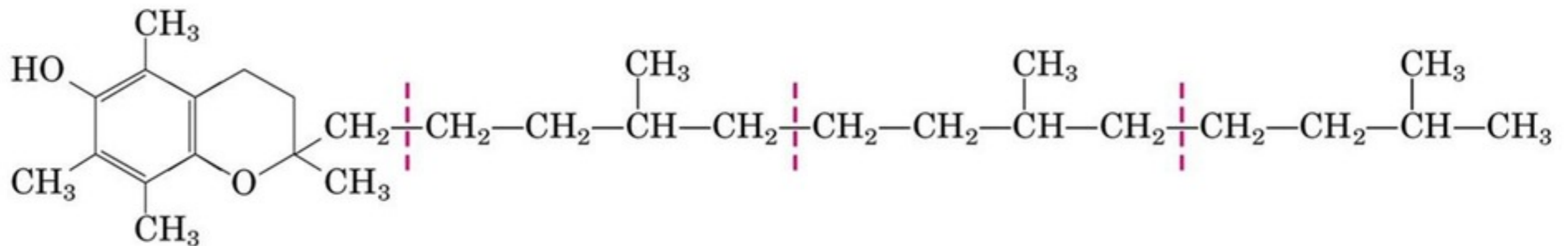
Insolúveis: A, D, E e K.

Todas são compostos isoprenóides, ou seja, derivadas do composto isopreno.



Exemplo: Vitamina E

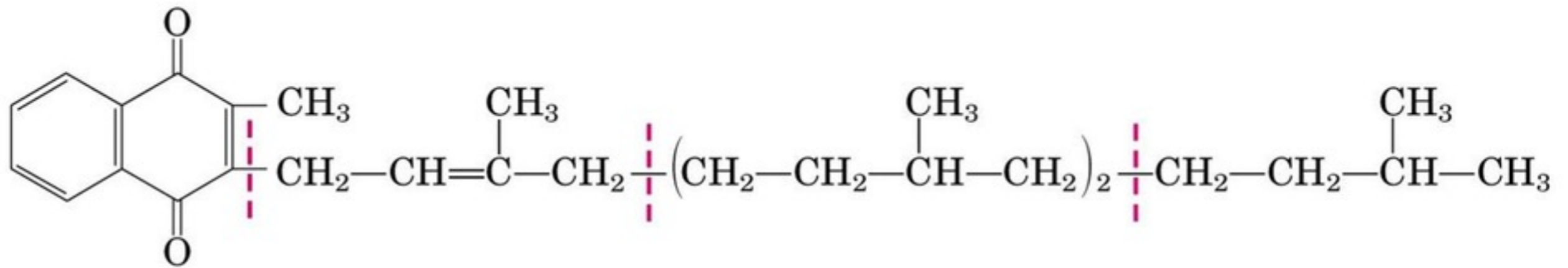
Grupo de lipídios chamados tocoferóis, possuem uma cadeia lateral isoprenóide e possuem ação antioxidante.



Hidrofóbicos: associam com a membrana celular, depósitos de lipídeos e lipoproteínas no sangue

