

# Bioquímica Geral

## RFM0004

# Proteínas: Função

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table 4-3

Solubilities of Some Gases in Water			
Gas	Structure*	Polarity	Solubility in water (g/L) <sup>†</sup>
Nitrogen	$\text{N}\equiv\text{N}$	Nonpolar	0.018 (40 °C)
Oxygen	$\text{O}=\text{O}$	Nonpolar	0.035 (50 °C)
Carbon dioxide	$\begin{array}{c} \delta^- \quad \delta^- \\ \longleftarrow \quad \longrightarrow \\ \text{O}=\text{C}=\text{O} \end{array}$	Nonpolar	0.97 (45 °C)
Ammonia	$\begin{array}{c} \text{H} \quad \text{H} \quad \text{H} \\ \diagdown \quad   \quad / \\ \text{N} \\ \downarrow \delta^- \end{array}$	Polar	900 (10 °C)
Hydrogen sulfide	$\begin{array}{c} \text{H} \quad \text{H} \\ \diagdown \quad / \\ \text{S} \\ \downarrow \delta^- \end{array}$	Polar	1,860 (40 °C)

\*The arrows represent electric dipoles; there is a partial negative charge ( $\delta^-$ ) at the head of the arrow, a partial positive charge ( $\delta^+$ ; not shown here) at the tail.

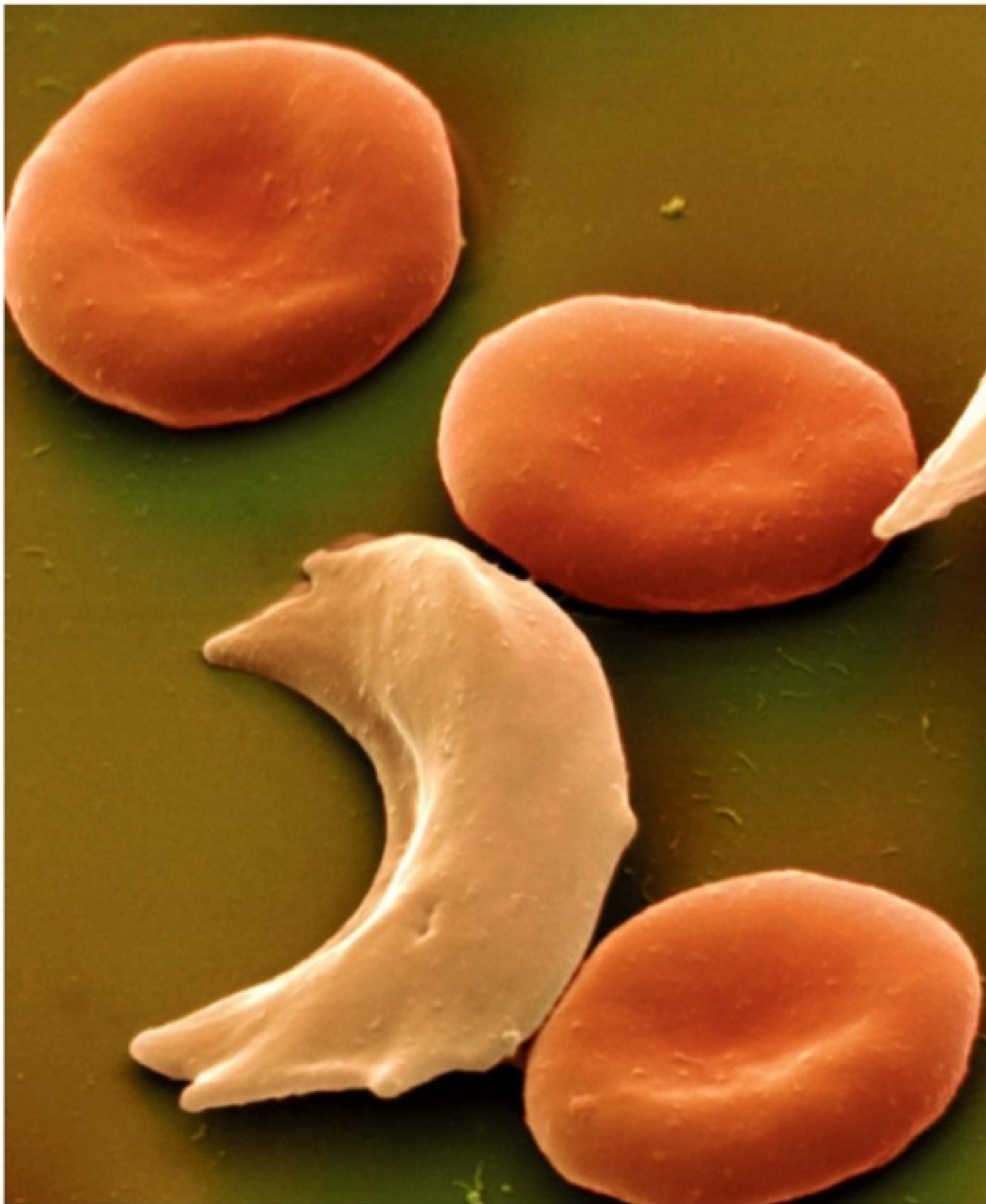
<sup>†</sup>Note that polar molecules dissolve far better even at low temperatures than do nonpolar molecules at relatively high temperatures.



*“Structure and function are really information processing being implemented by the physical and chemical properties made available by biological molecules, cells, networks of cells and so forth (...).”*

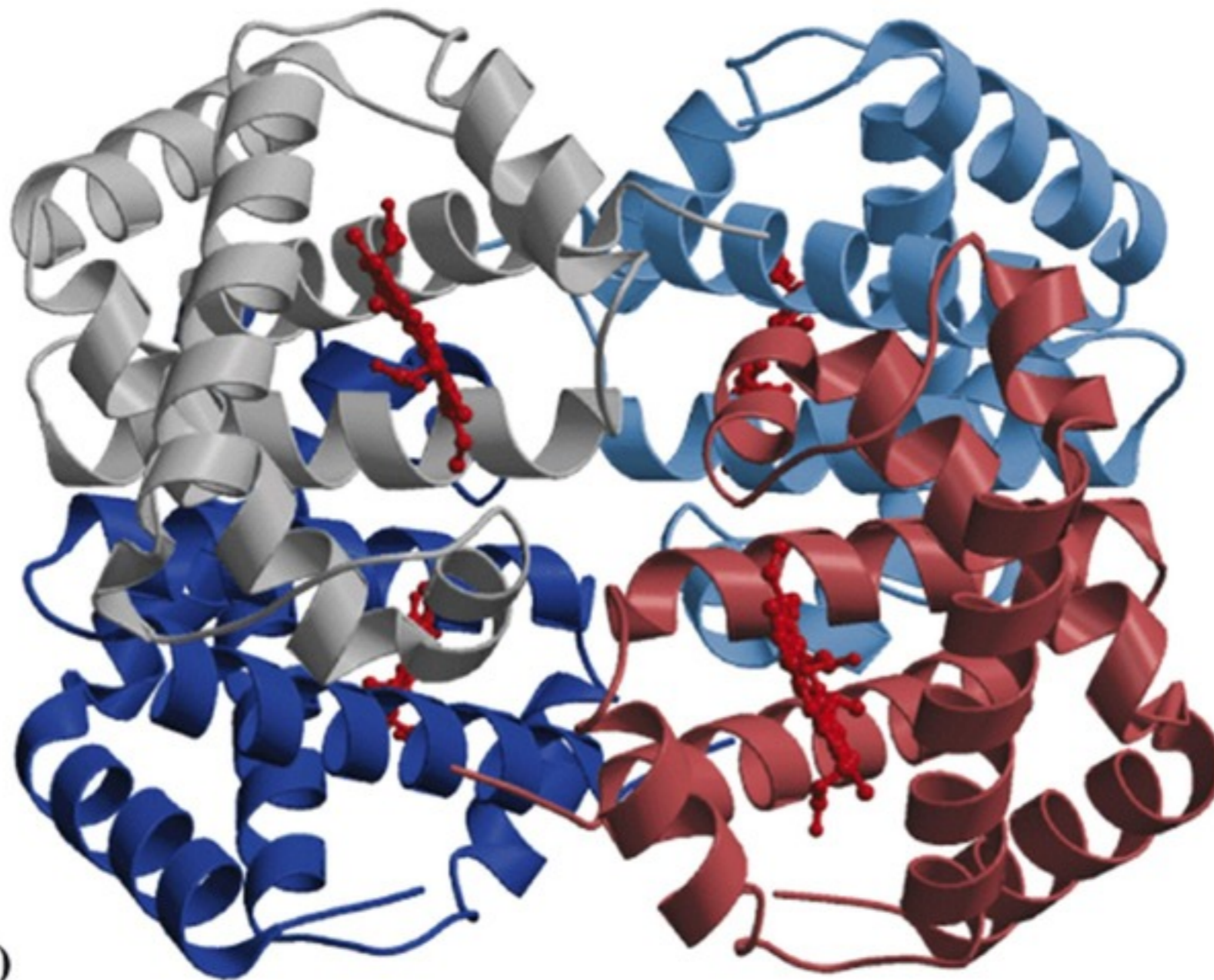
Read Montague –  
“Your brain is (almost) perfect – how we make decisions”

**Chapter 7 Opener part 1**  
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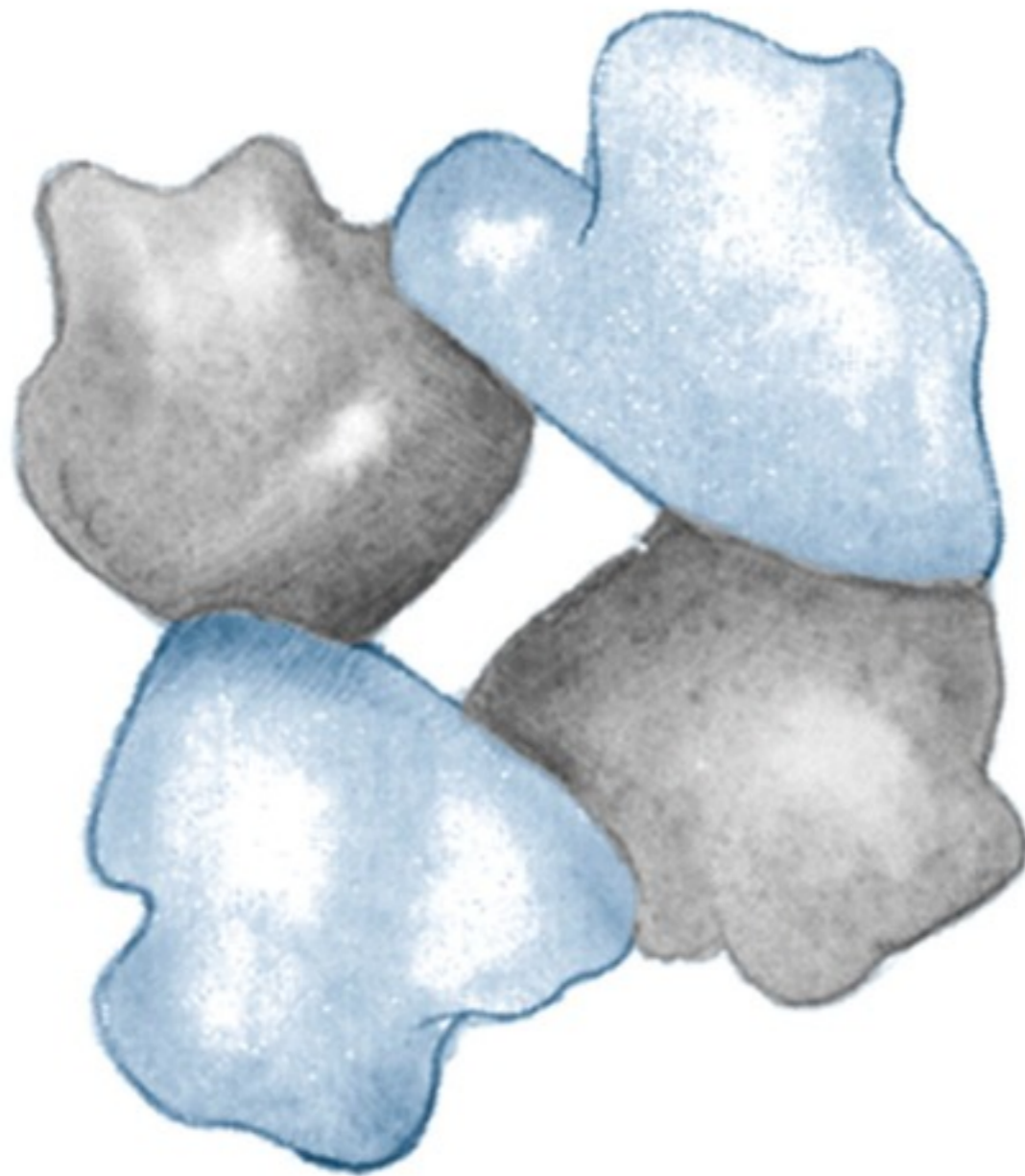
**Figure 7-23**  
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# Hemoglobina

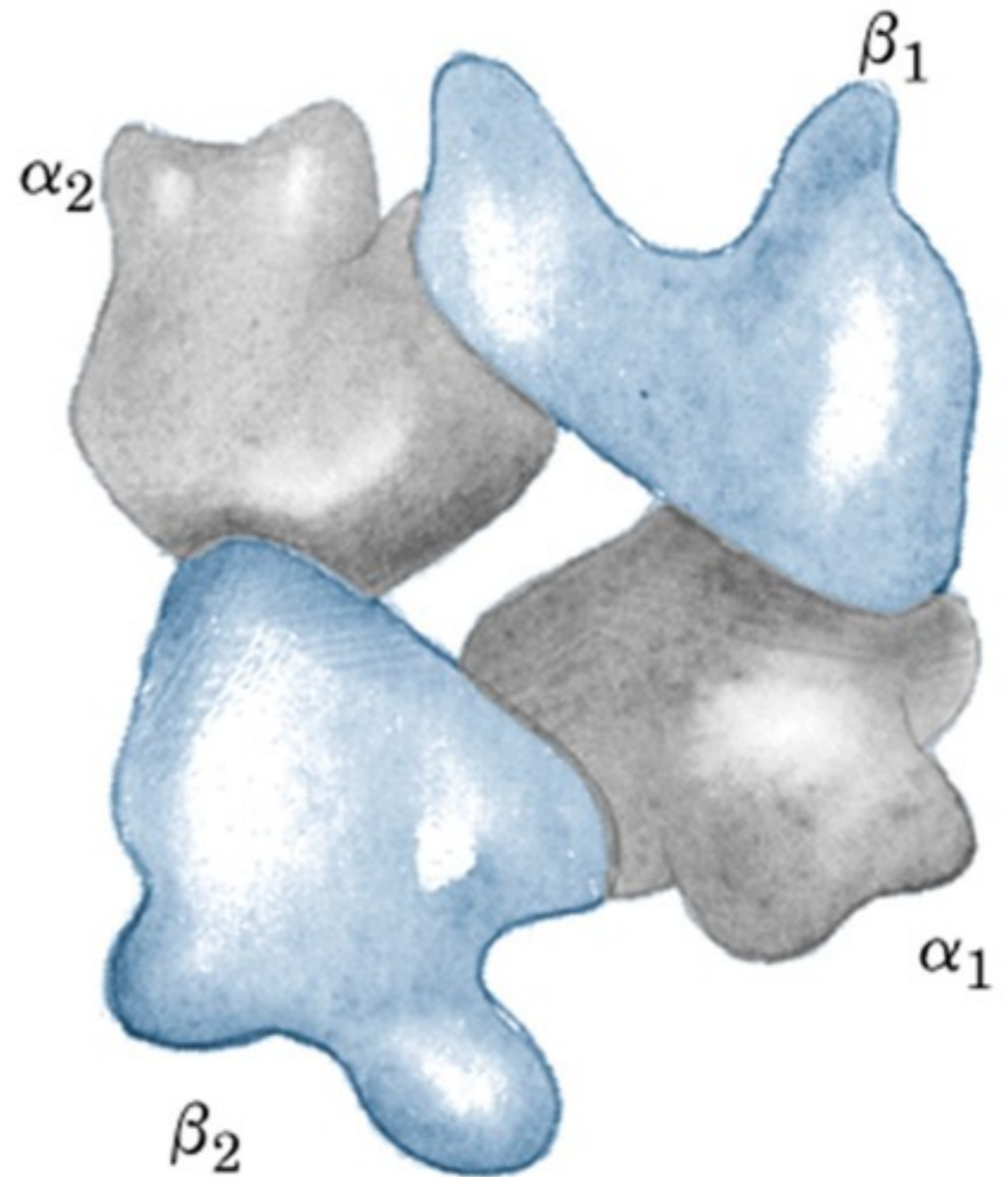


(a)

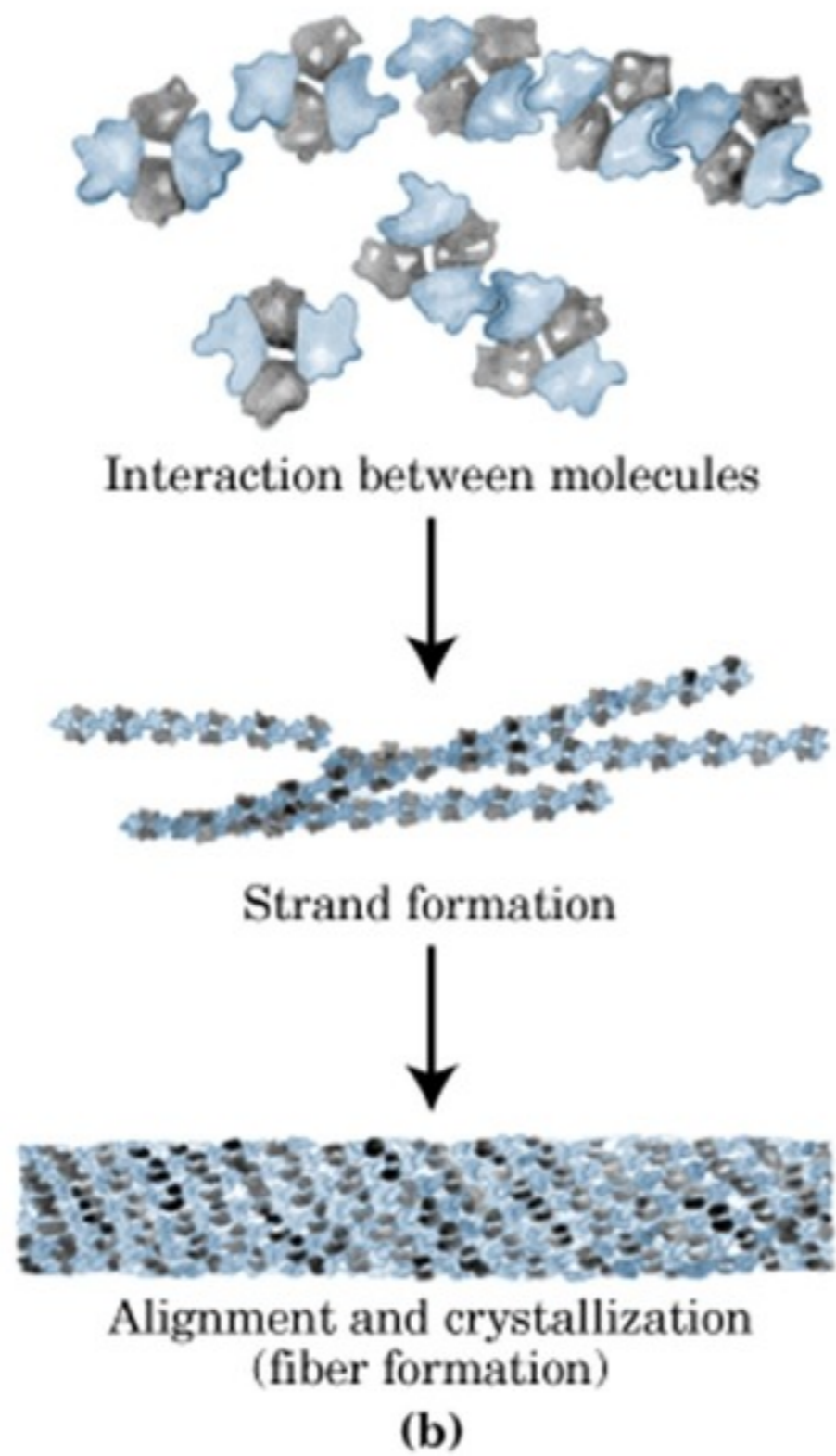
Hemoglobin A



Hemoglobin S



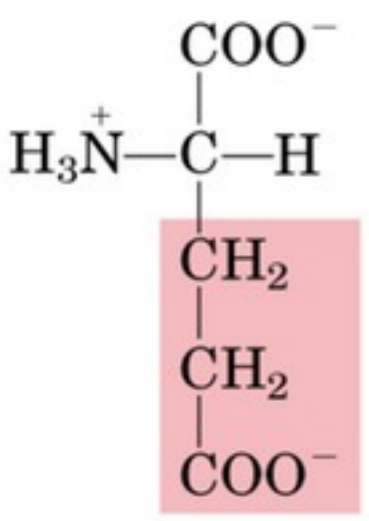
(a)



