



Des. e Prod. de medicamento - 2023

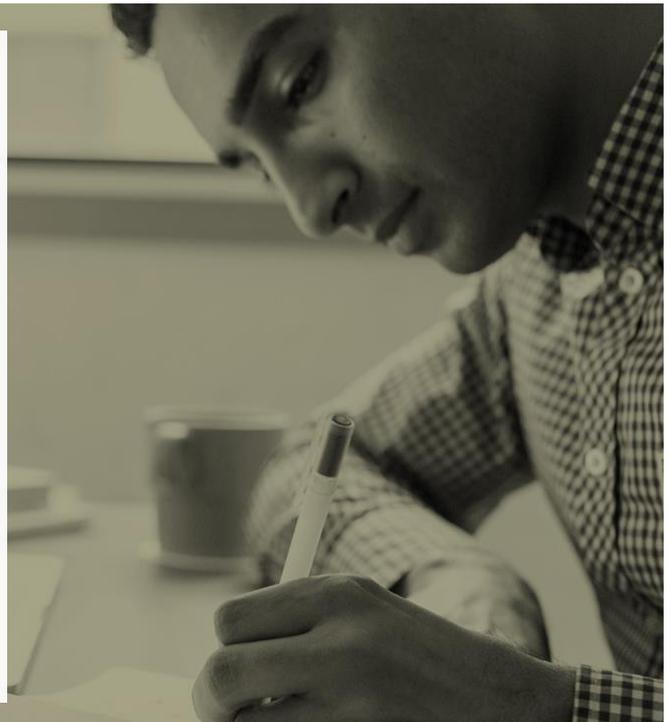
PROF. FERNANDO BARROSO
FCFRP/USP

APLICAÇÃO DE
BIOINFORMÁTICA EM
CIÊNCIAS FARMACÊUTICAS 1

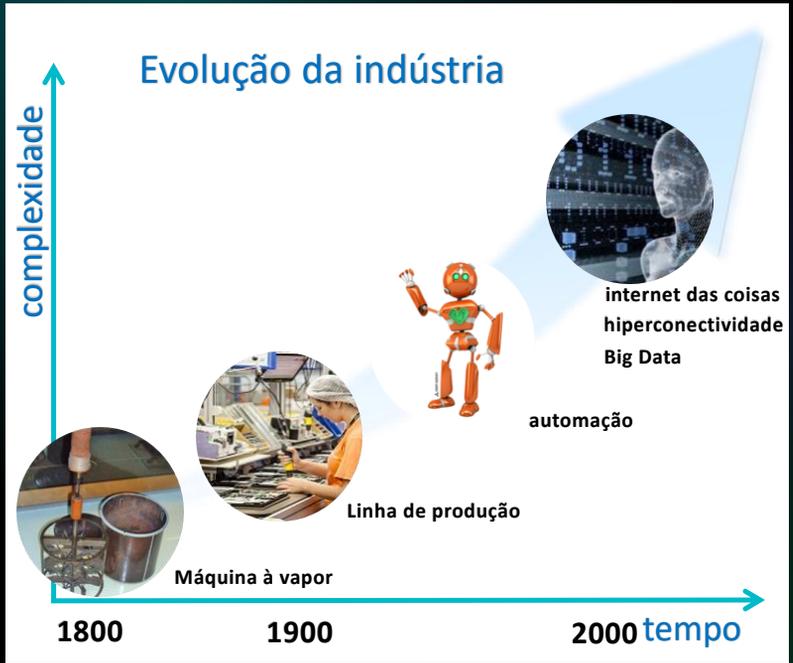
05.10.2023

Estrutura dessa aula

- Contexto
 - Motivação
- 1.1. Introdução a Bioinformática e seus aspectos estruturais;
 - 1.2. Interações fundamentais e os campos de forças para simulações moleculares
 - 1.3. Conceitos básicos de simulações moleculares
 - a) Dinâmica molecular
 - b) "Docking" molecular
 - 1.4. Modelagem molecular e o desenvolvimento racional de fármacos

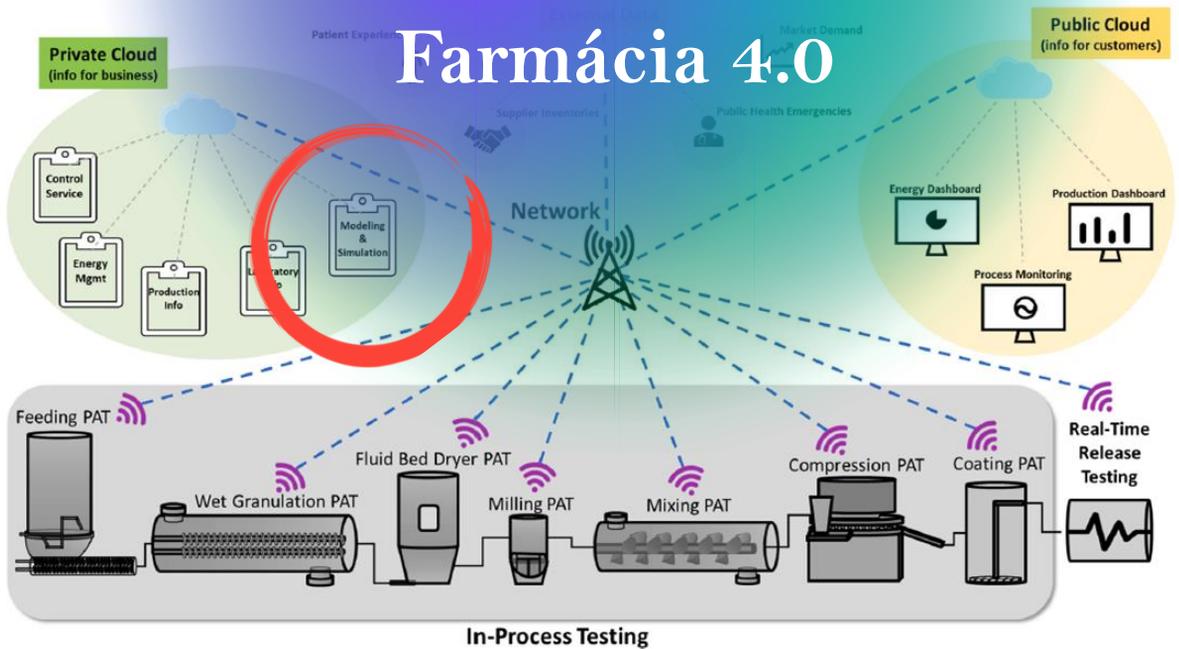


Farmácia 4.0

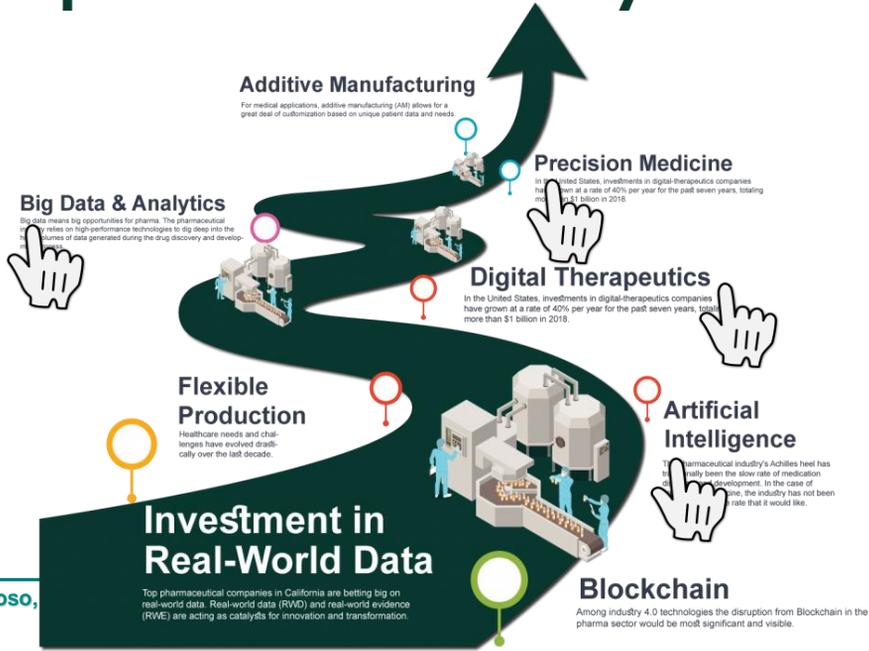


Arden et al.

International Journal of Pharmaceutics 602 (2021) 120554

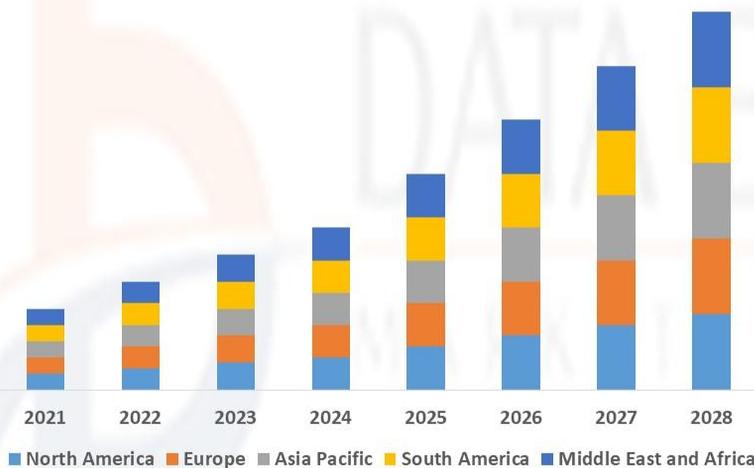


Top 10 Pharma Industry Trends



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Global Molecular Modelling Market is Expected to Account for USD 7,244.69 Million by 2028



Global Molecular Modelling Market, By Regions, 2021 to 2028



DATA BRIDGE MARKET RESEARCH



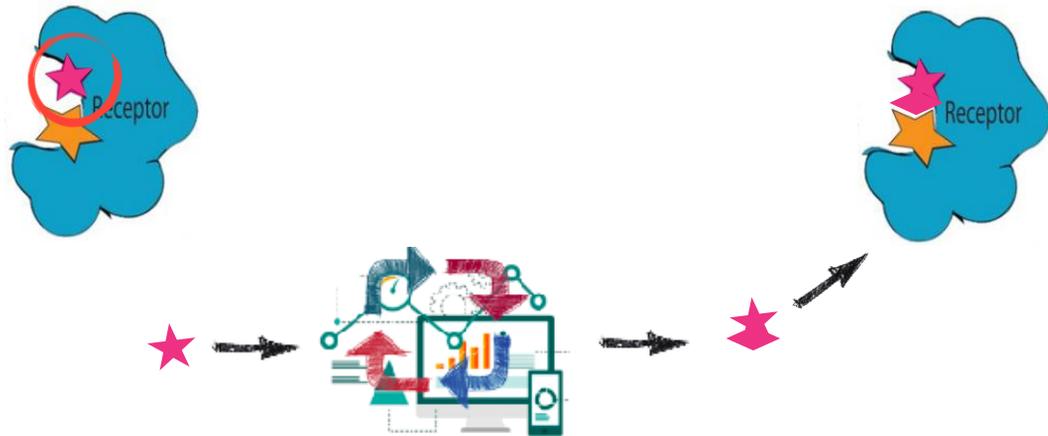


Motivação

1. Qual o sítio de ligação?



2. Como otimizar um ligante?



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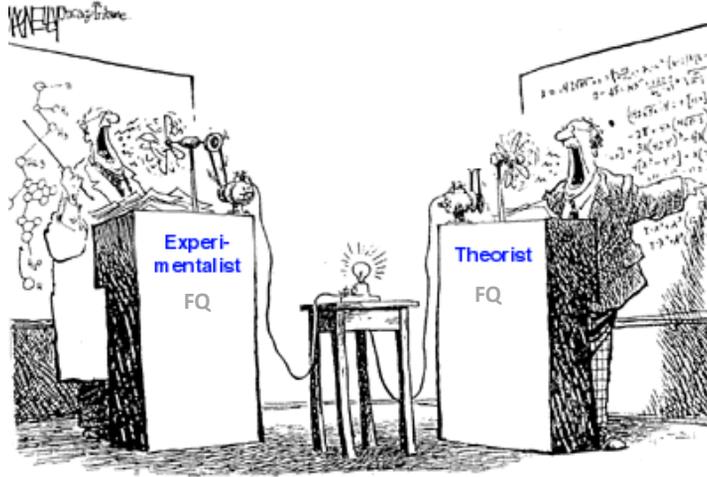
3. Qual o melhor ligante?