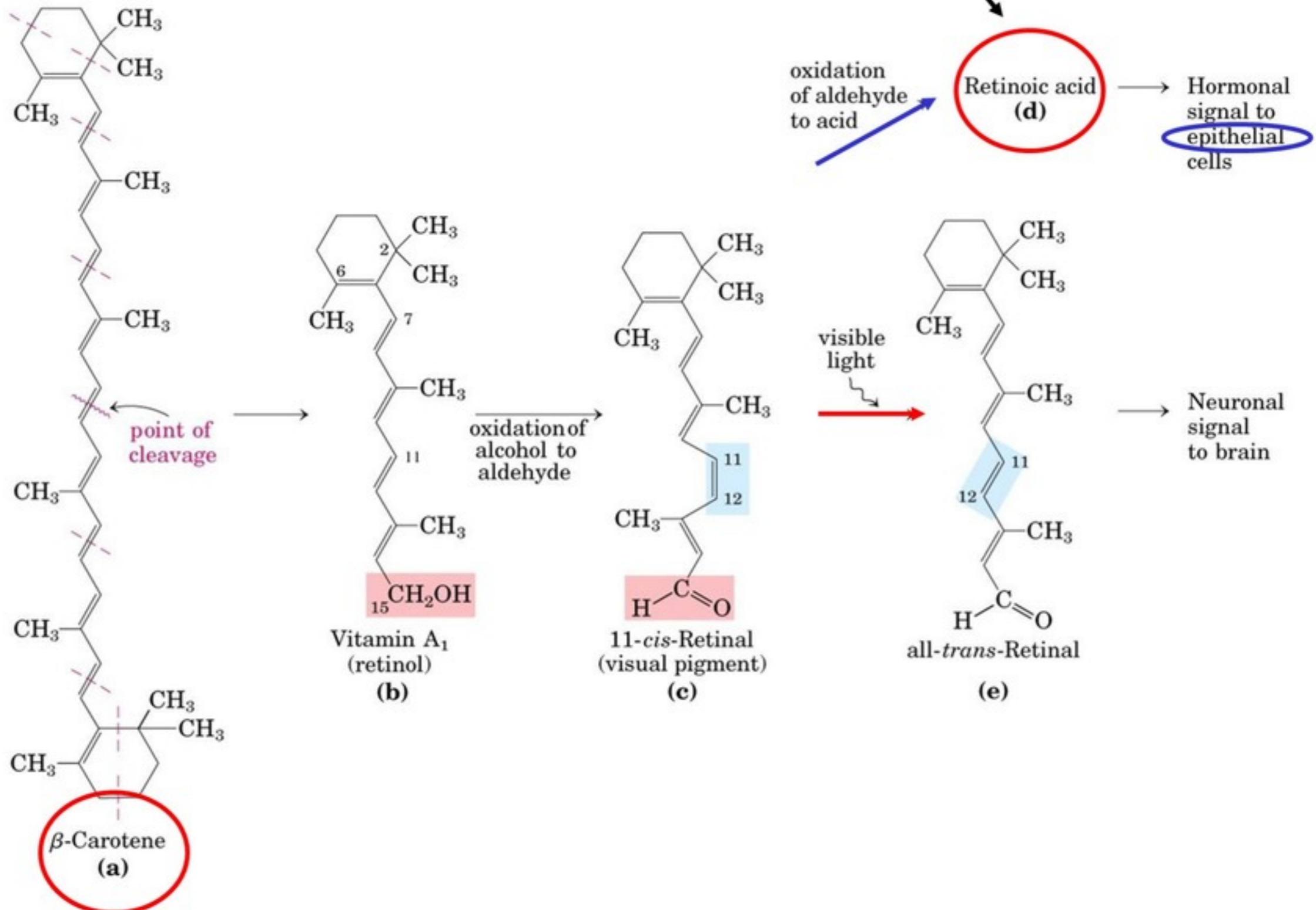
Exemplo: Vitamina K

Participa das reações de formação de pró-trombina, e logo participa do processo de coagulação.