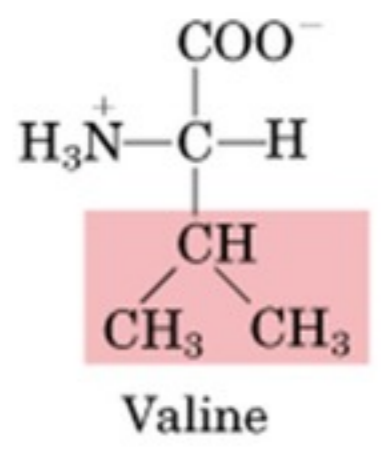


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Glutamate



Valine

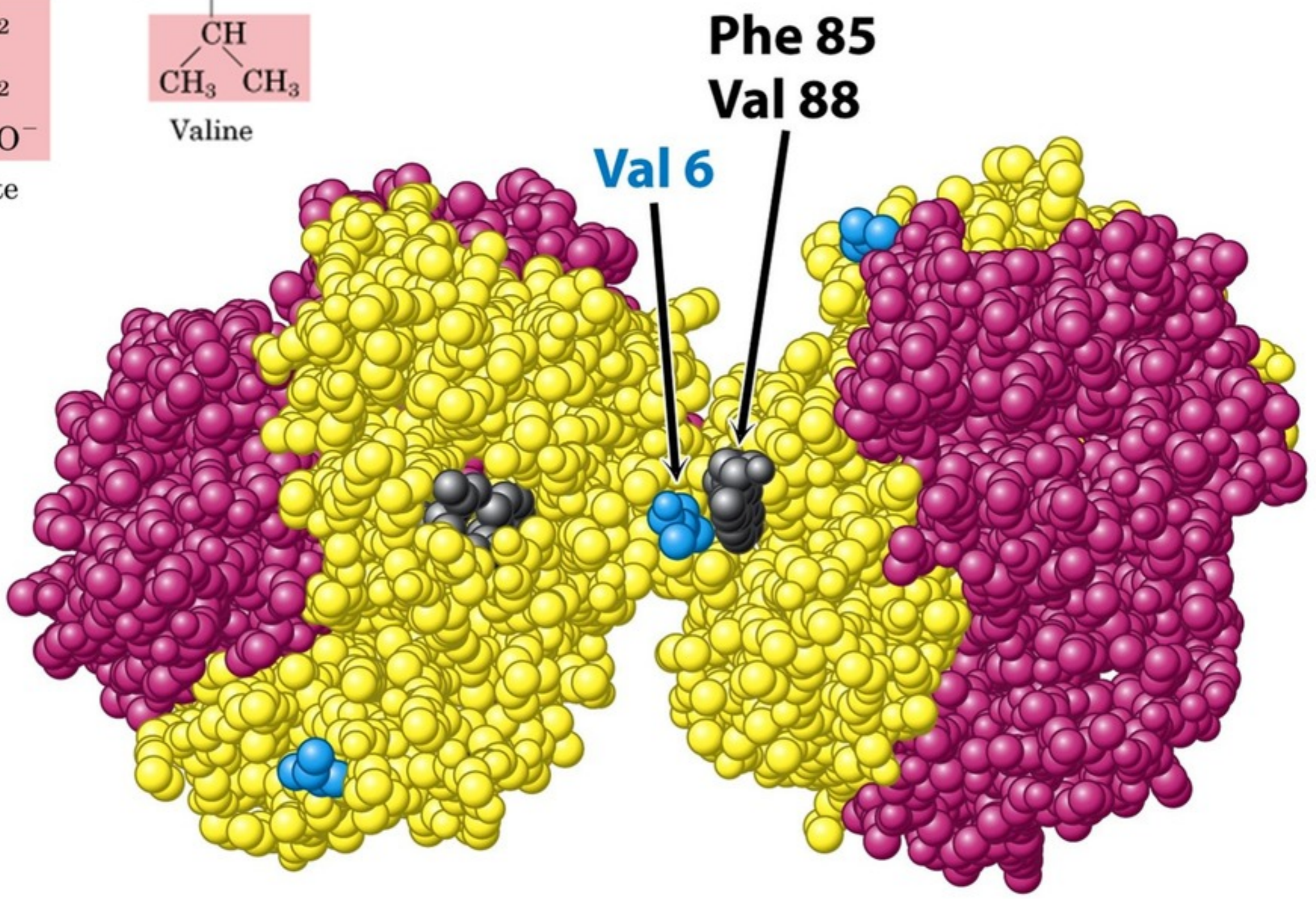


Figure 7-25  
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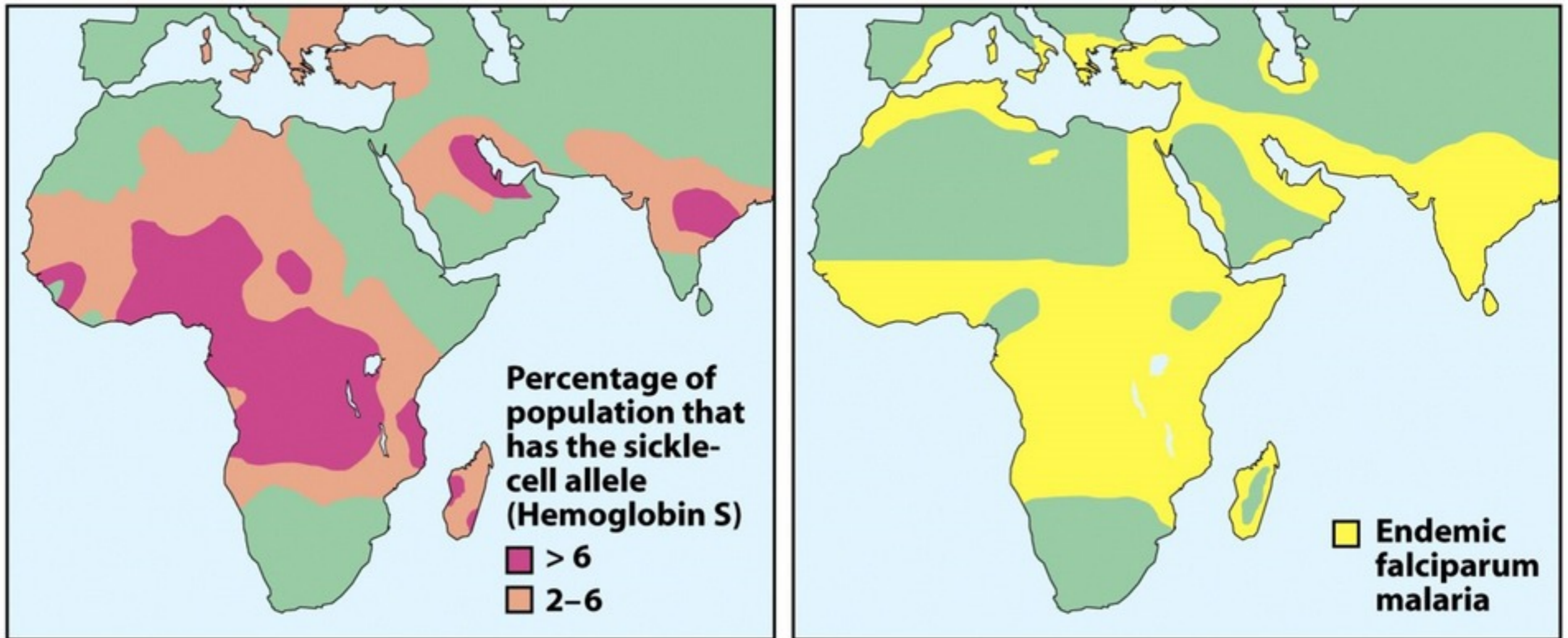
Hb-S: substituição de glutamato por valina na posição 6 da cadeia β



Como a substituição da glutamina pela valina está relacionada com os efeitos fisiológicos observados na anemia falciforme?

**Collaborate!**

**Hemoglobina S**



**Figure 7-26**  
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## ESTRUTURA e FUNÇÃO

Ligação reversível de outra molécula = LIGANTE

Sítio de ligação

Especificidade

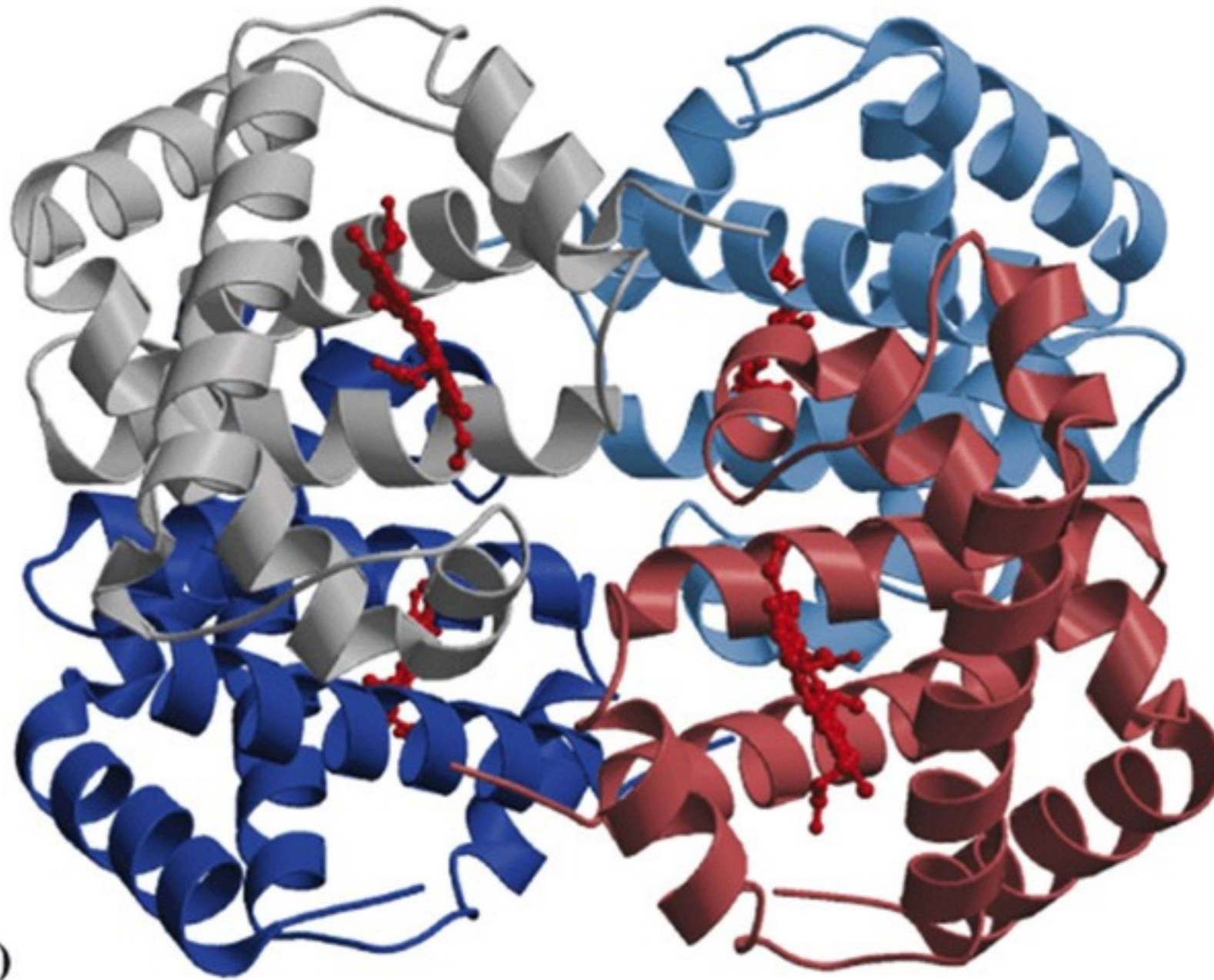
Conformação

Flexibilidade

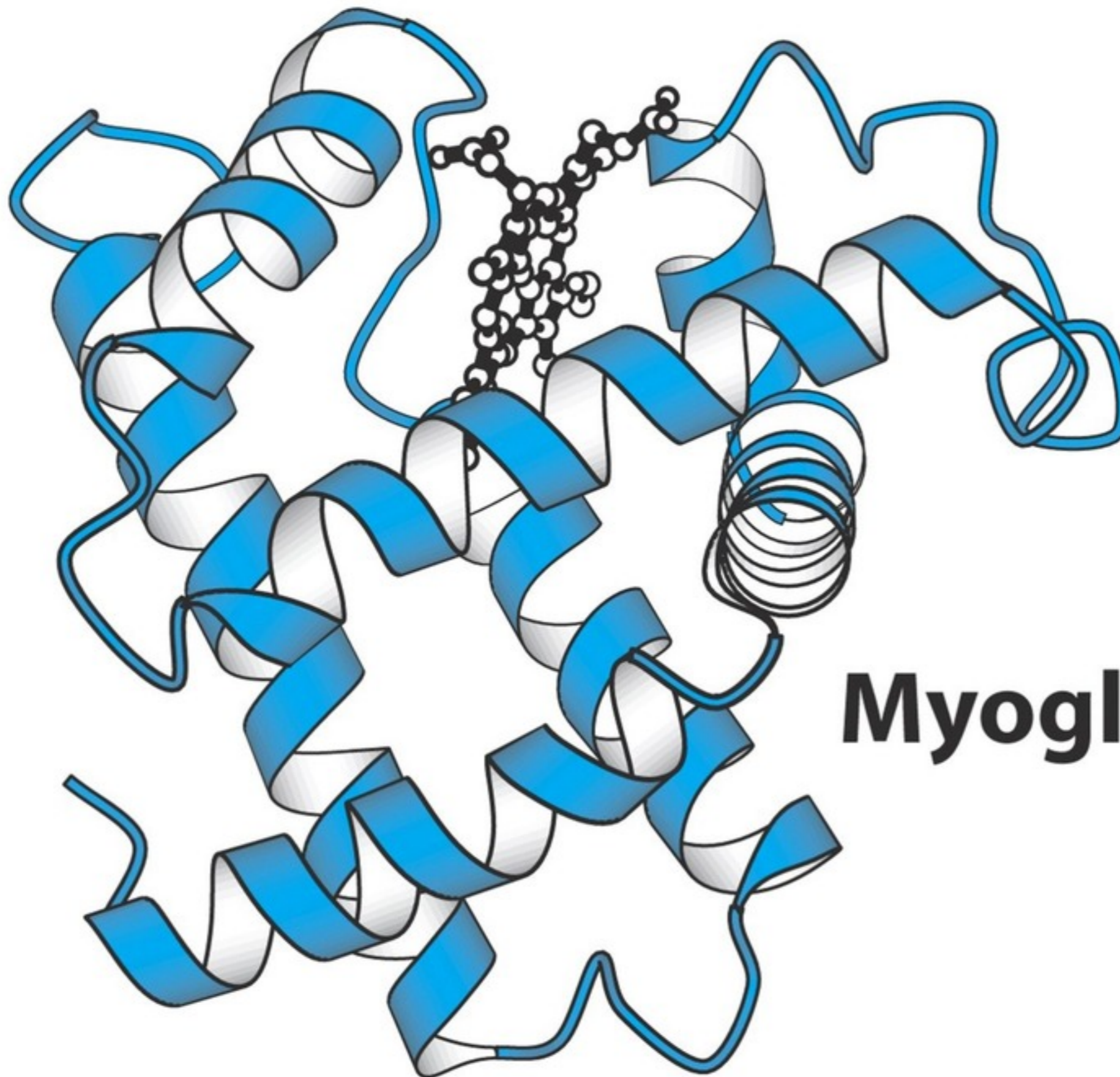
Alterações estruturais

Regulação da interação entre ligantes e proteínas

# Hemoglobina



(a)



# Myoglobin

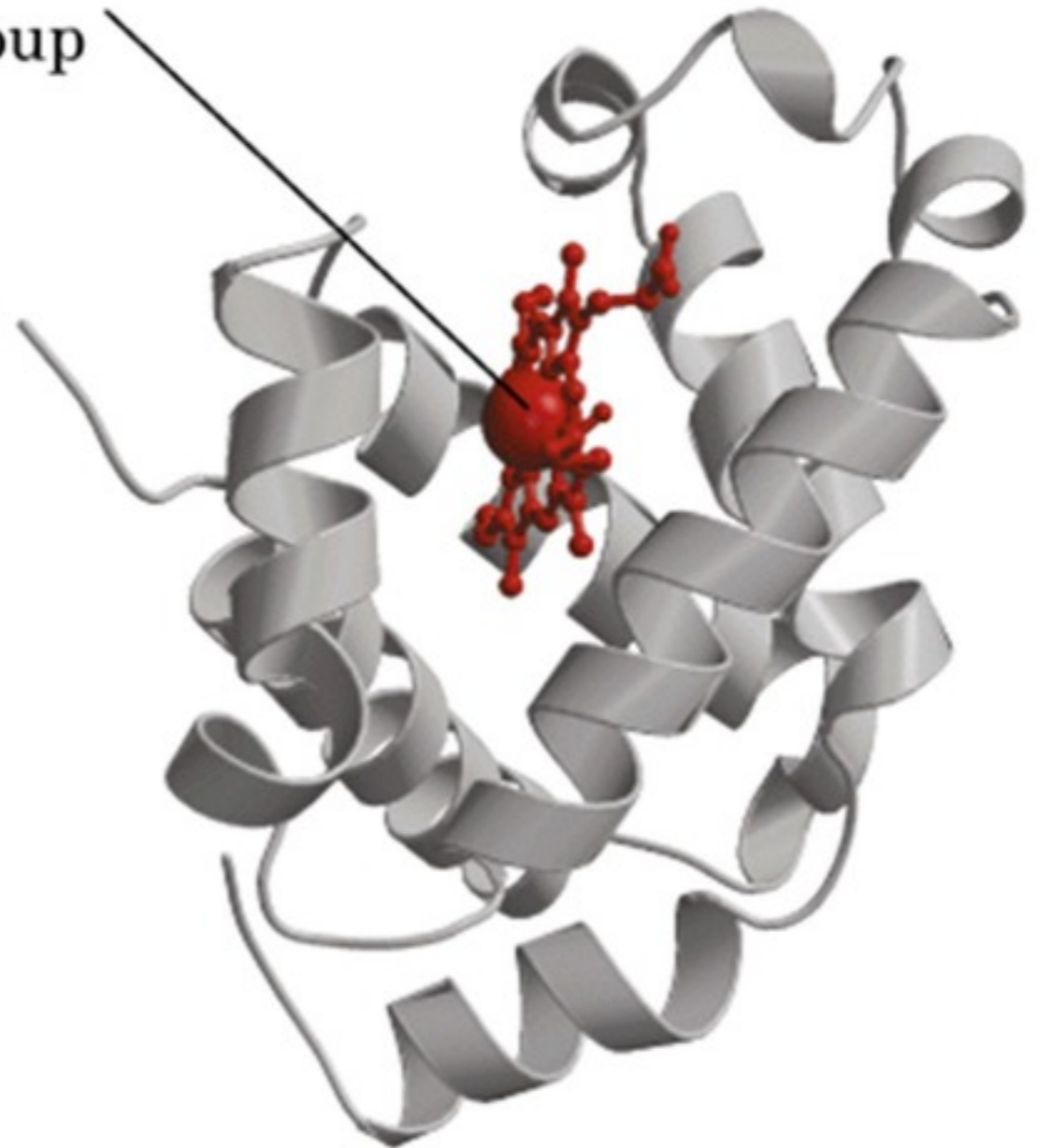
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Grupo prostético

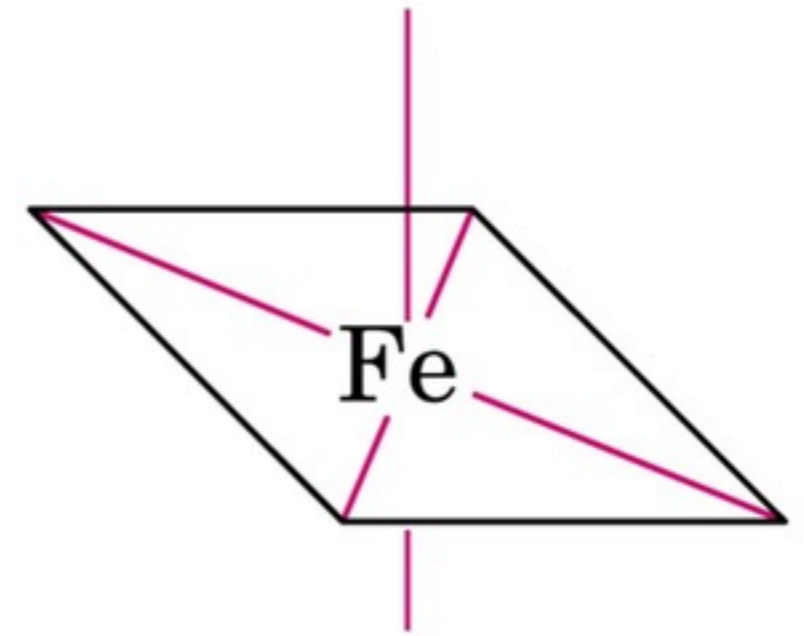
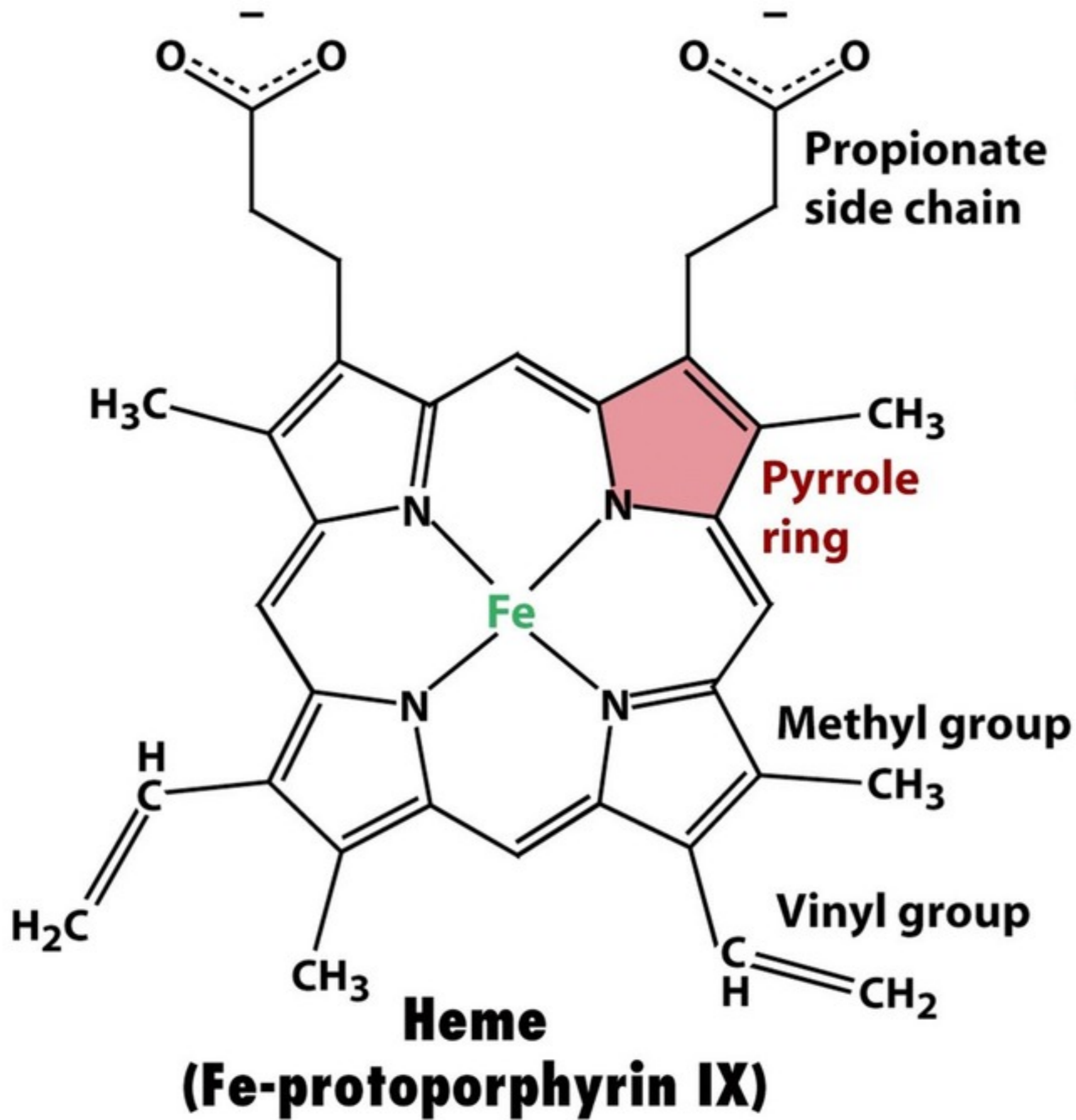
Heme  
group