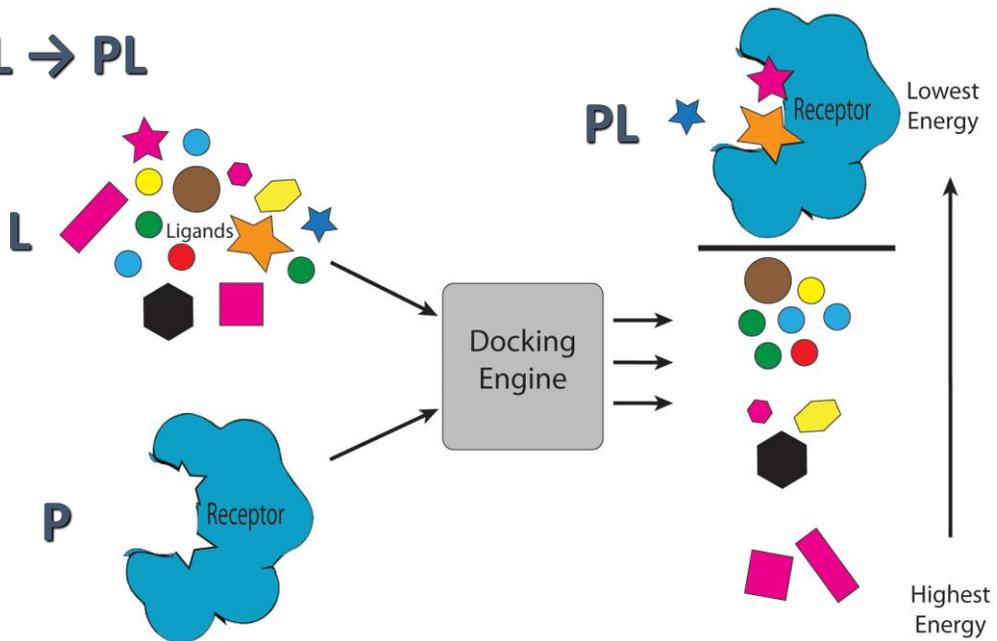
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Físico-Química

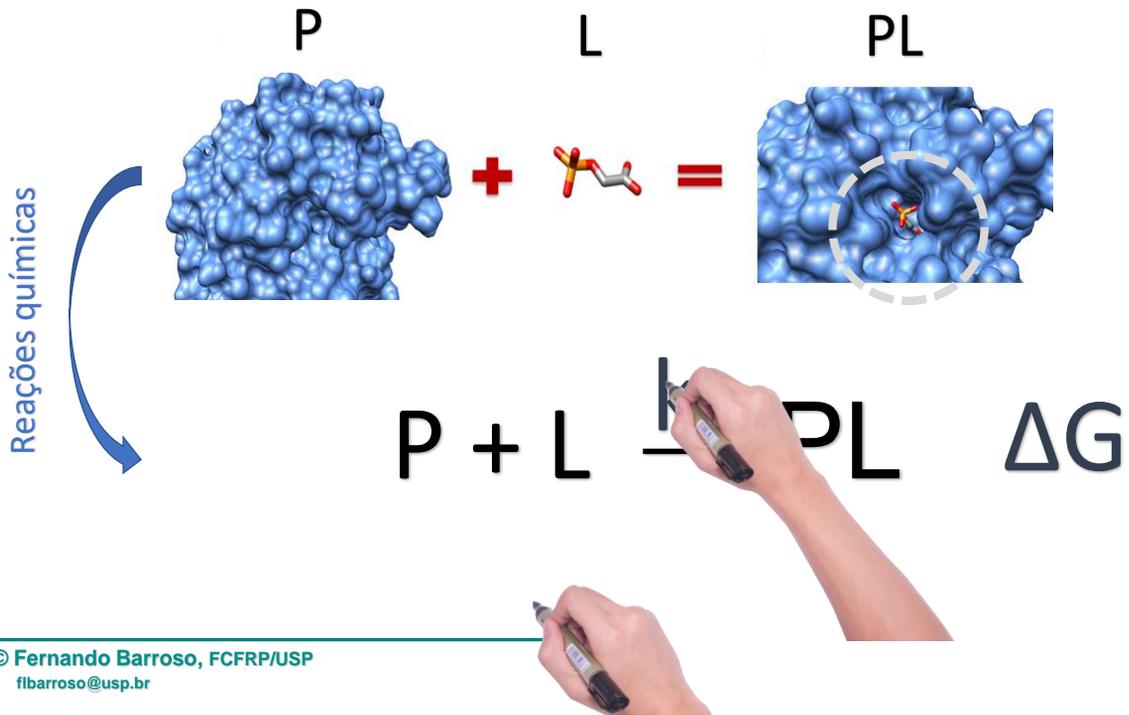


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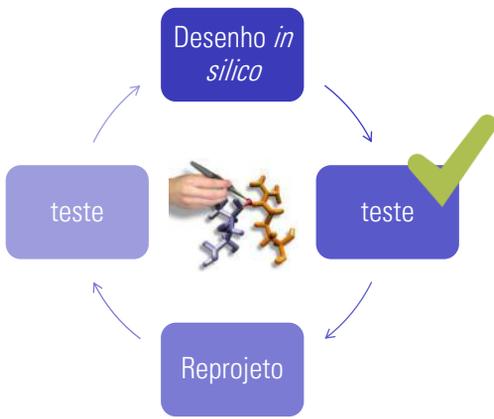


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Como prever isso?



PROTEIN/DRUG DESIGN



O "docking" molecular já nos trouxe algum remédio?



Has Molecular Docking Ever Brought us a Medicine?

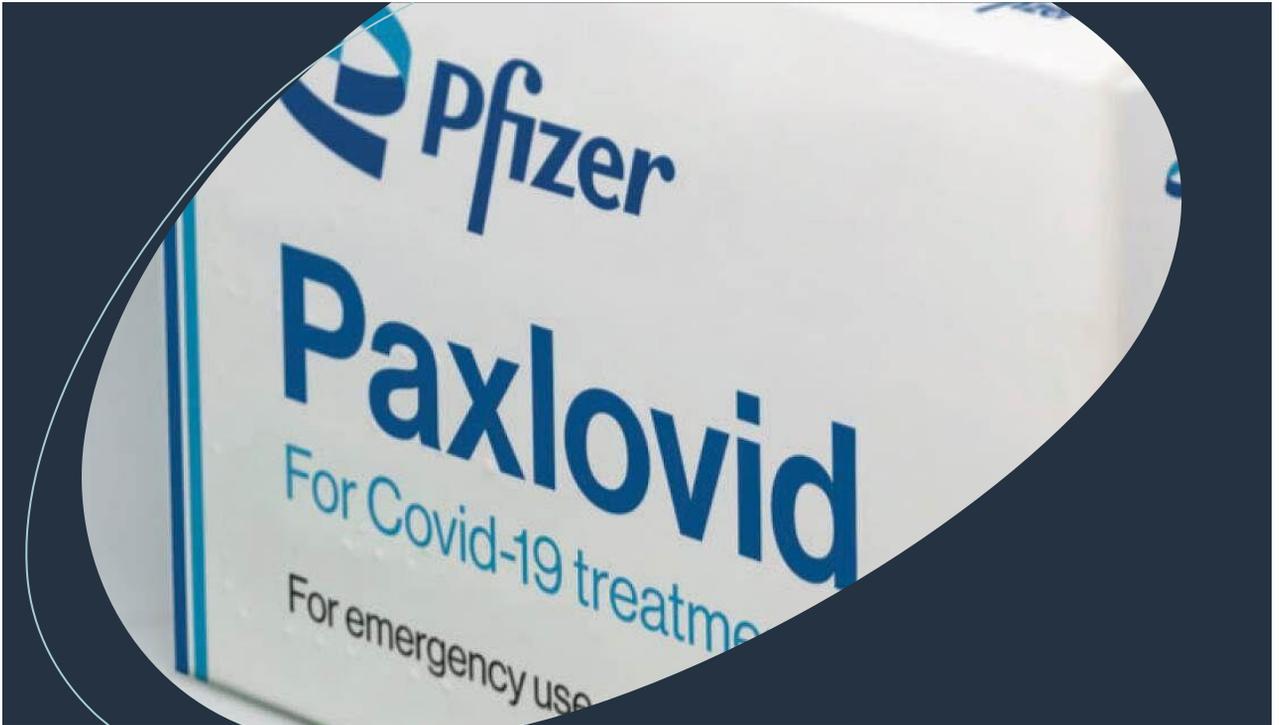
Mark Andrew Phillips, Marisa A. Stewart,
Darby L. Woodling and Zhong-Ru Xie

Additional information is available at the end of the chapter

<http://dx.doi.org/10.5772/intechopen.72898>

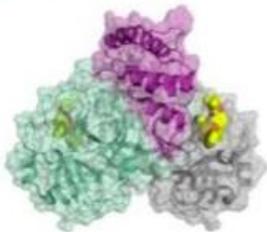
- ✓ **Zanamivir** (influenza)
- ✓ **Raltegravir, Nelfinavir & Saquinavar** (HIV)
- ✓ **Dorzolamide** (Glaucoma)



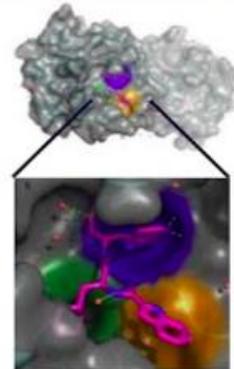


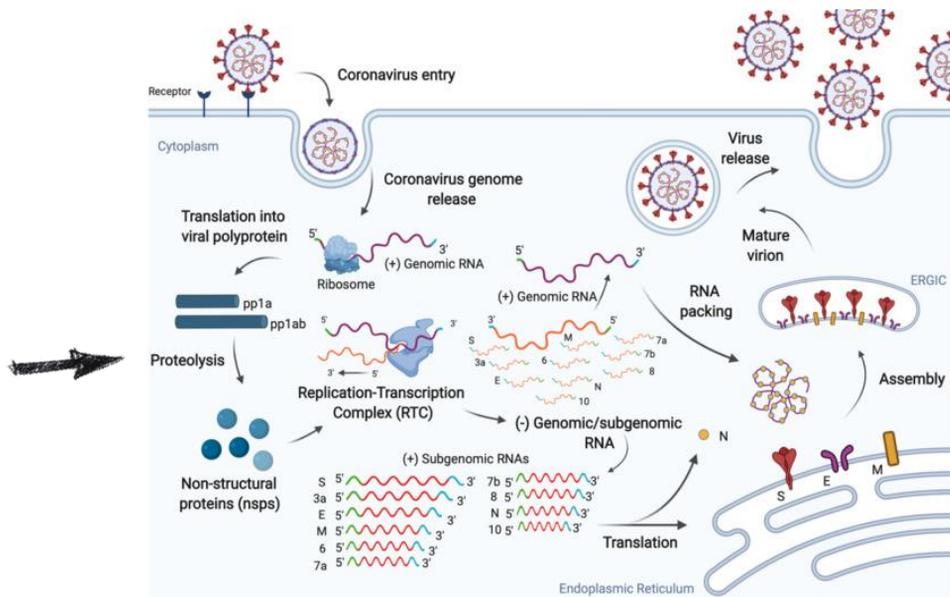
Main protease (Mpro) as a SARS-CoV-2 drug target

- Essential enzyme for processing the polyproteins translated from viral RNA
- Highly conserved across coronavirus family
- Highly structurally enabled
- >400 public PDB structures as of November 11, 2021, many with ligands bound



Structure of SARS-CoV-2 Mpro
The active form of Mpro is a homo-dimer with orthogonally positioned catalytic sites





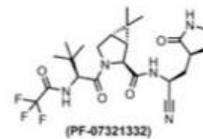
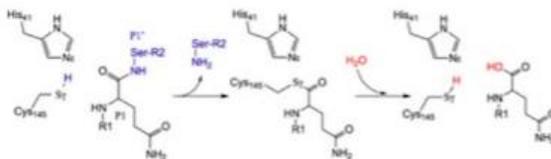
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10.1101/2020.12.05.409821