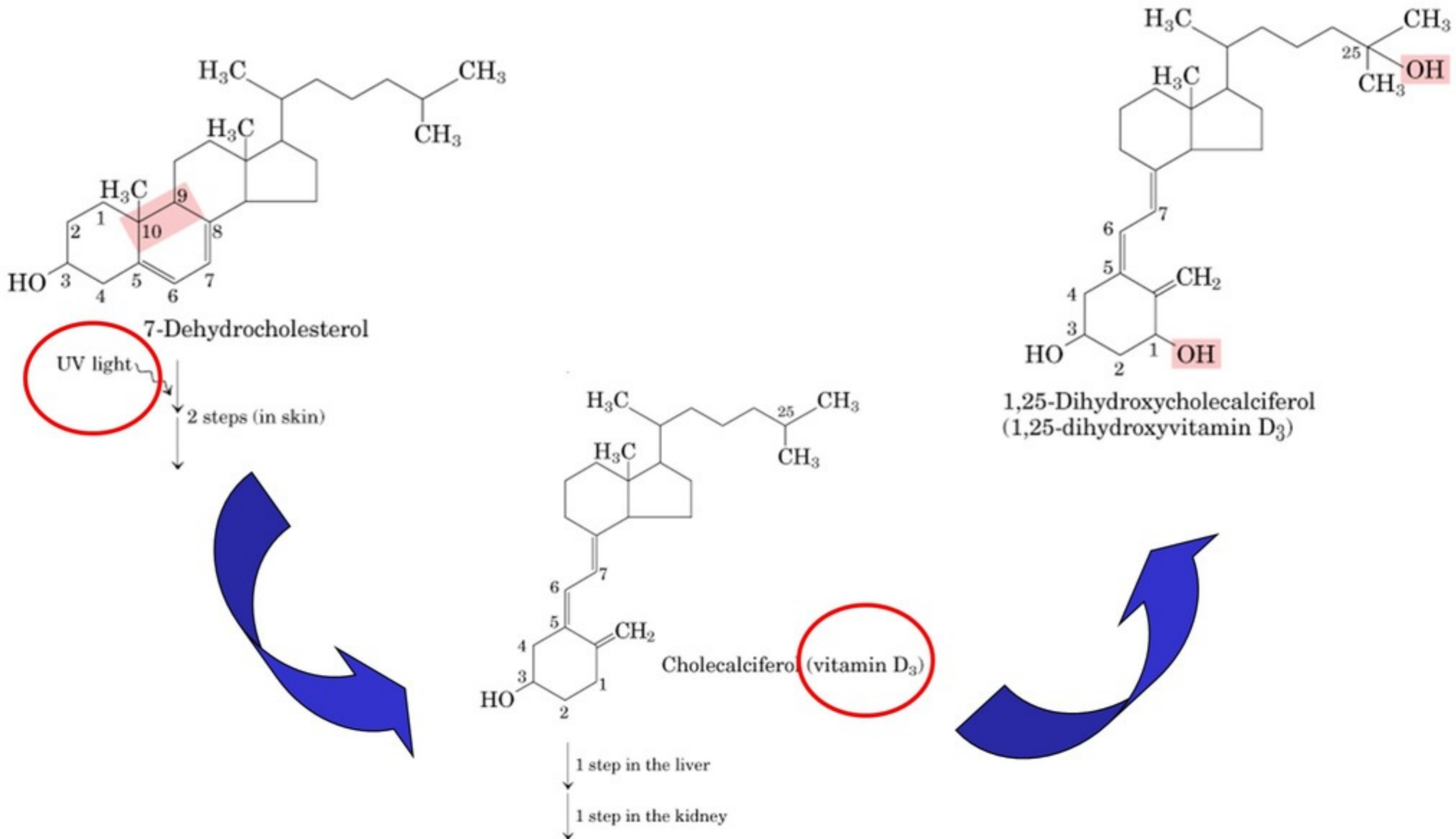
Exemplo: Vitamina A

Também chamado de retinol, possui atividade como hormônio e como o pigmento visual de vertebrados. Seu derivado, ácido retinóico, é utilizado no tratamento de pele.



Exemplo: Vitamina D₃

Também chamada de colecalciferol, ela participa da regulação de absorção de cálcio.



