PQI-3401 – Engenharia de Reações Químicas II São Paulo, SP, janeiro/2021

ANALYSIS OF NON-IDEAL REACTORS

ANÁLISE DE REATORES NÃO IDEAIS

Reinaldo Giudici



UNIVERSIDADE DE SÃO PAULO ESCOLA POLITÉCNICA DEPARTAMENTO DE ENGENHARIA QUÍMICA

Analysis of Nonideal Reactors OBJECTIVES

- To study deviations from the ideal reactors (CSTR, PFR)
 - Identification/diagnosis (qualitative)
 - Models for nonideal reactors (quantitative)

- Ideal reactors
 - Limiting behavior (total mixing/no-mixing)
 - Easier to math. model
 - Fairly good approx. for many real systems

Ref.

H.S. Fogler, Elements of Chemical Reaction Engineering, 3rd ed., Prentice-Hall, 1999. (Cap. 13 & 14)

O. Levenspiel, Chemical Reaction Engineering, 3rd ed., Wiley, 1999

Deviations of the ideal reactor behavior







4

Factors affecting nonideal behavior

- the **RTD** or **residence time distribution** of material which is flowing through the vessel
- the state of aggregation of the flowing material, its tendency to clump and for a group of molecules to move about together



• the earliness and lateness of mixing of material in the vessel.





RTD Residence Time Distribution



How to measure the RTD?



Stimulus-response experiment using a tracer

Tracer requirements:

Physical properties similar to the reacting mixture Completely soluble in the reacting mixture Should not adsorb on the reactor walls and other surfaces Easily detectable Usually nonreactive



How to measure the RTD ?





The pulse experiment (input = pulse)



10 O. Levenspiel, Chemical Reaction Engineering, 3rd ed., Wiley, 1999.

The pulse experiment





11 O. Levenspiel, Chemical Reaction Engineering, 3rd ed., Wiley, 1999.

The step experiment (input = step)





O. Levenspiel, Chemical Reaction Engineering, 3rd ed., Wiley, 1999.



Relationship between C(t) = response to pulse F(t) = response to stepand E(t) = RTD

PULSE EXPERIMENT

all tracer entered at t = 0then amount of tracer exiting the reactor at time t has RT = t

 $\begin{pmatrix} \text{fraction of tracer} \\ \text{exiting the reactor} \\ \text{at time } t \end{pmatrix} = \begin{pmatrix} \text{fraction of} \\ \text{tracer that} \\ \text{has } \mathbf{RT} = t \end{pmatrix}$

C(t) = E(t)

STEP EXPERIMENT tracer exiting the reactor at time t_1 entered in the reactor at a time between 0 and t_1 (fraction of fraction of tracer exiting the reactor |=|tracer at time = t_1 with $\mathbf{RT} \leq \mathbf{t}_1$ $F(t_1) = \int_{-\infty}^{t_1} E(t) dt$ $E(t) = \frac{dF(t)}{L}$



15 O. Levenspiel, Chemical Reaction Engineering, 3rd ed., Wiley, 1999.

Characteristics of the RTD curve

$$\begin{pmatrix} \text{mean} \\ \text{residence} \\ \text{time} \end{pmatrix} = t_m = \frac{\int_0^\infty t.E(t).dt}{\int_0^\infty E(t).dt} = \int_0^\infty t.E(t).dt$$
$$\text{variance} = \sigma^2 = \int_0^\infty (t - t_m)^2.E(t).dt$$
$$\text{skewness} = s^3 = \frac{1}{\sigma^{3/2}} \int_0^\infty (t - t_m)^3.E(t).dt$$

for a fluid of constant density (constant volumetric flow rate)

$$\begin{pmatrix} \text{mean} \\ \text{residence} \\ \text{time} \end{pmatrix} = t_m = \tau = \frac{V}{v_o}$$



RTD of the ideal reactors



RTD of the ideal reactors



Assignment #1

Derive the RTD curve (E(t)) for the ideal reactors (PFR and CSTR)

Use "though experiment" and mass balances

Using RTD to detect flow nonidealities



19

Using RTD to detect flow nonidealities





Figure 12.3 Misbehaving plug flow reactors.

O. Levenspiel, Chemical Reaction Engineering, 3rd ed., Wiley, 1999.

Using RTD to detect flow nonidealities





Figure 12.4 Misbehaving mixed flow reactors.