Main protease (Mpro) as a SARS-CoV-2 drug target

Mpro is a protease that uses a nucleophilic cysteine to cleave peptide bonds



- First oral Mpro inhibitor to enter clinical development
- Peptidomimetic with cysteine modifying warhead

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Schrödinger

MedRxiv 2021 DOI: 10.1101/2021.07.28.21261232

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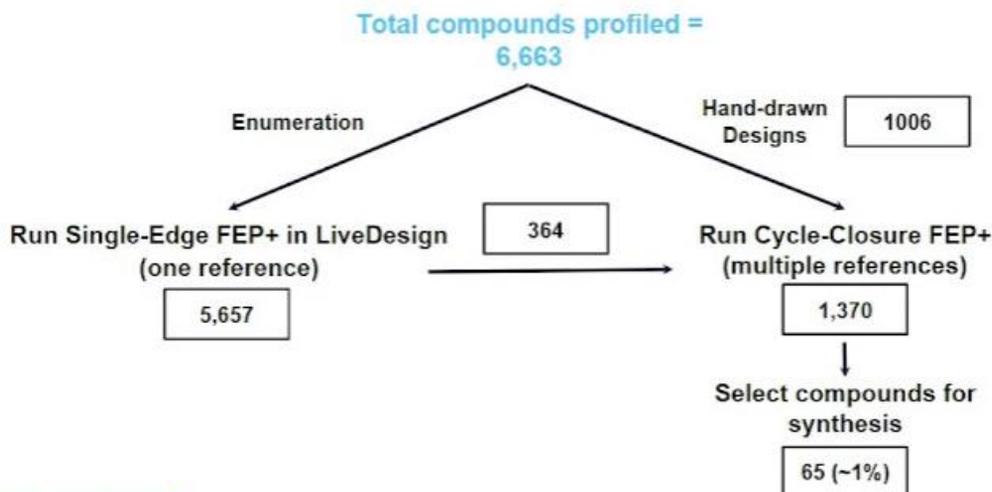
Building virtual libraries using reaction enumeration in LiveDesign

Compound 11a (SLZE ligand)
W. Dai et al., Science
10.1126/science.abb4489 (2020)

Search for commercially available reagents

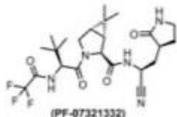
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FEP+: Workflow and throughput



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Lead compounds possess pan-coronavirus activity and similar potency to Pfizer oral Mpro inhibitor



Prioritized for synthesis based on
FEP+ after reaction enumeration

Designed based on WaterMap

Str Name	(PF-07321332)	Lead 1	Lead 2	Lead 3
Enzyme IC ₅₀ (nM)				
Mpro_SARS-CoV-2	12.6	16.2	21.6	12.1
Mpro_229E	67.8	48.4	26.4	21.8
Mpro_HKU1	17.2	23.8	16	7.74
Mpro_MERS	146	276	208	88.7
Mpro_NL63	218	180	40	37.6
Mpro_OC43	32.3	59.8	35.1	18.9
Mpro_SARS-CoV-1	21.3	27.8	30	11.9
CathepsinB(h)	>50000	38000	>50000	>50000
CPE EC ₅₀ (μM) / Parent Cell line CC ₅₀ (μM)				
SARS-CoV-2 (HeLa-ACE2)	0.059 / >40	0.11 / >40	0.12 / >40	0.097 / >40
OC43 (Huh7)	0.11 / >50	0.28 / >50	0.29 / >50	0.12 / >50
229E (MRC5)	0.10 / >50	0.11 / >50	0.06 / >50	0.07 / >50

Schrödinger

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Como fazemos isso?

Algumas definições e conceitos

Formalismos para usarmos o docking molecular

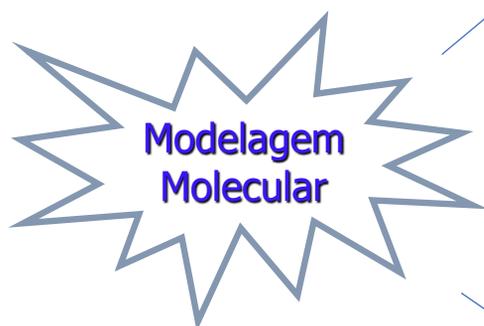


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Nomenclatura

docking

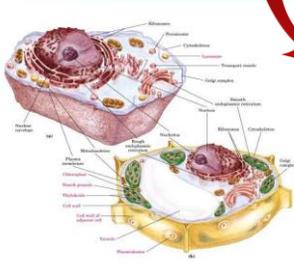
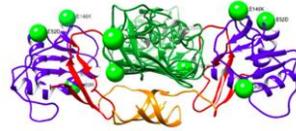


- Química Teórica
- Química Computacional
- Biologia Computacional
- Bioquímica Computacional
- Biofísica Computacional
- Bioinformática
- Biocomputação
- Simulação Computacional
- Simulação Molecular
- Experimento em computador
- Experimentos "in silico"
- Computer-aid.....etc.

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Bioinformática Estrutural

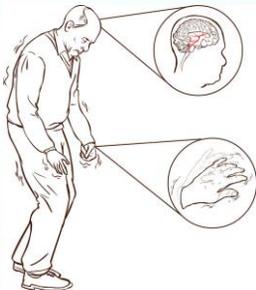
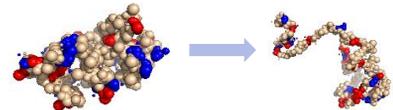


... está relacionada com o **entendimento da biologia** em termos de suas **moléculas** (no sentido da físico-química) e aplicando **técnicas de "informática"** para entender e organizar as informações associadas com estas moléculas, **em larga-escala**". (M. Gerstein)

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Bioinformática Estrutural

desenvolvimento de proteínas

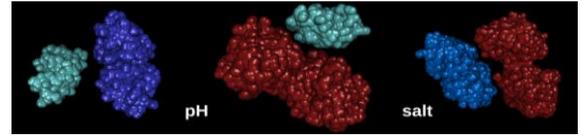


✓ Base molecular para o entendimento das doenças

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Barroso Research lab

Bioinformática Estrutural



- ✓ Base molecular para o entendimento das doenças

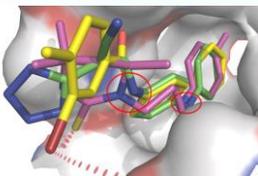
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Barroso Research lab

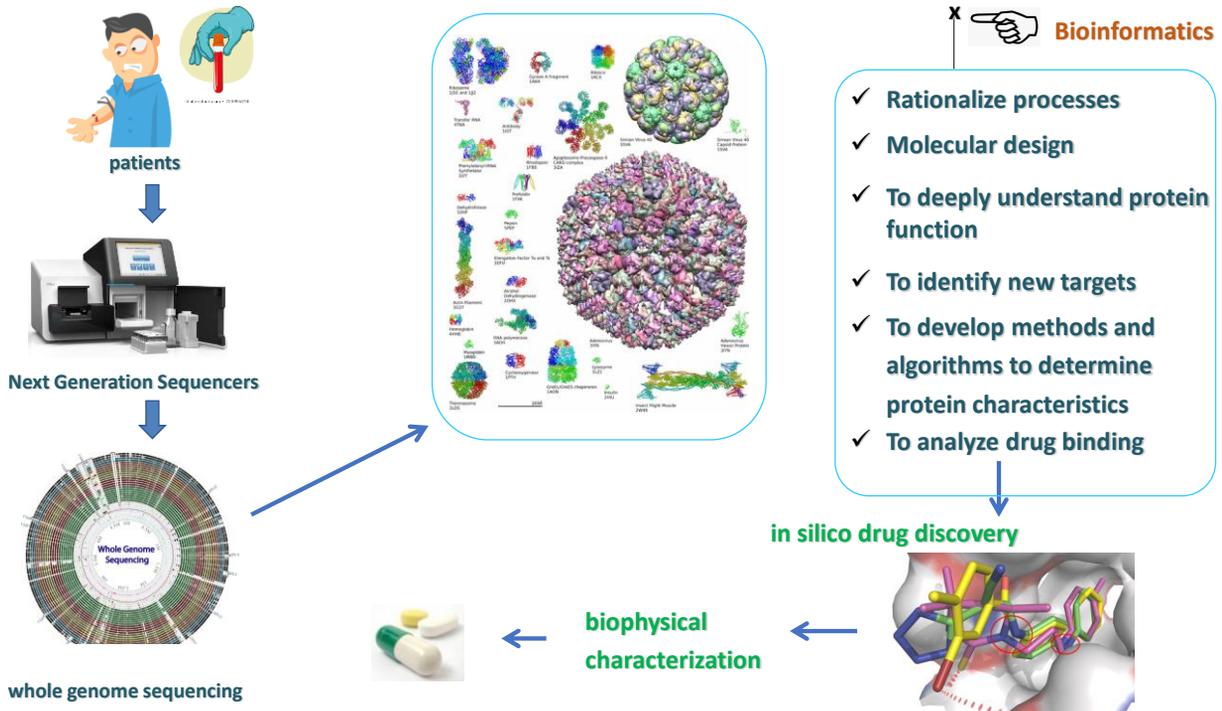
Bioinformática Estrutural



- ✓ Base molecular para o entendimento das doenças
- ✓ Engenharia de novas moléculas



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Algumas vantagens...

- Não tem risco de envenenamento, mau-cheiro ou explosão.
- Não consome reagentes (caros)
- Os experimentos podem ser feitos sob condições drásticas (alta pressão, elevada temperatura, etc...)



Algumas vantagens...

- É fácil trocar a(s) molécula(s)
 - o meio ambiente
 - temperatura, pH, concentrações,...

..... **BASTA TROCAR ALGUNS PARÂMETROS NO INPUT!**



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Algumas vantagens...

- Informações diretas no nível molecular
- Experimento livre de contaminação e impurezas



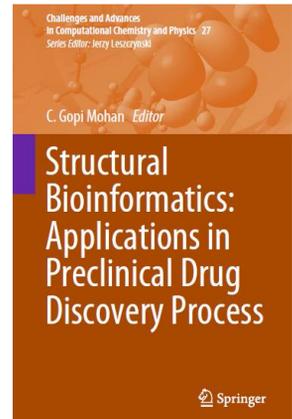
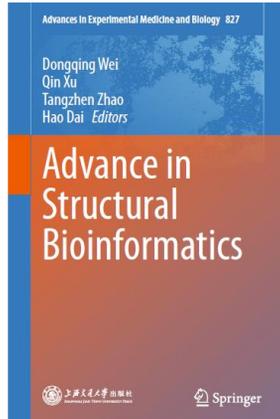
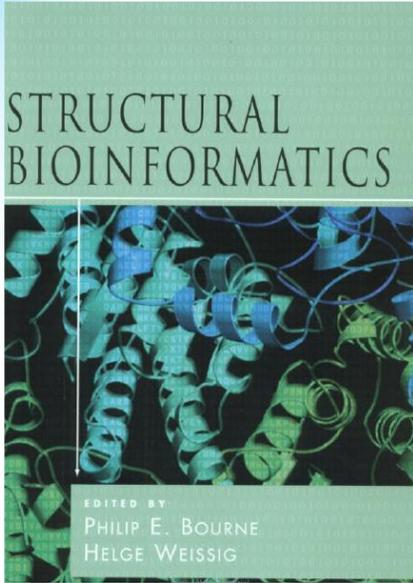
"As simulações são ficções que aspiram a imitar a realidade. Bonitas imagens e até mesmo alguns bons números não garantem uma boa ciência" (Peter Steinbach, Center of Molecular Modeling, NIH, EUA)