Myoglobin



$\beta$  subunit of  
hemoglobin



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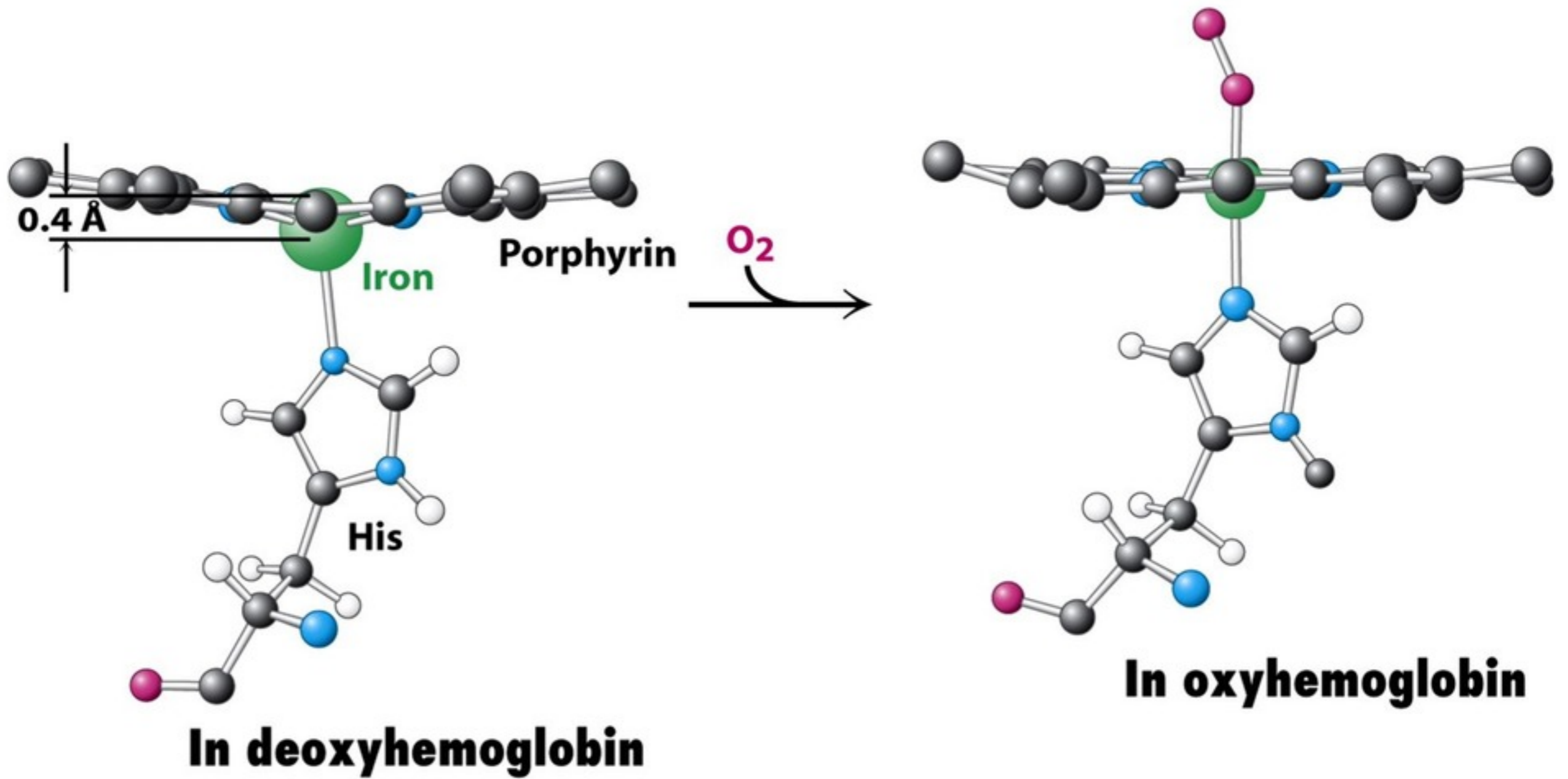


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# Poll

A mioglobina se liga a X oxigênios e a hemoglobina a Y. X =? e Y=?

1 e 2

2 e 4

1 e 1

1 e 4

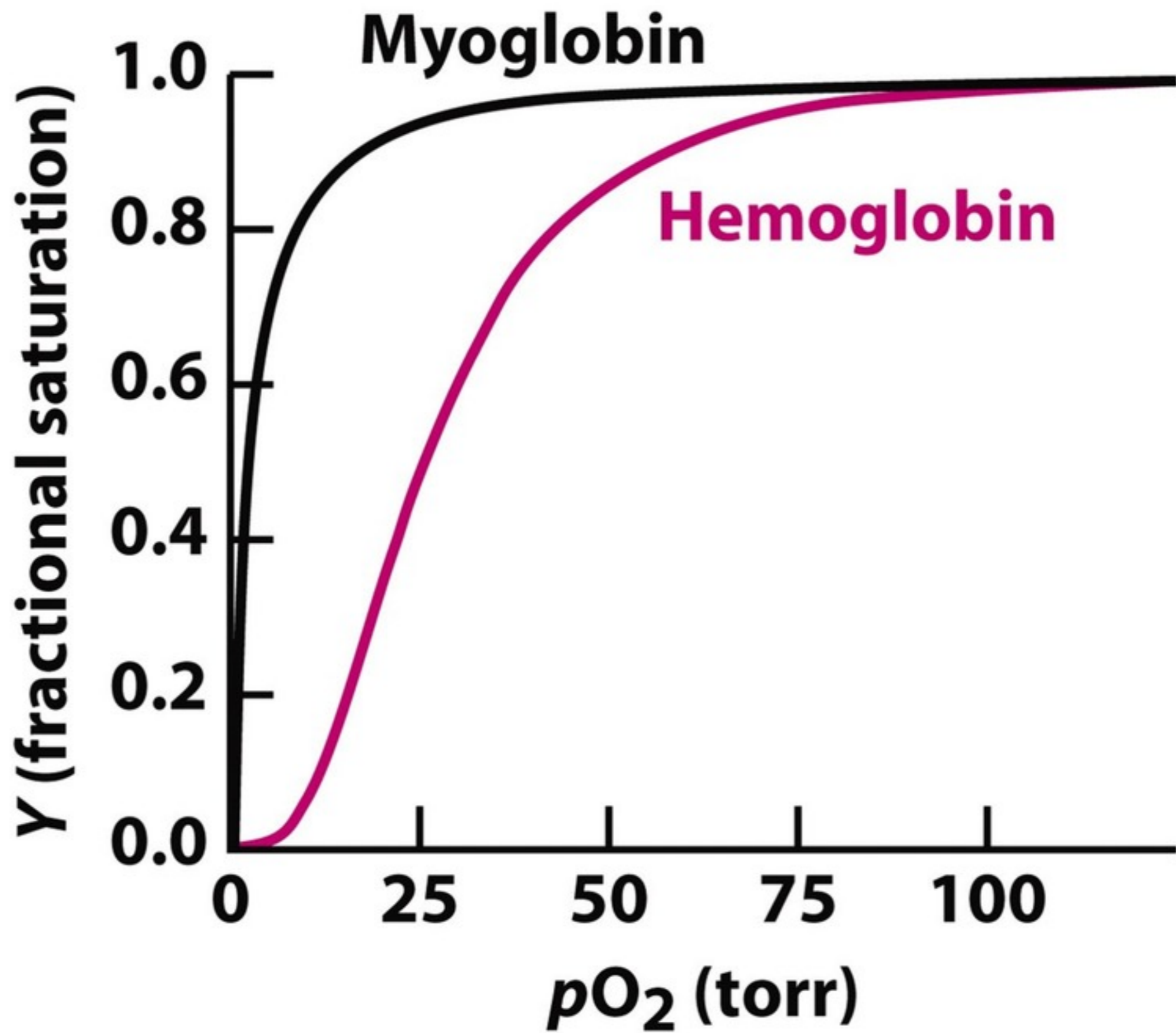
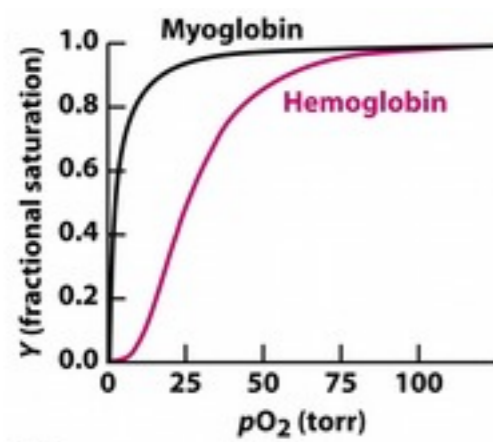


Figure 7-7  
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# Poll



Qual das proteínas tem maior afinidade pelo oxigênio?

- Mioglobina
- Hemoglobina

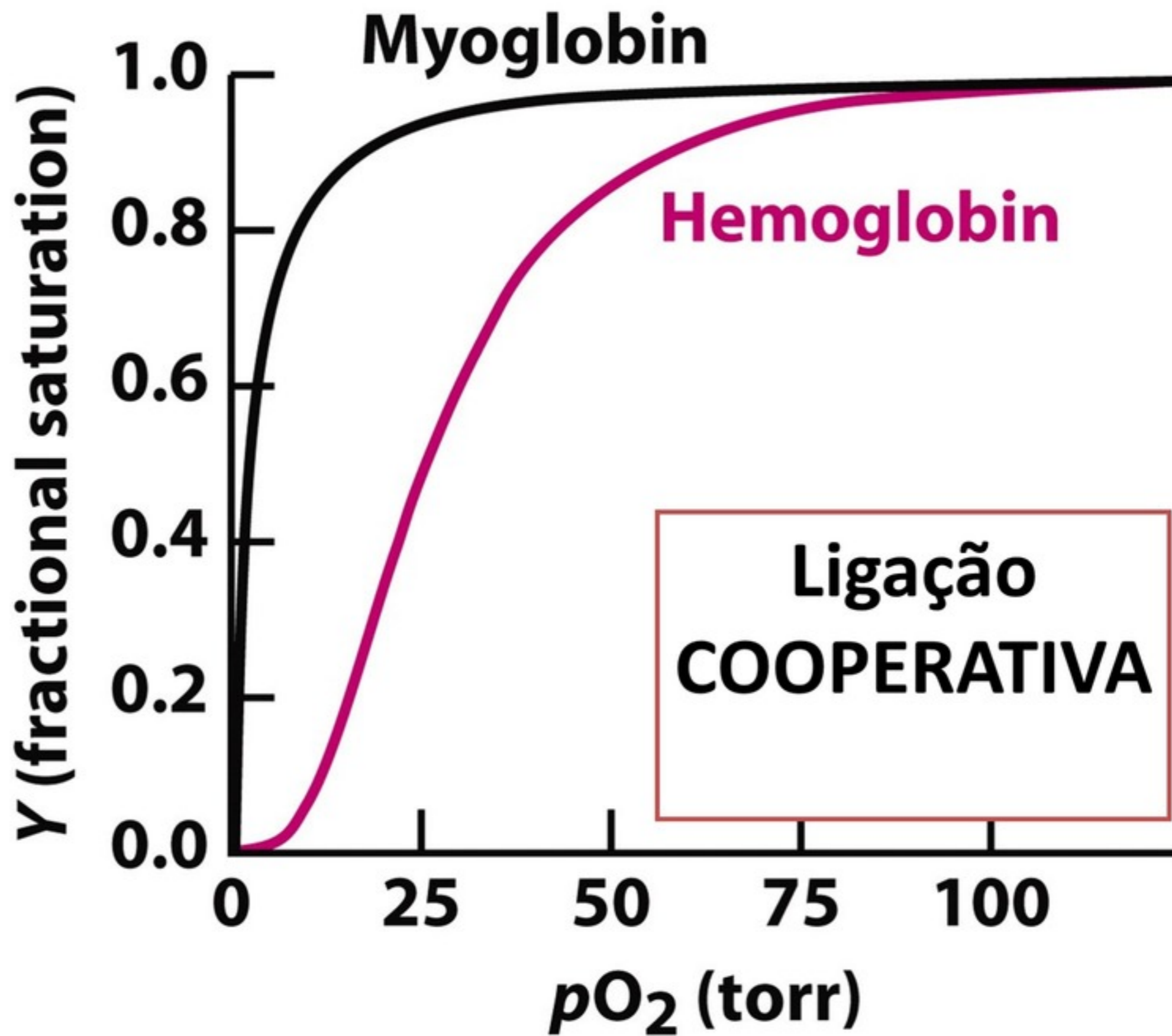
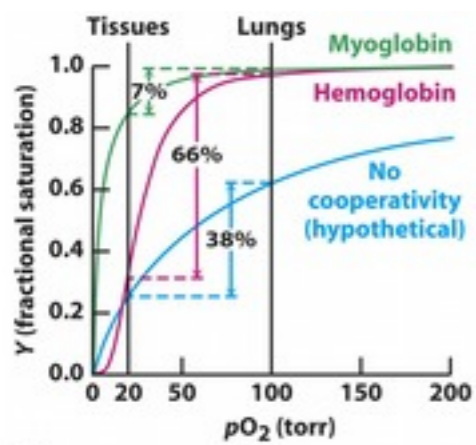


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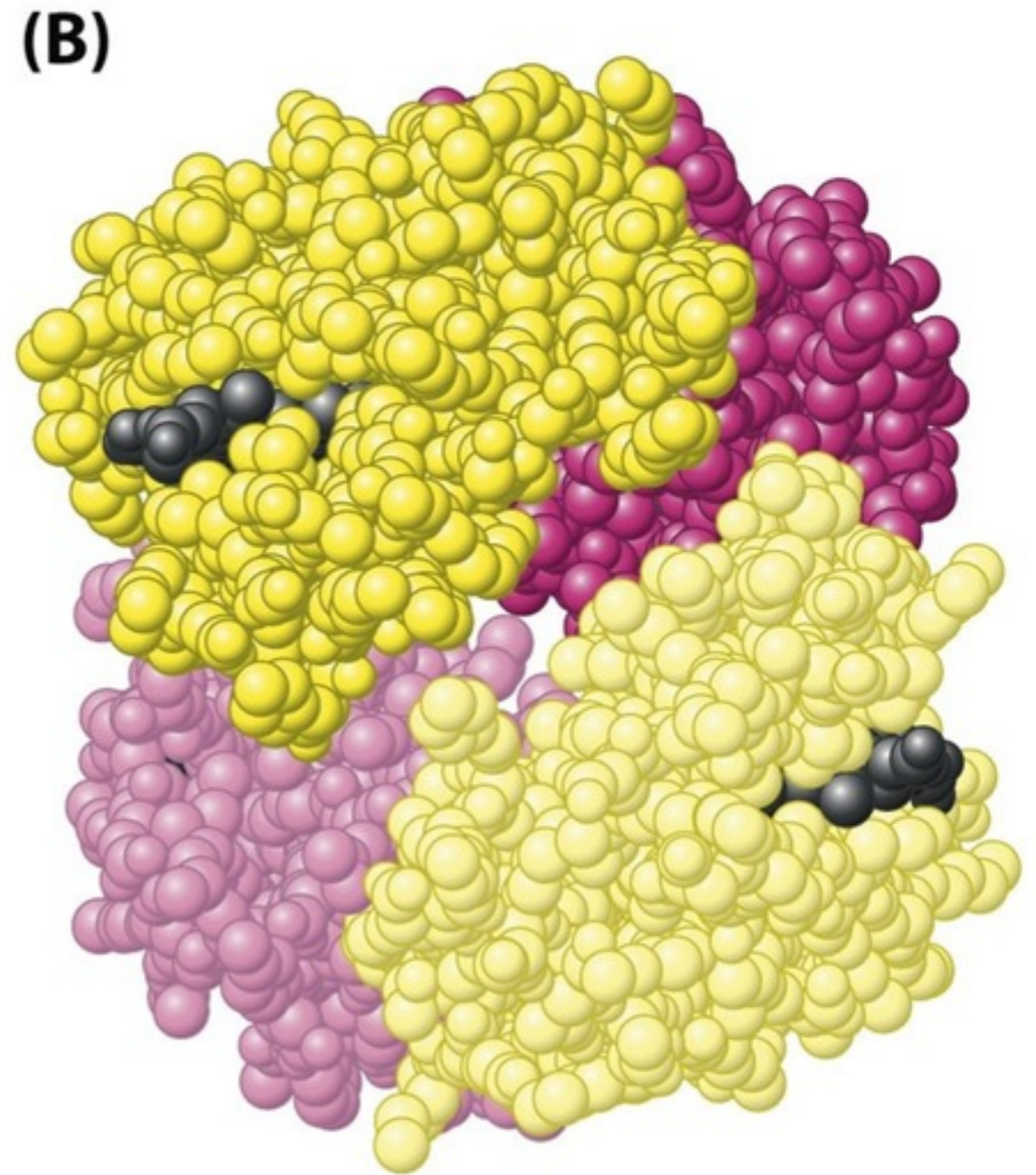
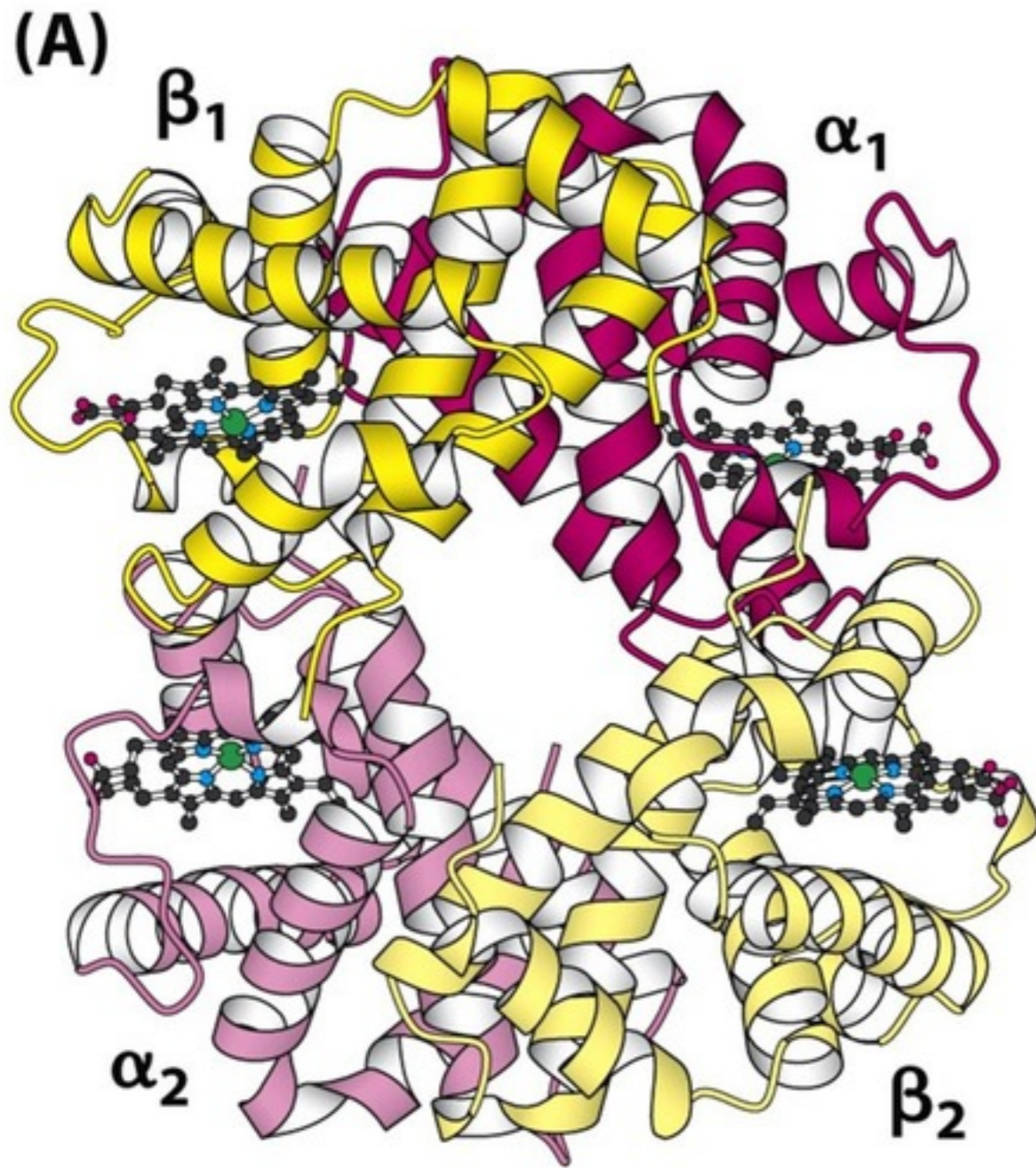
# Poll





Qual das proteínas é um transportador mais eficiente de oxigênio?