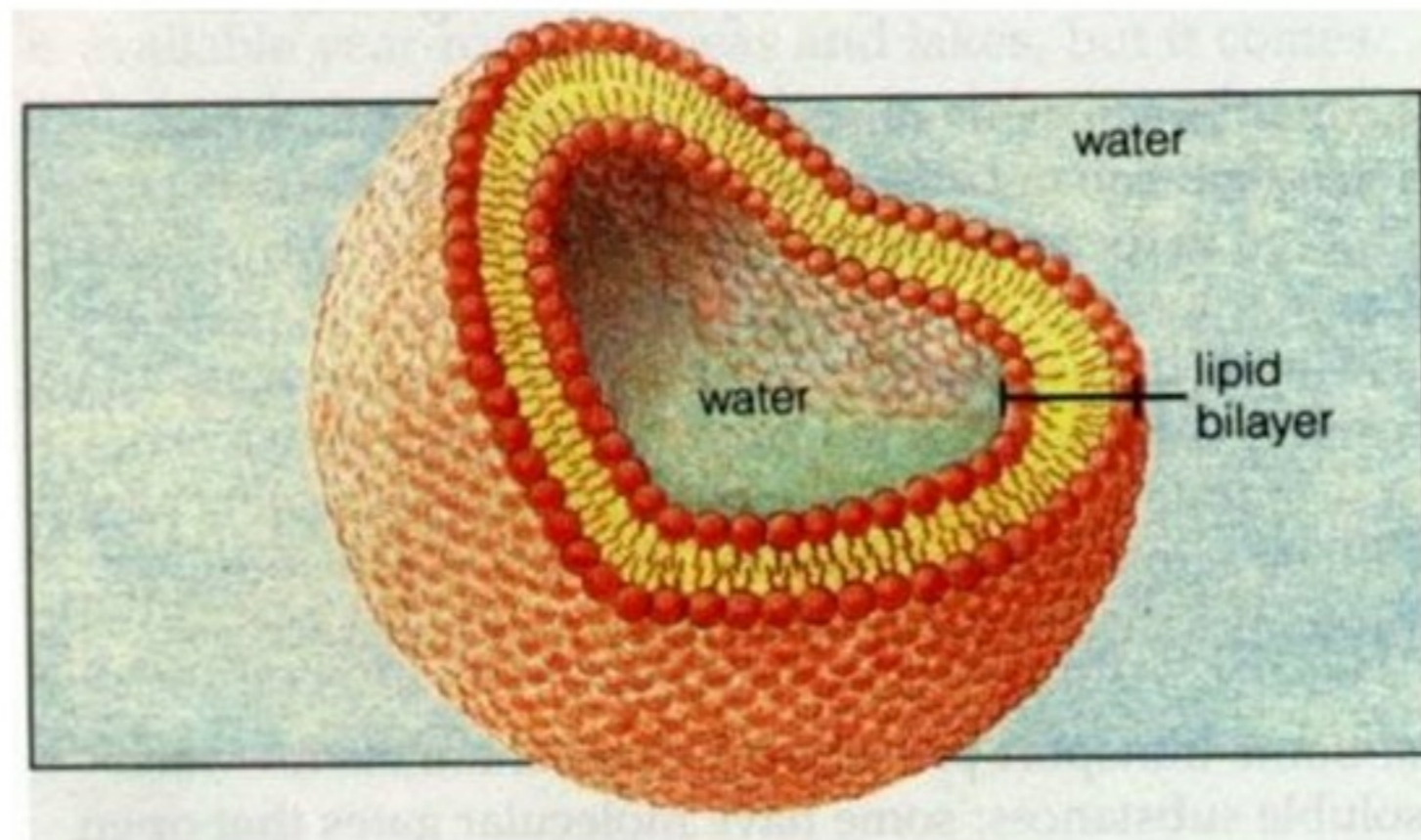
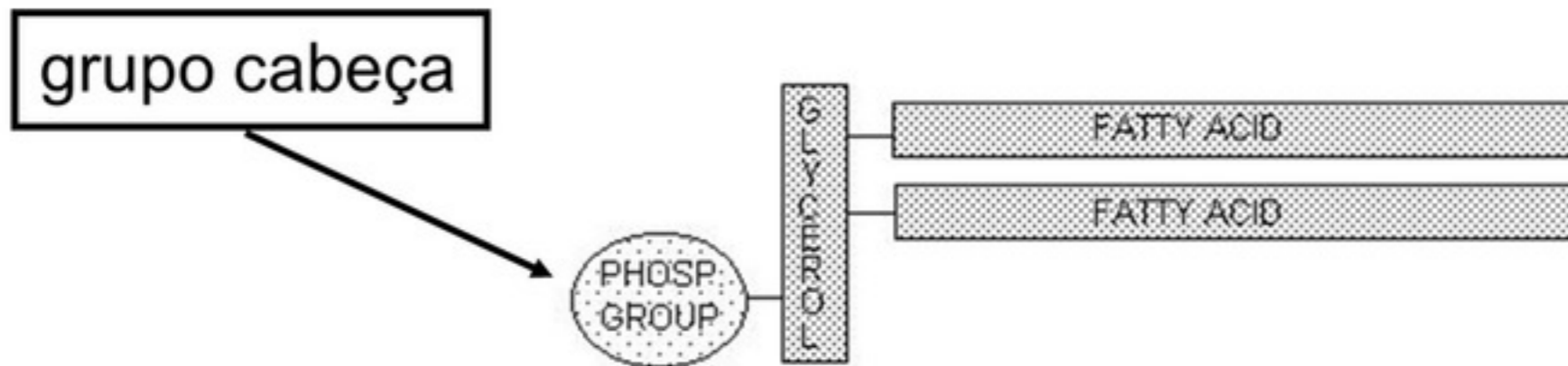


Before vitamin D treatment



After 14 months of vitamin D treatment

Organização membranar



A composição da membrana vai depender da função da célula ou da organela em questão.

