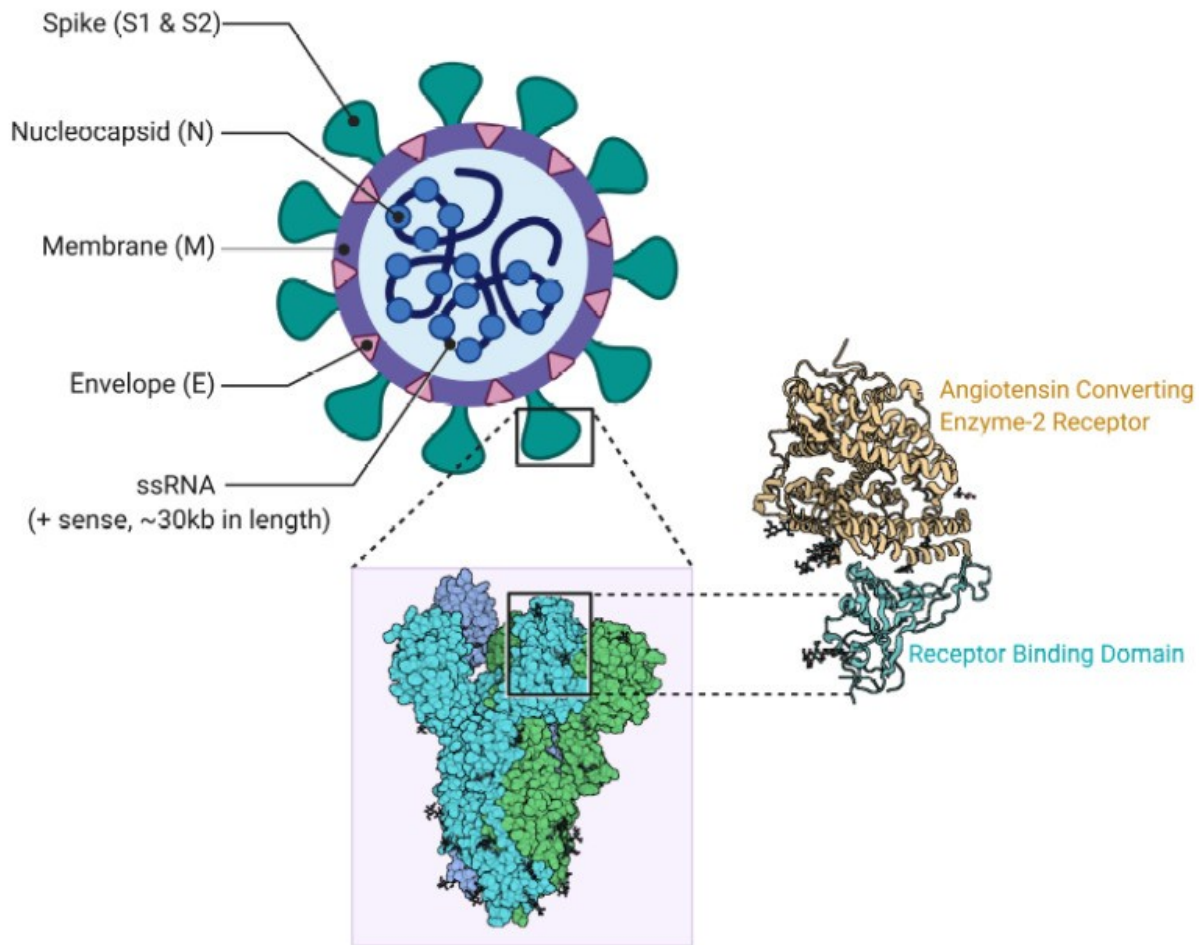




# Aminoácidos

Contribuição slides : Evelyn Ayumi Onga  
(PAE 2020)  
Prof<sup>a</sup> Dr<sup>a</sup> Tie Koide

# SARS-CoV 2 Structure



# Covid: Variante brasileira corresponde a 90% dos casos em SP, diz estudo




Análise identificou a variante P.1 correspondente em 90% dos sequenciamentos genéticos realizados; 21 linhagens já foram encontradas em São Paulo

Imagem: Divulgação/Governo do Estado de São Paulo



Colaboração para o UOL  
28/04/2021 13h31 | Atualizada em 28/04/2021 14h25

Um estudo do Instituto Adolfo Lutz concluiu que a variante brasileira, a P.1, [que surgiu em Manaus](#), corresponde a 90% dos casos analisados de infecção pelo novo [coronavírus](#) no estado de São Paulo.



# VARIANTE 'BRASILEIRA' P1 PODE TER CARGA VIRAL ATÉ 10 VEZES MAIOR QUE OUTRAS CEPAS, DIZ CIENTISTA DA FIOCRUZ

Descoberta consta em estudo de epidemiologia genômica assinado por 29 pesquisadores brasileiros da área de virologia

Época abril 2021

## **Covid-19: Variante brasileira faz explodir intubações de jovens**

**A maioria dos casos do novo coronavírus em São Paulo se concentra entre pessoas de 20 a 54 anos**

## **Brasil, um possível celeiro de novas variantes do coronavírus**

Grande número de casos, medidas de restrição não respeitadas, falta de sequenciamento do vírus e vacinação lenta fazem do país um caldeirão de mutações

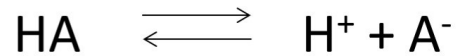
Veja saúde abril 2021

# Ácidos e Bases

Definição de Brønsted & Lowry

Ácido: capaz de doar  $H^+$

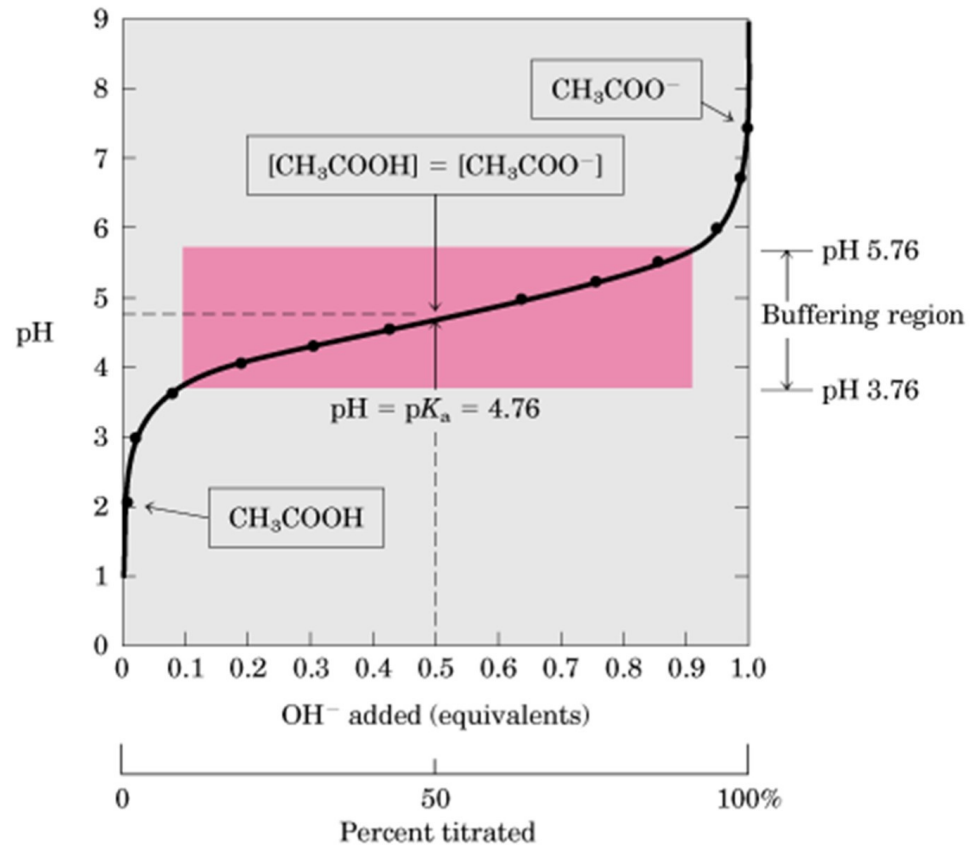
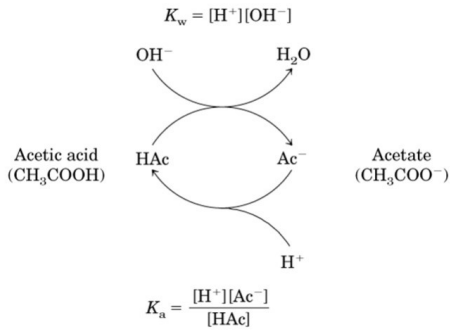
Base: capaz de receber  $H^+$



$$K_{eq} = \frac{[H^+][A]}{[HA]} = K_a$$

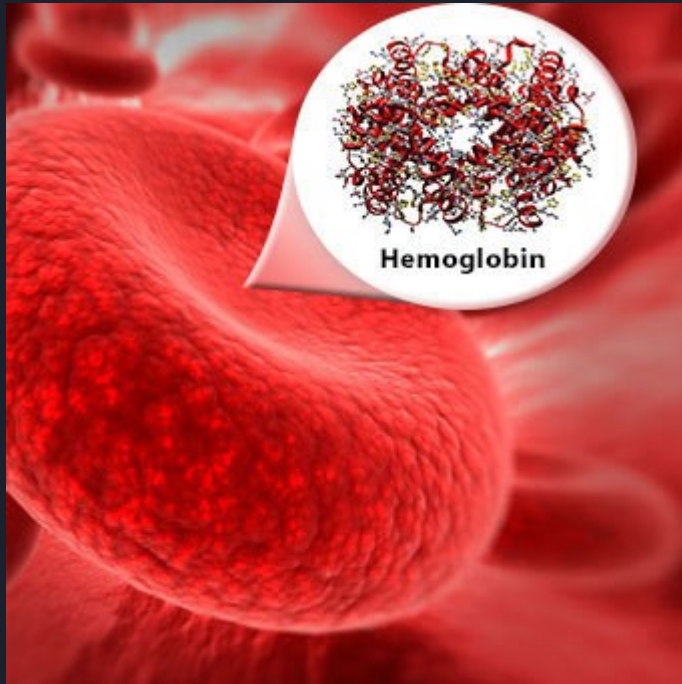
Constante de dissociação

$$pK_a = \log \frac{1}{K_a} = -\log K_a$$

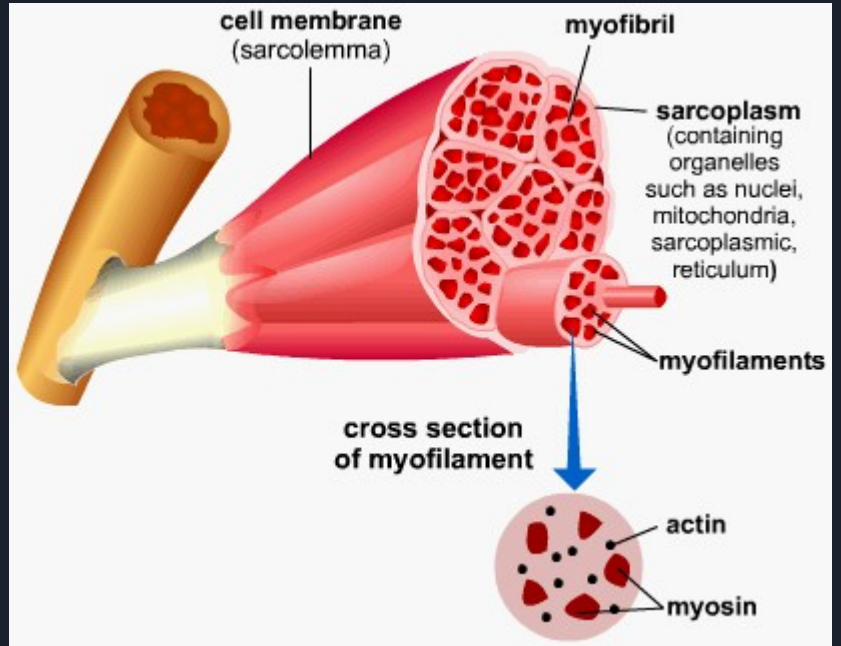


$$\text{pH} = \text{p}K_a + \log \frac{[\text{A}^-]}{[\text{HA}]}$$

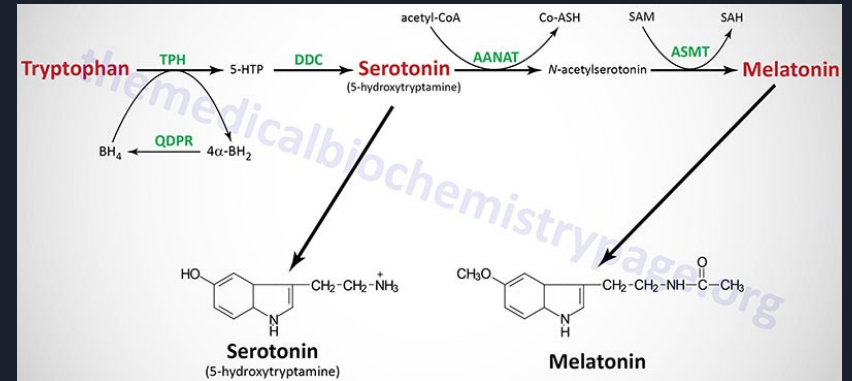
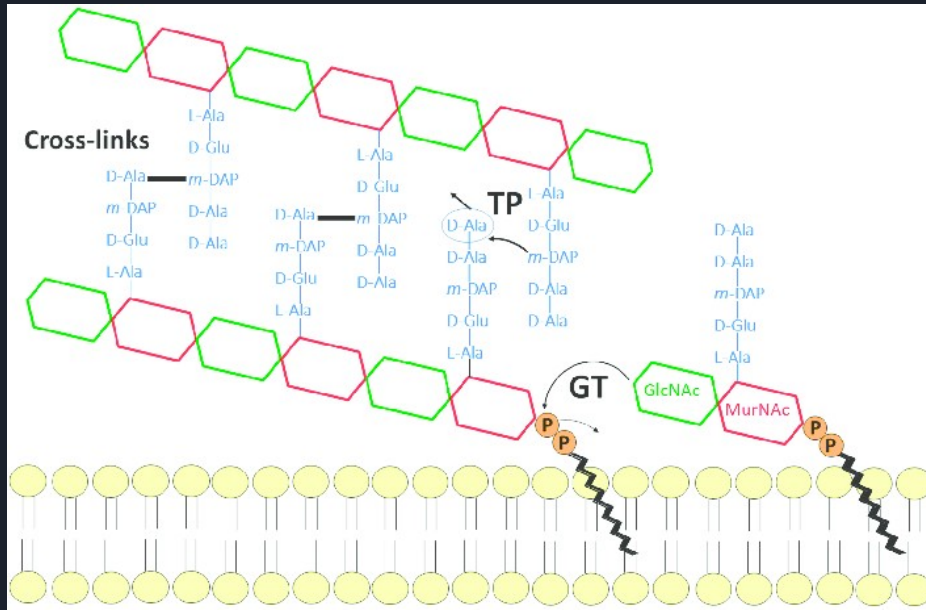
Um ácido fraco está 50% dissociado num pH igual ao seu pKa