- mioglobina
- hemoglobina
- hemoglobina sem cooperatividade



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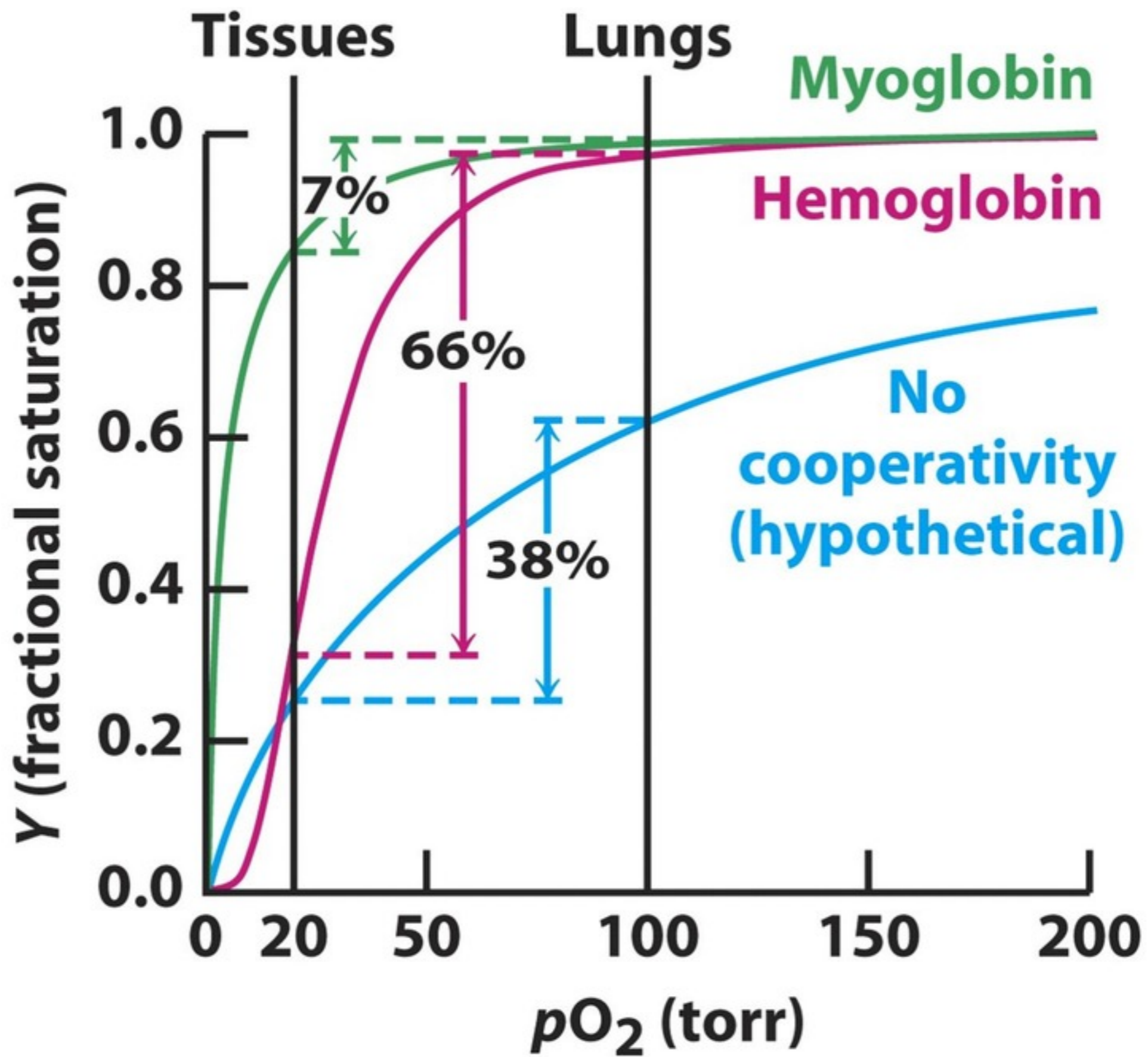
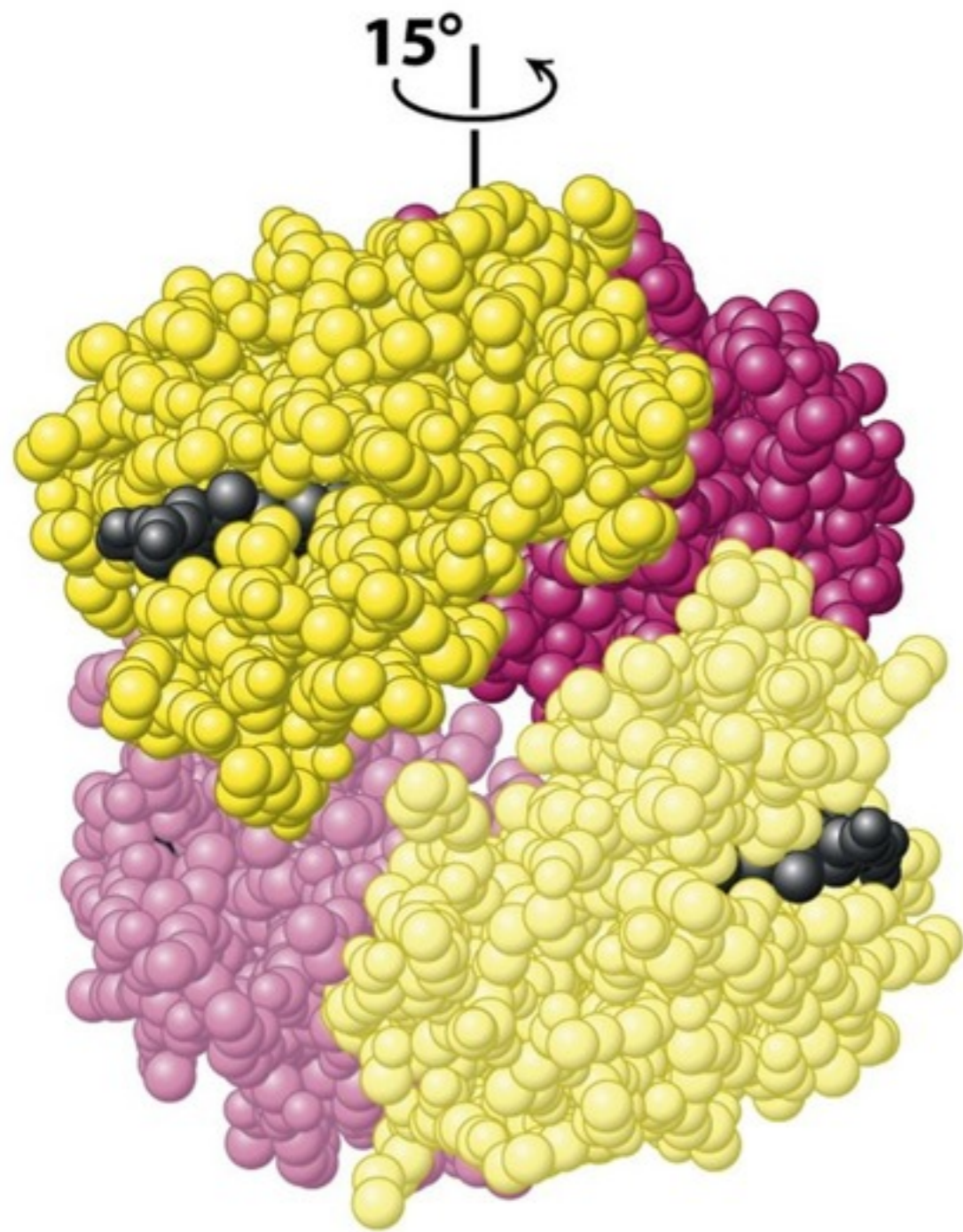
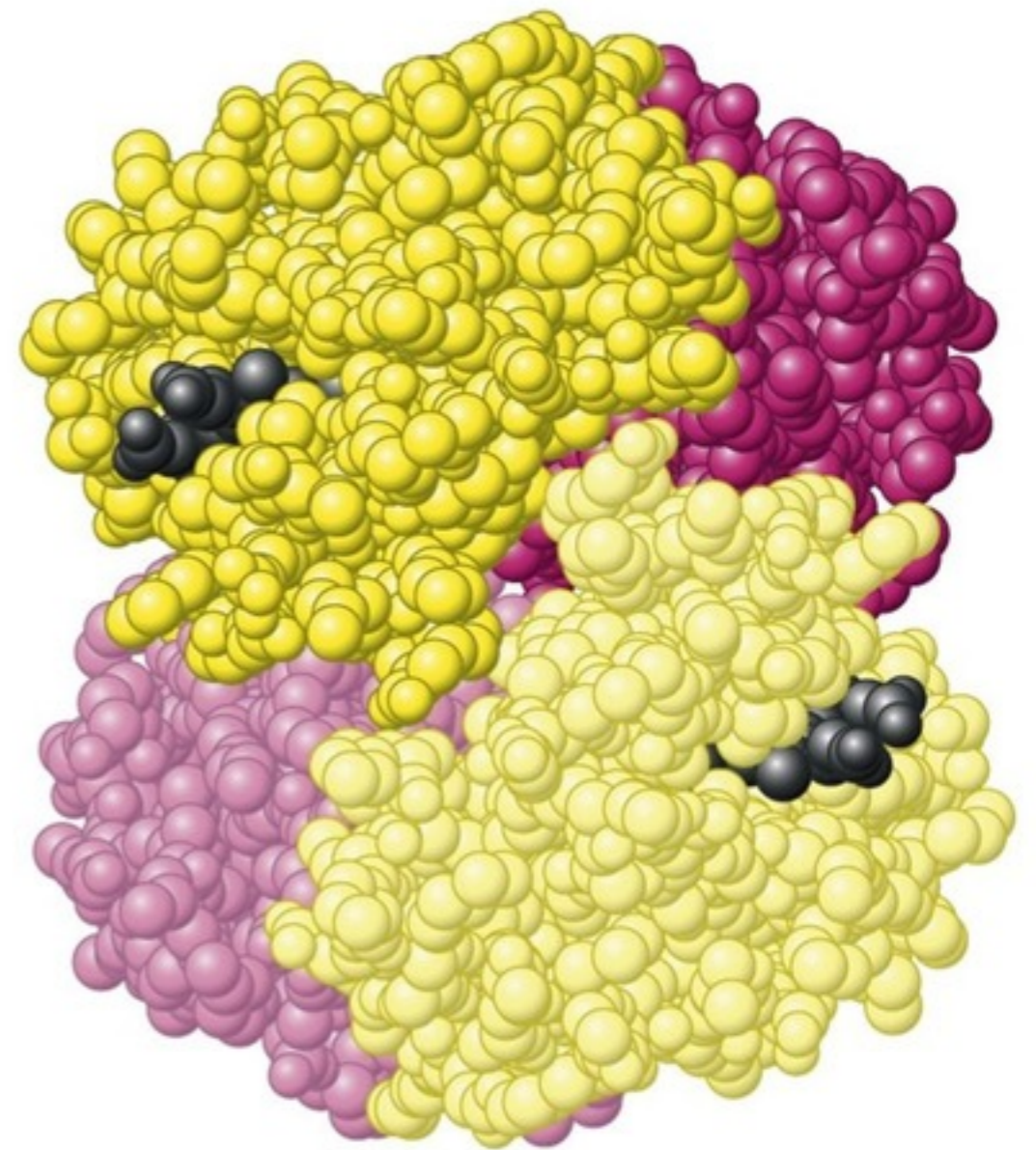


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**Deoxyhemoglobin**



**Oxyhemoglobin**

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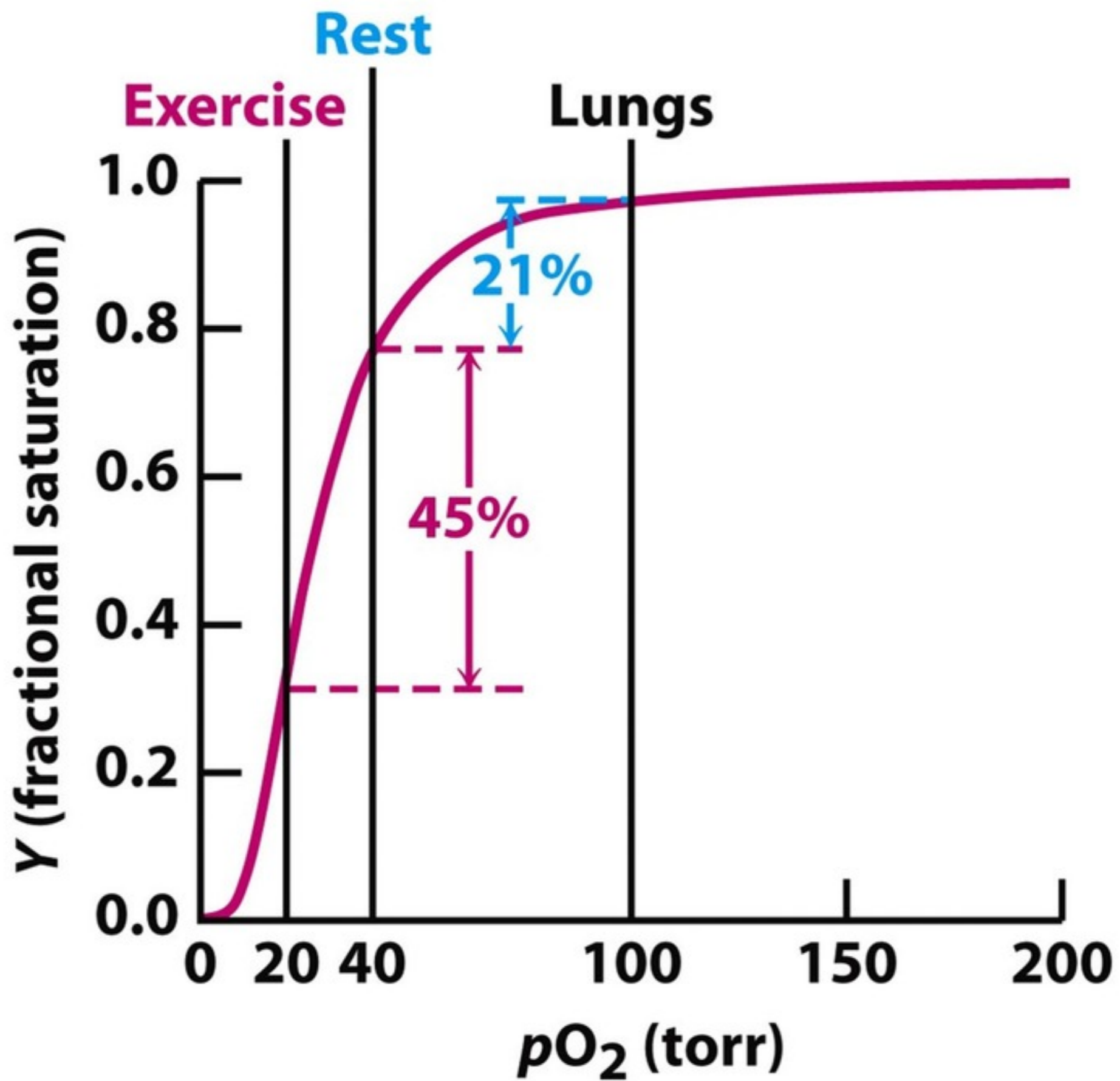
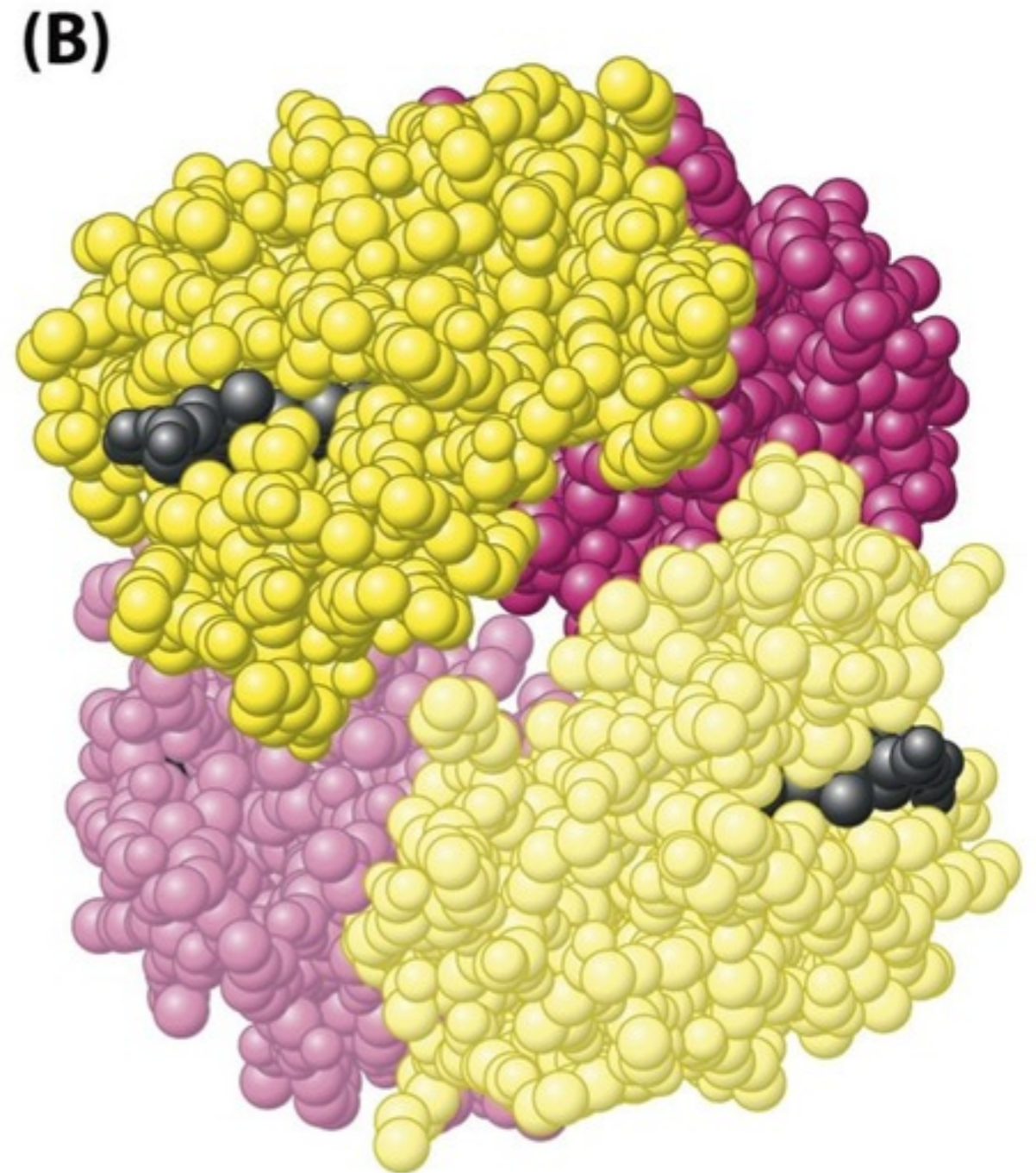
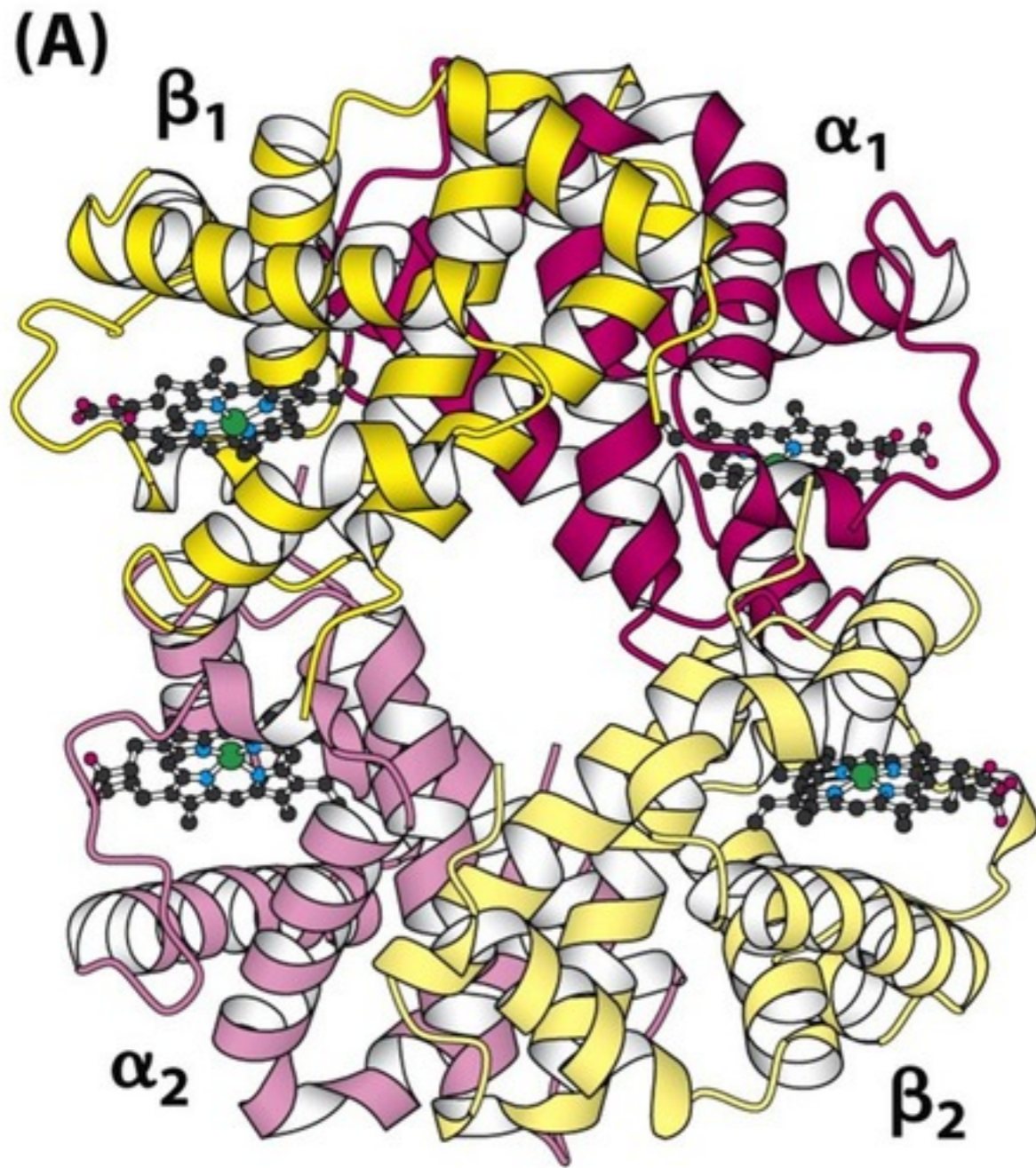