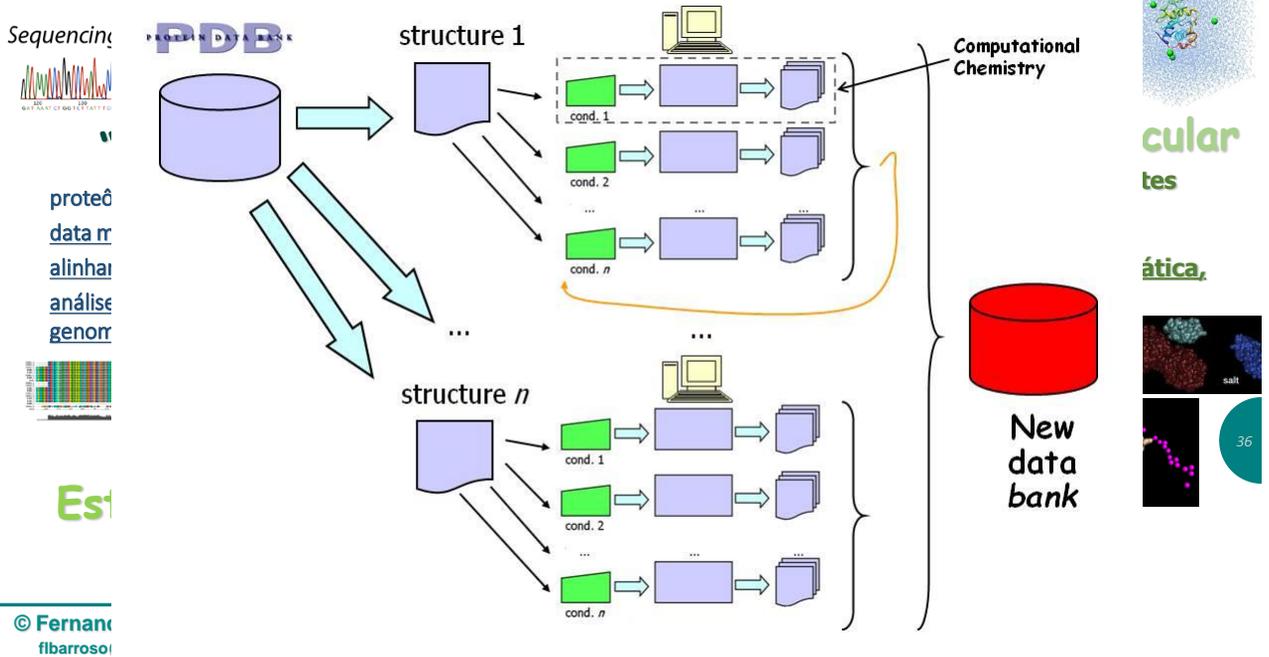
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Bons livros



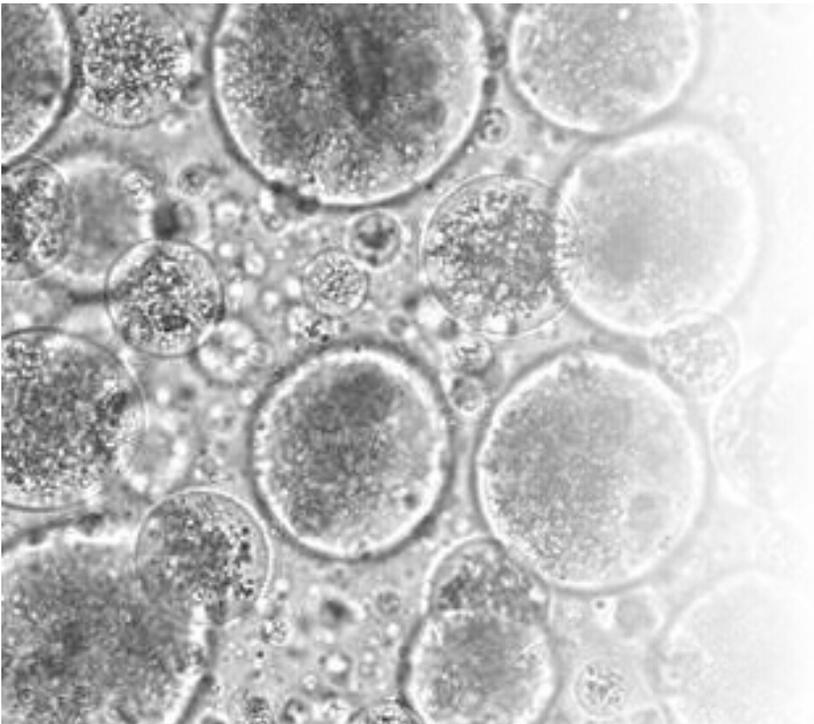
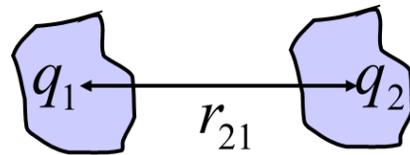
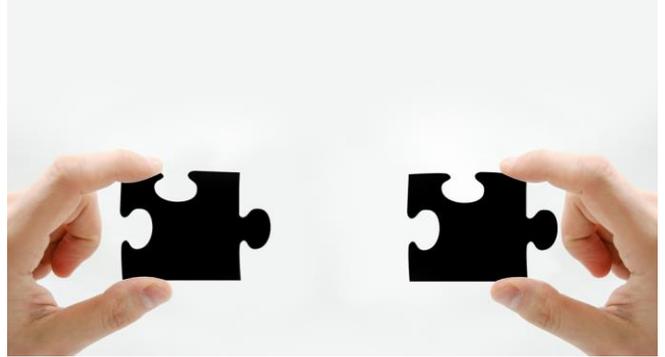
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Docking

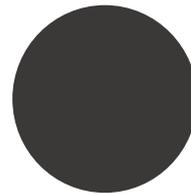
O que controla a distância entre essas moléculas?



soluto



1 nm

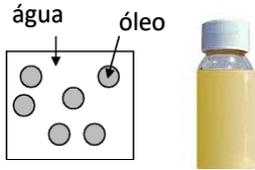


1 μm

Coloides

são misturas heterogêneas

Sistema de dispersão farmacêutico



Emulsão

Uma emulsão é uma **dispersão** de pelo menos **dois líquidos imiscíveis**, um dos quais está disperso como gotículas no outro líquido, e **estabilizado por um agente emulsionante**.

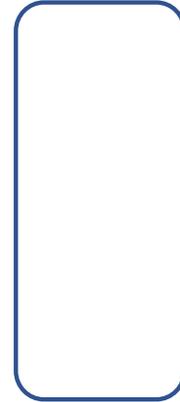
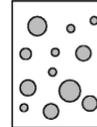
É termodinamicamente instável!

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Estabilidade?

Como evitar a separação de fases?

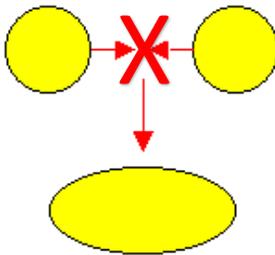
emulsão



É irreversível!
Precisa ser evitada!

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Sistema de dispersão farmacêutico



Evitar a associação!

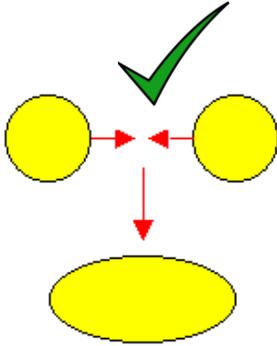
O que controla a distância entre as (macro)partículas?

- ✓ Forças intermoleculares
- ✓ Como interpretá-las?
- ✓ Como quantificá-las?

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Docking molecular



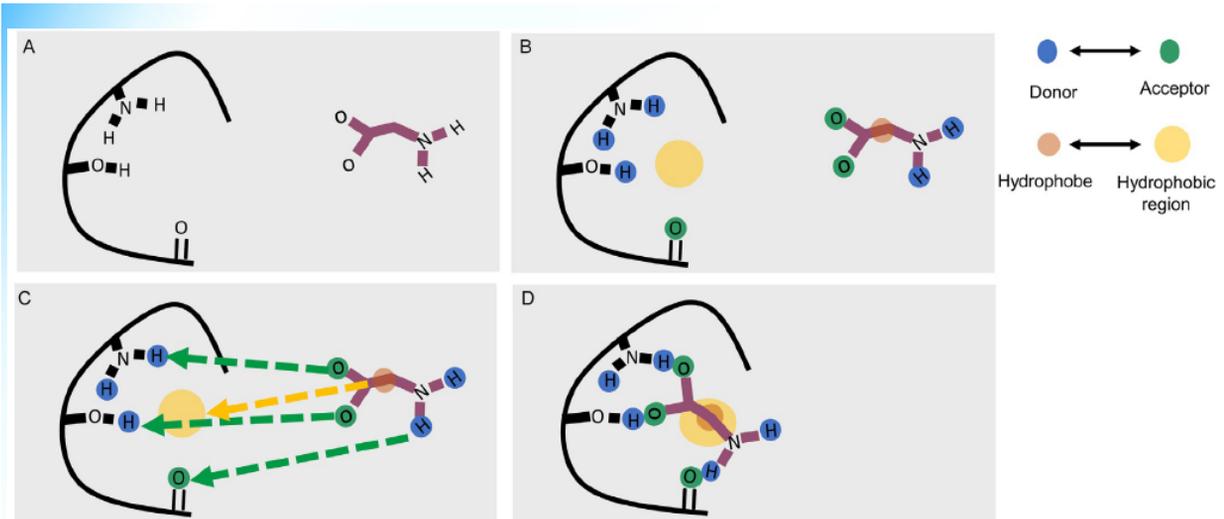
O que controla a distância entre as (macro)partículas?

- ✓ Forças intermoleculares
- ✓ Como interpretá-las?
- ✓ Como quantifica-las?

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Garantir a associação!

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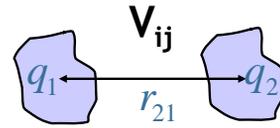
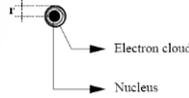
Fig. 1 Rigid docking approach: (A) protein and ligand. (B) The initial pose generation through pharmacophore point matching. Donor–acceptor pharmacophore point and hydrophobic pharmacophore points are added to the protein and the ligand. (C) The searching algorithm tries to match the protein and the ligand fitting points by matching donor with acceptors and hydrophobic atoms with hydrophobic cavities. (D) Different solutions are found.

<https://doi.org/10.1016/bs.pmch.2021.01.004>

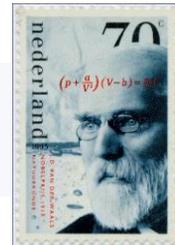
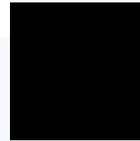
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Interações intermoleculares

$V_{ij} =$ repulsão eletrônica
 + transferência de carga
 + interação elétrica multipolo-multipolo
 + interação elétrica multipolo-multipolo induzido
 + dispersão



Forças de van der Waals



sempre **atrativas**

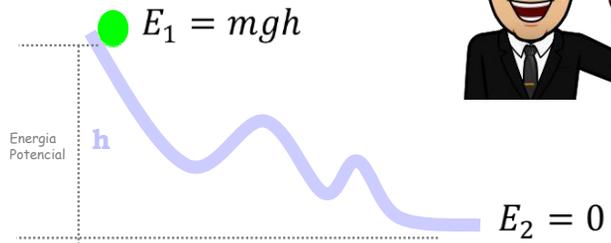
Não podem ser alteradas!

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- ✓ Formalismo **Newtoniano**
- ✓ Formalismo **Hamiltoniano**

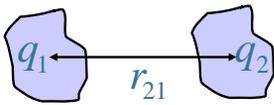


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Força (partícula)

Energia (campo)

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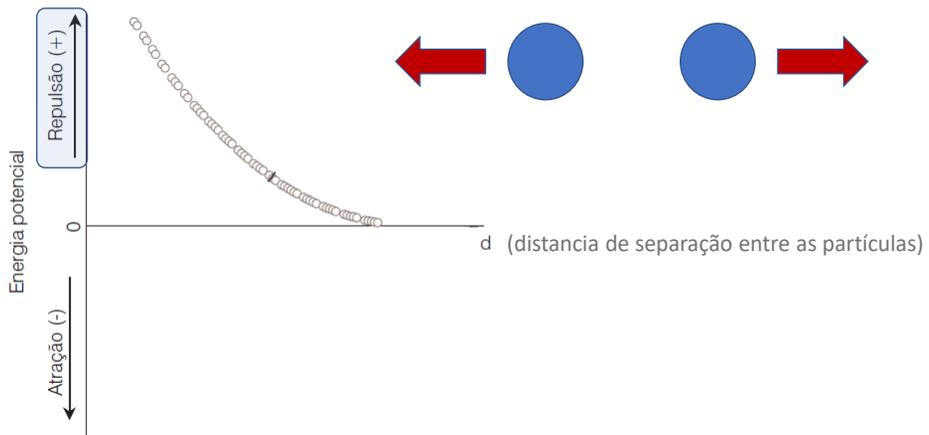
Propriedades de campo vs de partícula

Saiba as diferenciar!