Transporte

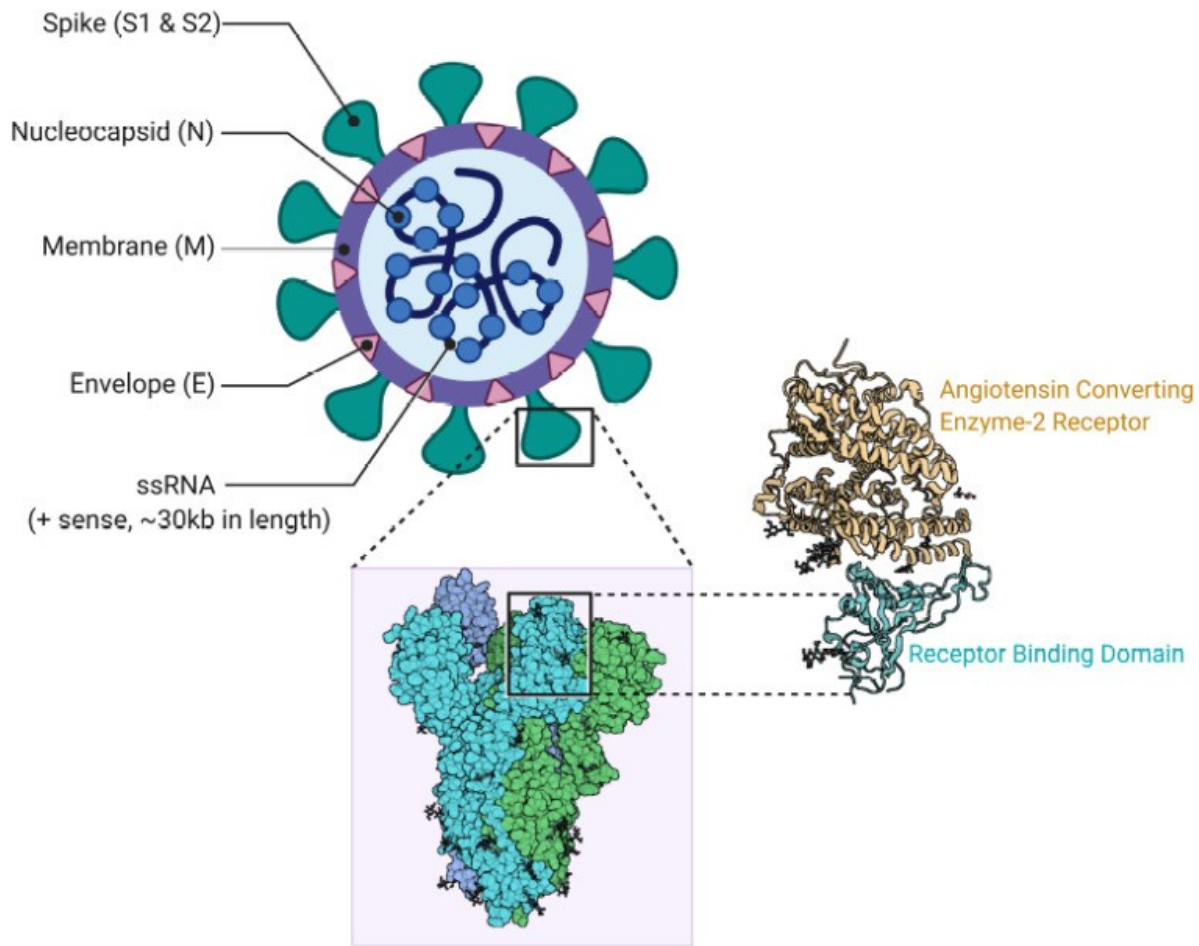


Estrutura  
Contração muscular

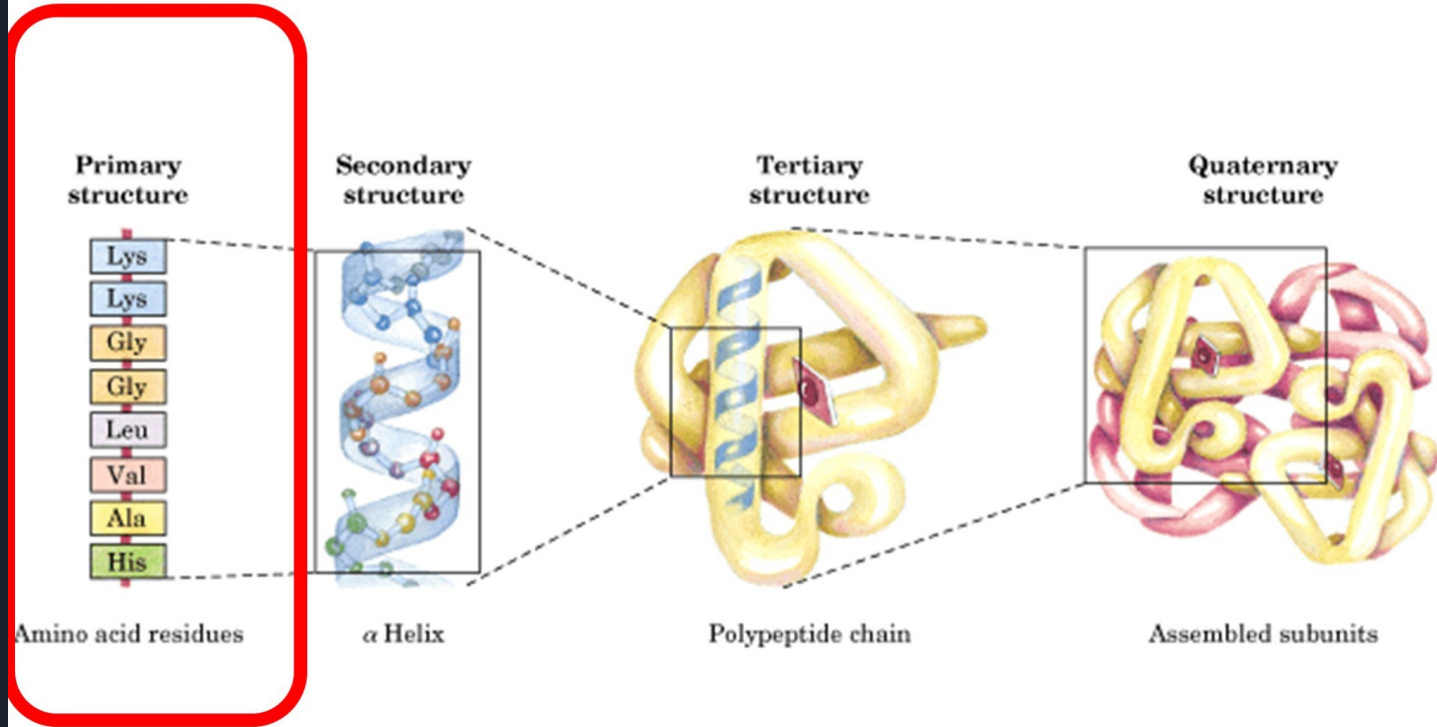


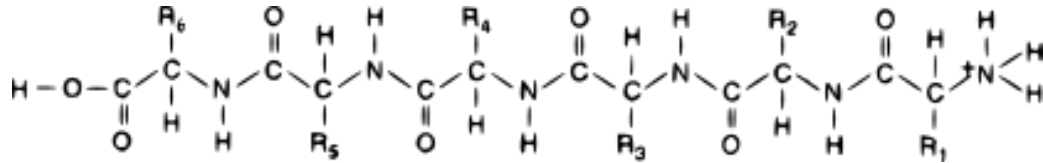


# SARS-CoV 2 Structure

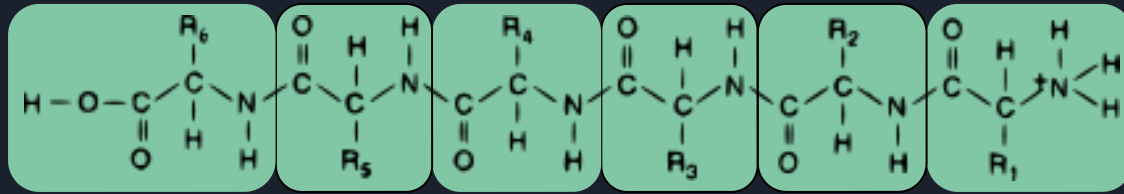


# Proteínas: polímeros de aminoácidos





Aminoácidos - Unidades monoméricas que formam as proteínas (polímeros).



Aminoácidos - Unidades monoméricas que formam as proteínas (polímeros de aminoácidos).

Formação de proteína hipotética, com 20 aa, um de cada tipo =  $2.4 \times 10^{18}$  moléculas diferentes

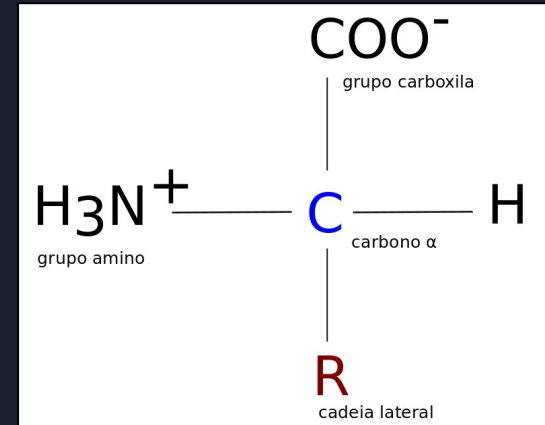
table 5-2

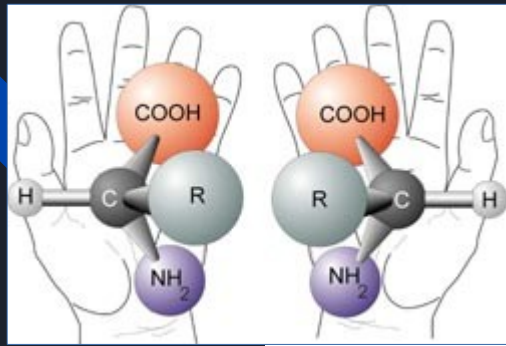
<b>Molecular Data on Some Proteins</b>			
	<b>Molecular weight</b>	<b>Number of residues</b>	<b>Number of polypeptide chains</b>
Cytochrome <i>c</i> (human)	13,000	104	1
Ribonuclease A (bovine pancreas)	13,700	124	1
Lysozyme (egg white)	13,930	129	1
Myoglobin (equine heart)	16,890	153	1
Chymotrypsin (bovine pancreas)	21,600	241	3
Chymotrypsinogen (bovine)	22,000	245	1
Hemoglobin (human)	64,500	574	4
Serum albumin (human)	68,500	609	1
Hexokinase (yeast)	102,000	972	2
RNA polymerase ( <i>E. coli</i> )	450,000	4,158	5
Apolipoprotein B (human)	513,000	4,536	1
Glutamine synthetase ( <i>E. coli</i> )	619,000	5,628	12
Titin (human)	2,993,000	26,926	1

# 1. Conceito e estrutura geral de um AMINOÁCIDO

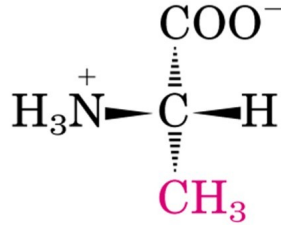
Substâncias que apresentam grupos funcionais **amino** e **ácido carboxílico**.

- **Cadeia lateral** ou **grupo R** variável: pode conter grupos funcionais, carbonos ou hidrogênio.

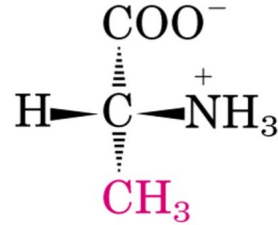




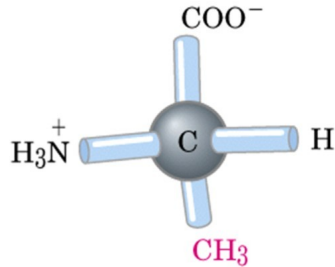
## As proteínas são constituídas de L-aminoácidos



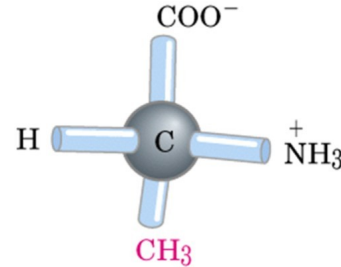
L-Alanine



D-Alanine



L-Alanine

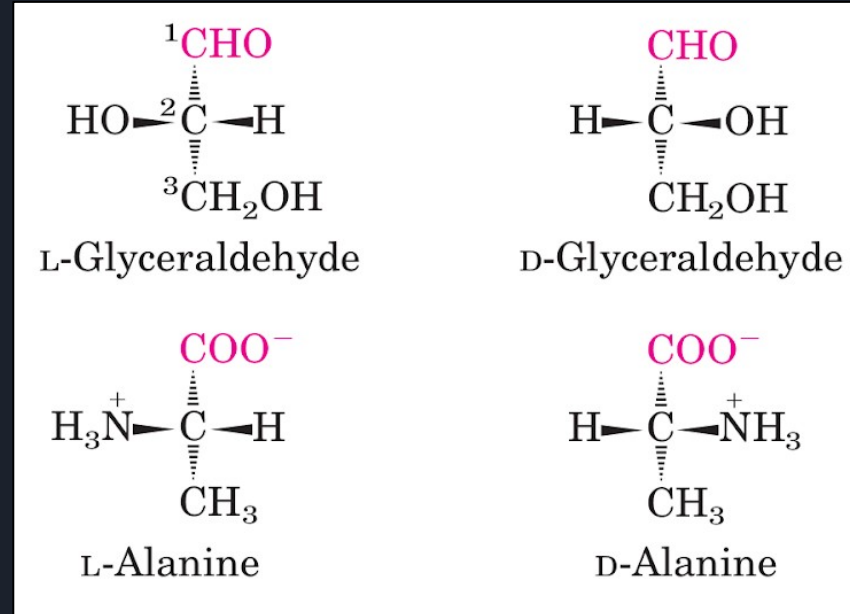


D-Alanine

# Isomeria

A maioria dos aminoácidos, com exceção à glicina, apresentam **isomeria óptica** e existem como um par de **enantiômeros**.

- Aminoácidos são L-enantiômeros nas proteínas.







# Nomenclatura

- 20 aminoácidos mais comuns nas proteínas.
- Nomeados por semelhança com os objetos de pesquisa.

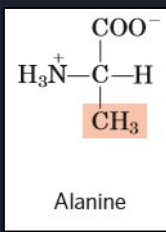
Trivial name <sup>a</sup>	Symbols <sup>b</sup>
Alanine	Ala A
Arginine	Arg R
Asparagine	Asn <sup>d</sup> N <sup>d</sup>
Aspartic acid	Asp <sup>d</sup> D <sup>d</sup>
Cysteine	Cys C
Glutamine	Gln <sup>d</sup> Q <sup>d</sup>
Glutamic acid	Glu <sup>d</sup> E <sup>d</sup>
Glycine	Gly G
Histidine	His H
Isoleucine	Ile I
Leucine	Leu L
Lysine	Lys K
Methionine	Met M
Phenylalanine	Phe F
Proline	Pro P
Serine	Ser S
Threonine	Thr T
Tryptophan	Trp W
Tyrosine	Tyr Y
Valine	Val V

# Nomenclatura

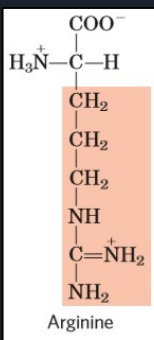
- 20 aminoácidos mais comuns nas proteínas.
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Trivial name <sup>a</sup>	Symbols <sup>b</sup>
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Arginine	Arg R
Asparagine	Asn <sup>d</sup> N <sup>d</sup>
Aspartic acid	Asp <sup>d</sup> D <sup>d</sup>
Cysteine	
Glutamine	
Glutamic acid	
Glycine	
Histidine	
Isoleucine	
Leucine	
Lysine	Lys K
Methionine	Met M
Phenylalanine	Phe F
Proline	Pro P
Serine	Ser S
Threonine	Thr T
Tryptophan	Trp W
Tyrosine	Tyr Y
Valine	Val V