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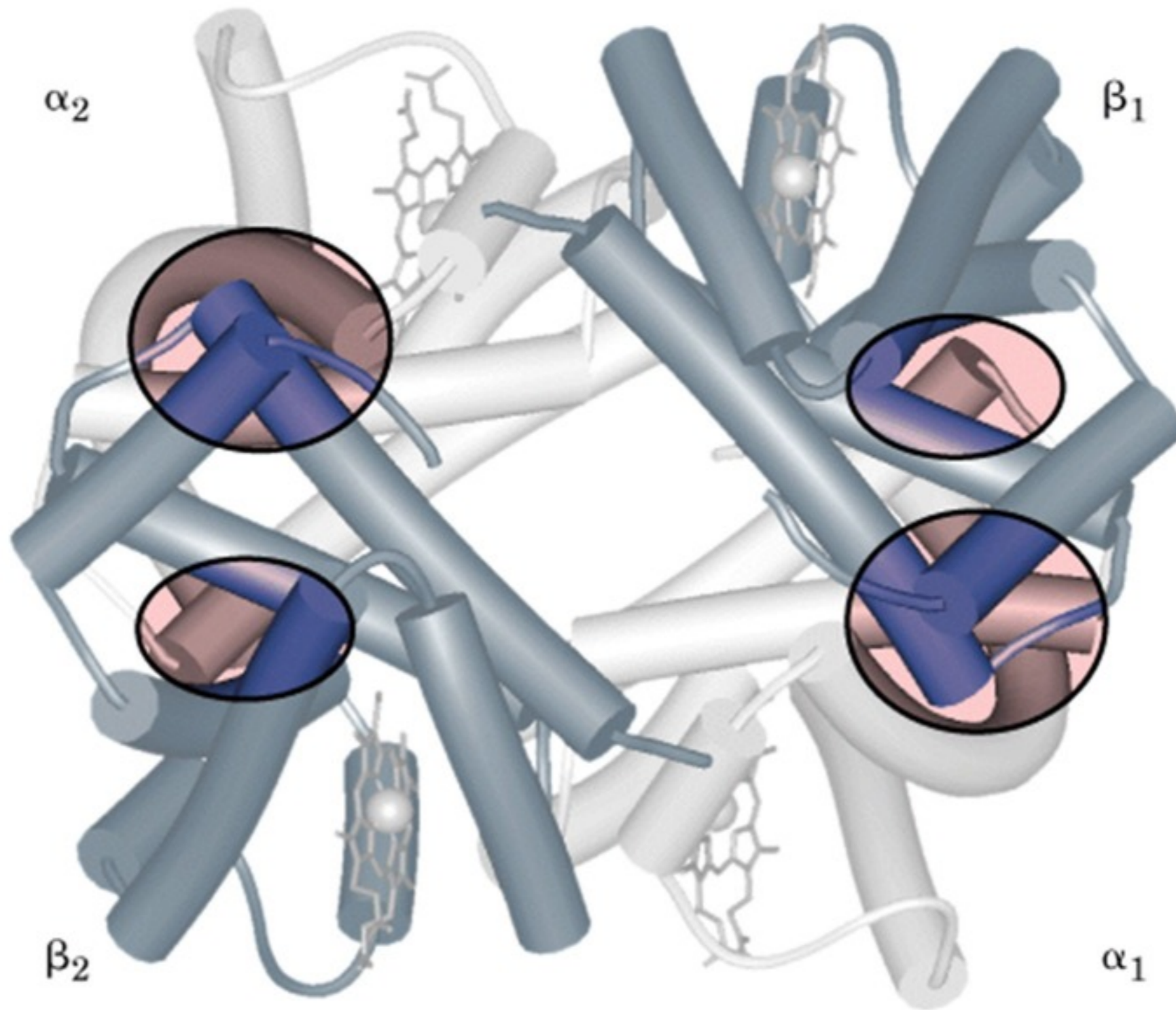
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**Collaborate!**

## Mioglobina e hemoglobina

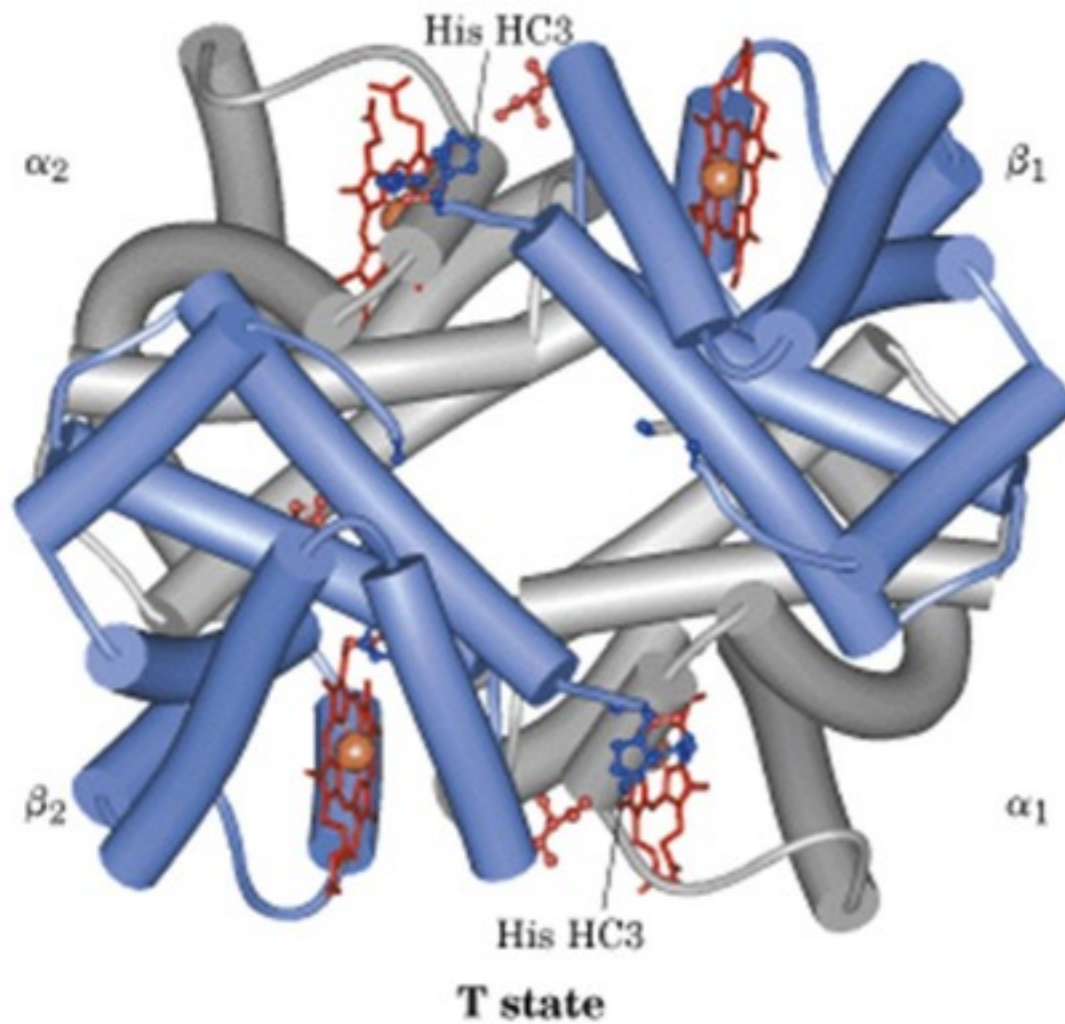
# Interações $\alpha_1 \beta_1$ e $\alpha_2 \beta_2$





**SEM O<sub>2</sub>**

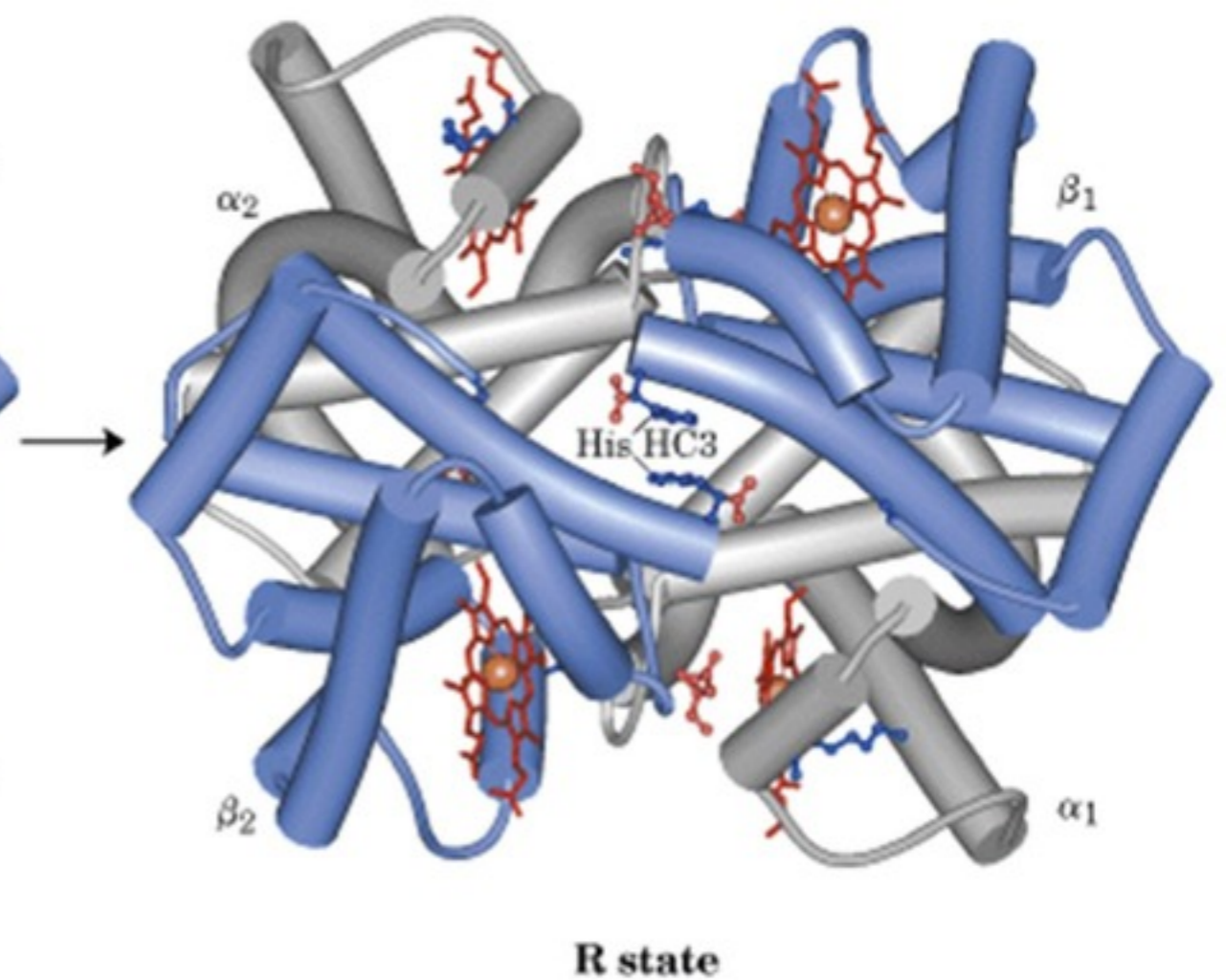
**desoxihemoglobina**



**Tenso:** estabilizado por mais ligações iônicas

**COM O<sub>2</sub>**

**oxihemoglobina**



**Relaxado:** pares iônicos que estabilizam estado T são rompidos, novos são formados

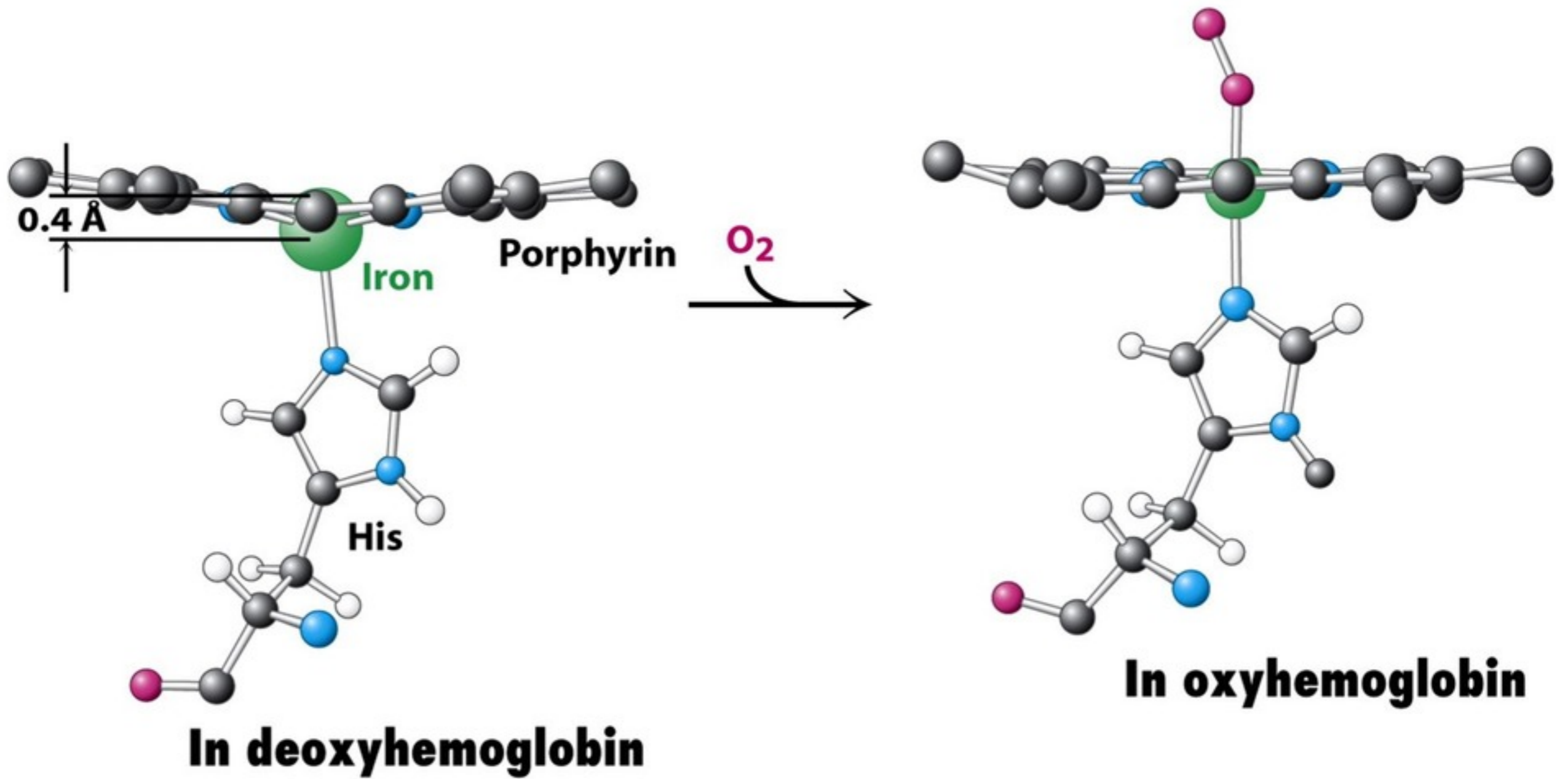
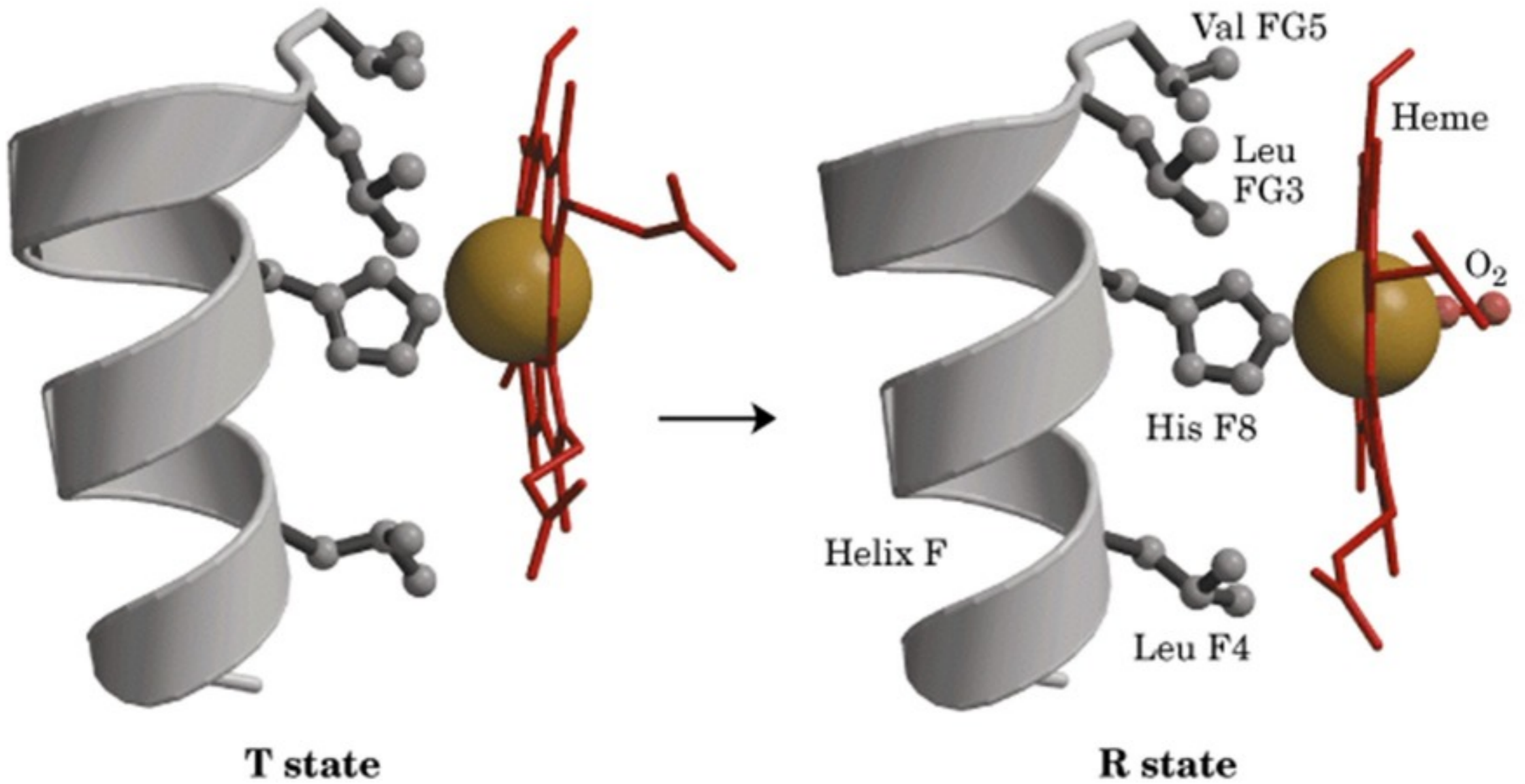
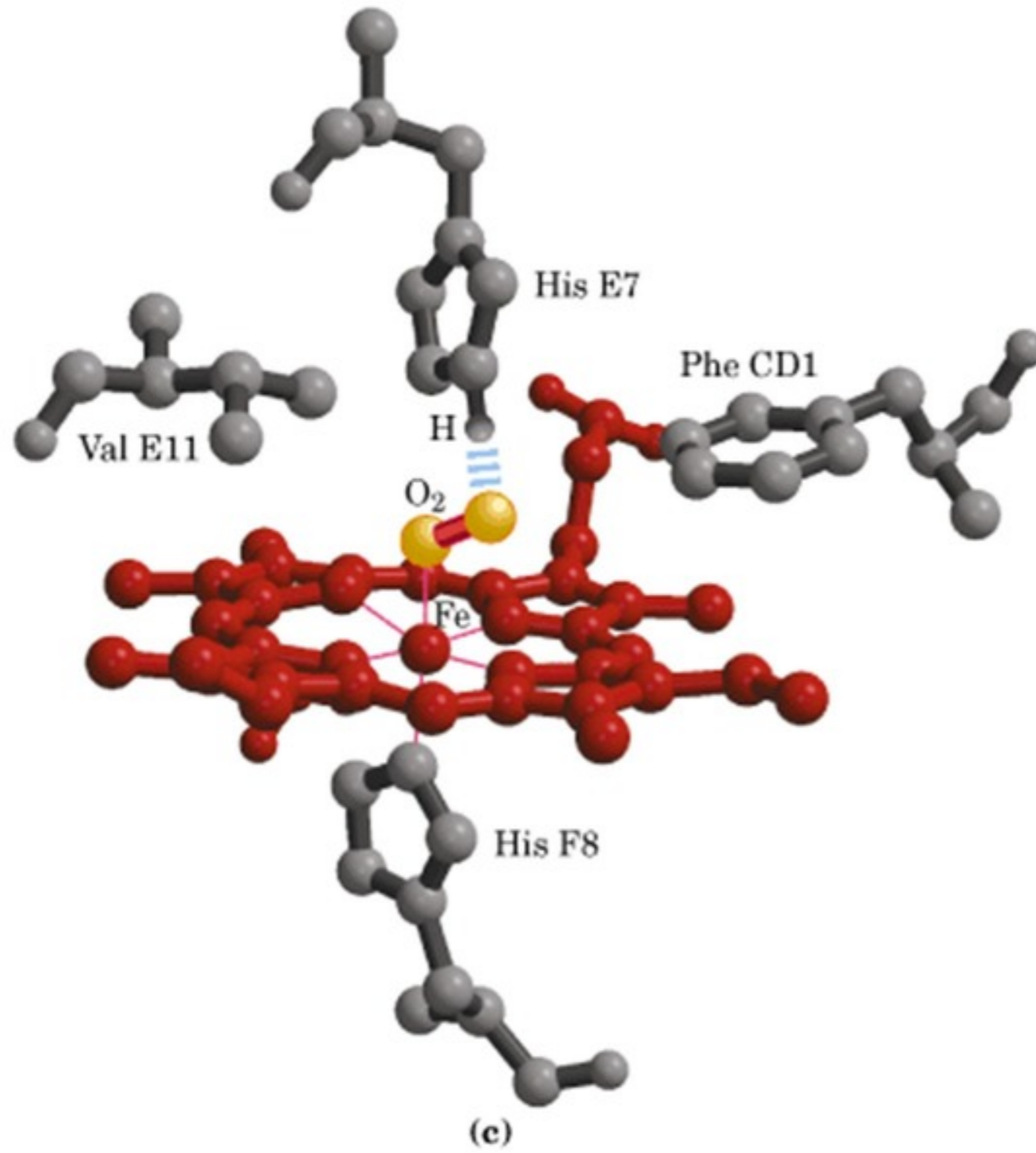