	Propriedades de partícula	Inter-relações	Propriedades de campo
Grandeza vetorial	<p><i>Força</i></p> $\mathbf{F} = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{ r ^2} \hat{\mathbf{r}}$	$\mathbf{F} = q\mathbf{E}$	<p><i>Campo elétrico</i></p> $\mathbf{E} = \frac{1}{4\pi\epsilon_0} \frac{q}{ r ^2} \hat{\mathbf{r}}$
Inter-relações	$\mathbf{F} = -\nabla U$		$\mathbf{E} = -\nabla V$
Grandeza escalar	<p><i>Energia Potencial</i></p> $U = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{ r }$	$U = qV$	<p><i>Potencial</i></p> $V = \frac{1}{4\pi\epsilon_0} \frac{q}{ r }$

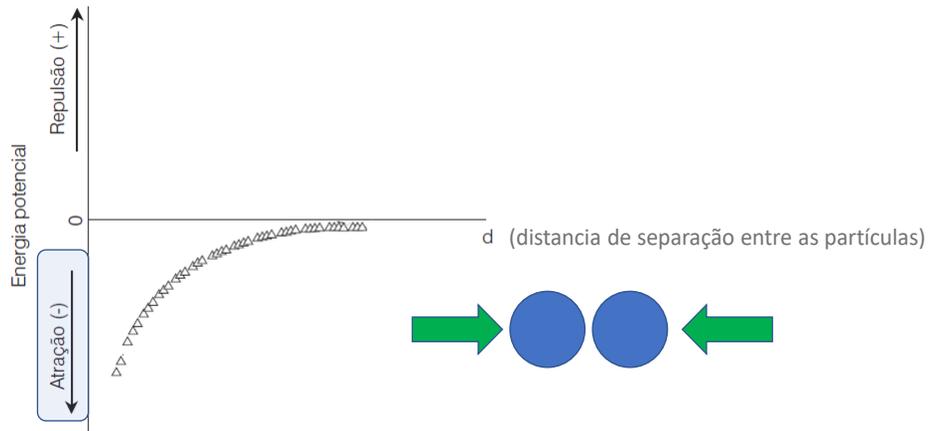
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Interpretação: energia potencial de interação



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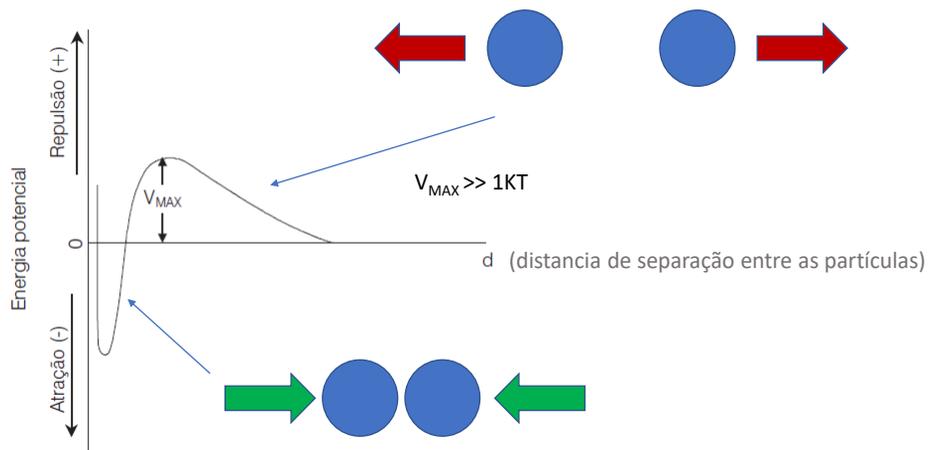
Interpretação: energia potencial de interação



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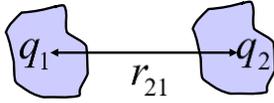
Interpretação: energia potencial de interação



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Potencial vs Potencial de interação



$$U_i(r_{21}) = E_1 + E_2 + \frac{1}{2}(q_1\phi_2(r_{21}) + q_2\phi_1(r_{21}))$$

- ✓ Energia total de interação
- ✓ Energia configuracional
- ✓ Energia interna do sistema

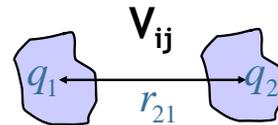
$$U_t = \frac{1}{2}(q_1\phi_2(r_{21}) + q_2\phi_1(r_{21}))$$

$$U_t = \frac{1}{2} \sum_{i=1}^N \sum_{j=1}^N q_i \phi_j(r_{ji})$$

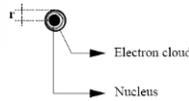
} potencial
} "potencial" de interação
} ou
} energia potencial
} ou
} energia de interação

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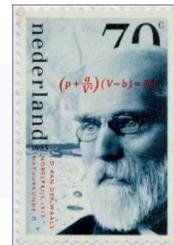
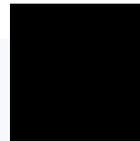
Interações intermoleculares



- $V_{ij} =$
- repulsão eletrônica
 - +
 - transferência de carga
 - +
 - interação elétrica multipolo-multipolo
 - +
 - interação elétrica multipolo-multipolo induzido
 - +
 - dispersão



Forças de van der Waals



sempre atrativas

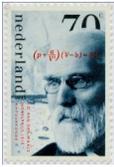
Não podem ser alteradas!

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Como quantifica-las?

$$u_{vdW}(r) = \frac{A}{r_{ij}^{12}} - \frac{B}{r_{ij}^6}$$

Para átomos



Forças de van der Waals

Forças intermoleculares

Lenard-Jones

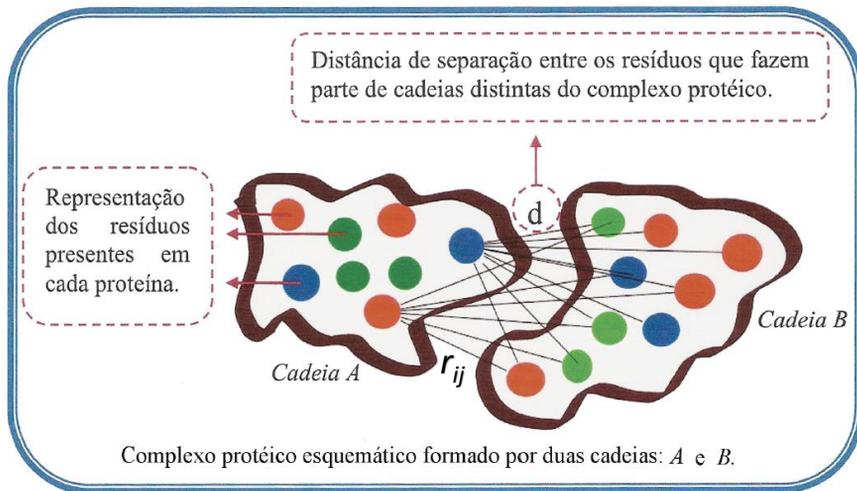
$$A_{ij} = 4\epsilon_{ij}\sigma_{ij}^{12}$$

$$B_{ij} = 4\epsilon_{ij}\sigma_{ij}^6$$

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$$u_{vdW}(r) = \frac{A}{r_{ij}^{12}} - \frac{B}{r_{ij}^6}$$



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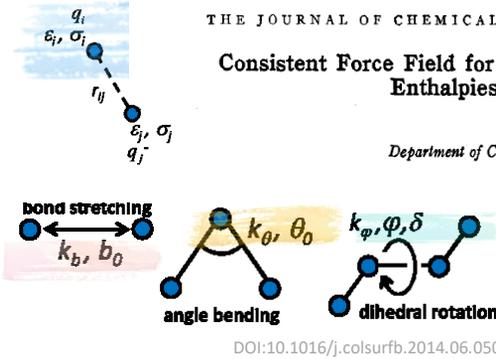
Campo de força

Consistent Force Field for Calculations of Conformations, Vibrational Spectra, and Enthalpies of Cycloalkane and n-Alkane Molecules

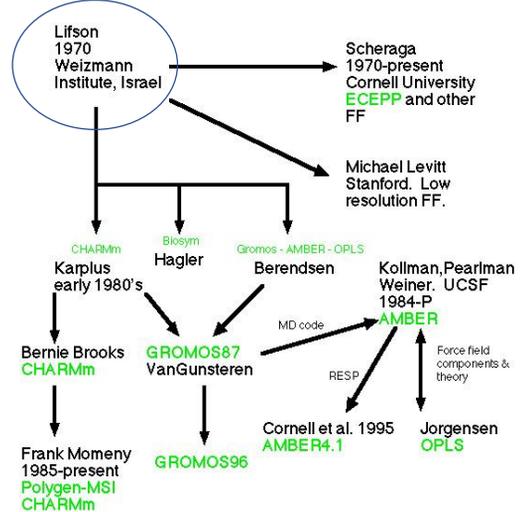
S. LIFSON AND A. WARSHEL

Department of Chemical Physics, Weizmann Institute of Science, Rehovot, Israel

(Received 13 May 1968)



$$U = \frac{1}{2} \sum_{\text{bonds}} k_b (b - b_0)^2 + \frac{1}{2} \sum_{\text{angles}} k_\theta (\theta - \theta_0)^2 + \frac{1}{2} \sum_{\text{dihedrals}} k_\phi (1 + \cos(n\phi - \delta)) + \sum_{\text{non-bonded pairs}} \left[\frac{A}{r^{12}} - \frac{B}{r^6} \right] + \frac{q_1 q_2}{\epsilon \epsilon_0 r}$$



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Institut Pasteur

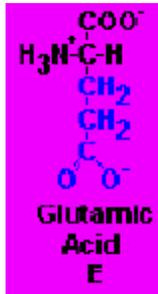
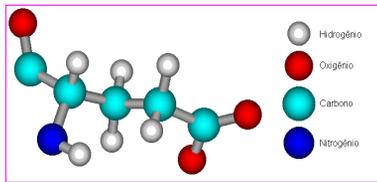
SIRAH
Southamerican Initiative for a Rapid and Accurate Hamiltonian



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G 57.05 GLY 6.064 Glycine	A 71.09 ALA 6.107 Alanine	V 99.14 VAL 6.002 Valine	L 113.16 LEU 6.036 Leucine	I 113.16 ILE 6.038 Isoleucine	P 112.7 PRO 6.3 Proline	F 147.18 PHE 5.91 Phenylalanine	W 186.12 TRP 5.88 Tryptophan	D 115.09 ASP 4.5 Aspartic Acid	K 128.17 LYS 9.47 Lysine	R 173.4 ARG 10.76 Arginine
C 103.15 CYT 9.1-9.5 Cysteine	M 131.19 MET 5.74 Methionine	S 87.08 SER 5.68 Serine	T 101.11 THR 5.60 Threonine	N 114.11 ASN 5.41 Asparagine	Q 128.14 GLN 5.65 Glutamine	H 137.14 HIS 6.2 Histidine	Y 163.18 TYR 9.7 Tyrosine	E 129.12 GLU 5.08 Glutamic Acid	1-letter Code Molecular Weight (dalton) pI (at 25°C) Molecular Type Nonpolar Basic Acidic Purine Pyrimidine Solvent SIRAH Representation Bead Name Molecular Weight (dalton) pI (at 25°C) Molecular Type Nonpolar Basic Acidic Purine Pyrimidine Solvent SIRAH Representation Bead Name	
A 331.2 ADA Adenine	G 347.2 GUA Guanine	T 322.2 THY Thymine	C 307.2 CYT Cytosine	Na+ 22.989 NaW Sodium ion	K+ 39.102 KW Potassium ion	Cl- 35.453 CW Chloride ion	WT4 Water model	The SIRAH Force Field www.sirahff.com		

Campos de força

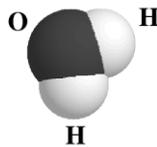


Atom	Gromos	Parse	Charmm	Reduced
N	-0.328	-0.400	-0.470	0,000
CA	0.050	0.000	0.070	0,000
C	0.355	0.550	0.510	0,000
O	-0.380	-0.550	-0.510	0,000
CB	0.010	0.000	-0.180	0,000
CG	-0.155	0.000	-0.280	0,000
CD	0.365	0.100	0.620	0,000
OE1	-0.575	-0.550	-0.760	-0.500
OE2	-0.575	-0.550	-0.760	-0.500
H	0.233	0.400	0.310	0,000
HÁ	0.000	0.000	0.090	0,000
HB1	0.000	0.000	0.090	0,000
HB2	0.000	0.000	0.090	0,000
HG1	0.000	0.000	0.090	0,000
HG2	0.000	0.000	0.090	0,000

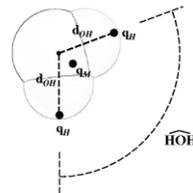
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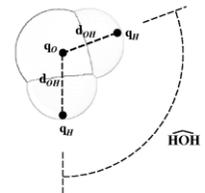
Campos de força



TIP4P



SPC



Model	q_O e	q_M e	q_H e	d_{OH} Å	\widehat{HOH} deg	$C_6(O,O)$ $(kJ\text{mol}^{-1}\text{nm}^6)$	$C_{12}(O,O)$ $(kJ\text{mol}^{-1}\text{nm}^{12})$
TIP3P	-0.834		0.417	0.9572	104.52	2.4889×10^{-3}	2.4352×10^{-6}
TIP4P		-1.04*	0.52	0.9572	104.52	2.5543×10^{-3}	2.5145×10^{-6}
SPC	-0.82		0.41	1.0	109.47	2.6171×10^{-3}	2.6331×10^{-6}
SPC/E	-0.8476		0.4238	1.0	109.47	2.6171×10^{-3}	2.6331×10^{-6}

Table 5.1: Popular water model parameters.