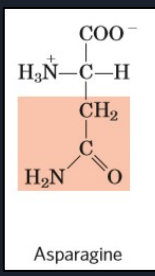




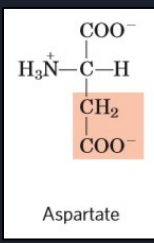
Alanine



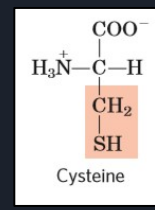
Arginine



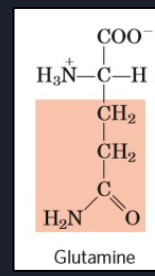
Asparagine



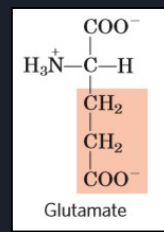
Aspartate



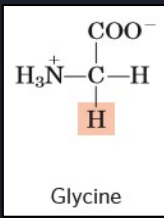
Cysteine



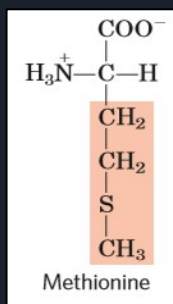
Glutamine



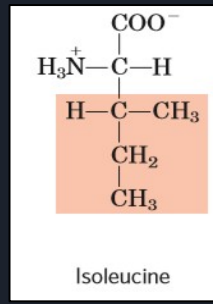
Glutamate



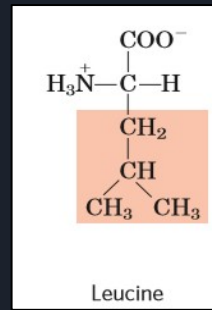
Glycine



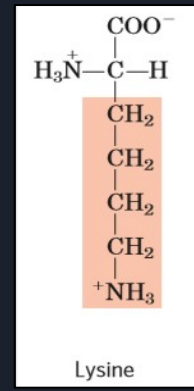
Methionine



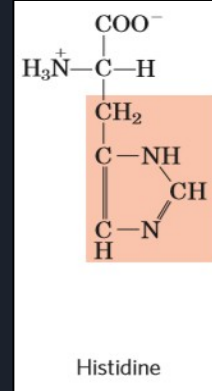
Isoleucine



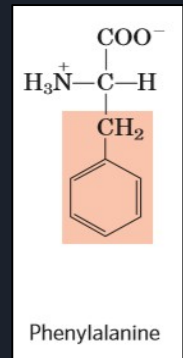
Leucine



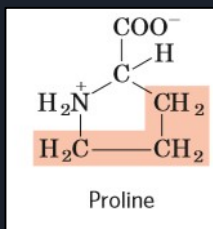
Lysine



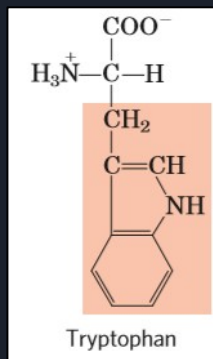
Histidine



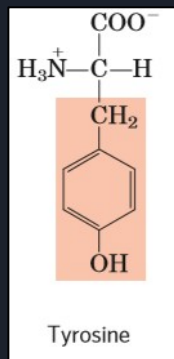
Phenylalanine



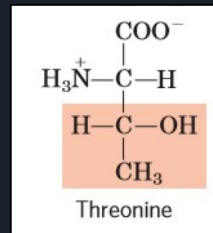
Proline



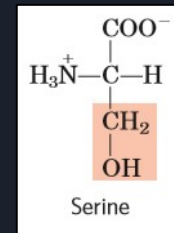
Tryptophan



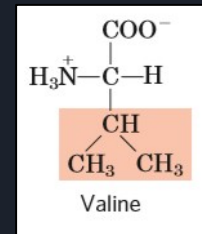
Tyrosine



Threonine

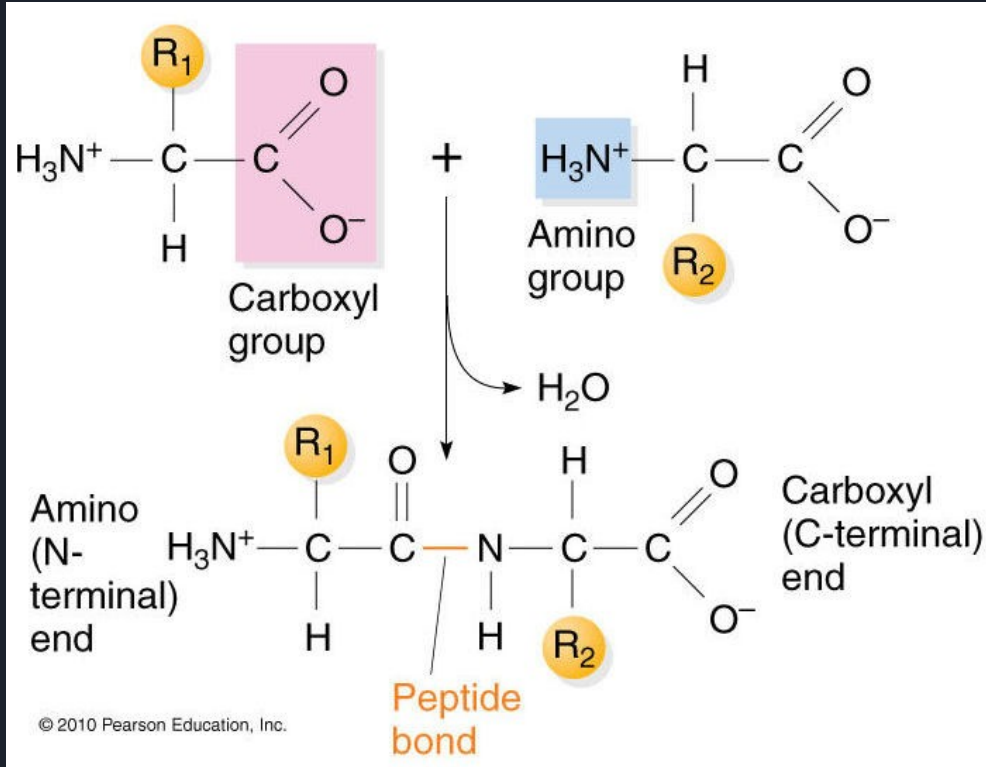


Serine

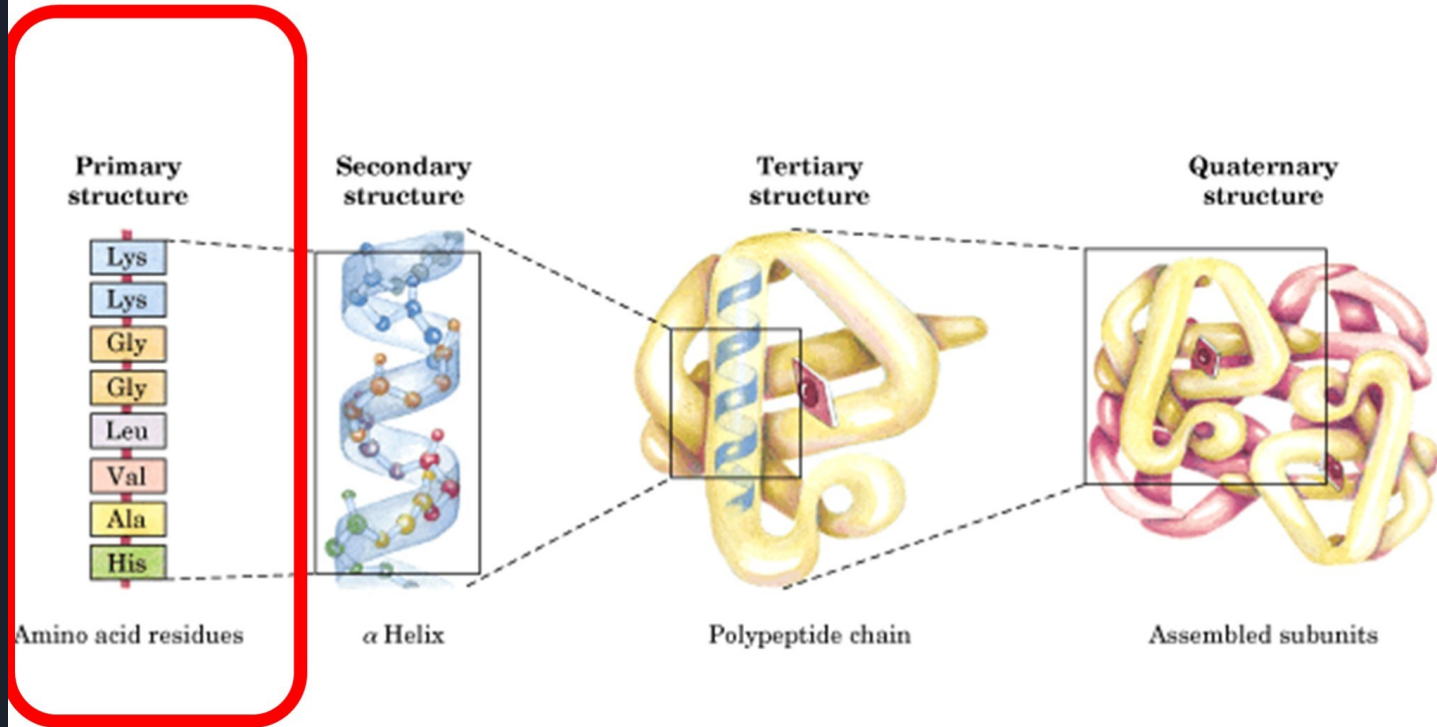


Valine

# Por que caracterizar as cadeias laterais?



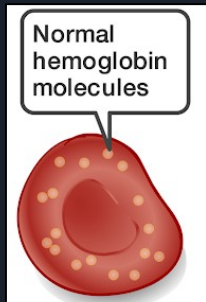
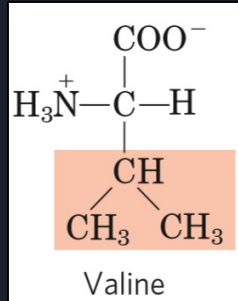
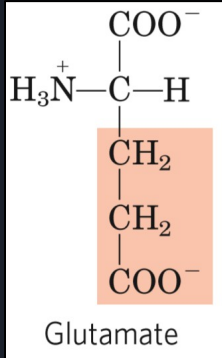
# Proteínas: polímeros de aminoácidos



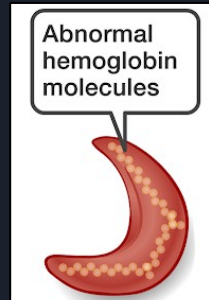
# Por que caracterizar as cadeias laterais?

Proteína **GLU** Proteína normal

**VAL** Proteína mutante



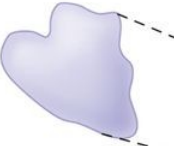
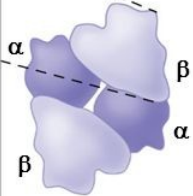
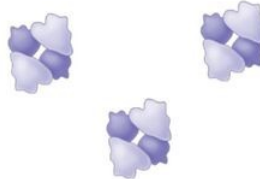
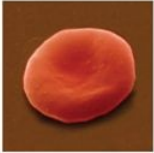
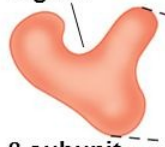
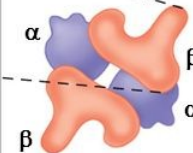
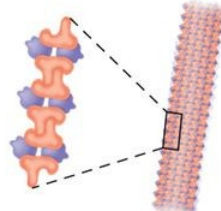

Eritrócito



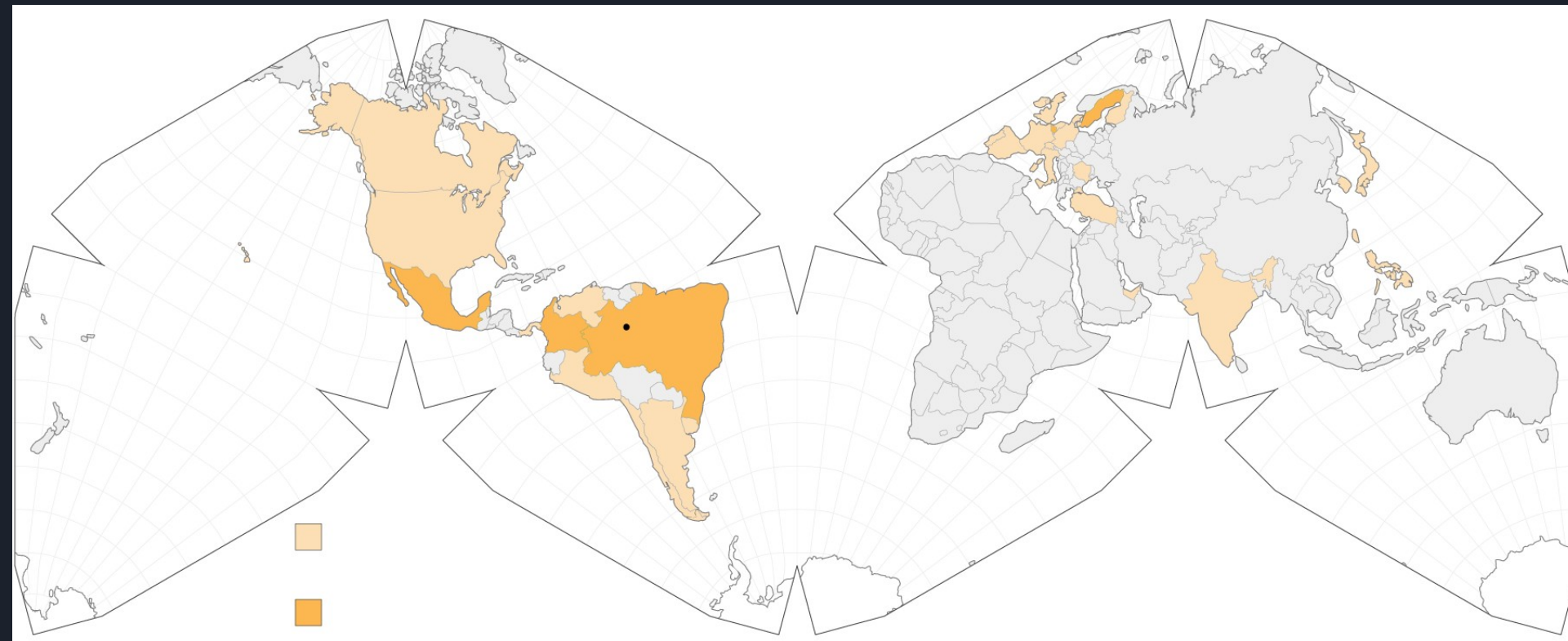
## Anemia Falciforme

- Falta de oxigenação nos tecidos;
- Tendência a infecções;
- Problemas neurológicos, cardiovasculares, pulmonares e renais;

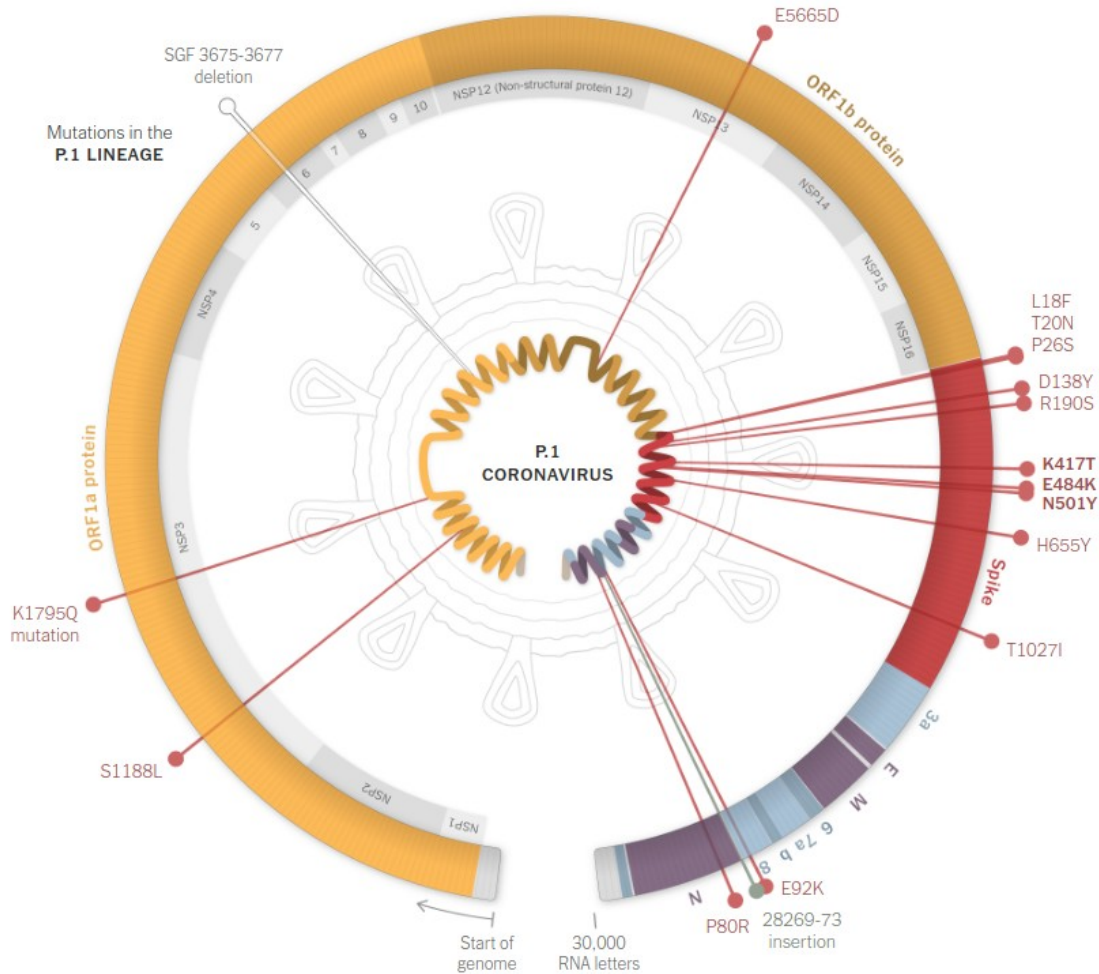
# Por que caracterizar as cadeias laterais?

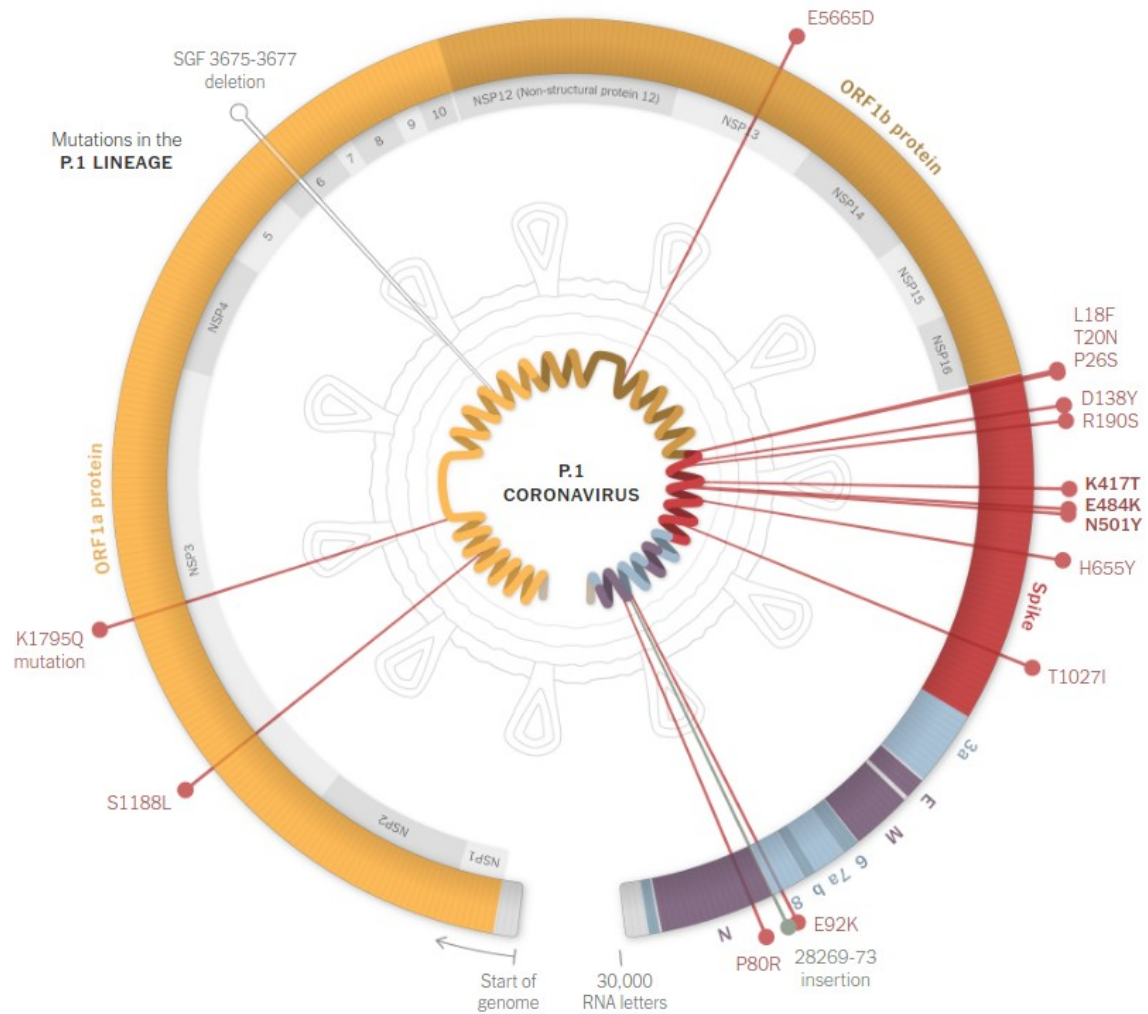
	Primary Structure	Secondary and Tertiary Structures	Quaternary Structure	Function	Red Blood Cell Shape
Normal hemoglobin	1 Val 2 His 3 Leu 4 Thr 5 Pro 6 Glu 7 Glu	 <p><math>\beta</math> subunit</p>	 <p>Normal hemoglobin</p>	<p>Molecules do not associate with one another; each carries oxygen.</p> 	 <p>10 <math>\mu</math>m</p>
Sickle-cell hemoglobin	1 Val 2 His 3 Leu 4 Thr 5 Pro 6 Val 7 Glu	 <p>Exposed hydrophobic region</p> <p><math>\beta</math> subunit</p>	 <p>Sickle-cell hemoglobin</p>	<p>Molecules crystallize into a fiber; capacity to carry oxygen is reduced.</p> 	 <p>10 <math>\mu</math>m</p>