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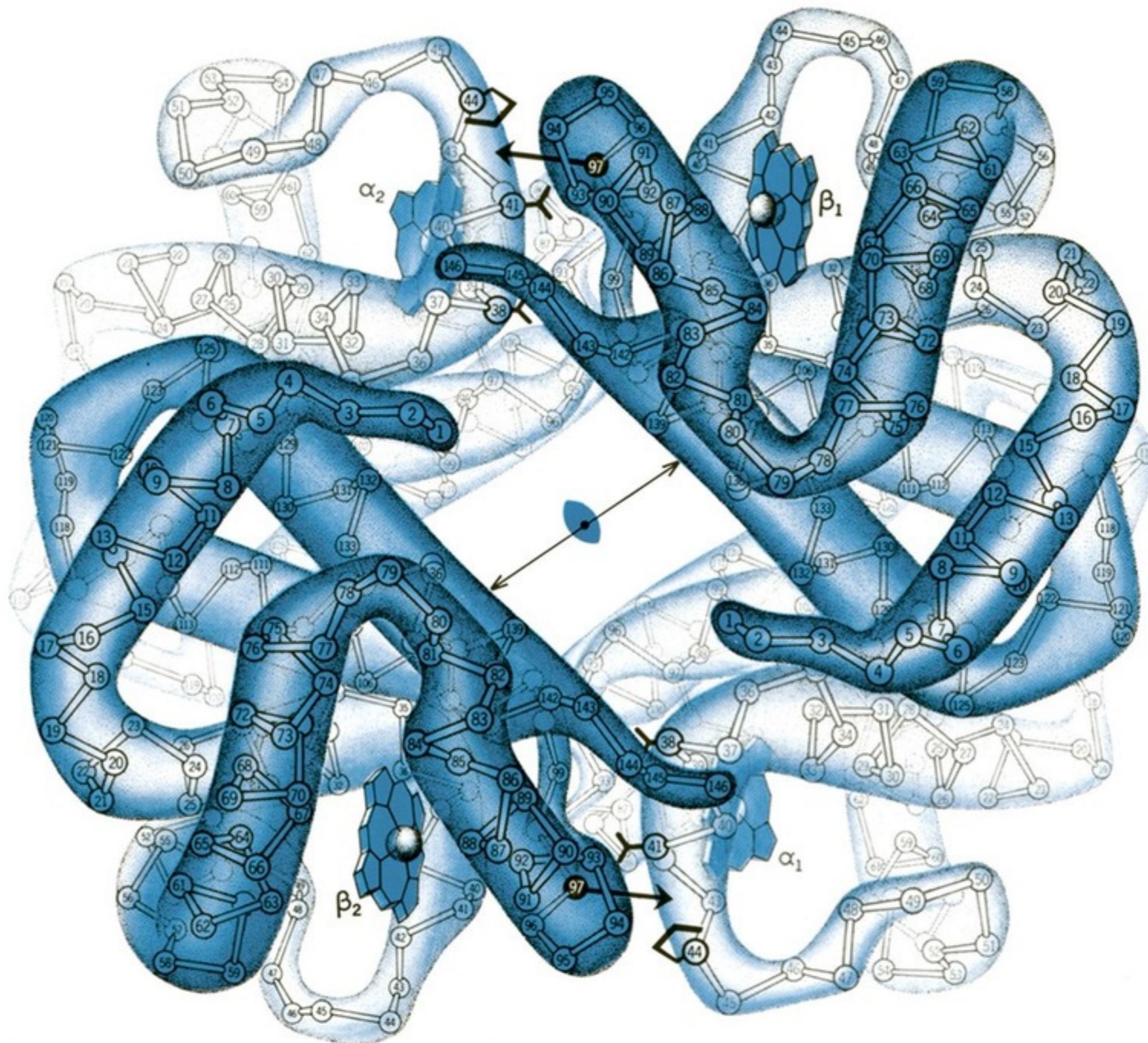


Figure 7-5 part 1 Fundamentals of Biochemistry, 2/e

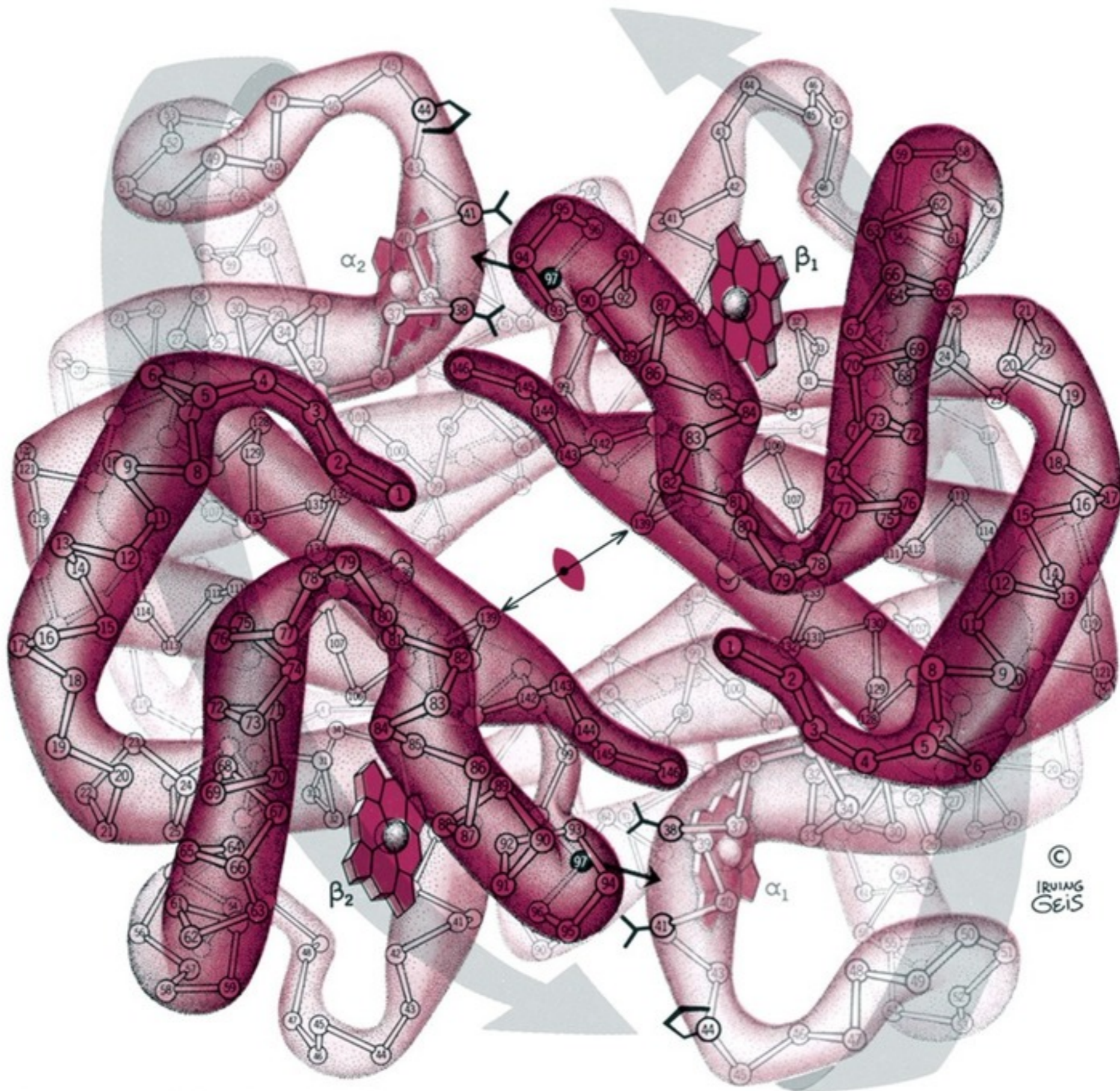
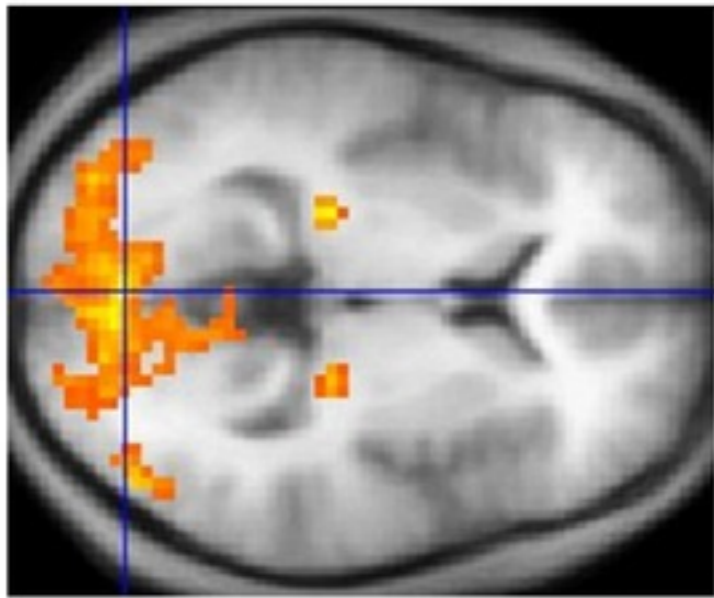


Figure 7-5 part 2 Fundamentals of Biochemistry, 2/e



Scientific american mind, mar 2011

IT IS THIS IDEA THAT MODERN FMRI MACHINES TAKE ADVANTAGE OF TO SCAN YOUR THOUGHTS.

hope you're not claustrophobic, too!

FMRI WORKS BY MEASURING THE OXYGEN CONSUMED BY ACTIVE NEURONS.

WHEN NEURONS TAKE IN OXYGEN FROM THE BLOODSTREAM, THE MAGNETIC PROPERTY OF THE HEMOGLOBIN CHANGES.

POWERFUL MAGNETS IN THE MACHINE ALIGN THE HEMOGLOBIN MOLECULES, THEN CAUSE THEM TO SPIN AND EMIT ENERGY.

BY MEASURING THIS ENERGY, THE MACHINE CAN TELL WHICH AREAS ARE MORE ACTIVE WHEN WE THINK, FEEL OR PLAN SPECIFIC ACTIONS.

BUT IT'S NOT FOOLPROOF ...

# Proteína alostérica:

Ligante em um sítio afeta as propriedades de ligação de outro sítio

*(allos = outra, stereos = estrutura)*

Ligação cooperativa de oxigênio na hemoglobina:  
Ligação alostérica



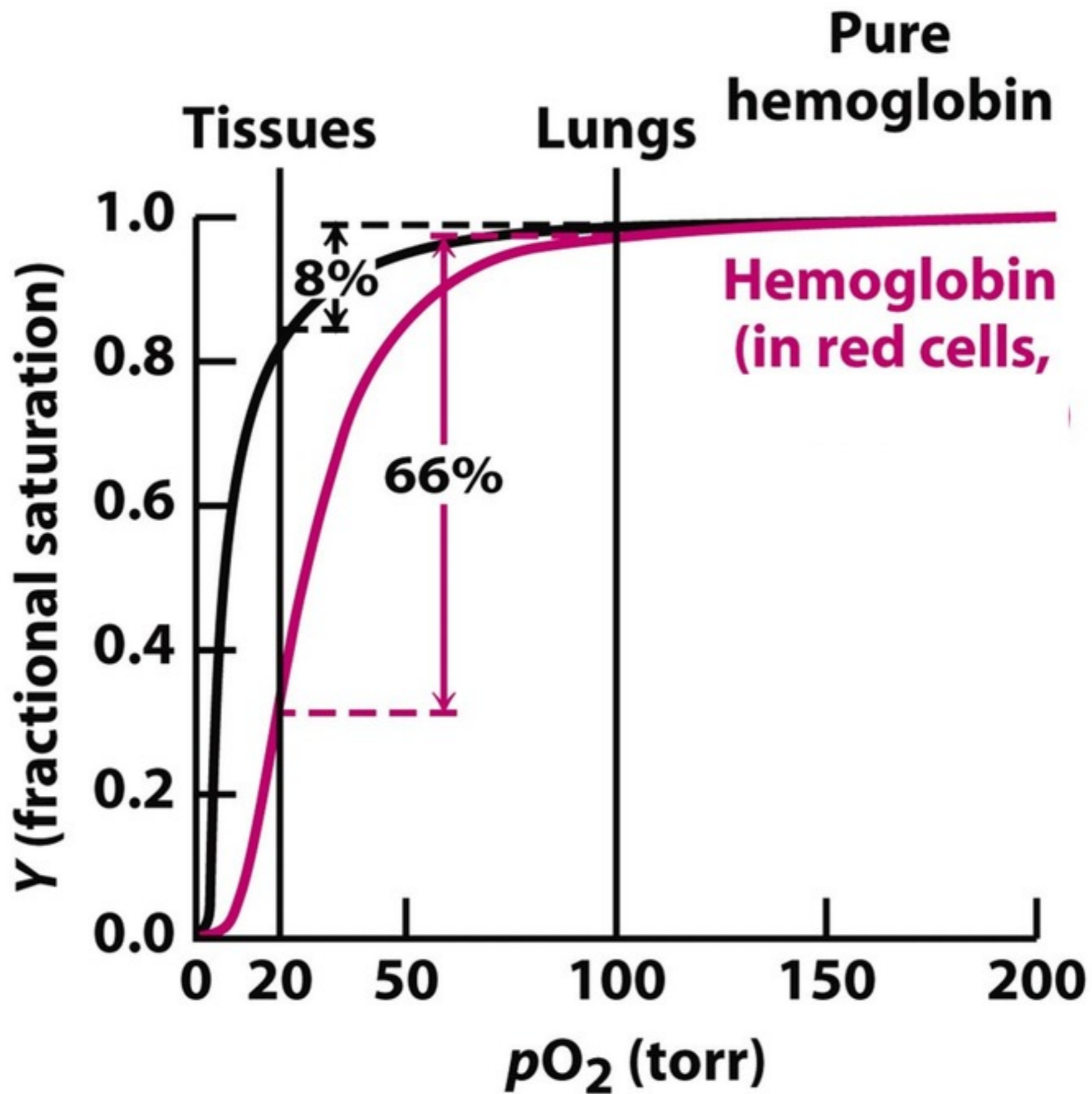
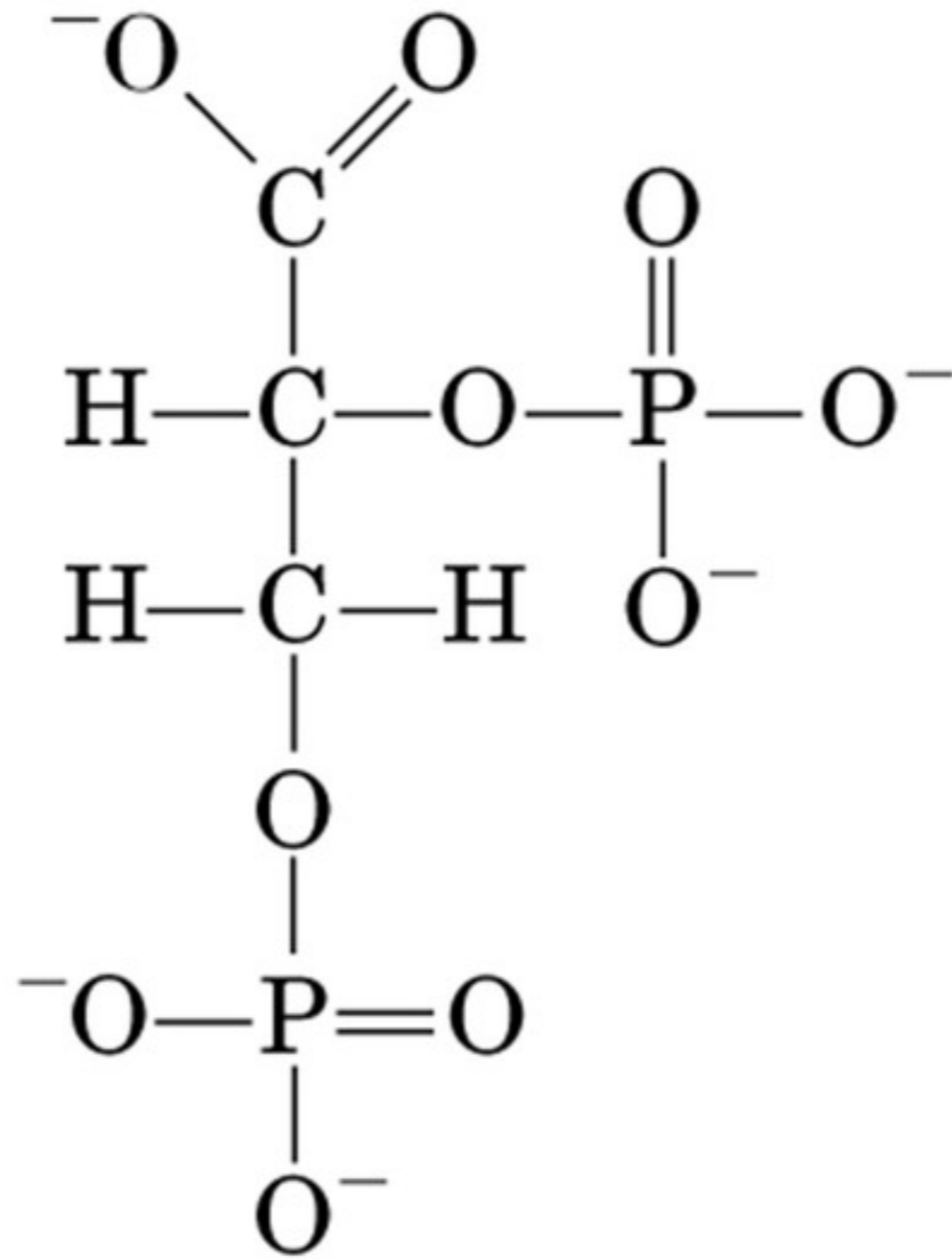


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2,3-Bisphosphoglycerate

# 2,3 BPG: Efetuator alostérico

(*allos = outra, stereos = estrutura*)

