



ΔH	SiO2	=	30833,33 kJ	HT-H298	
ΔH	Fe2O3	=	546250 kJ	HT-H299	
ΔH	N2	=	-687500 kJ	N2	(H127-H298)-(H827-H298)= H127-H827
PT			-110416,7 kJ	(26.415,47)	kcal

$$\Delta H_{aqu,min} + \Delta H_{resfr,N2} + PT = 0$$

$$\Delta H_{aqu,SiO2} + \Delta H_{aqu,Fe2O3} + \Delta H_{resfr,N2} + PT = 0$$