# Distribuição mundial da variante P1 SARS-Cov2

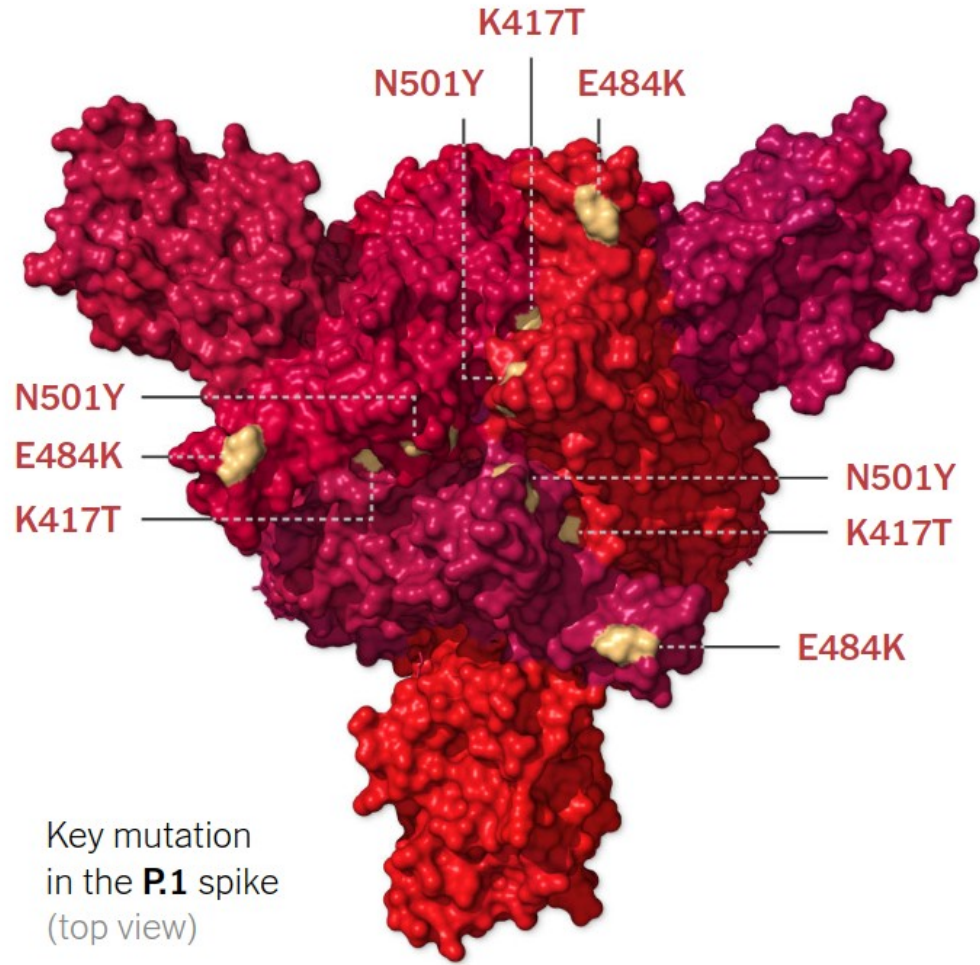








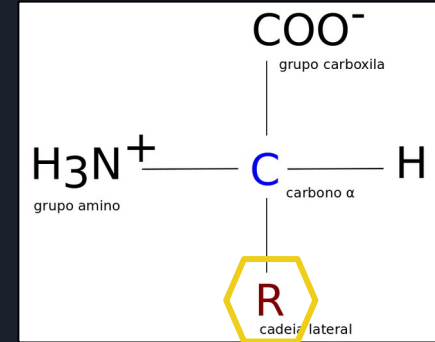
Trivial name <sup>a</sup>	Symbols <sup>b</sup>
Alanine	Ala A
Arginine	Arg R
Asparagine	Asn <sup>d</sup> N <sup>d</sup>
Aspartic acid	Asp <sup>d</sup> D <sup>d</sup>
Cysteine	Cys C
Glutamine	Gln <sup>d</sup> Q <sup>d</sup>
Glutamic acid	Glu <sup>d</sup> E <sup>d</sup>
Glycine	Gly G
Histidine	His H
Isoleucine	Ile I
Leucine	Leu L
Lysine	Lys K
Methionine	Met M
Phenylalanine	Phe F
Proline	Pro P
Serine	Ser S
Threonine	Thr T
Tryptophan	Trp W
Tyrosine	Tyr Y
Valine	Val V



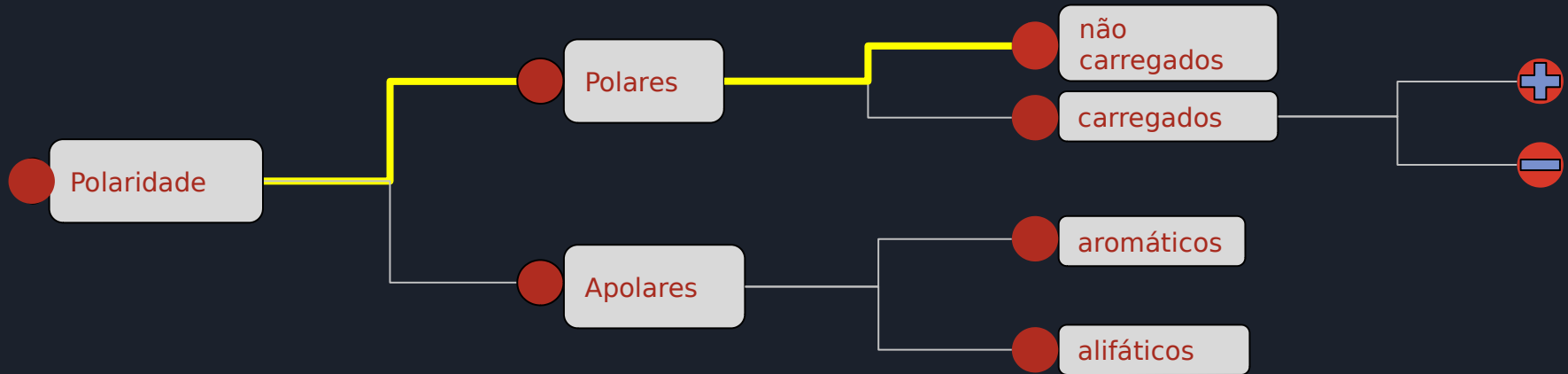
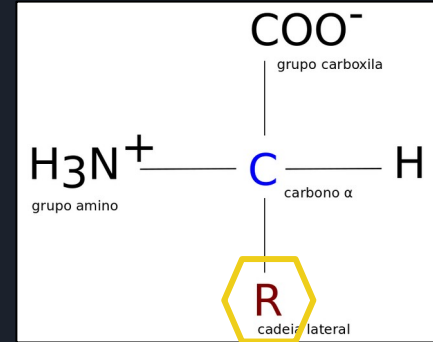
- N501Y, which helps the virus latch on more tightly to human cells. This mutation also appears in the B.1.1.7 and B.1.351 lineages.
- K417T, which is the same site as the K417N mutation in the B.1.351 lineage. It may also help the virus latch on tighter.
- E484K, which may help the virus evade some kinds of antibodies.

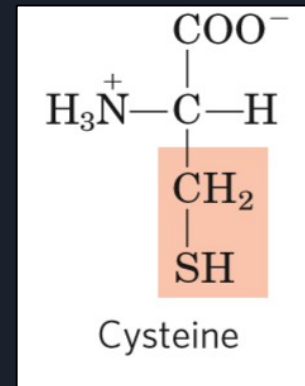
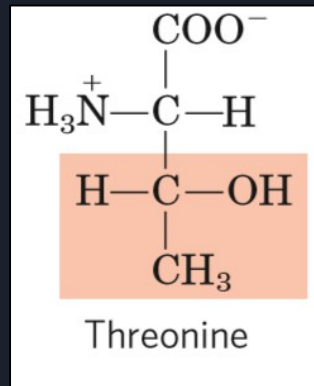
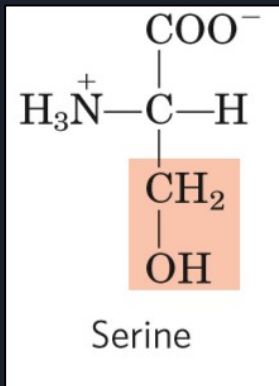
<https://www.nytimes.com/interactive/2021/health/coronavirus-variant-tracker.html#P1>

# Classificação dos aminoácidos



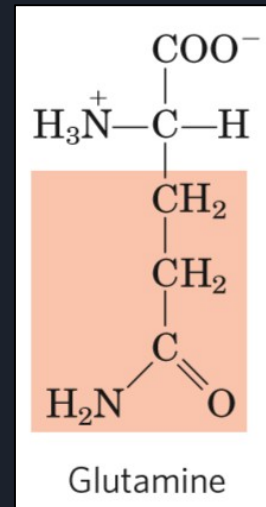
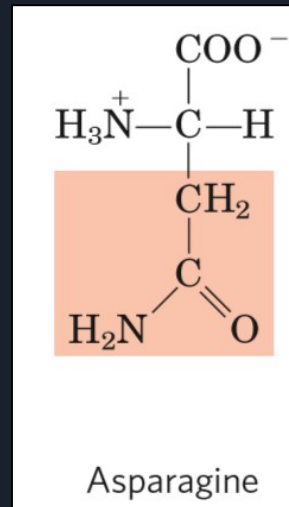
# Classificação dos aminoácidos



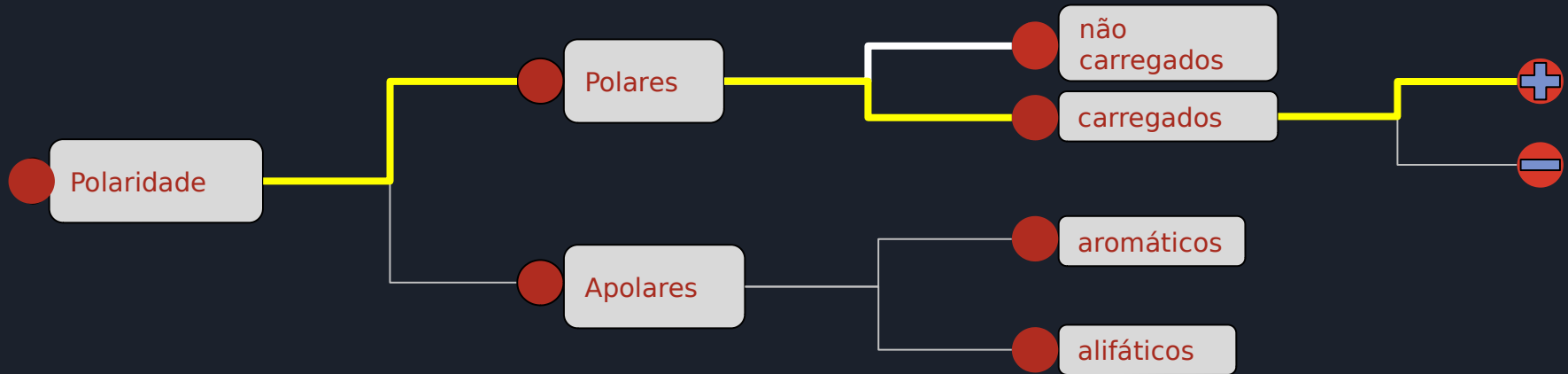
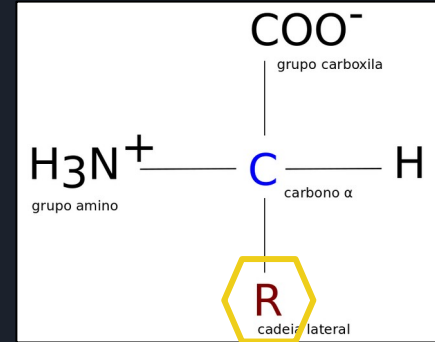


## Polares não carregados

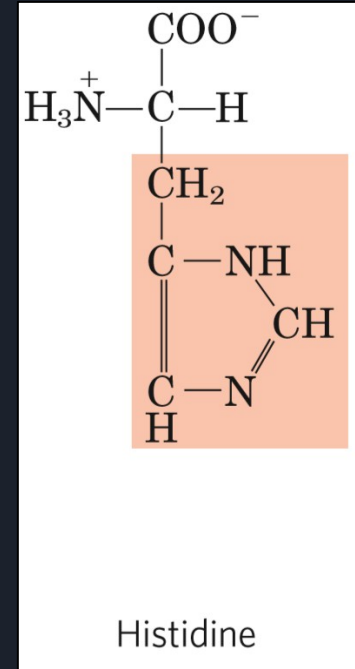
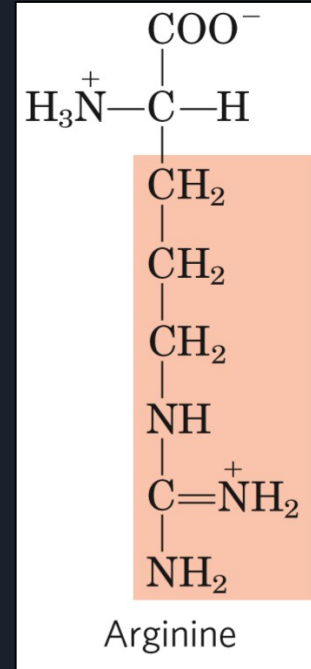
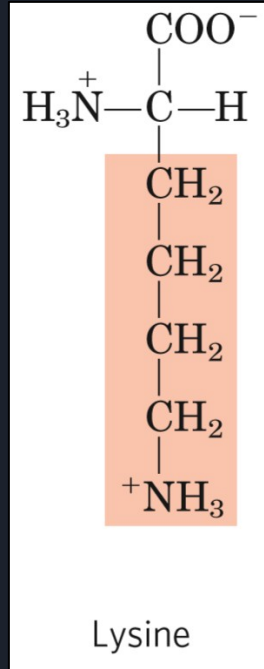
Hidrofílicos, capazes de formar ligações de hidrogênio com a água.



# 4. Classificação dos aminoácidos



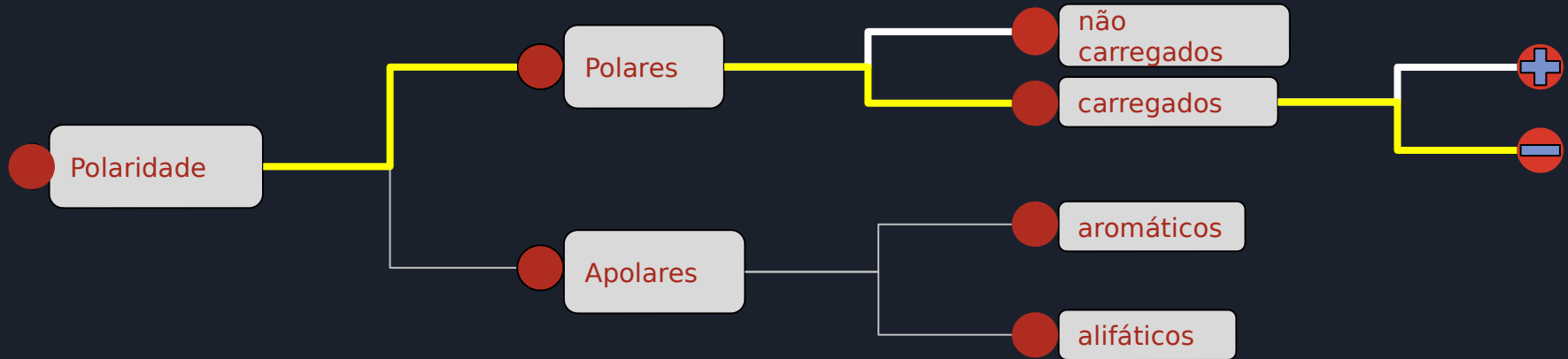
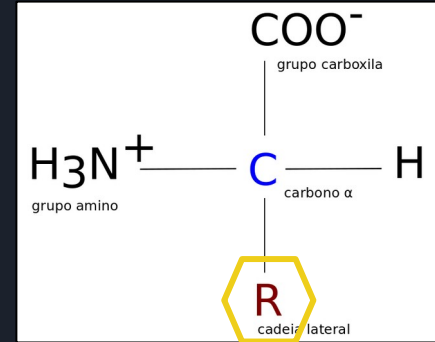
# Polares carregados positivamente



Mais hidrofílicos que os polares não carregados.

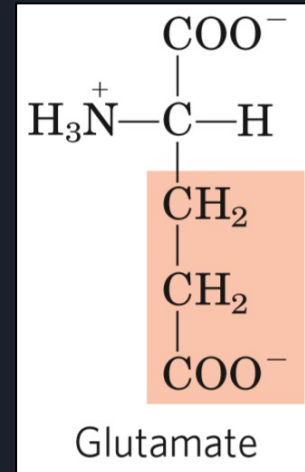
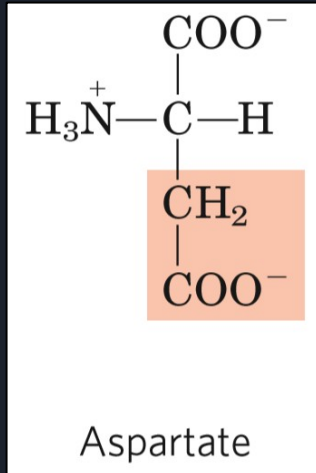


# 4. Classificação dos aminoácidos

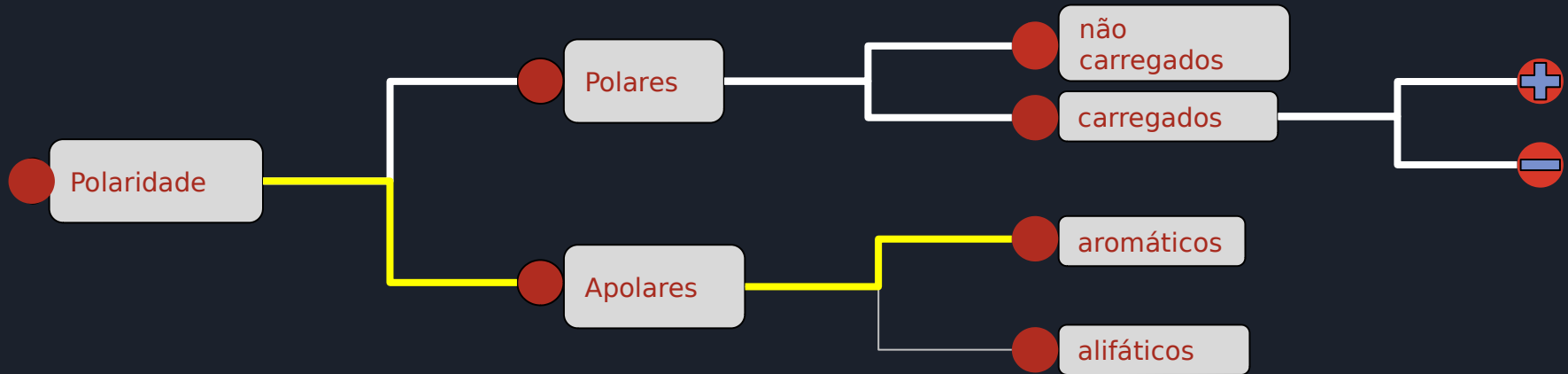
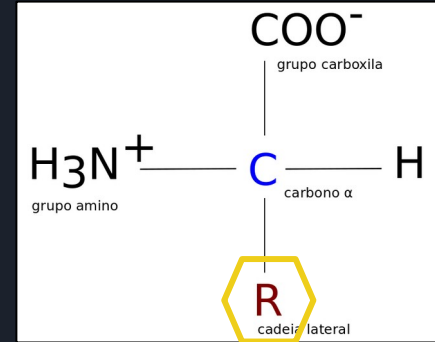


# Polares carregados negativamente

Hidrofilicos.