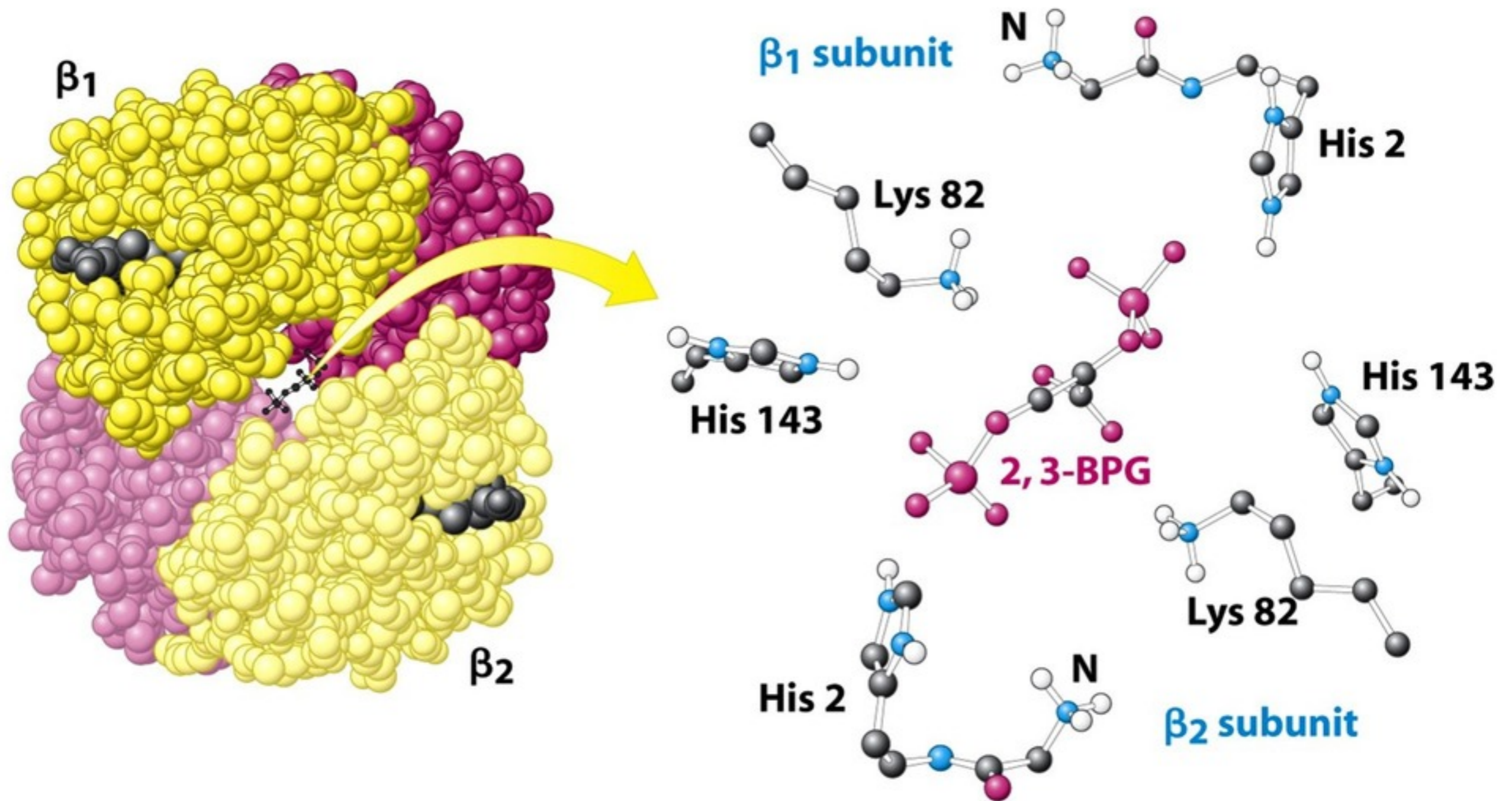


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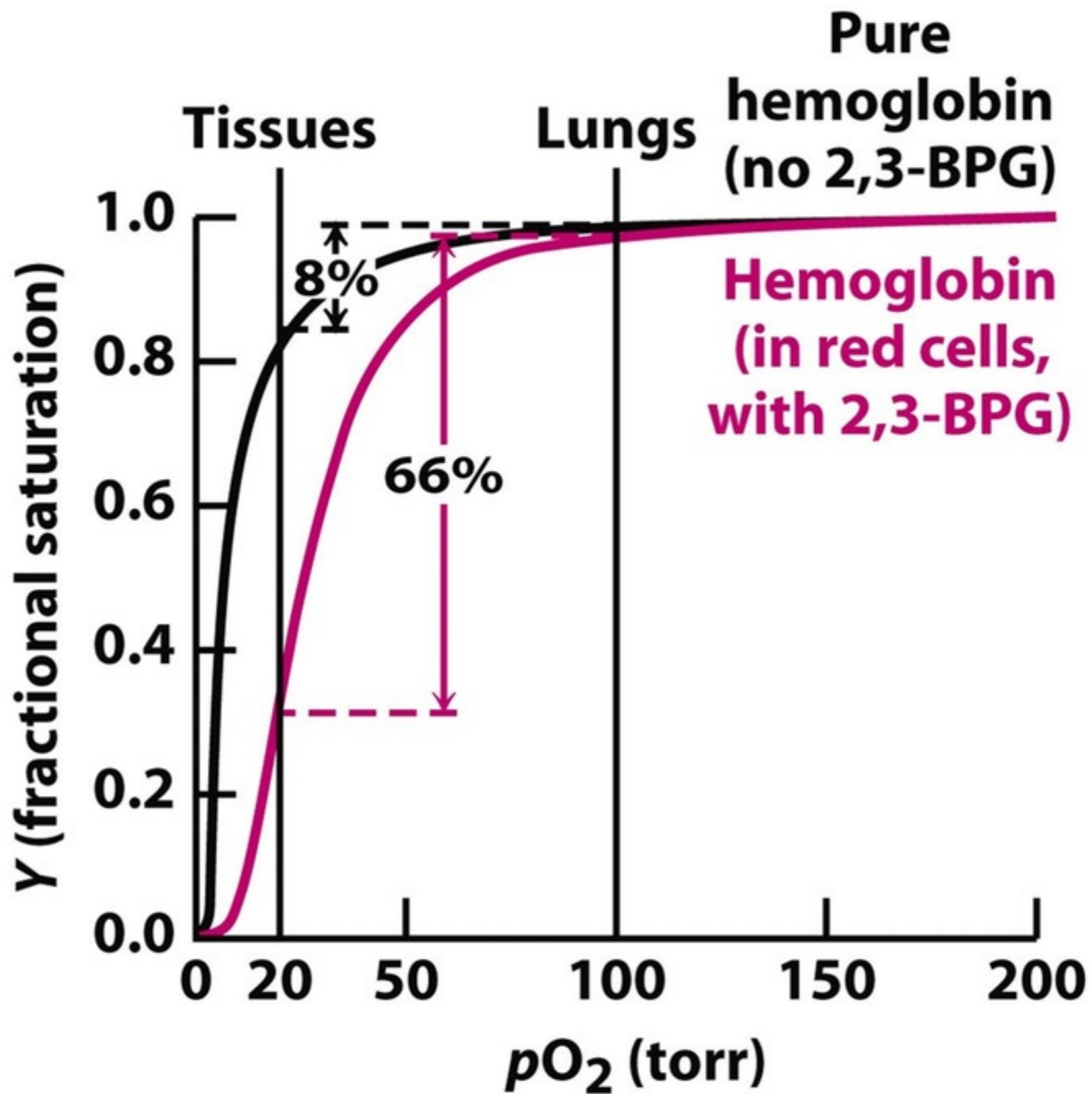


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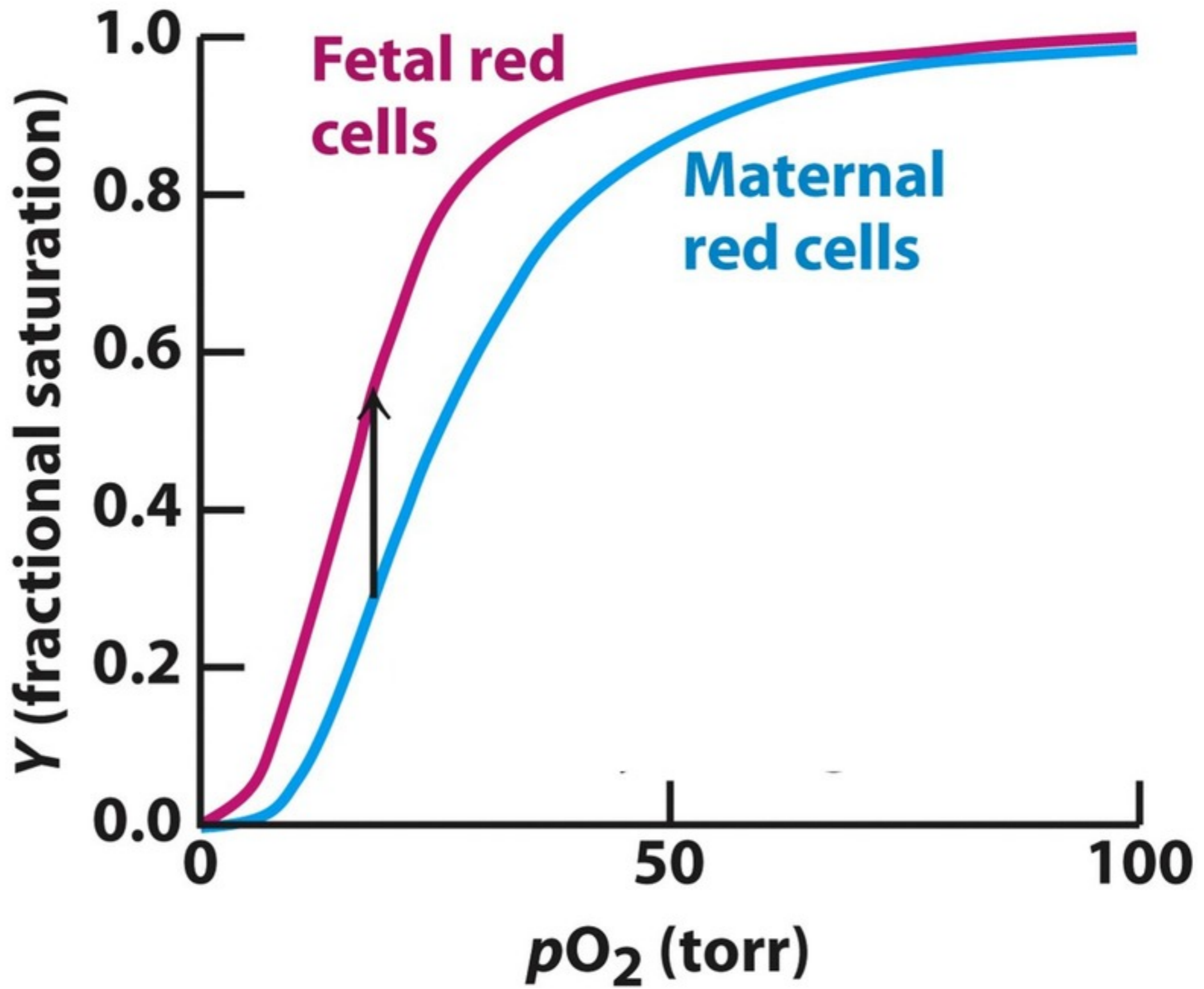
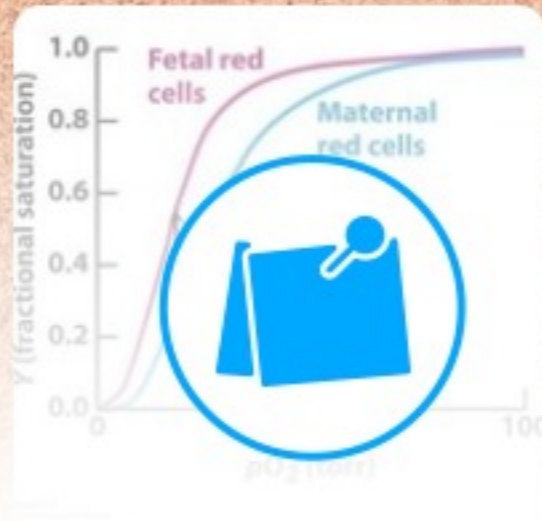
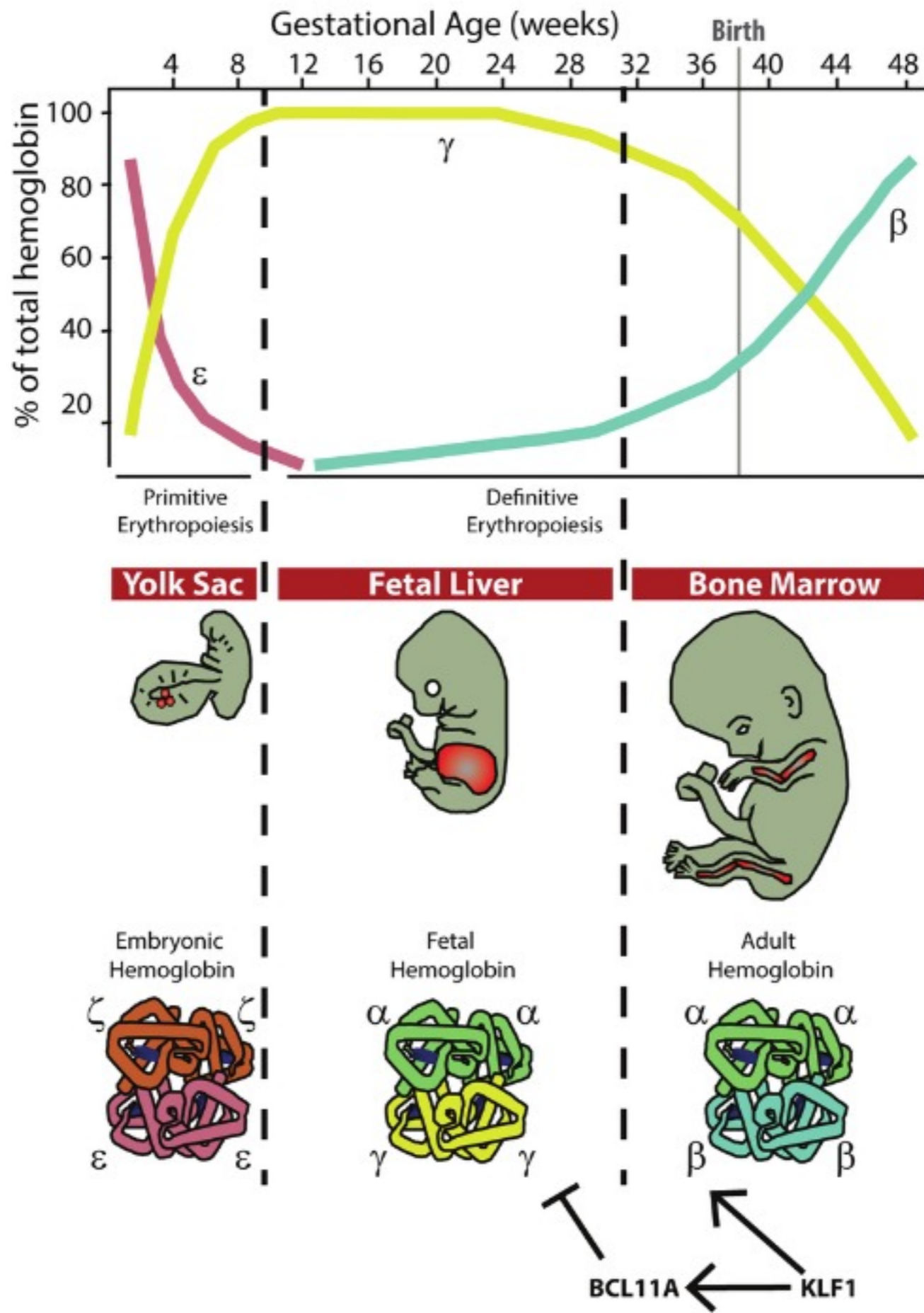


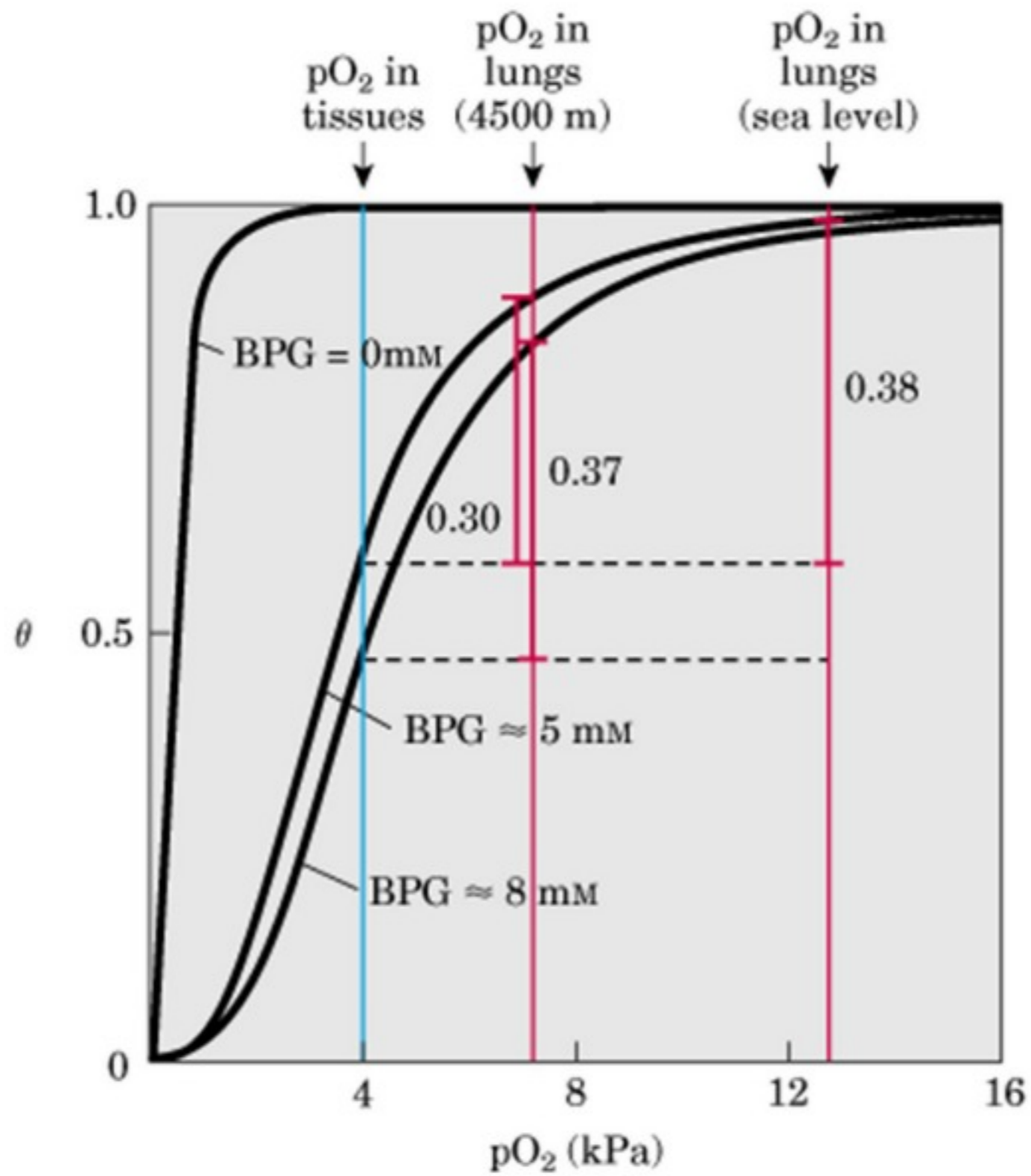
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**Collaborate!**

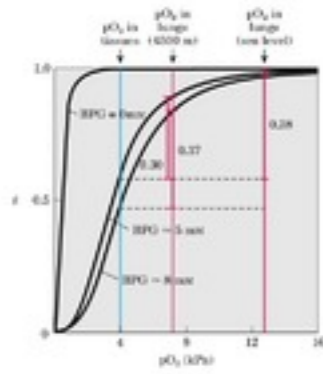
## Hemoglobina fetal e materna







# Open Ended Question



A concentração de BPG de um indivíduo normal a nível do mar é de 5mM. Ao ser transportado para altas altitudes, após algumas horas a concentração de BPG começa a aumentar, atingindo ~8mM. Qual é o efeito deste ajuste nos níveis de BPG e a sua relevância fisiológica?

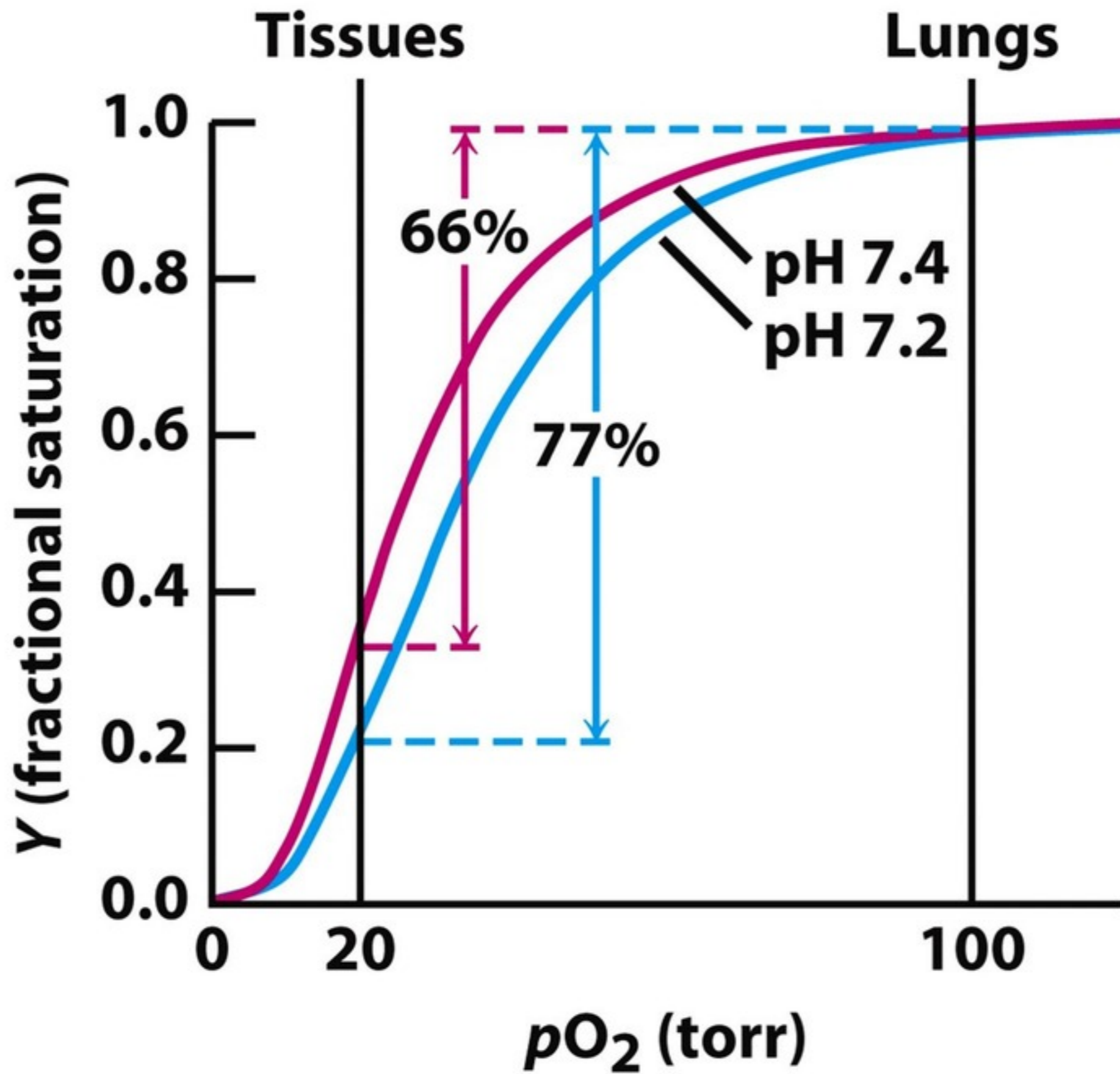


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# Hemoglobina liga $H^+$ e $CO_2$