* This single negative charge is located along the \widehat{HOH} bisector with $d_{OM} = 0.15 \text{ Å}$.

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Campos de força

17618 *J. Phys. Chem. B*, Vol. 110, No. 35, 2006

Hess and van der Vegt

TABLE 2: Water Model Properties at 298 K, 1 atm^a

	r_c (nm)	LRC	μ (D)	ϵ_r	ρ (kg m ⁻³)	α_p (10 ⁻⁴ K ⁻¹)	$\partial\alpha_p/\partial T$ (10 ⁻⁶ K ⁻²)	κ_T (GPa ⁻¹)	$k_B T^2 \alpha_p$ (kJ mol ⁻¹)
SPC	1.40	no	2.27	66	973.8	7.4	6.2	0.53	0.55
SPC/E	1.40	no	2.35	71	995.5	5.0	7.5	0.47	0.37
SPC/E	0.85	yes	2.35	71	998.9	5.0	6.8	0.46	0.37
TIP3P	0.85	yes	2.35	98	986.0	8.9	5.3	0.57	0.66
TIP4P	0.85	yes	2.18	52	994.1	5.6	8.9	0.51	0.41
TIP4P-Ew	0.85	yes	2.32	65	995.7	3.2	10.5	0.47	0.23
experiment				78	997.2	2.56	9.7	0.468	0.19

^a The dipole moment μ , dielectric constant ϵ_r , density ρ , thermal expansion coefficient α_p and its temperature derivative, isothermal compressibility κ_T , and the liberation correction in the hydration enthalpy are given. r_c is the LJ cutoff radius. LRC is the long-range dispersion correction. All quantities, except for κ_T , were determined from 800 ns NPT Simulations: ϵ_r from dipole fluctuations,⁴⁹ temperature derivatives at 298 K from finite differences using the temperatures 278, 298, and 323 K. κ_T was determined from a finite difference using 10 ns simulations with $\pm 1\%$ density differences. The experimental data were taken from Wagner et al.⁴⁷

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Eur Biophys J (2005) 34: 273–284
DOI 10.1007/s00249-004-0448-6

ARTICLE

Chris Oostenbrink · Theerza A. Soares
Nico F. A. van der Vegt · Wilfred F. van Gunsteren

Validation of the 53A6 GROMOS force field

Table 3 Comparison of structural parameters obtained from experimental and simulation structures of the DNA dodecamer

	45A4	53A6	NMR	X-ray
Local base-pair parameters				
Shear (nm)	-0.002	0.007	0.0	-0.018
Stretch (nm)	-0.017	-0.017	-0.039	-0.015
Stagger (nm)	-0.035	-0.033	-0.022	-0.002
Buckle (°)	0.0	-0.6	-0.1	2.5
Propeller (°)	-7.1	-7.1	-12.3	-19.0
Opening (°)	-2.0	-2.2	2.1	3.0
Sugar-ring puckering				
C3'-endo (%)	5	4	0	4
C4'-exo (%)	4	4	0	0
O4'-endo (%)	20	19	0	12
C1'-exo (%)	44	44	63	42
C2'-endo (%)	23	25	37	42
C3'-exo (%)	2	2	0	0
C2'-exo (%)	1	1	0	0
Pseudorotation				
Phase (°)	118.8	121.1	132.1	133.7
Amplitude (°)	42.5	42.4	29.3	41.3
Local base-pair helical parameters				
X-displacement (nm)	-0.349	-0.321	-0.128	-0.052
Y-displacement (nm)	0.004	-0.009	0.001	-0.022
Rise (nm)	0.278	0.300	0.329	0.329
Inclination (°)	15.5	14.2	4.6	5.0
Tip (°)	-0.2	-0.1	0.0	1.3
Helical twist (°)	33.4	33.3	35.1	37.8
Local base-pair step parameters				
Shift (nm)	-0.002	0.003	0.0	0.002
Slide (nm)	-0.109	-0.092	-0.046	0.004
Rise (nm)	0.342	0.357	0.338	0.341
Tilt (°)	0.1	0.1	0.0	-0.9
Roll (°)	8.4	7.8	2.5	3.4
Twist (°)	30.4	30.5	34.6	36.1

Averages are over all bases, sugars, base pairs and inter-base-pair parameters observed in 4 ns of simulation (parameter sets 45A3 and 53A6), five NMR model structures (Jandra et al. 2000) or the X-ray structure (Shui et al. 1998). Local base-pair parameters, helical parameters and step parameters were calculated using the three DNA definitions (Lu and Olson 2003; Olson et al. 2001). Sugar puckering, pseudorotation phase and pucker amplitudes are according to Altona et al. (1968) and Altona and Sundaralingam (1972)

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Ferramentas disponíveis



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Click2Drug contains a comprehensive list of computer-aided drug design (CADD) software, databases and web servers. These tools are classified according to their application field, trying to cover the whole drug design pipeline. If you think that an interesting tool is missing in this list, please contact us.

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AutoDock

<https://autodock.scripps.edu/>



AutoDock Vina

<https://vina.scripps.edu/>



https://dock.compbio.ucsf.edu/DOCK_6/index.htm



GOLD

<https://www.ccdc.cam.ac.uk/solutions/csd-discovery/components/gold/>



<https://www.rosettacommons.org/software/servers>

Mais populares

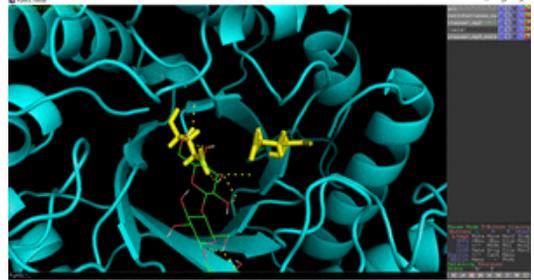
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Exemplo de resultados

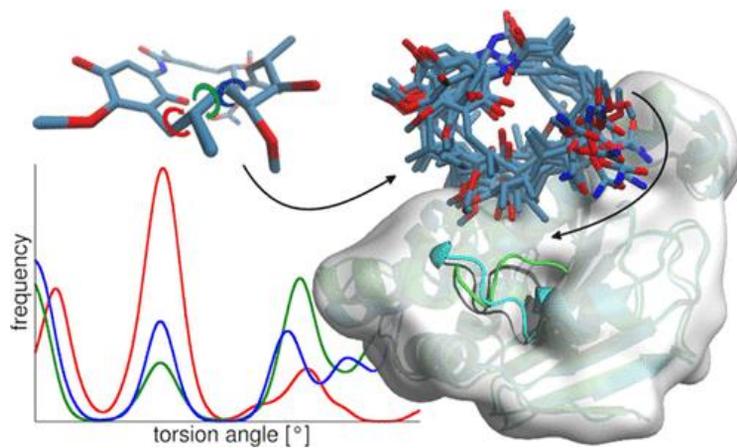


AutoDock Vina

mode	affinity (kcal/mol)	dist from best mode rmsd l.b. rmsd u.b.	
1	-7.5	0.000	0.000
2	-7.3	1.764	3.961
3	-7.2	3.178	10.371
4	-6.9	1.668	2.928
5	-6.7	3.877	7.199
6	-6.6	1.571	9.263
7	-6.6	3.307	10.436
8	-6.5	2.670	5.733
9	-6.4	1.168	2.462



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Diferentes opções

1) Modelo (rígido ou flexível)

2) "Scoring function"

$$f(x)$$

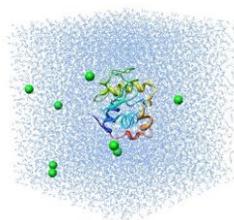
Modelagem

Qual campo de força?

simplificação
redução # sítios



precisão
relevância prática



Solução do modelo

Dinâmica molecular?
Dinâmica de Langevin?
Algoritmo genético?
Monte Carlo?
Poisson-Boltzmann?

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Table 1. Molecular docking software.

Software	Posing	Scoring	Availability	Reference
Vina	Iterated Local Search + BFGS Local Optimiser	Empirical/ Knowledge-Based	Free (Apache License)	Trott, 2010 [3]
AutoDock4	Lamarckian Genetic Algorithm, Genetic Algorithm or Simulated Annealing	Semiempirical	Free (GNU License)	Morris, 2009; Huey, 2007 [31,32]
Molegro/MolDock	Differential Evolution (Alternatively Simplex Evolution and Iterated Simplex)	Semiempirical	Commercial	Thomsen, 2006 [9]
Smina	Monte Carlo stochastic sampling + local optimisation	Empirical (customisable)	Free (GNU License)	Koes, 2013 [33]
Plants	Ant Colony Optimisation	Empirical	Academic License	Korb, 2007; Korb, 2009 [34,35]
ICM	Biased Probability Monte Carlo + Local Optimisation	Physics-Based	Commercial	Abagyan, 1993; Abagyan, 1994 [36,37]
Glide	Systematic search + Optimisation (XP mode also uses anchor-and-grow)	Empirical	Commercial	Friesner, 2004 [38]
Surflex	Fragmentation and alignment to idealised molecule (Protomol) + BFGS optimisation	Empirical	Commercial	Jain, 2003; Jain 2007 [39,40]
		Physics-based (GoldScore), Empirical		Jones 1997



Docking

Como escolher o melhor método?