# 4. Classificação dos aminoácidos

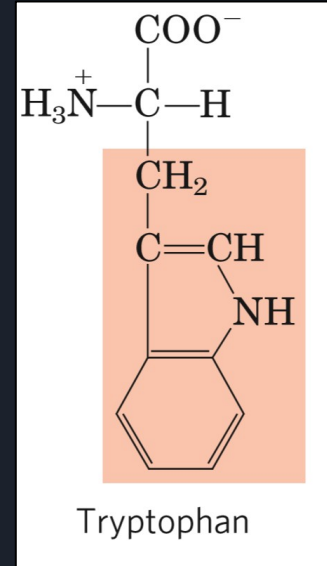
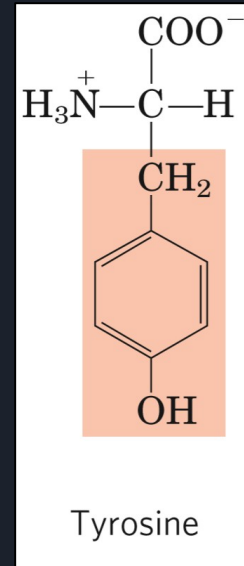
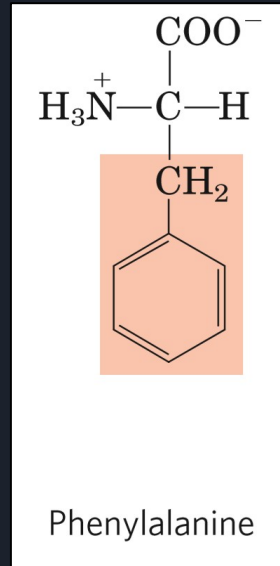


# Apolares aromáticos

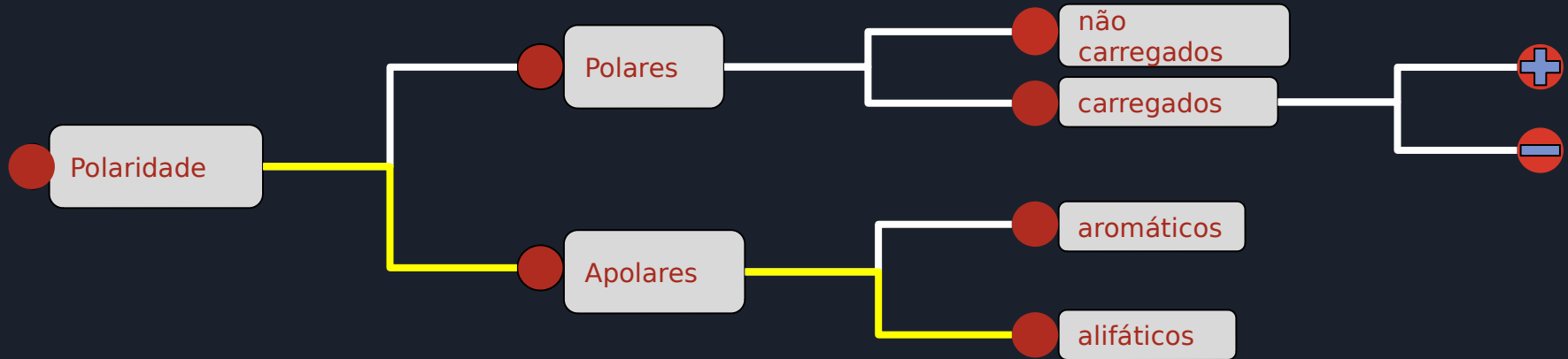
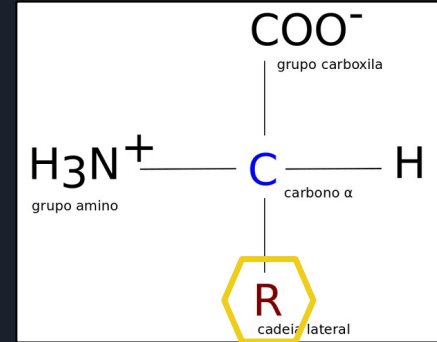
Participam de interações hidrofóbicas.

Tyr e Trp mais polares que Phe.

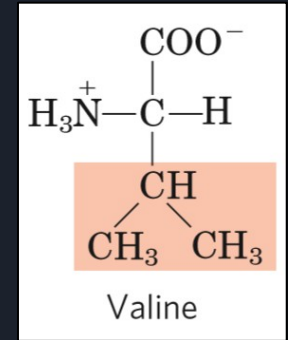
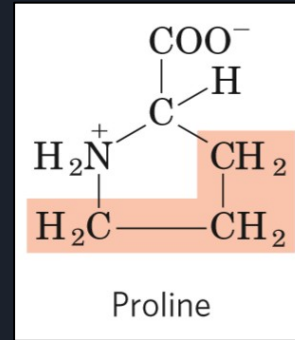
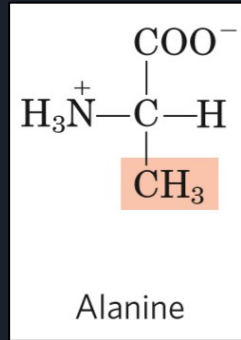
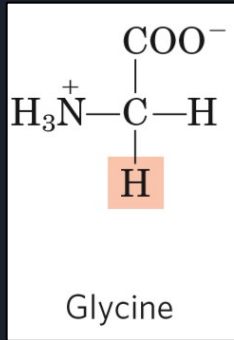
Podem absorver luz UV.



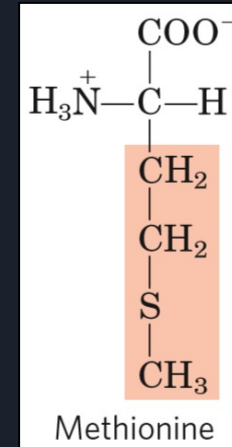
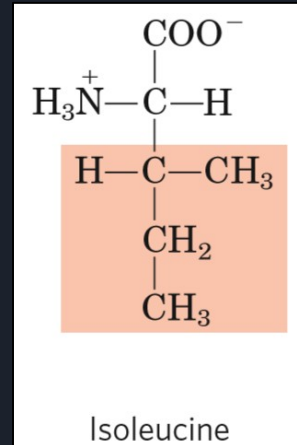
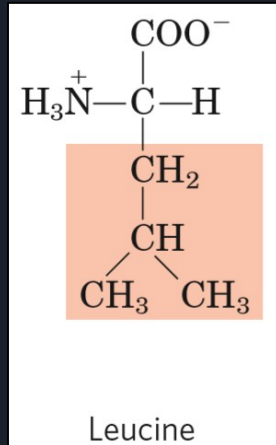
# Classificação dos aminoácidos



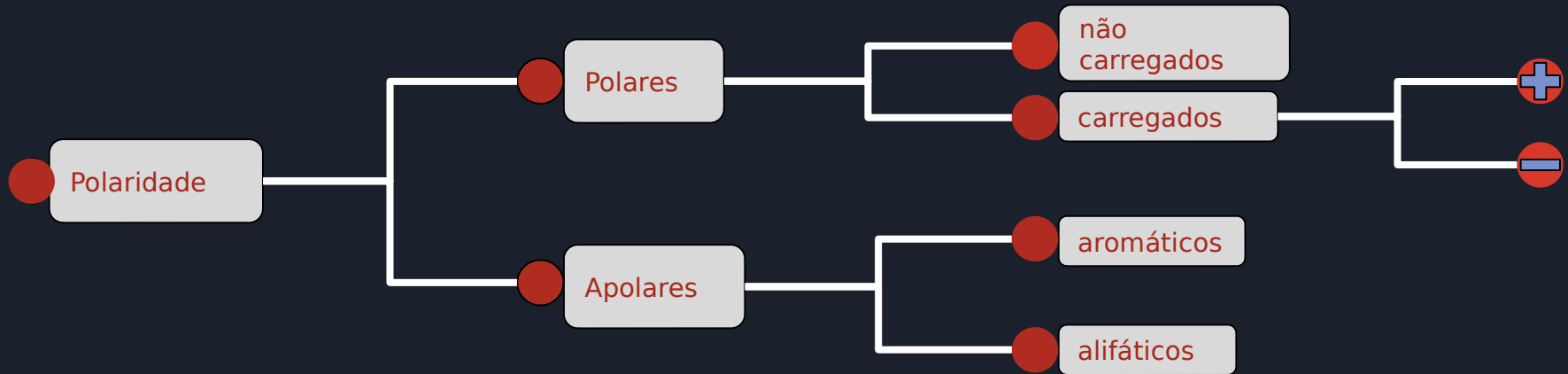
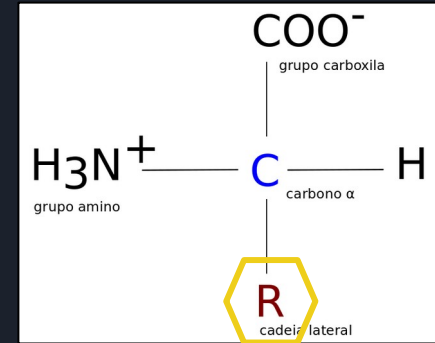
# Apolares alifáticos



Participam de interações hidrofóbicas.

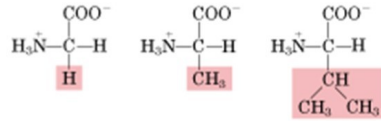


# Classificação dos aminoácidos



# Os 20 aminoácidos

## Nonpolar, aliphatic R groups



Glycine

Alanine

Valine

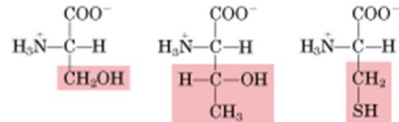


Leucine

Methionine

Isoleucine

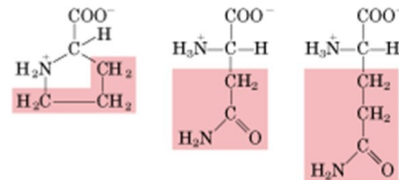
## Polar, uncharged R groups



Serine

Threonine

Cysteine

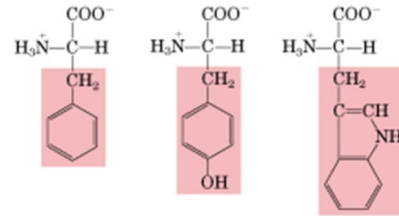


Proline

Asparagine

Glutamine

## Aromatic R groups

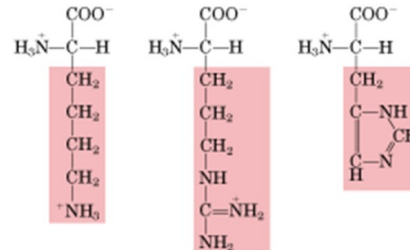


Phenylalanine

Tyrosine

Tryptophan

## Positively charged R groups

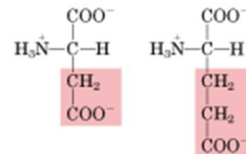


Lysine

Arginine

Histidine

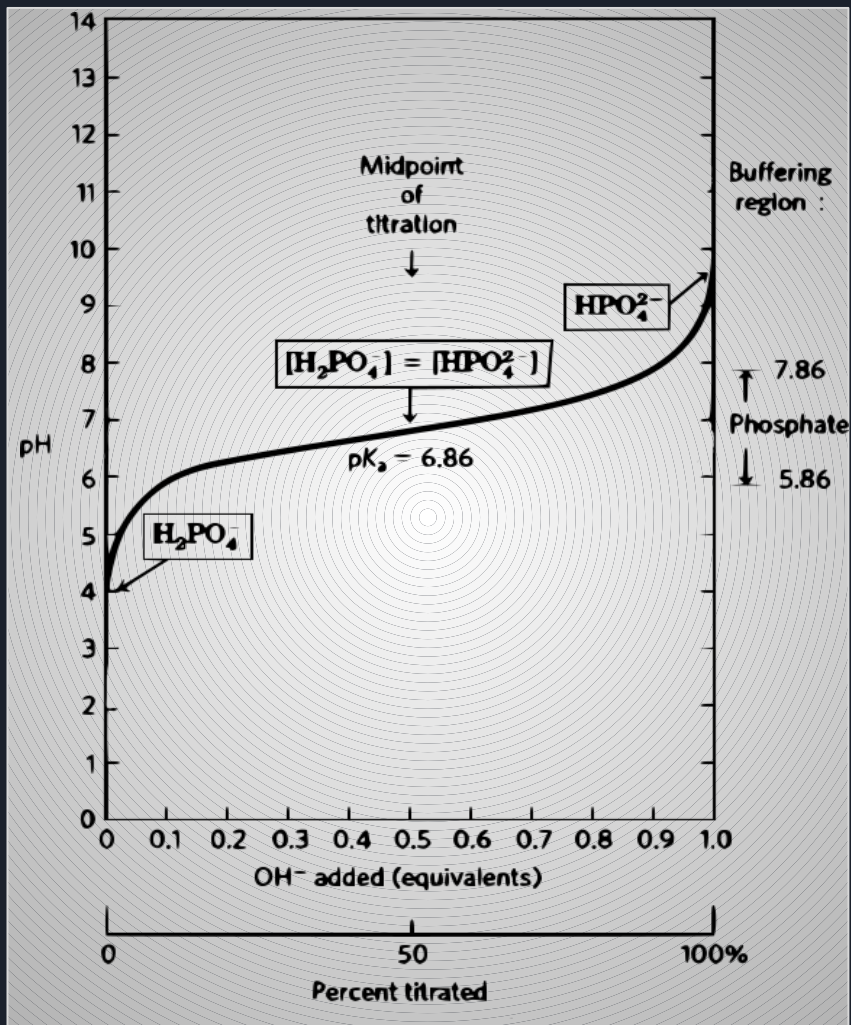
## Negatively charged R groups



Aspartate

Glutamate



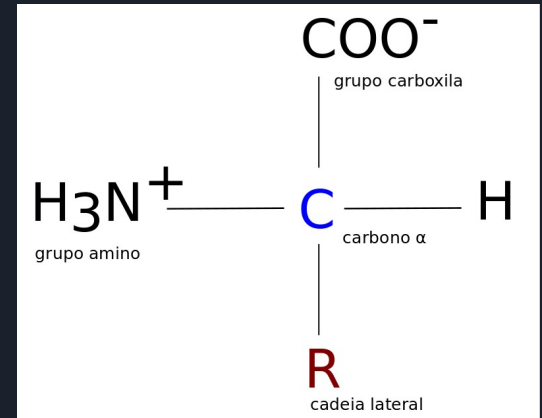


$$\text{pH} = \text{p}K_a + \log \frac{[\text{A}^-]}{[\text{HA}]}$$

Equação de Henderson-Hasselbalch

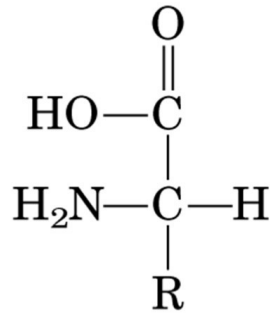
# Ionização dos aminoácidos

- Aminoácidos podem ter carga positiva, negativa ou neutra.
- Zwitterions - ácido ou base.
- pKa - constante de dissociação e faixa de tamponamento.

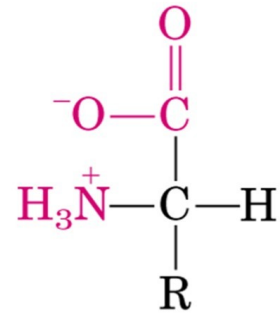


aa não carregado em pH  
 próx. 7.