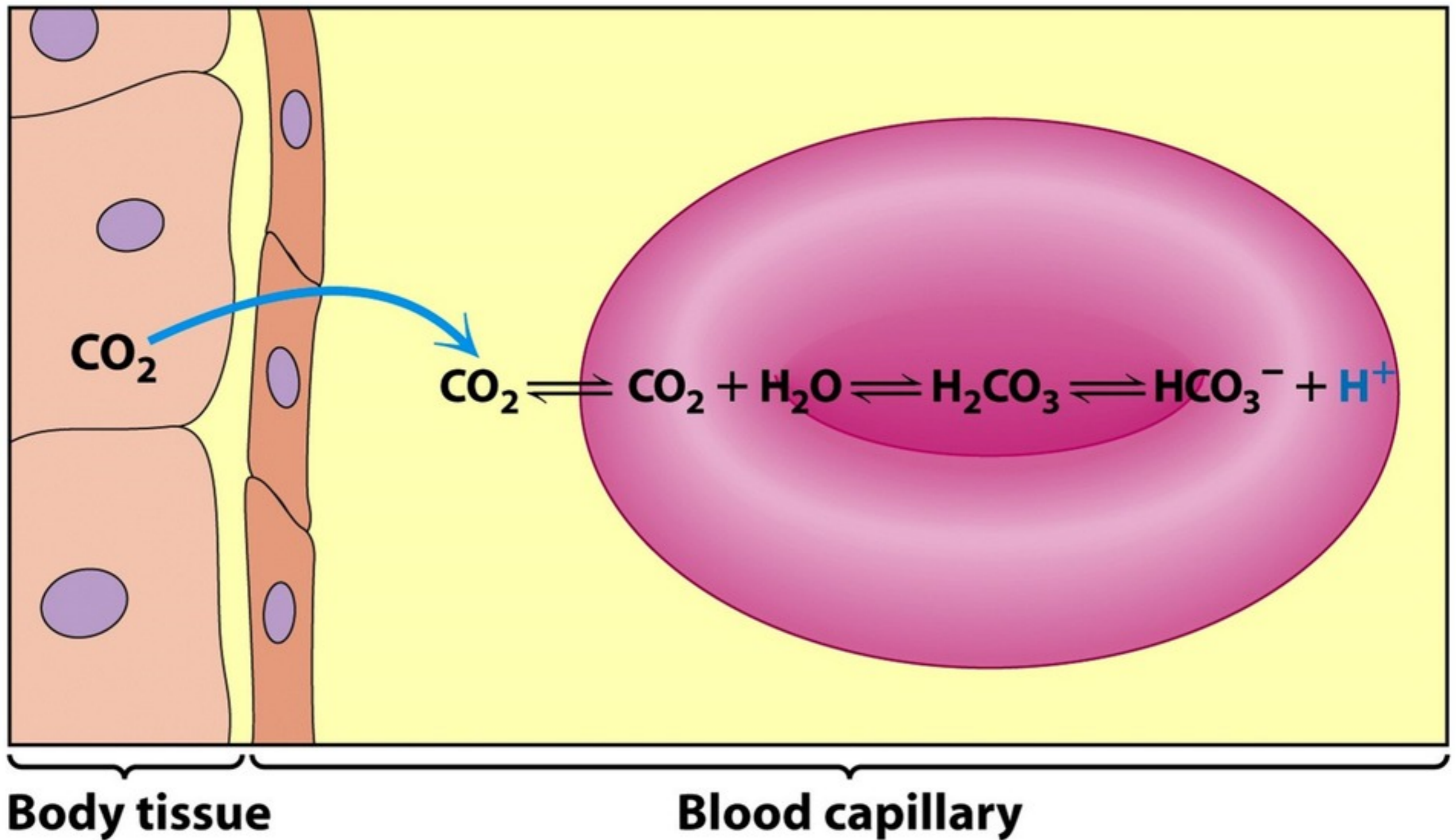


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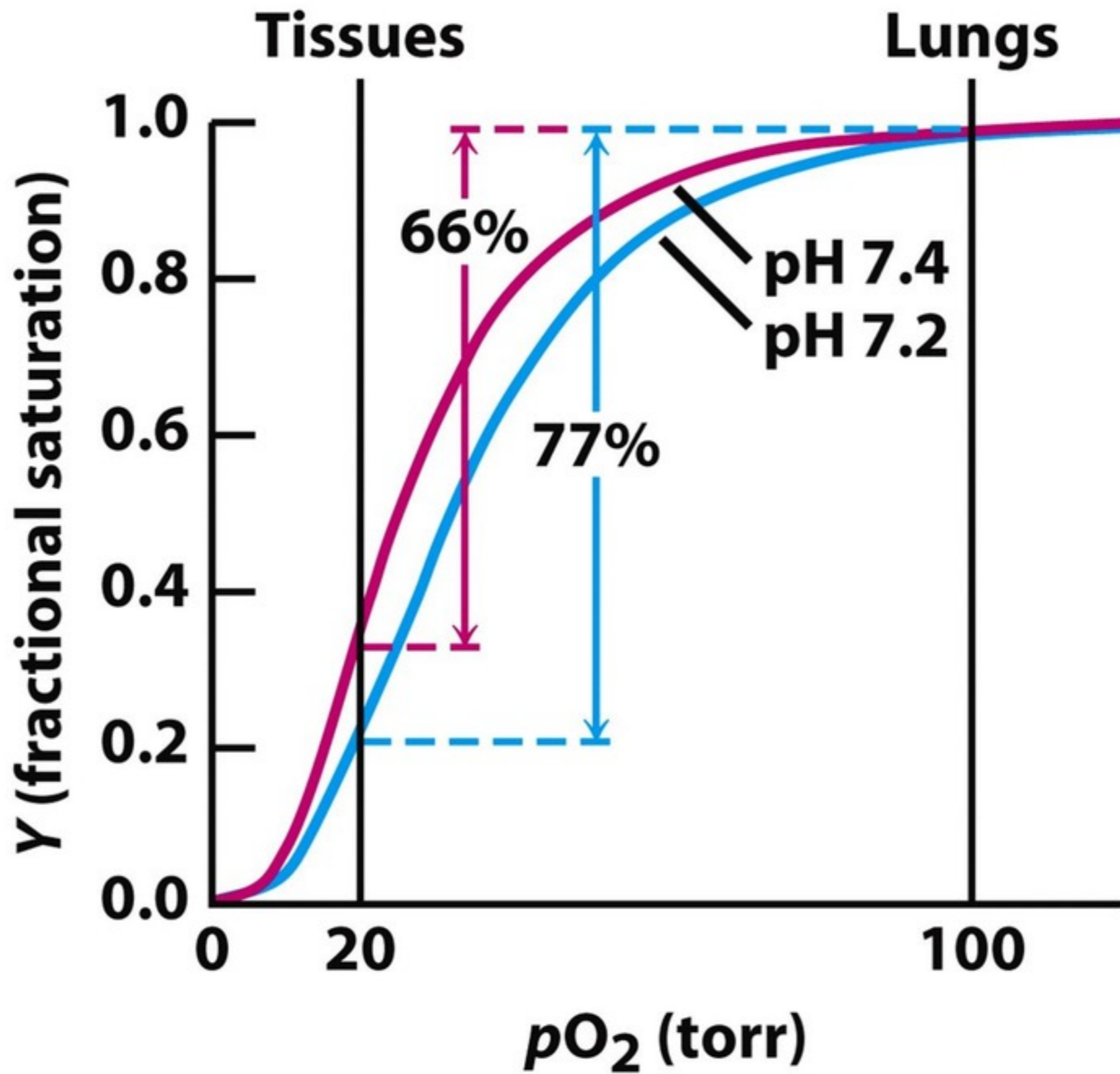


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# Efeito Bohr:

Efeito do pH e CO<sub>2</sub> na liberação de O<sub>2</sub>

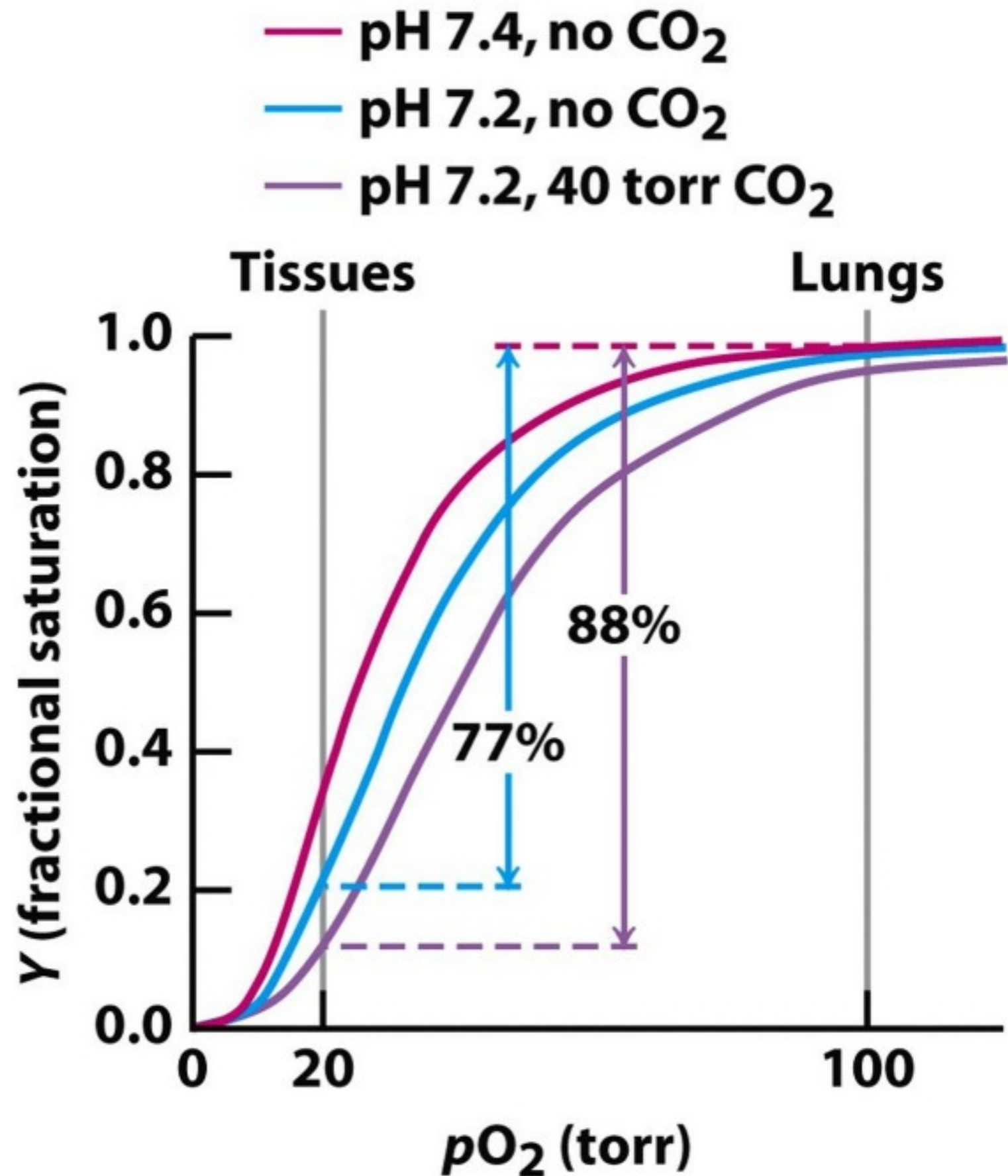


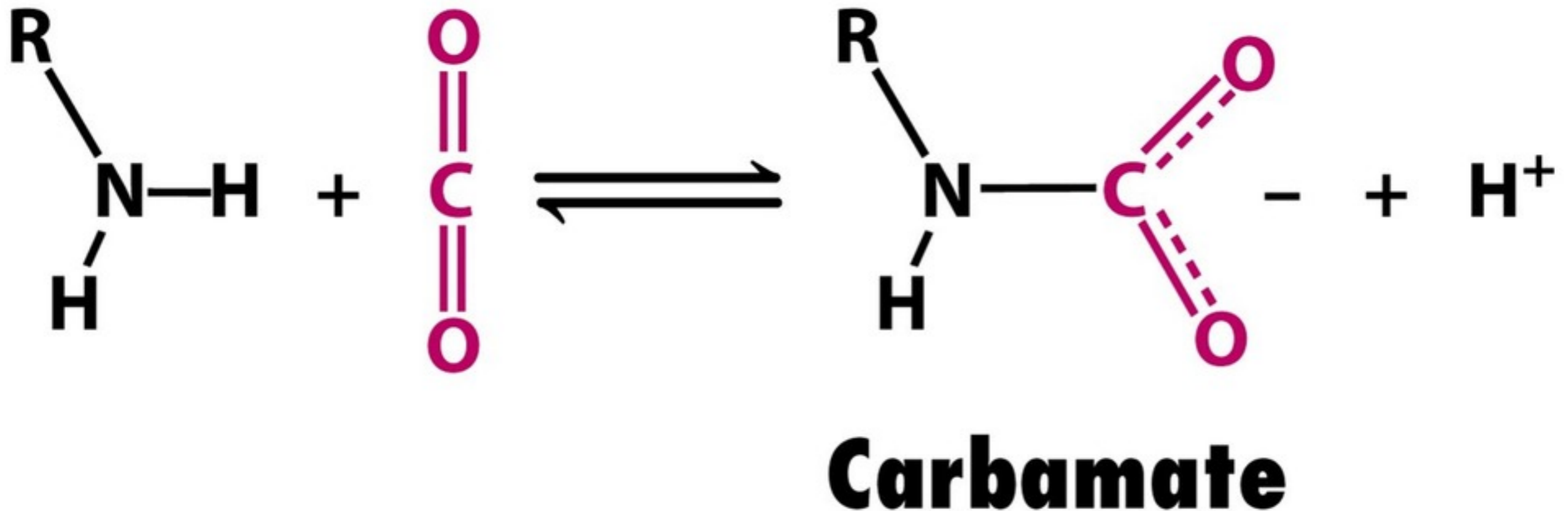
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## O<sub>2</sub>, H<sup>+</sup> e CO<sub>2</sub> se ligam a hemoglobina

O<sub>2</sub>: grupo heme

H<sup>+</sup>: resíduos de aa da proteína – His146 – estabiliza deoxihemoglobina

CO<sub>2</sub>: aminas terminais da cadeia de globina – estabiliza desoxihemoglobina



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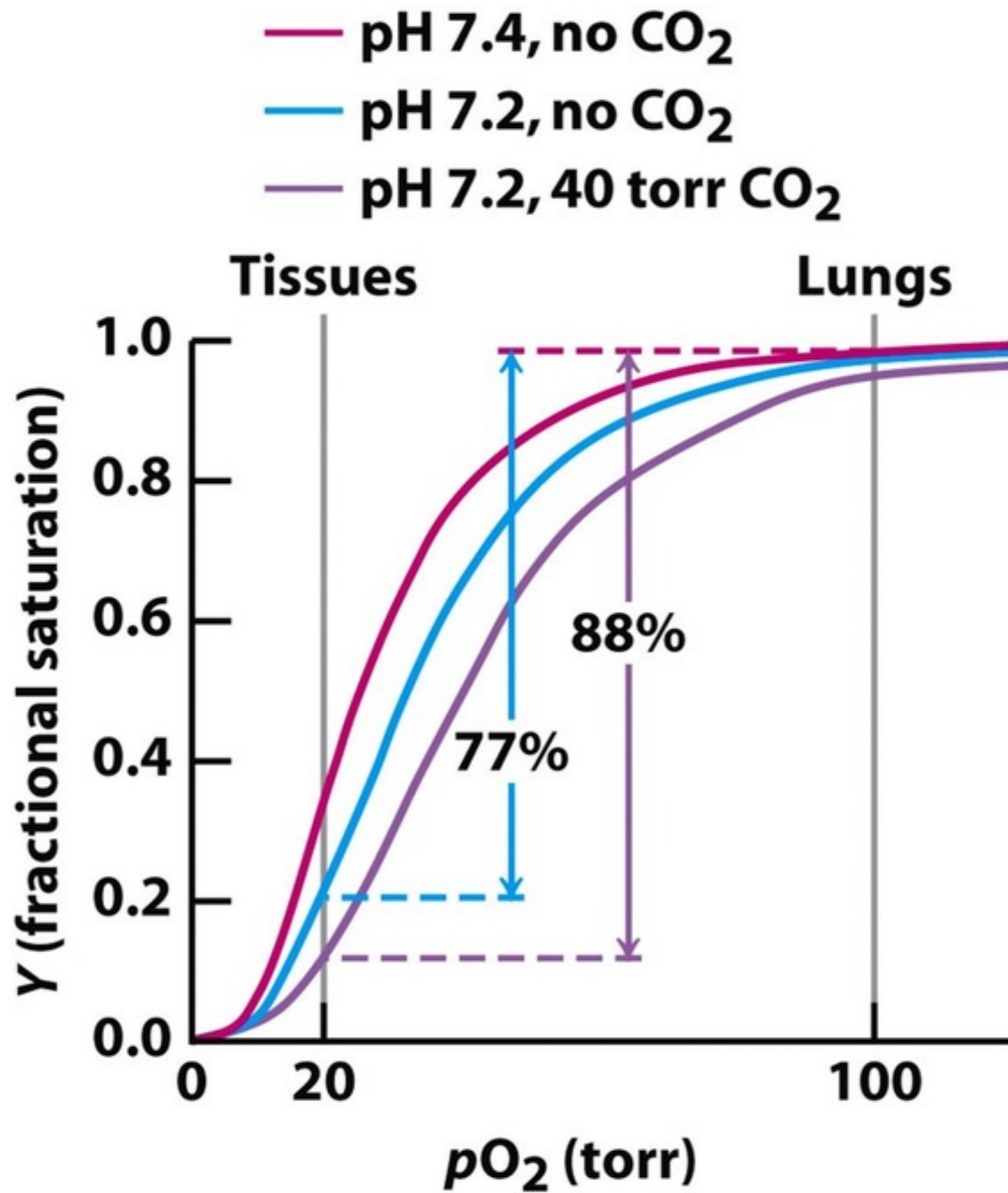


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# Open Ended Question

O que acontece com a afinidade da hemoglobina pelo oxigênio quando há ligação de  $H^+$  e  $CO_2$ ? Quando  $CO_2$  é excretado e o pH do sangue aumenta, o que acontece com a afinidade da hemoglobina pelo oxigênio?

Oxi-Hb  $\rightarrow$  desoxi-Hb  
 Captação de prótons

desoxi-Hb  $\rightarrow$  oxi-Hb  
 Liberação de prótons

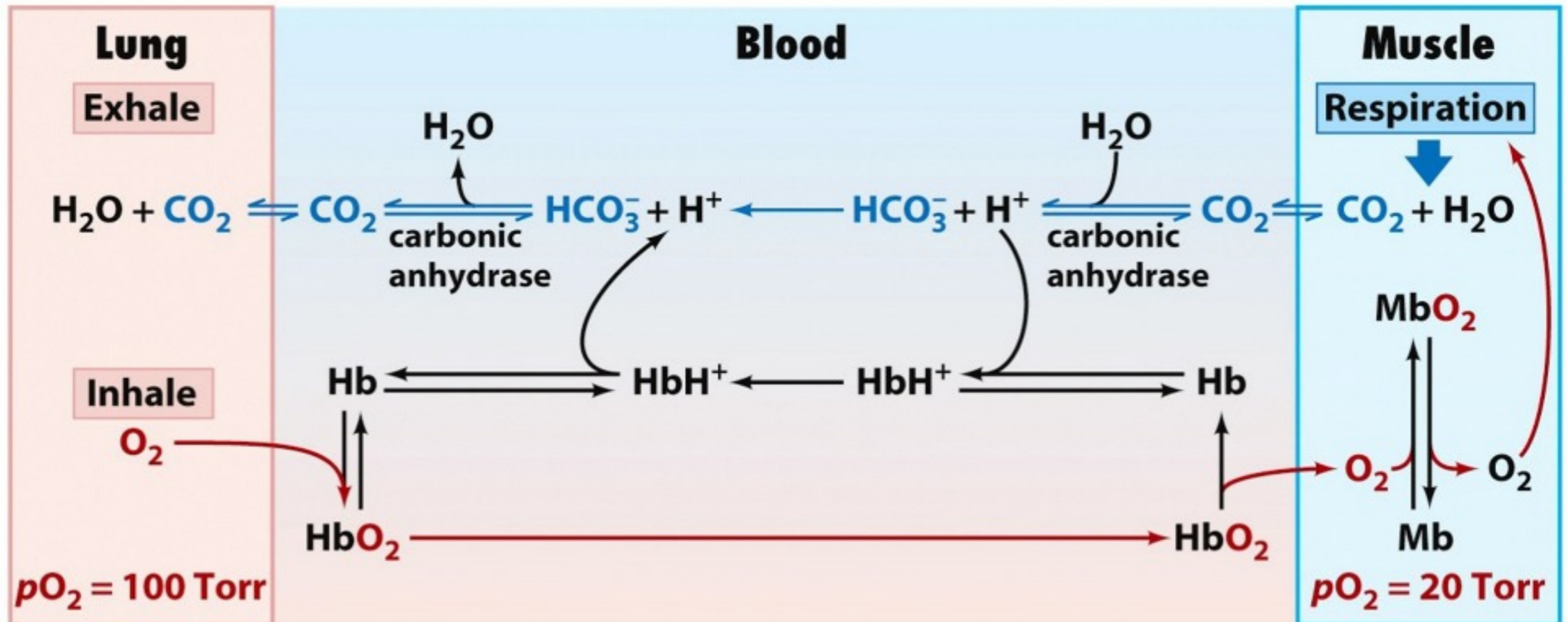


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