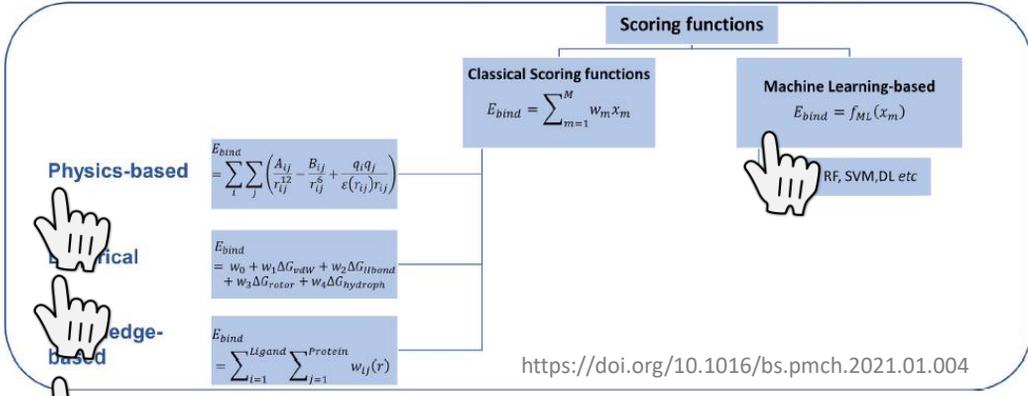
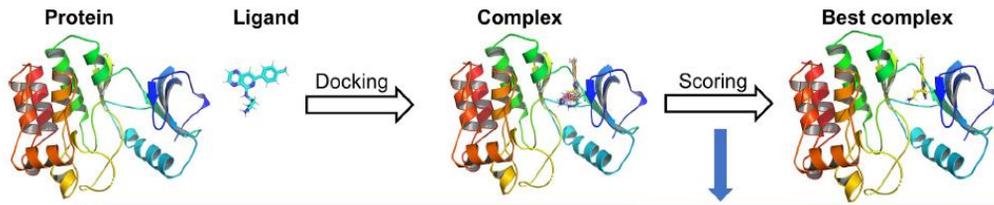


		Correlation		Entire set		Small		Large		Hydrophilic		Hydrophobic		Proteins	
		Pearson	Spearman	RMSD	% Pairs	RMSD	% Pairs	RMSD	% Pairs	RMSD	% Pairs	RMSD	% Pairs	RMSD	% Pairs
Top score															
Best docking program	Program	eHits	eHits	GOLD	GOLD	GOLD	GOLD	GOLD	GOLD	GOLD	GOLD	eHits	eHits	GOLD	GOLD
	Results	0.38	0.29	2.68	58.45	1.96	67.11	3.50	48.45	2.30	65.67	2.91	48.33	4.03	46.50
Second best program	Program	Surflex	Surflex	eHits	eHits	eHits	eHits	eHits	eHits	eHits	eHits	GOLD	GOLD	AutoDock	AutoDock
	Results	0.33	0.22	2.76	54.02	1.96	64.80	3.59	37.71	2.62	50.96	3.06	50.28	4.06	46.86
Averaged results*	Results	0.23	0.16	3.63	45.13	2.50	56.69	4.53	34.84	3.26	49.19	4.02	42.10	5.45	36.80
MetaPose	Results	0.42	0.43	2.57	62.67	1.75	70.60	3.11	52.25	2.08	67.31	2.63	57.90	3.59	50.00
MetaScore	Results	0.48	0.47	3.43	52.70	2.17	62.47	3.70	42.81	2.67	56.65	3.19	48.46	4.36	28.78
VoteDock	Results	0.49	0.50	2.20	68.67	1.58	76.18	2.82	59.04	1.90	72.49	2.28	62.70	3.26	57.00
Best pose															
Best docking program	Program	eHits	eHits	GOLD	GOLD	eHits	eHits	GOLD	GOLD	GOLD	GOLD	eHits	eHits	GOLD	GOLD
	Results	0.29	0.38	1.66	73.83	1.20	83.30	2.01	66.38	1.43	79.05	1.74	72.42	2.63	61.77
Second best program	Program	Surflex	Surflex	eHits	eHits	GOLD	GOLD	Surflex	Surflex	eHits	eHits	Surflex	Surflex	LigandFit	LigandFit
	Results	0.22	0.30	1.69	73.80	1.31	81.07	2.42	64.82	1.64	72.30	2.04	69.04	2.71	49.87
Averaged results*	Results	0.16	0.19	2.22	65.63	1.54	76.92	2.99	53.54	2.02	68.46	2.52	62.44	3.64	48.60
MetaPose	Results	0.30	0.34	1.54	82.53	0.97	87.83	1.86	74.62	1.29	84.64	1.59	77.89	1.86	74.62
MetaScore	Results	0.39	0.44	1.29	82.12	0.93	89.91	1.65	72.35	1.17	84.52	1.41	77.73	1.94	65.84
VoteDock	Results	0.40	0.45	1.64	78.91	1.35	83.75	2.09	70.19	1.46	80.01	1.83	74.05	2.34	65.17

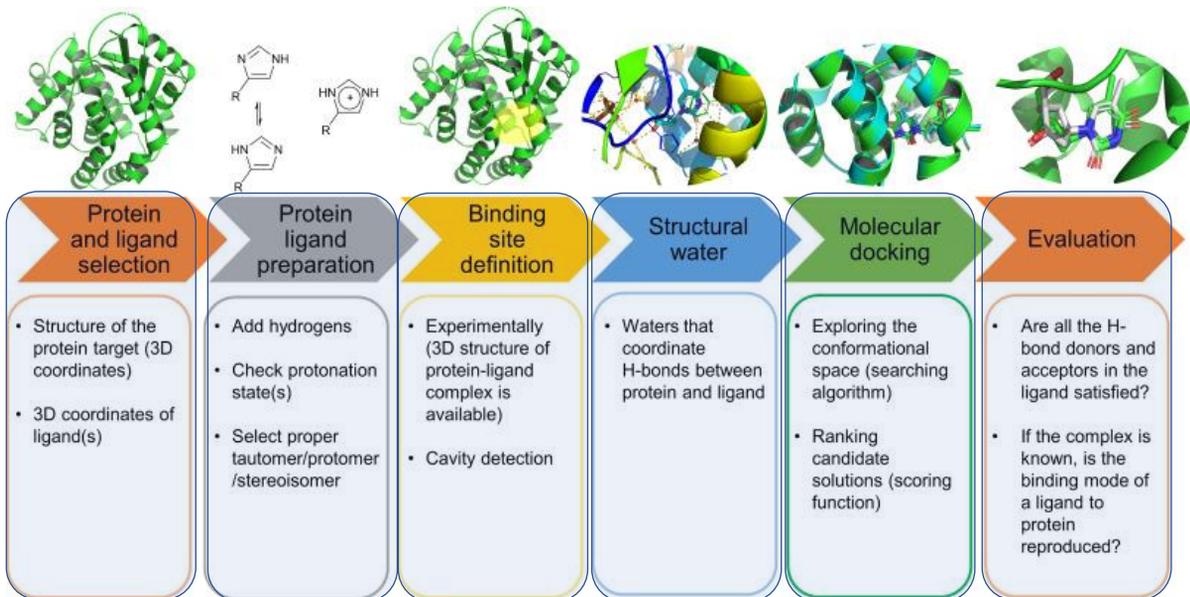
Consenso!

10.1002/jcc.21642

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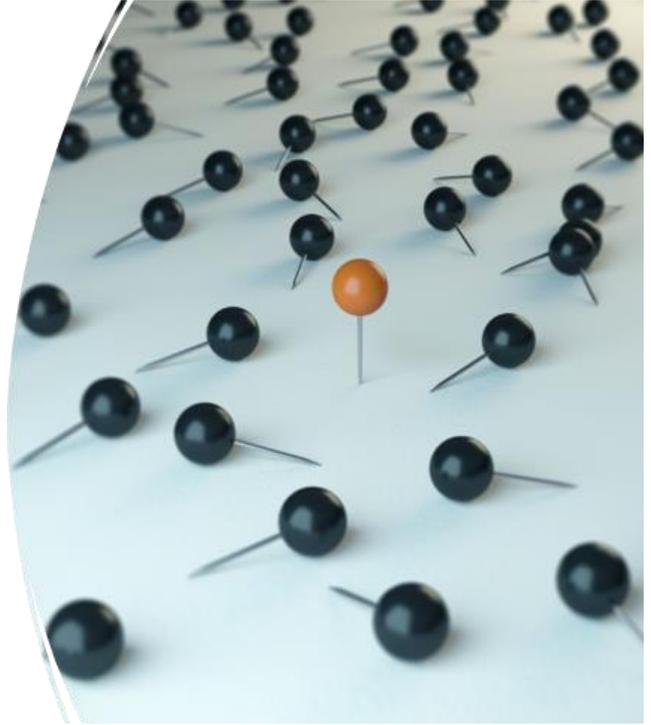
<https://doi.org/10.1016/bs.pmch.2021.01.004>

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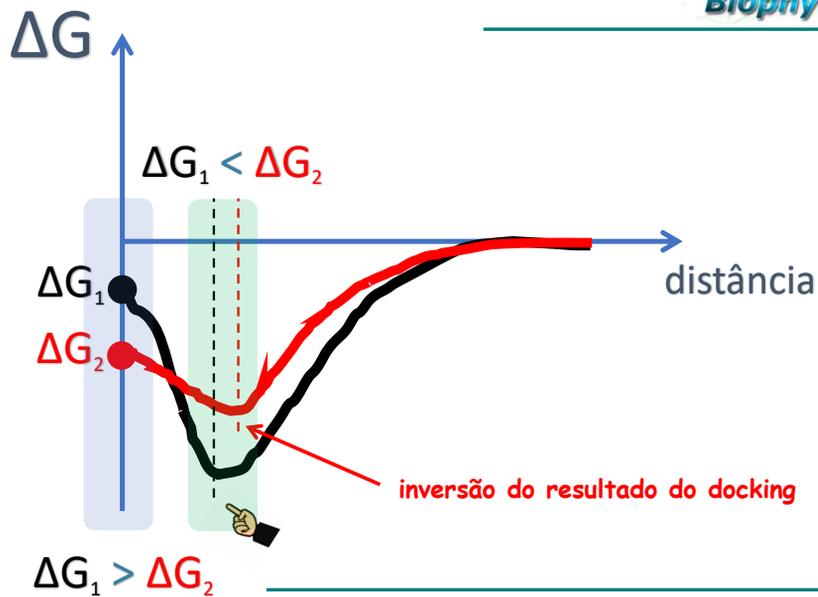


Docking

Outro importante
detalhe...



Laboratory of Computational
Biophysical Chemistry



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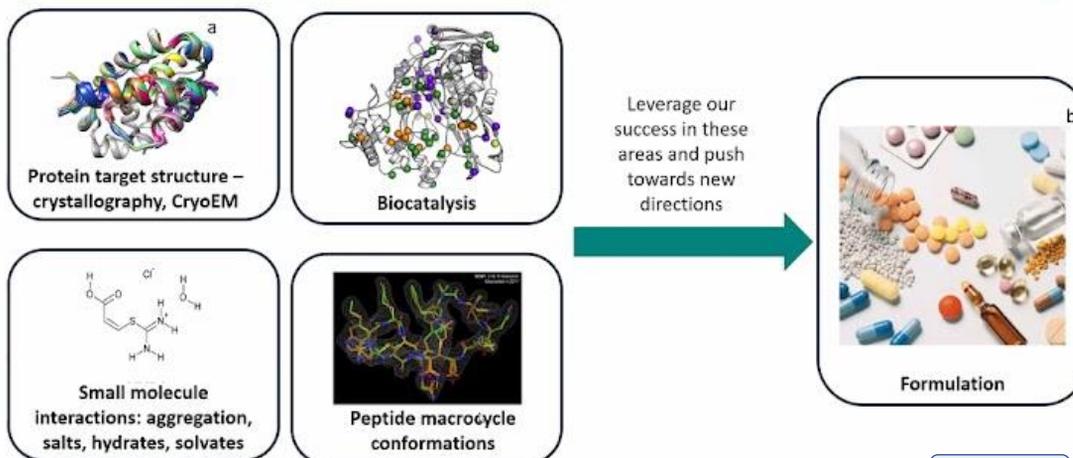
Dinâmica molecular

Resolução numérica de equações de movimento



Fernando Barroso, FCFRP/USP

Molecular dynamics across the discovery/development pipeline



a: <https://news.mit.edu/>
 b: <https://pharmaxchange.info/>



Fernando Barroso, FCFRP/USP

Studies in Molecular Dynamics. I. General Method*

B. J. ALDER AND T. E. WAINWRIGHT

Lawrence Radiation Laboratory, University of California, Livermore, California

(Received February 19, 1959)

A method is outlined by which it is possible to calculate exactly the behavior of several hundred interacting classical particles. The study of this many-body problem is carried out by an electronic computer which solves numerically the simultaneous equations of motion. The limitations of this numerical scheme are enumerated and the important steps in making the program efficient on the computers are indicated. The applicability of this method to the solution of many problems in both equilibrium and nonequilibrium statistical mechanics is discussed.

Computer "Experiments" on Classical Fluids. I. Thermodynamical Properties of Lennard-Jones Molecules*

LOUP VERLET†

Belfer Graduate School of Science, Yeshiva University, New York, New York

(Received 30 January 1967)

The equation of motion of a system of 864 particles interacting through a Lennard-Jones potential has been integrated for various values of the temperature and density, generally, to a fluid state. The equilibrium properties have been calculated and are shown to agree very well with the corresponding properties of argon. It is concluded that, to a good approximation, the equilibrium state of argon can be described through a two-body potential.