# Ionização de aminoácidos



Nonionic  
form

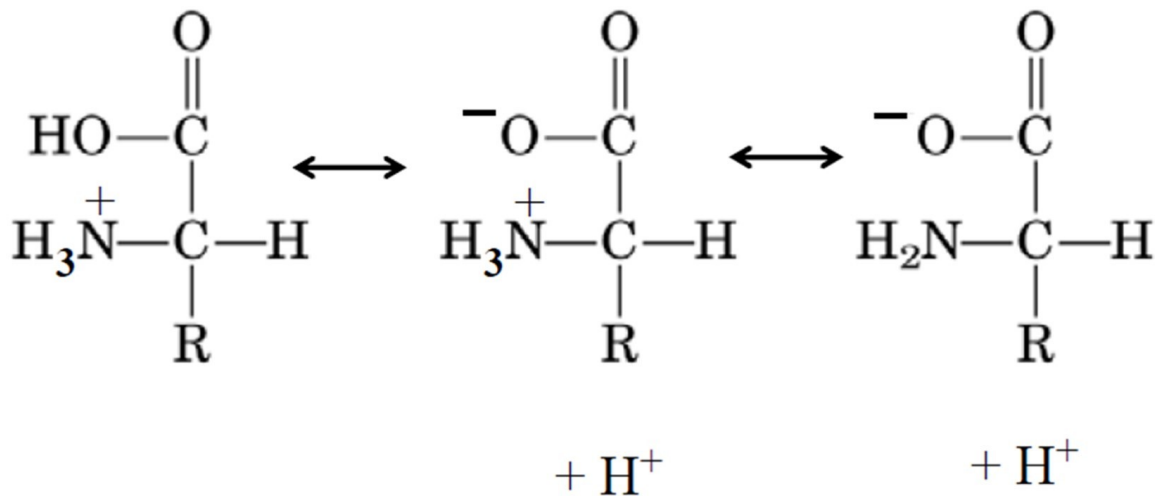


Zwitterionic  
form

**Forma anfótera ou  
zwitteriônica**

## Aminoácidos podem funcionar como ácidos ou bases

**zwitterion**



Carga líquida em cada uma das formas?

ácido

básico

pH



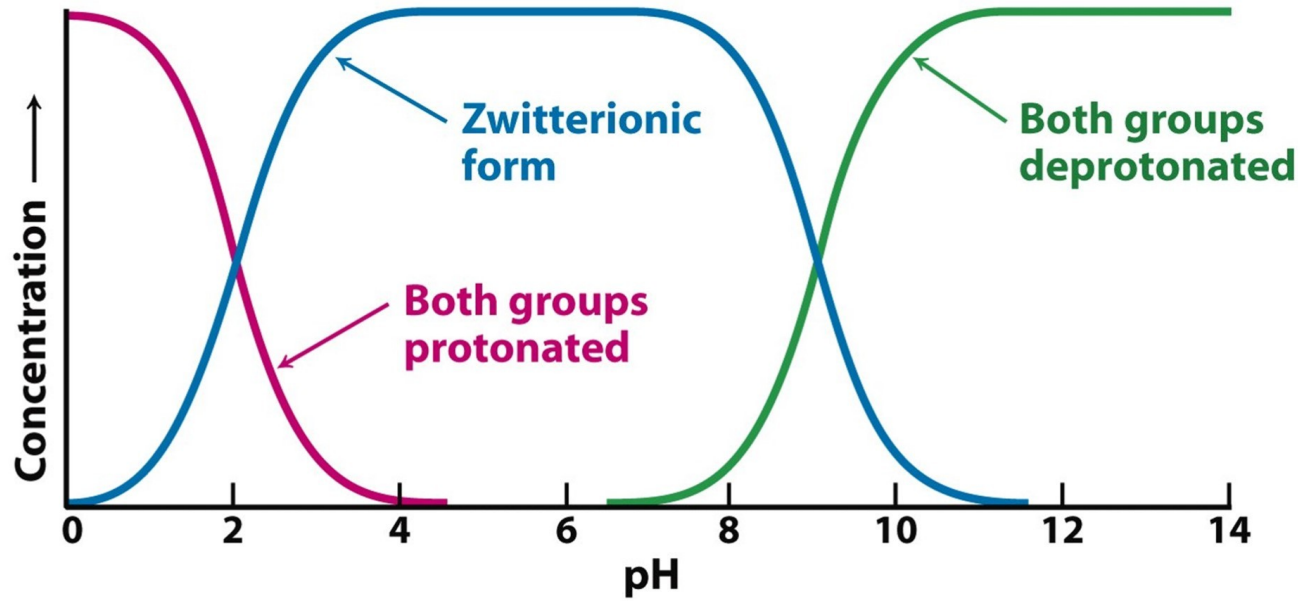
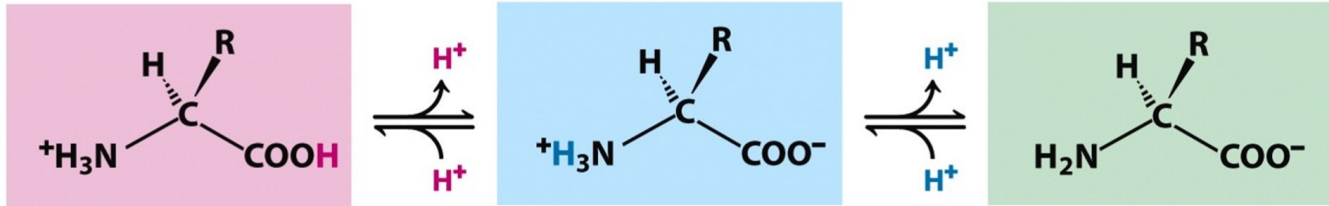
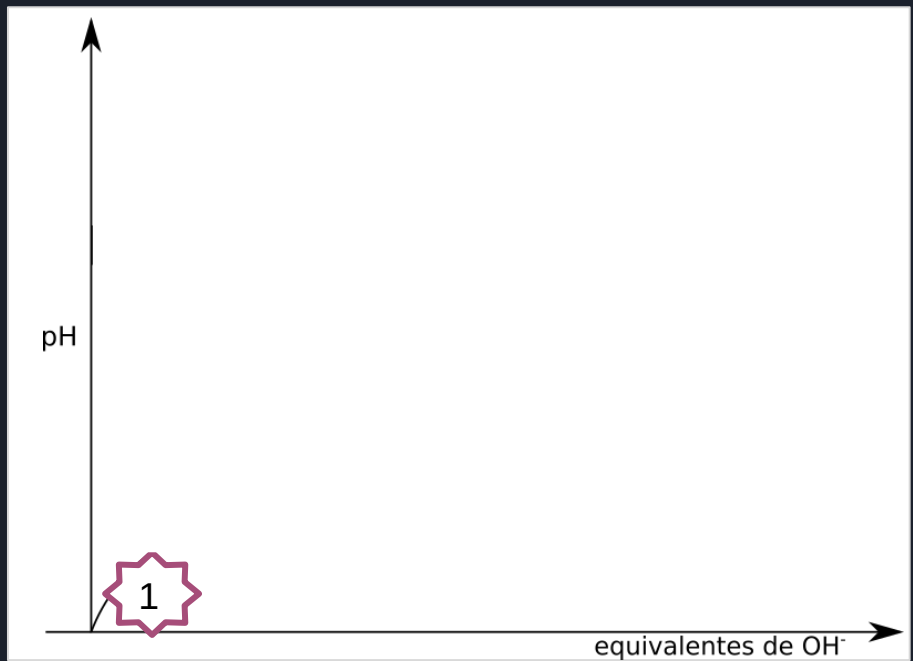
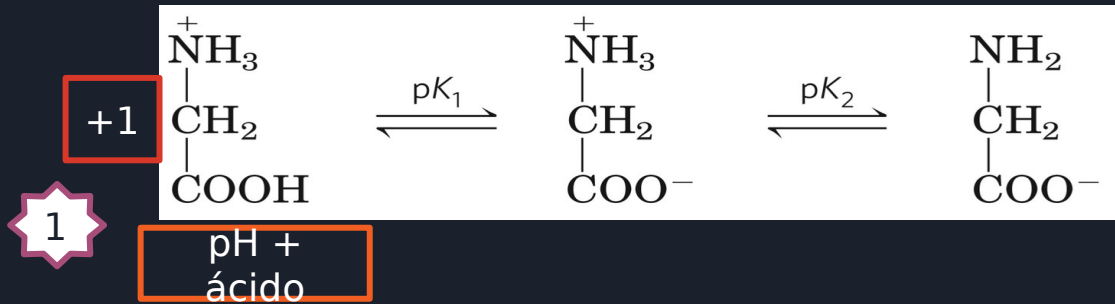
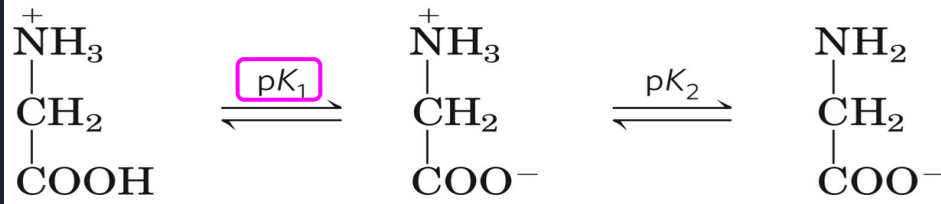


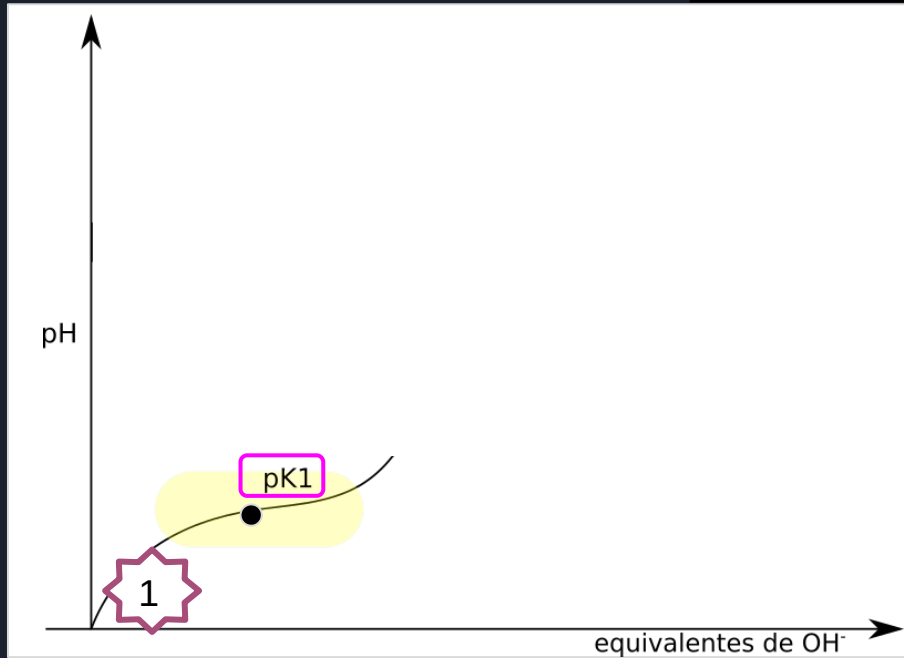
Figure 2-6  
*Biochemistry, Sixth Edition*  
 © 2007 W.H. Freeman and Company



# Ionização da Glicina

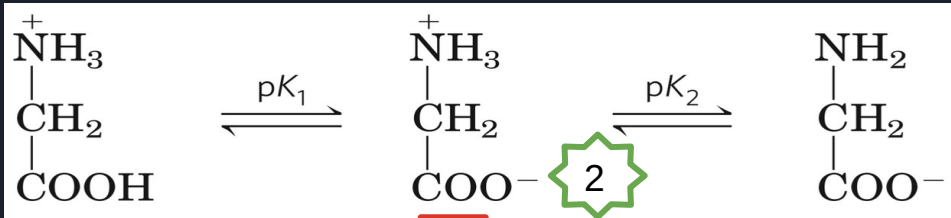


1



## Ionização da Glicina

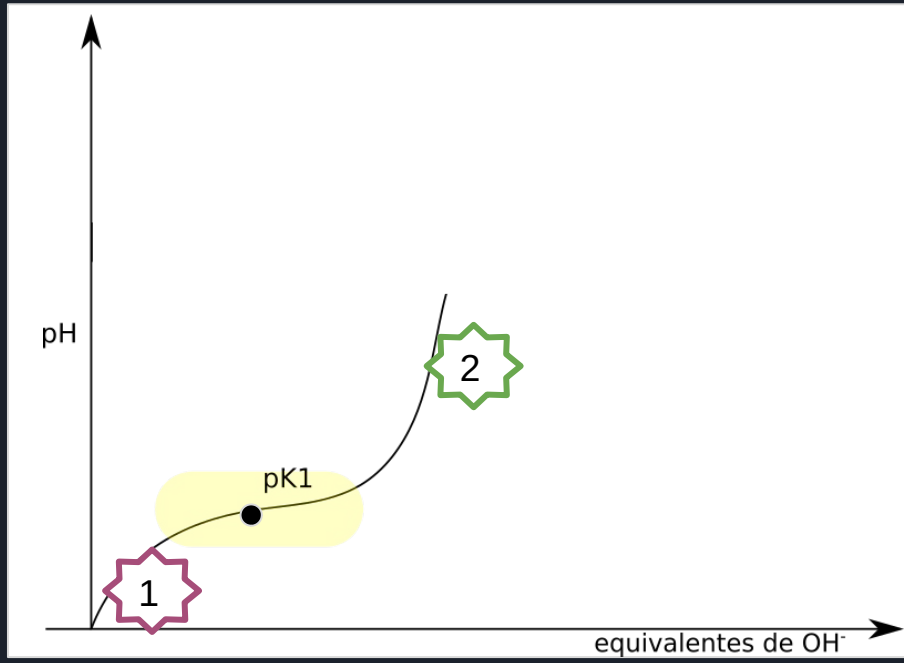
Faixas de tamponamento:  
 1. De  $\text{p}K_1 - 1$  até  $\text{p}K_1 + 1$



1

0

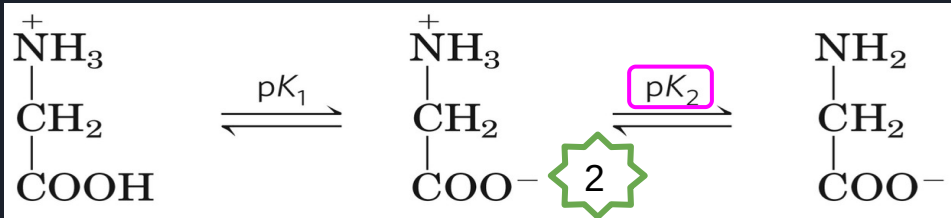
pH  
neutro



## Ionização da Glicina

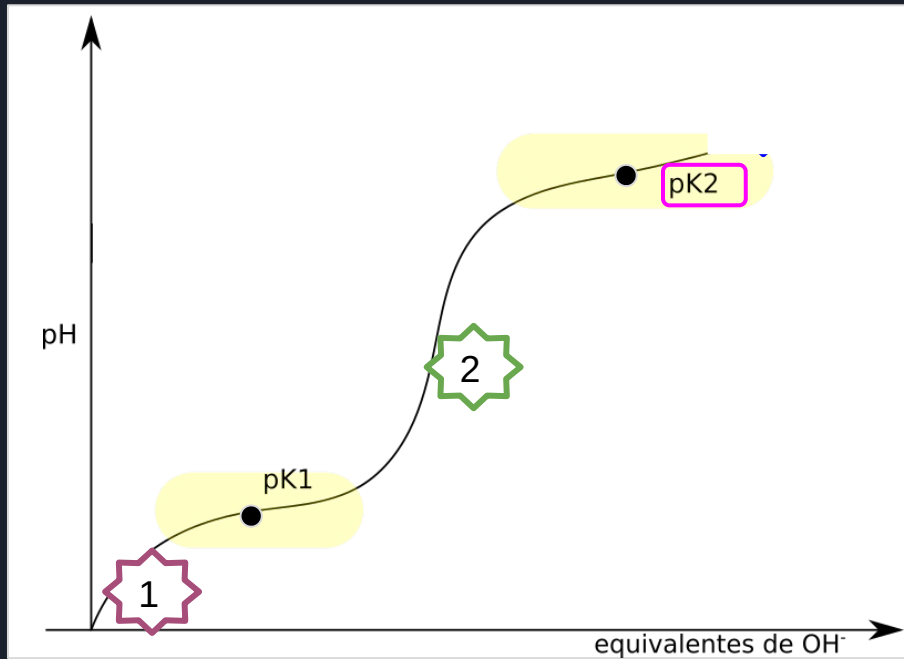
- Faixas de tamponamento:
1. De  $\text{p}K_1 - 1$  até  $\text{p}K_1 + 1$





1

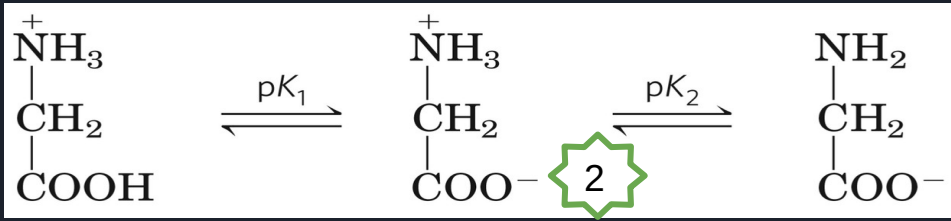
2



## Ionização da Glicina

Faixas de tamponamento:

1. De  $\text{p}K_1 - 1$  até  $\text{p}K_1 + 1$
2. De  $\text{p}K_2 - 1$  até  $\text{p}K_2 + 1$



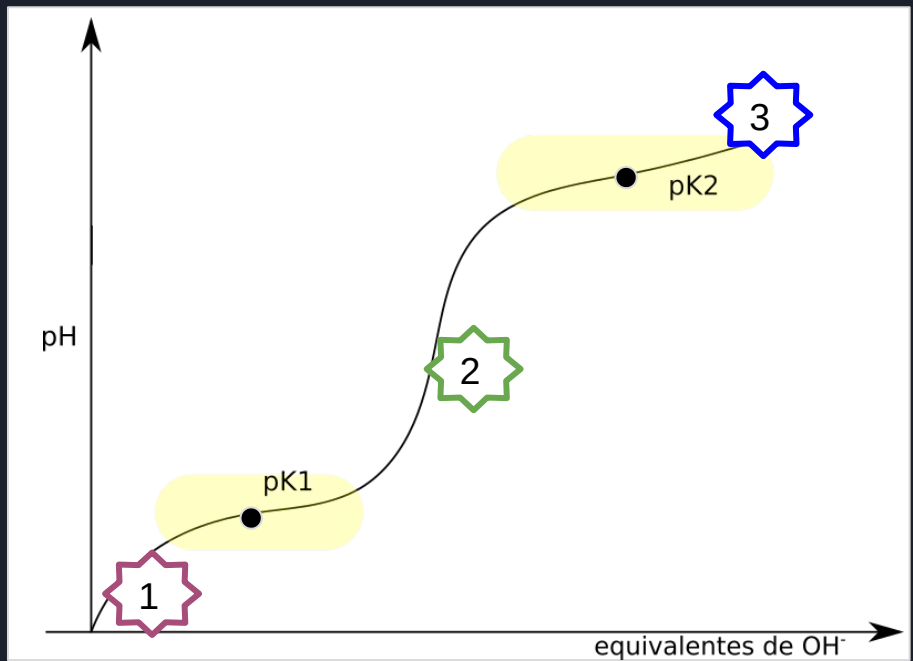
1

2

-1

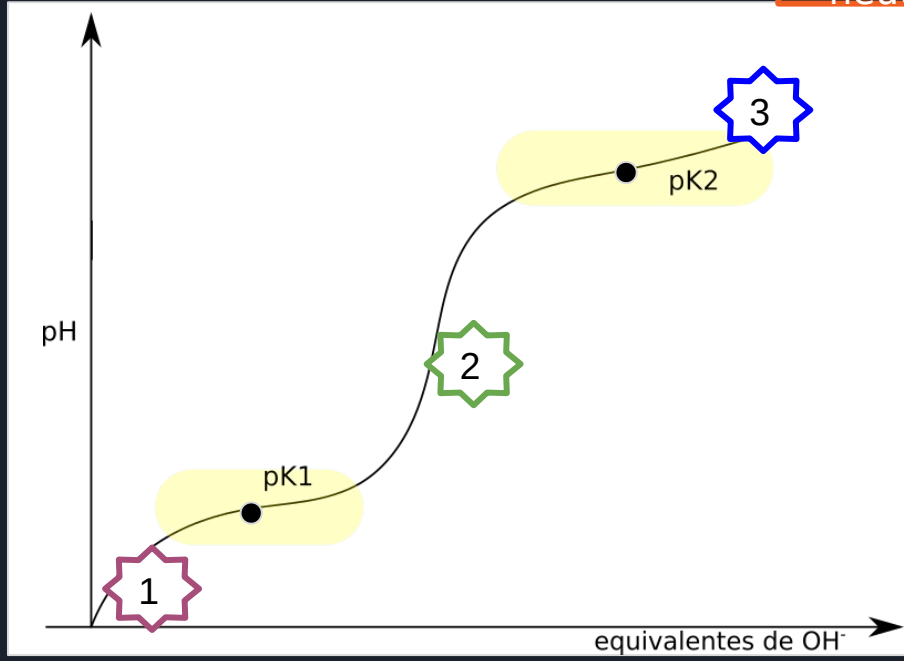
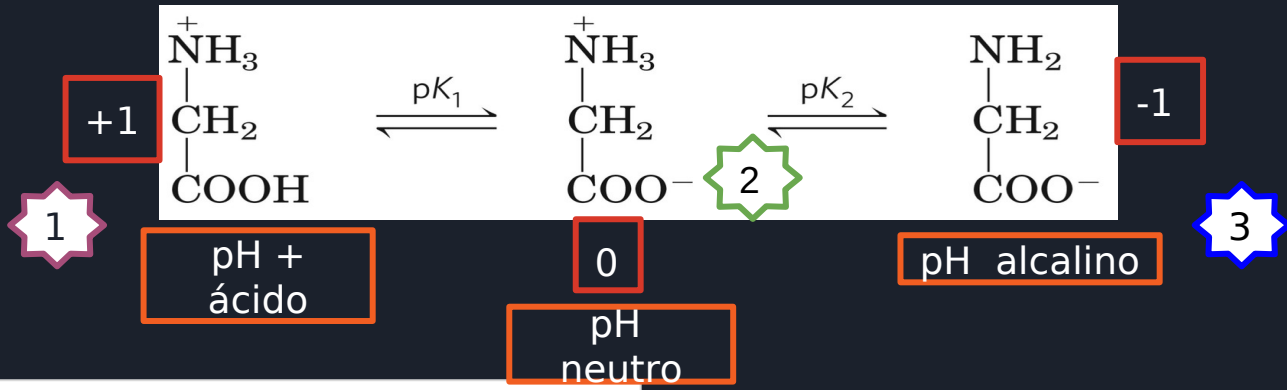
3

pH alcalino



## Ionização da Glicina

- Faixas de tamponamento:
1. De  $\text{p}K_1 - 1$  até  $\text{p}K_1 + 1$
  2. De  $\text{p}K_2 - 1$  até  $\text{p}K_2 + 1$



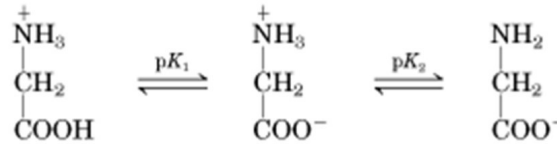
## Ionização da Glicina

Faixas de tamponamento:

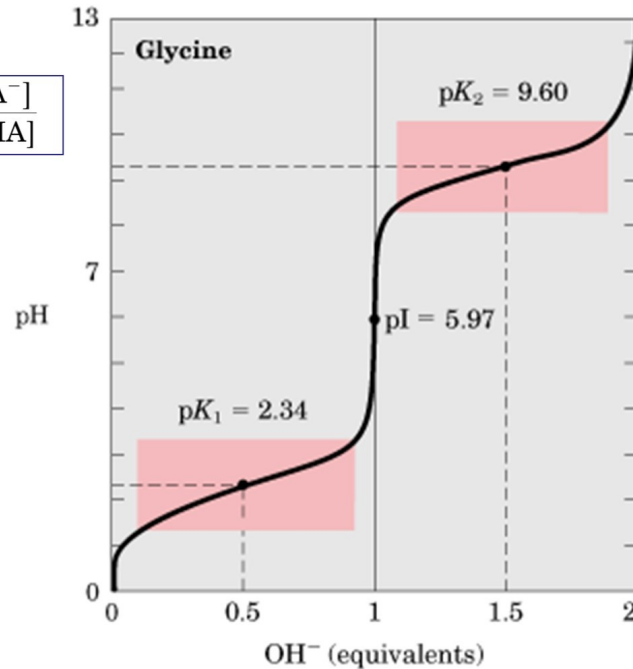
1. De  $pK_1 - 1$  até  $pK_1 + 1 \Rightarrow 1$  a  $3$
2. De  $pK_2 - 1$  até  $pK_2 + 1 \Rightarrow 8$  a  $10$

$pK_a$  amino  $\cong 9$   
 $pK_a$  ácido carboxílico  $\cong 2$

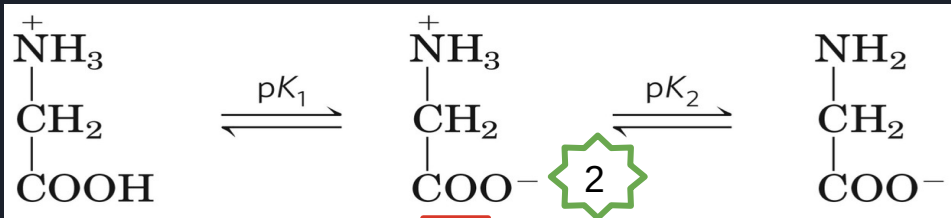
# Titulação de aminoácidos



$$\text{pH} = \text{p}K_a + \log \frac{[\text{A}^-]}{[\text{HA}]}$$



**pl = ponto isoelétrico**  
pH em que a carga líquida das moléculas de uma solução é igual a zero

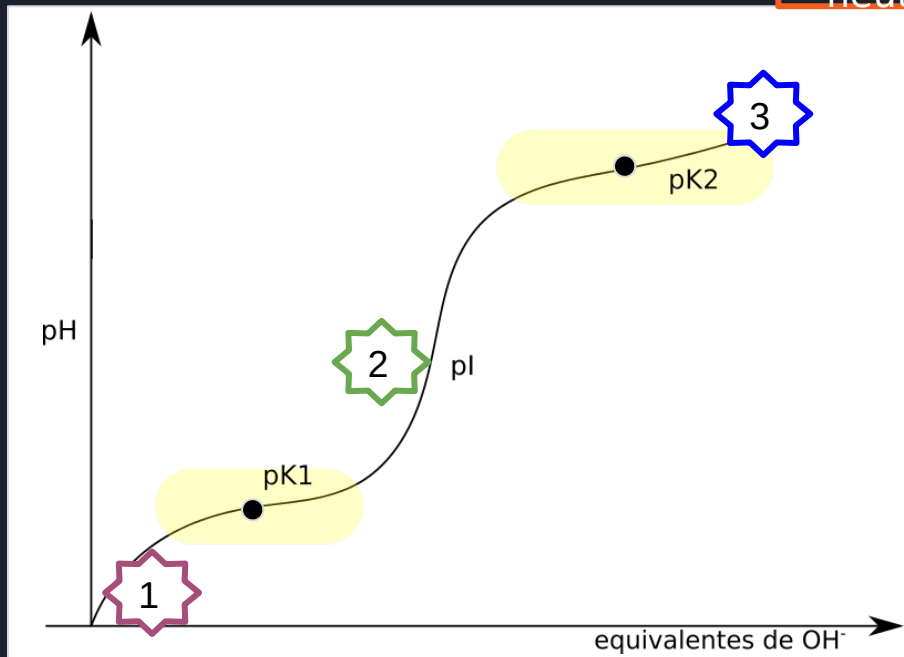


1

0

3

pH neutro



## Ponto Isoelétrico

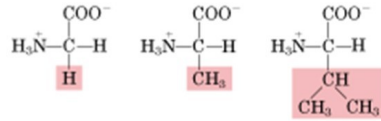
$$pI = \frac{pK_1 + pK_2}{2}$$

Média dos valores de pk que contribuem para a formação da molécula com soma de cargas igual a 0.

pka amino  $\cong$  9  
pka ácido carboxílico  $\cong$  2

# Os 20 aminoácidos

## Nonpolar, aliphatic R groups



Glycine

Alanine

Valine

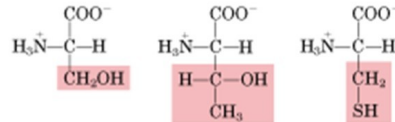


Leucine

Methionine

Isoleucine

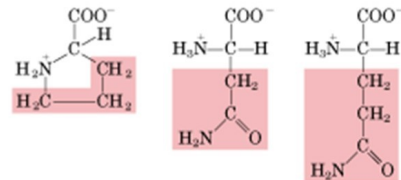
## Polar, uncharged R groups



Serine

Threonine

Cysteine

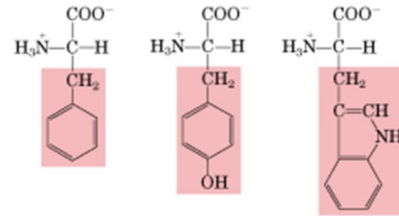


Proline

Asparagine

Glutamine

## Aromatic R groups

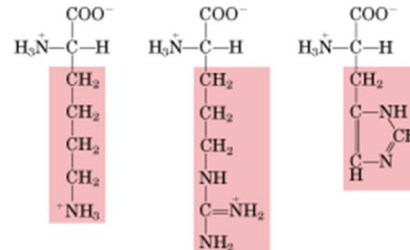


Phenylalanine

Tyrosine

Tryptophan

## Positively charged R groups

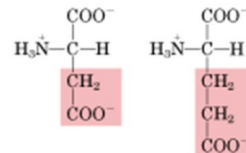


Lysine

Arginine

Histidine

## Negatively charged R groups



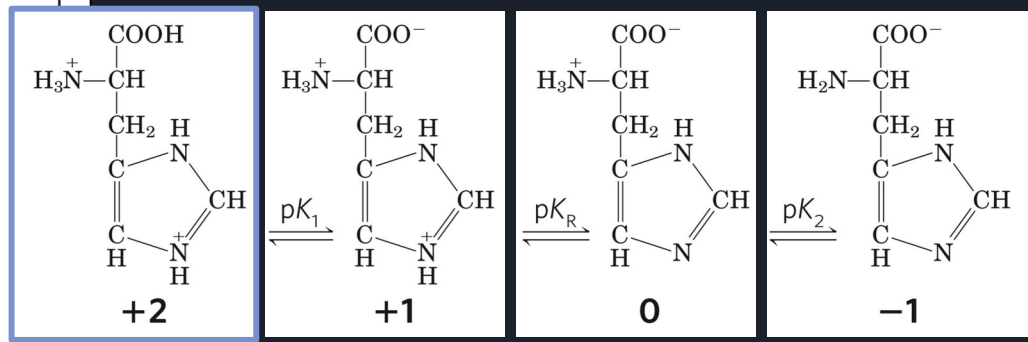
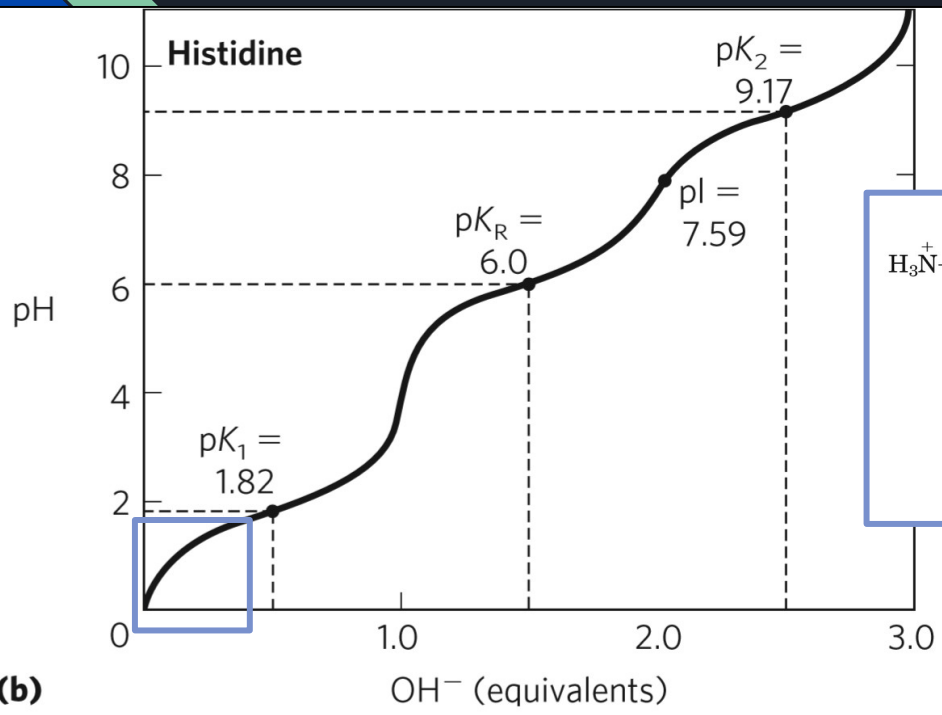
Aspartate

Glutamate

Amino acid	Abbreviation/ symbol	$M_r^*$	$pK_a$ values			pI
			$pK_1$ (—COOH)	$pK_2$ (—NH <sub>3</sub> <sup>+</sup> )	$pK_R$ (R group)	
<b>Nonpolar, aliphatic R groups</b>						
Glycine	Gly G	75	2.34	9.60		5.97
Alanine	Ala A	89	2.34	9.69		6.01
Proline	Pro P	115	1.99	10.96		6.48
Valine	Val V	117	2.32	9.62		5.97
Leucine	Leu L	131	2.36	9.60		5.98
Isoleucine	Ile I	131	2.36	9.68		6.02
Methionine	Met M	149	2.28	9.21		5.74
<b>Aromatic R groups</b>						
Phenylalanine	Phe F	165	1.83	9.13		5.48
Tyrosine	Tyr Y	181	2.20	9.11	10.07	5.66
Tryptophan	Trp W	204	2.38	9.39		5.89
<b>Polar, uncharged R groups</b>						
Serine	Ser S	105	2.21	9.15		5.68
Threonine	Thr T	119	2.11	9.62		5.87
Cysteine <sup>†</sup>	Cys C	121	1.96	10.28	8.18	5.07
Asparagine	Asn N	132	2.02	8.80		5.41
Glutamine	Gln Q	146	2.17	9.13		5.65
<b>Positively charged R groups</b>						
Lysine	Lys K	146	2.18	8.95	10.53	9.74
Histidine	His H	155	1.82	9.17	6.00	7.59
Arginine	Arg R	174	2.17	9.04	12.48	10.76
<b>Negatively charged R groups</b>						
Aspartate	Asp D	133	1.88	9.60	3.65	2.77
Glutamate	Glu E	147	2.19	9.67	4.25	3.22