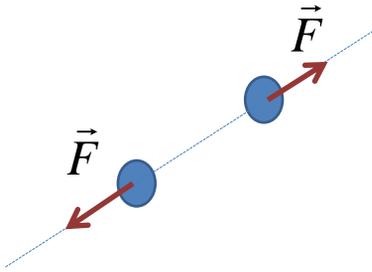
Table 9: History and extrapolated future of computer simulations of molecular dynamics. The future is deduced from extrapolation based on an observed increase of computing speed of a factor 10 every 5 years over the past decades (see Figure 31).

Table with 3 columns: Year, Molecular system (type, size), Length of the simulation [s]. Rows include 1957 first molecular dynamics simulation, 1964 atomic liquid (argon), 1971 molecular liquid (water), 1977 protein in a vacuum, 1983 protein in water, 1989 protein-DNA complex in water, 1997 polypeptide folding in solvent, 2001 micelle formation, 200x folding of a small protein, and a section 'And the future...' with rows for 2001 biomolecules in water, 2029 biomolecules in water, 2034 E. coli bacteria, 2056 mammalian cell, 2080 biomolecules in water, and 2172 human body.



Fernando Barroso, FCFRP/USP

Angew. Chem. Int. Ed. 2006, 45, 4064-4092



$$\vec{F} = m\vec{a} \qquad \vec{F} = -\nabla V(\vec{r})$$

$$\left. \begin{aligned} F_i &= m_i a_i \\ F_i &= -\nabla_i V \end{aligned} \right\} -\frac{dV}{dr_i} = m_i \frac{d^2 r_i}{dt^2}$$

$F_i = m_i a_i$

Posição
 $r_i(t)$

Energia Potencial

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$$\vec{F} = m \frac{dv}{dt} \xrightarrow{\text{Forças conservativas}} v(t) = \frac{F}{m}t + v_0$$

Equação de movimento

$$v(t) = \frac{dr}{dt} = \frac{F}{m}t + v_0$$

$$r(t) = \frac{F}{2m}t^2 + v_0t + r_0$$

Solução (F = cte)

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Numericamente:

$$r(t + \delta t) = r(t) + v(t)\delta t + \frac{1}{2}a(t)\delta t^2 + \dots$$

$$v(t + \delta t) = v(t) + a(t)\delta t + \frac{1}{2}b(t)\delta t^2 + \dots$$

$$a(t + \delta t) = a(t) + b(t)\delta t + \dots$$

- É fundamental a estabilidade numérica!
- Conservação de energia total e momento para grande δt

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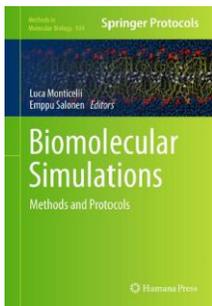
2.2. The Verlet Algorithm

The Verlet algorithm (10) is obtained by adding together the Taylor expansions of $\vec{r}(t + \Delta t)$ and $\vec{r}(t - \Delta t)$ about time t . It reads as follows:

$$\vec{r}_i(t + \Delta t) = 2\vec{r}_i(t) - \vec{r}_i(t - \Delta t) + \frac{1}{m}\vec{F}_i(t)(\Delta t)^2. \quad (37)$$

Não envolve velocidades, más,....

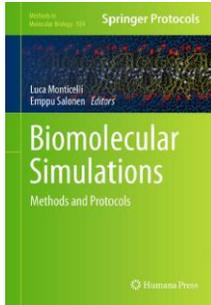
$$\vec{v}_i(t) = \frac{\vec{r}_i(t + \Delta t) - \vec{r}_i(t - \Delta t)}{2\Delta t}.$$



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2.3. The Leapfrog Algorithm



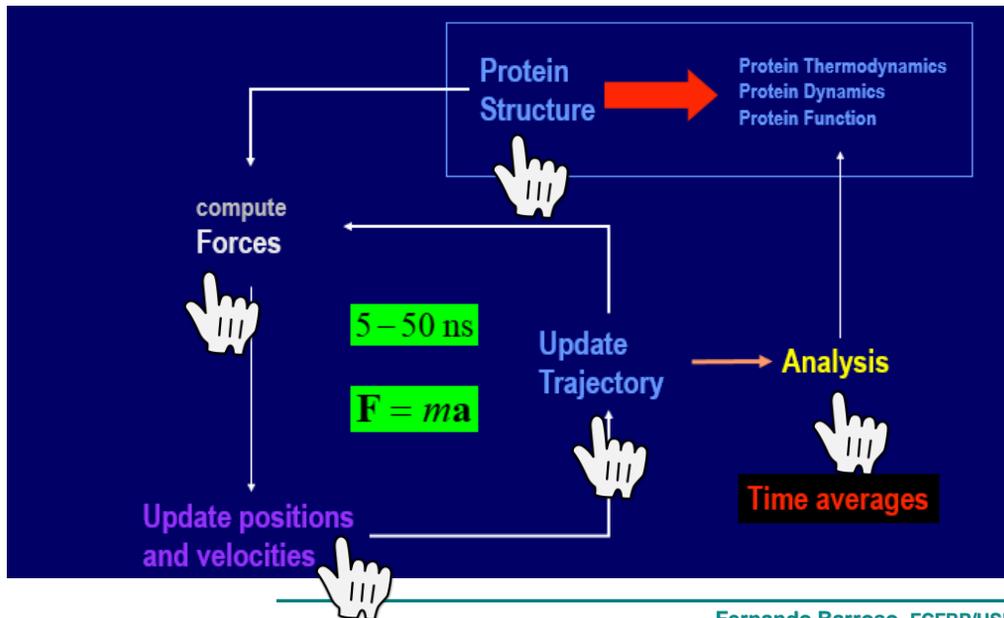
The leapfrog algorithm (12) is a modification of the Verlet algorithm, given by

$$\vec{v}_i\left(t + \frac{\Delta t}{2}\right) = \vec{v}_i\left(t - \frac{\Delta t}{2}\right) + \frac{1}{m_i} \vec{F}_i(t) \Delta t \quad (39)$$

$$\vec{r}_i(t + \Delta t) = \vec{r}_i(t) + \vec{v}_i\left(t + \frac{\Delta t}{2}\right) \Delta t. \quad (40)$$

$$\vec{v}_i = \frac{\vec{v}_i\left(t + \frac{\Delta t}{2}\right) + \vec{v}_i\left(t - \frac{\Delta t}{2}\right)}{2}$$

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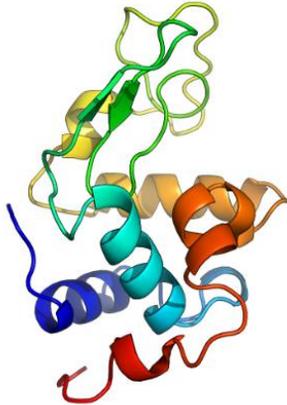


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GROMACS Tutorial

Lysozyme in Water

Justin A. Lemkul, Ph.D.
Virginia Tech Department of Biochemistry



- ✓ Gromacs
- ✓ Charmm
- ✓ Amber
- ✓ Discover, Insight
- ✓ Sigma, Tripos, ...
- ✓ NAMD
- ✓ Tinker
- ✓ Lammmps...



<http://www.mdtutorials.com/gmx/lysozyme/index.html>

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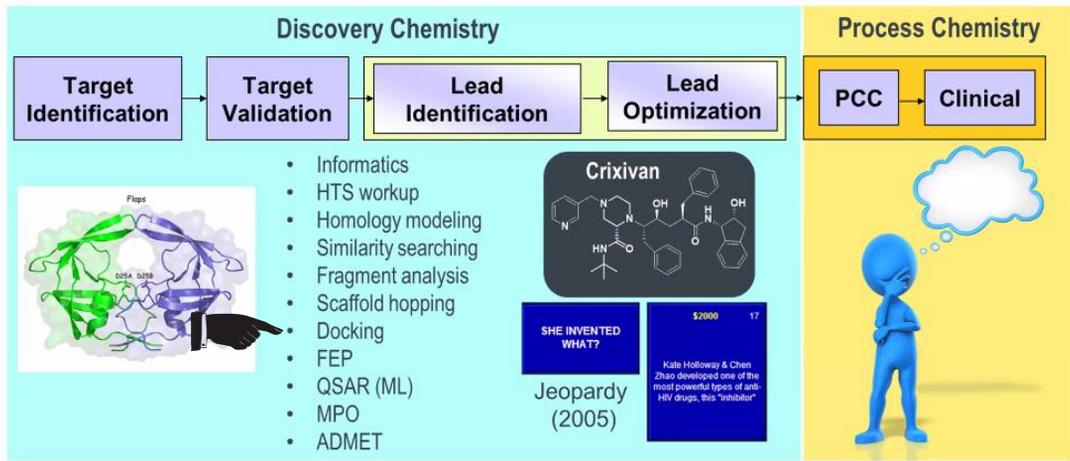


Bons livros para DM

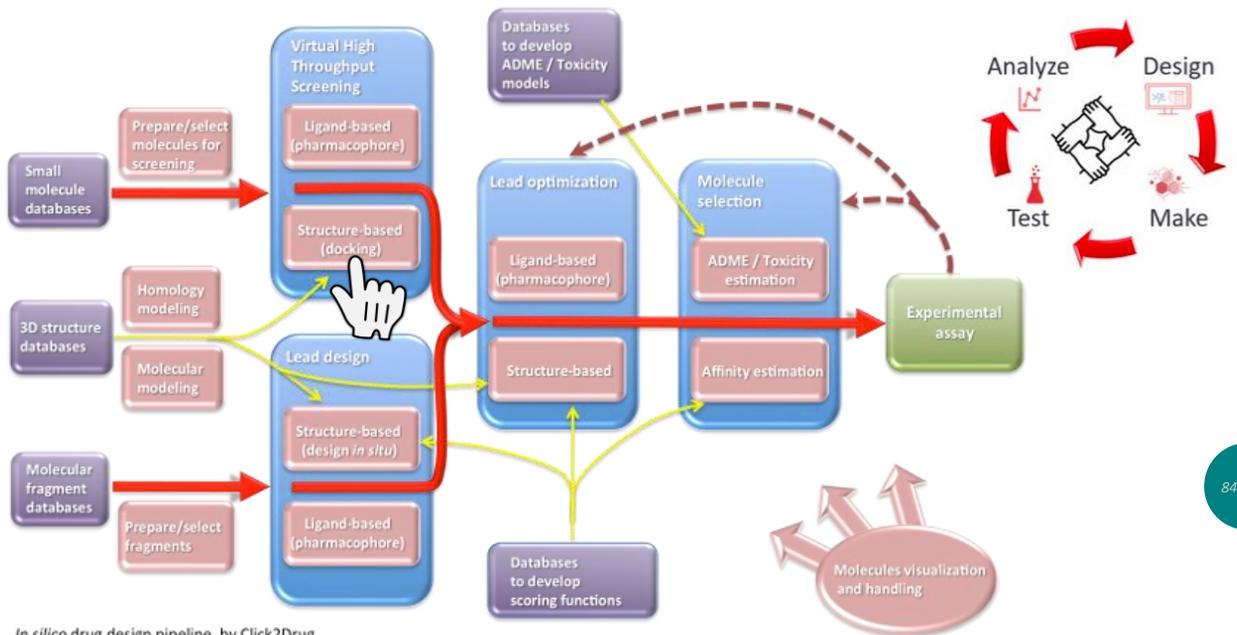
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Research pipeline



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In silico drug design pipeline, by Click2Drug

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(*)
$$W = - \int P_{ex} \cdot dV$$

$P_{ex} = cte$
(Proc. irreversível)

$$W = - P_{ex} \int dV = - P_{ex} \Delta V$$

$P_{ex} \neq cte$
(max trabalho)
 $P_{ex} = P_{in} - dP$
(Proc. REVERSÍVEL)

$$W = - \int P_{in} \cdot dV$$

$P(V) \neq cte$

Gás ideal
 $PV = nRT$

$$= - \int \left(\frac{nRT}{V} \right) \cdot dV$$

(Aula de Físico-química, agosto de 2019, FCFRP/USP)

Obrigado!

Fernando Barroso (flbarroso@usp.br)