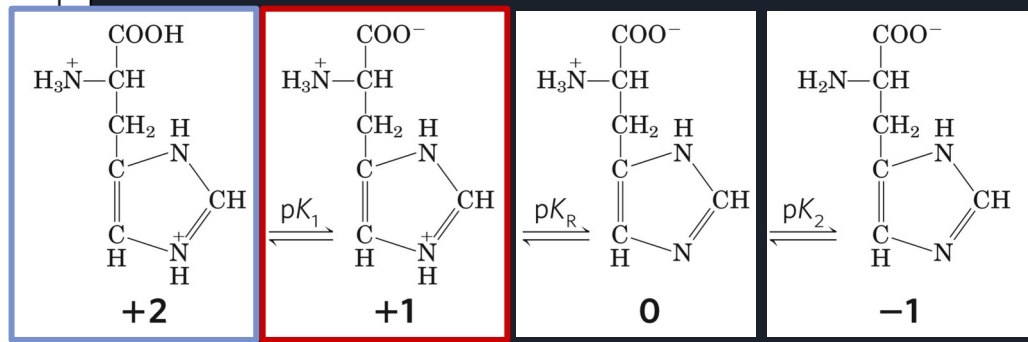
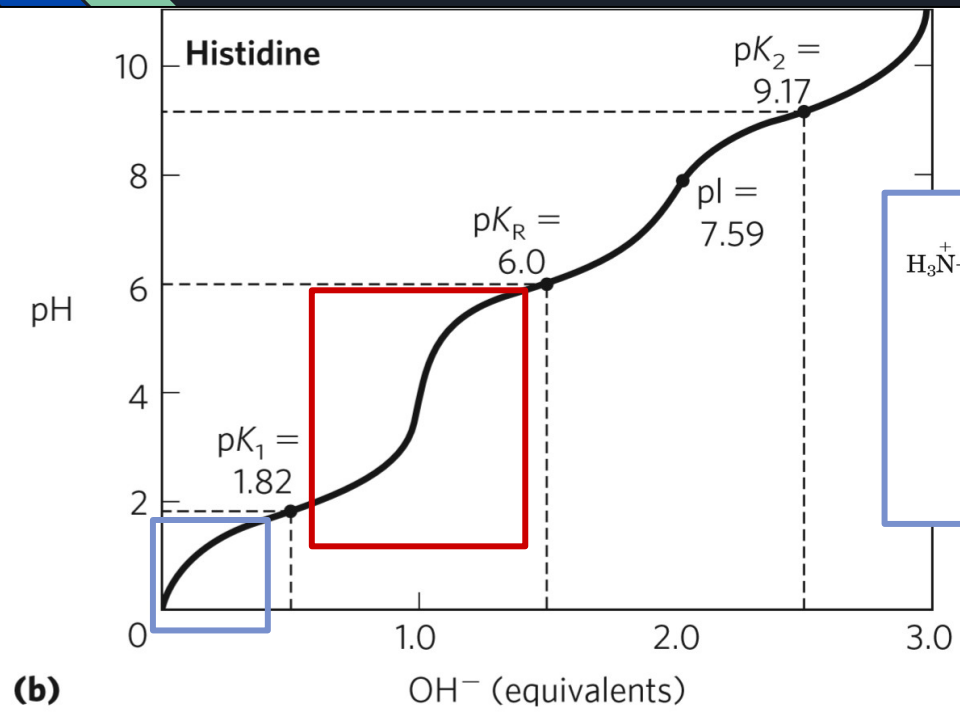
Amino acid	Abbreviation/ symbol	$M_r^*$	$pK_a$ values			pI
			$pK_1$ (—COOH)	$pK_2$ (—NH <sub>3</sub> <sup>+</sup> )	$pK_R$ (R group)	
<b>Nonpolar, aliphatic R groups</b>						
Glycine	Gly G	75	2.34	9.60		5.97
Alanine	Ala A	89	2.34	9.69		6.01
Proline	Pro P	115	1.99	10.96		6.48
Valine	Val V	117	2.32	9.62		5.97
Leucine	Leu L	131	2.36	9.60		5.98
Isoleucine	Ile I	131	2.36	9.68		6.02
Methionine	Met M	149	2.28	9.21		5.74
<b>Aromatic R groups</b>						
Phenylalanine	Phe F	165	1.83	9.13		5.48
Tyrosine	Tyr Y	181	2.20	9.11	10.07	5.66
Tryptophan	Trp W	204	2.38	9.39		5.89
<b>Polar, uncharged R groups</b>						
Serine	Ser S	105	2.21	9.15		5.68
Threonine	Thr T	119	2.11	9.62		5.87
Cysteine <sup>†</sup>	Cys C	121	1.96	10.28	8.18	5.07
Asparagine	Asn N	132	2.02	8.80		5.41
Glutamine	Gln Q	146	2.17	9.13		5.65
<b>Positively charged R groups</b>						
Lysine	Lys K	146	2.18	8.95	10.53	9.74
Histidine	His H	155	1.82	9.17	6.00	7.59
Arginine	Arg R	174	2.17	9.04	12.48	10.76
<b>Negatively charged R groups</b>						
Aspartate	Asp D	133	1.88	9.60	3.65	2.77
Glutamate	Glu E	147	2.19	9.67	4.25	3.22



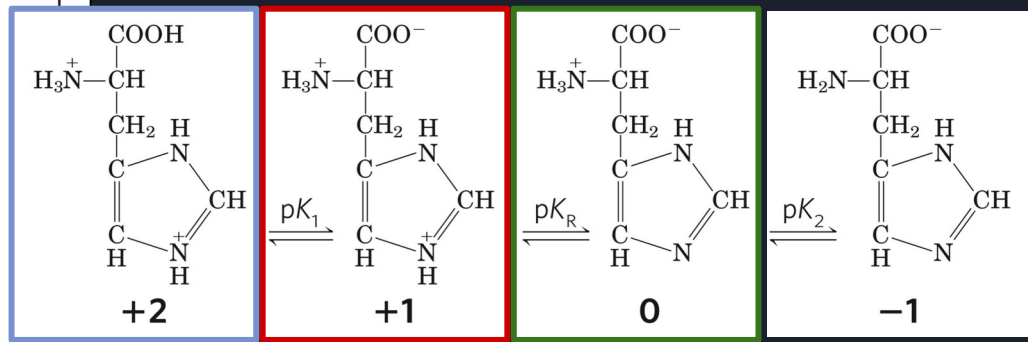
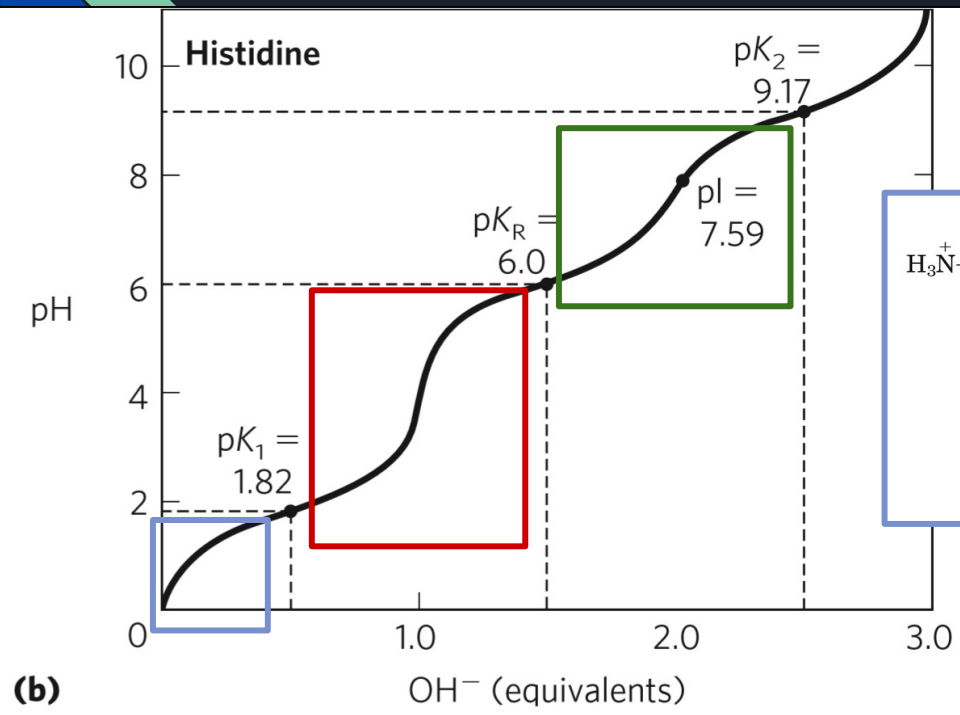


Faixas de tamponamento:

1. .
2. .
3. .

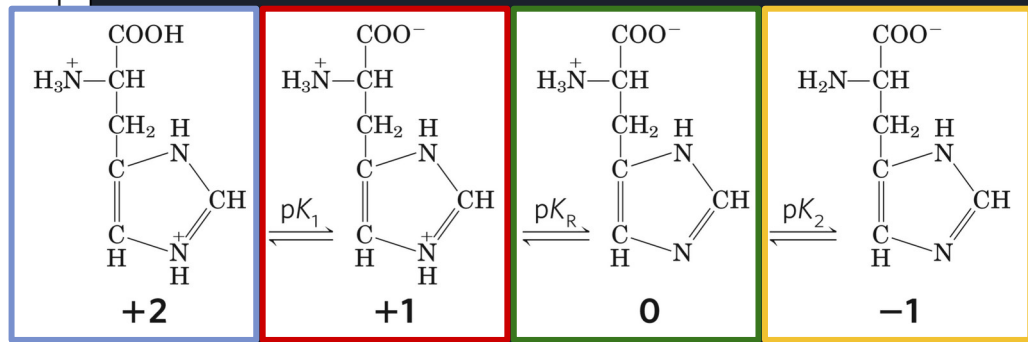
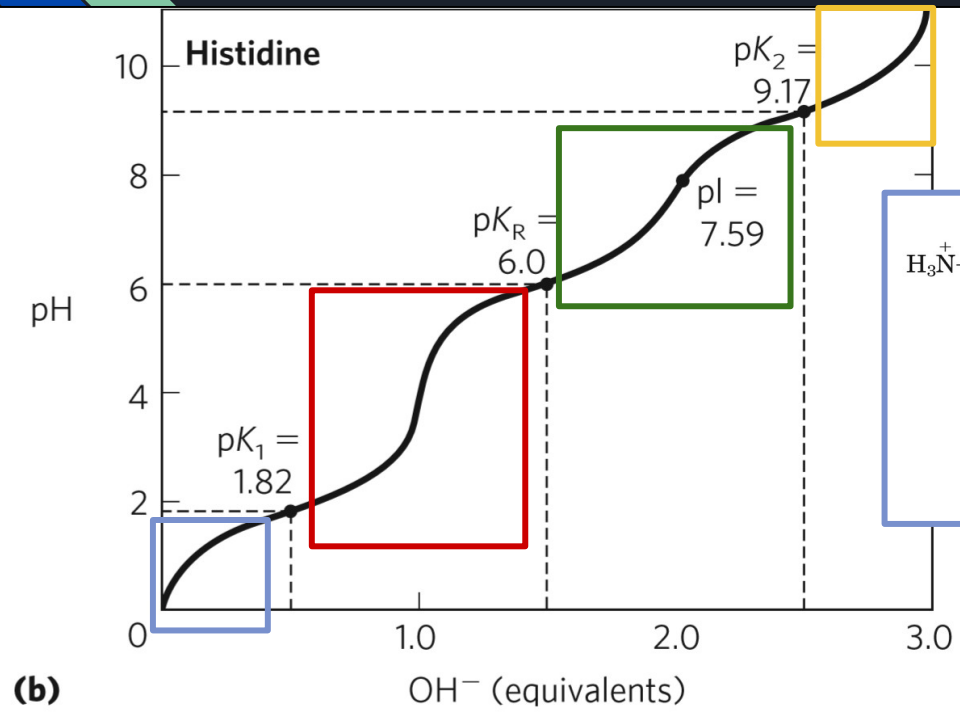


- Faixas de tamponamento:
1. De  $\text{pk}_1 - 1$  até  $\text{pk}_1 + 1 \Rightarrow 0,82$  a  $2,82$
  2. .
  3. .



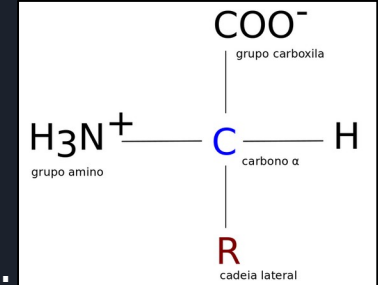
Faixas de tamponamento:

1. De  $pK_1 - 1$  até  $pK_1 + 1 \Rightarrow 0,82$  a  $2,82$
2. De  $pK_R - 1$  até  $pK_R + 1 \Rightarrow 5,0$  a  $7,0$
3. .



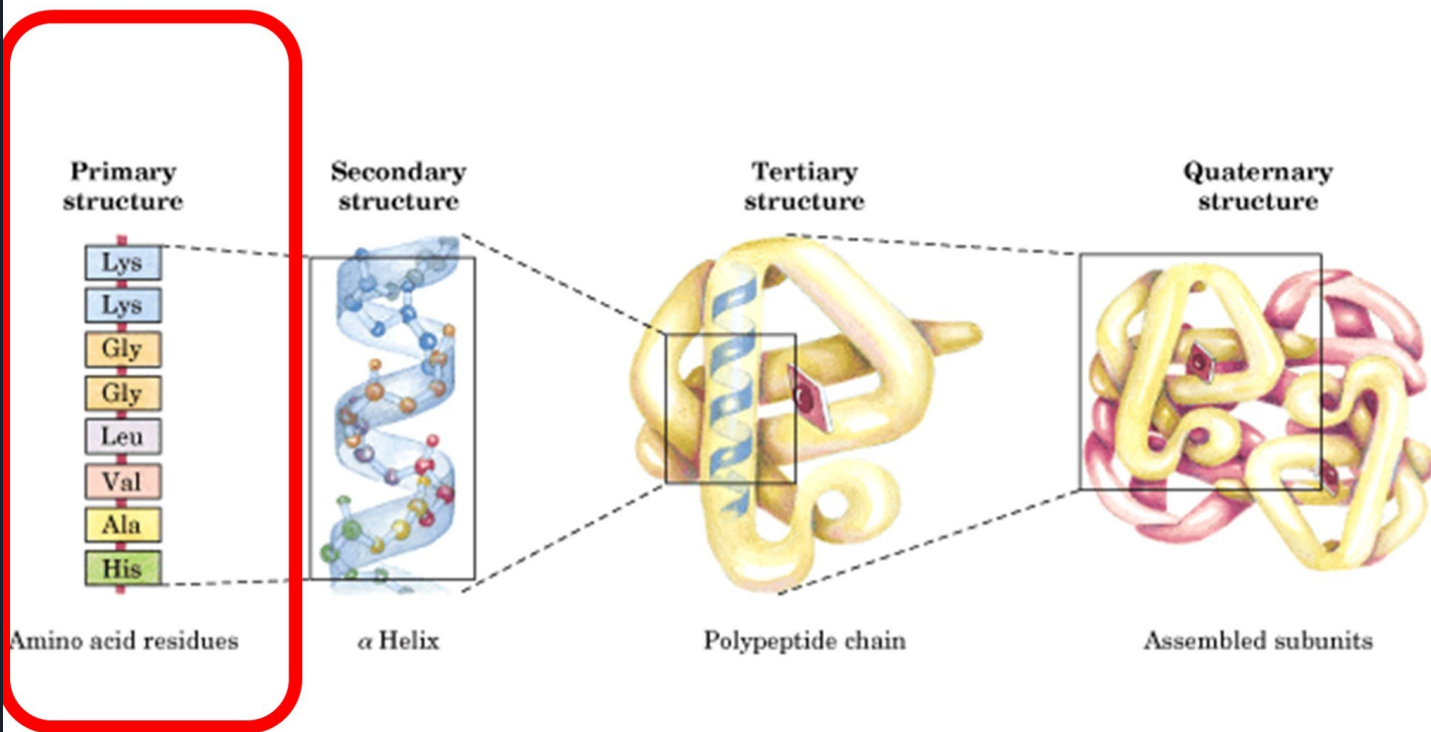
- Faixas de tamponamento:
1. De  $pK_1 - 1$  até  $pK_1 + 1 \Rightarrow 0,82$  a  $2,82$
  2. De  $pK_R - 1$  até  $pK_R + 1 \Rightarrow 5,0$  a  $7,0$
  3. De  $pK_2 - 1$  até  $pK_2 + 1 \Rightarrow 8,17$  a  $10,17$

## Vimos que os aminoácidos ...



- Estruturas comuns e **cadeia lateral** variável.
- Existem como enantiômeros D e L.
- Alguns ocorrem mas são menos comuns nas proteínas.
- **Polaridade** das cadeias laterais.
- Têm propriedades ácido-básicas que variam.
- Podem ser usados como **tampões** e têm curvas de titulação características.

# Proteínas: polímeros de aminoácidos



# Proteínas: pls característicos

## The Isoelectric Points of Some Proteins

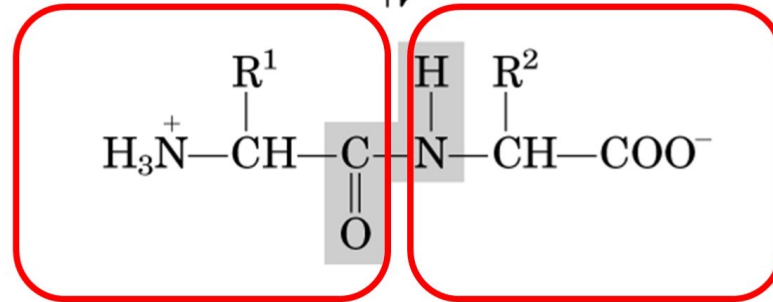
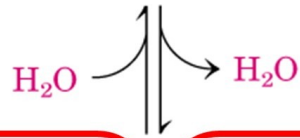
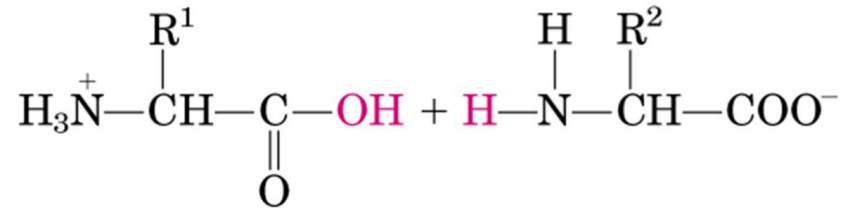
Protein	pI
Pepsin	~1.0
Egg albumin	4.6
Serum albumin	4.9
Urease	5.0
$\beta$ -Lactoglobulin	5.2
Hemoglobin	6.8
Myoglobin	7.0
Chymotrypsinogen	9.5
Cytochrome c	10.7
Lysozyme	11.0

## Peptídeos de importância biológica

Peptídeo	Número de aa	Efeito principal
Encefalina	5	Analgesia
Oxitocina	9	Contração da musculatura uterina no parto e glândulas mamárias na lactação
Vasopressina	9	Aumento da pressão sanguínea
Glucagon	29	Aumento da produção de glicose pelo fígado no jejum
Gramicidina	10	Antibiótico



# Ligação peptídica



Resíduo de aminoácido