21 O. Levenspiel, Chemical Reaction Engineering, 3rd ed., Wiley, 1999.



MODELS FOR NONIDEAL REACTORS

Laminar flow reactor



23



Fluid close to the



$$t(r) = \frac{L}{u(r)} = \frac{L\pi R^2}{2u_{medio}\pi R^2 \left[1 - (r/R)^2\right]} = \frac{V}{2v_0 \left[1 - (r/R)^2\right]} = \frac{\tau}{2\left[1 - (r/R)^2\right]}$$

$$dt = \frac{\tau}{2R^2} \frac{2.r.dr}{\left[1 - \left(r/R\right)^2\right]^2} \frac{(\tau/2)^2}{(\tau/2)^2} = \frac{4}{\tau R^2} \left[\frac{\tau/2}{1 - \left(r/R\right)^2}\right]^2 r.dr = \frac{4t^2}{\tau R^2} r.dr$$

 $\begin{cases} \text{fraction of fluid} \\ \text{flowing between} \\ \text{r and } \text{r} + \text{dr} \end{cases} = \frac{dv}{v_o} = \frac{u(r).2\pi . r.dr}{v_o} = \frac{L}{t} \frac{2\pi}{v_o} \frac{\tau R^2}{4t^2} dt = \frac{\tau^2}{2t^3} dt$

Fluid close to the
wall moves slowly
Fastest flowing fluid
element is in the center
(fraction of fluid
flowing between
r and r + dr) =
$$\frac{\tau^2}{2t^3} dt = \begin{pmatrix} \text{fraction of fluid} \\ \text{flowing with RT} \\ \text{between} \\ \text{t and t + dt} \end{pmatrix} = E(t).dt$$

minimum residence time (at the tube center) is

$$t_{\min} = \frac{L}{u_{\max}} = \frac{L}{2.u_{medio}} \left(\frac{\pi R^2}{\pi R^2}\right) = \frac{V}{2v_0} = \frac{\tau}{2}$$

$$E(t) = \begin{cases} 0 & \text{for } t < \frac{\tau}{2} \\ \frac{\tau^2}{2t^3} & \text{for } t < \frac{\tau}{2} \end{cases} \qquad F(t) = \int_0^t E(t)dt = \begin{cases} 0 & \text{for } t < \frac{\tau}{2} \\ 1 - \frac{\tau^2}{4t^2} & \text{for } t < \frac{\tau}{2} \end{cases}$$

Fluid close to the wall moves slowly Fastest flowing fluid element is in the center Tube wall

$$E(t) = \begin{cases} 0 & \text{for } t < \frac{\tau}{2} \\ \frac{\tau^2}{2t^3} & \text{for } t < \frac{\tau}{2} \end{cases}$$

$$F(t) = \int_{0}^{t} E(t)dt = \begin{cases} 0 & \text{for } t < \frac{\tau}{2} \\ 1 - \frac{\tau^{2}}{4t^{2}} & \text{for } t < \frac{\tau}{2} \end{cases}$$





Tanks-in-series Model





$$\theta = \frac{t}{\tau}$$
 $E_{\theta} = \tau . E(t)$ $E_{\theta} = \frac{N(N\theta)^{N-1}}{(N-1)!} \exp(-N\theta)$

 $\frac{\sigma^2}{\tau^2} = \frac{1}{N}, \quad N = \text{parameter that quantifies nonideality} \begin{cases} N = 1 & \text{CSTR} \\ 1 < N < \infty & \text{real reactor} \\ N \to \infty & \text{PFR} \end{cases}$



Tanks-in-series Model





Figure 14.3 Properties of the RTD curve for the tanks-in-series model.



Axial dispersion Model



Axially-Dispersed Tubular Reactor (ADPFR)

 $J_i = -D_A \frac{dC_i}{I}$



- No changes in radial and angular direction, only axial variations (z)
- Flow in direction $z (v_r=0, v_{\theta}=0)$ and some mixing in axial direction (similar to diffusion)



• Steady state

z=0

 $\frac{dF_i}{dV} = \frac{dF_i}{A_c.dz} = r_i \Longrightarrow \frac{dF_i}{dz} = A_c r_i$

 $F_{i} = \underbrace{A_{c} u C_{i}}_{\text{convective}} + \underbrace{A_{c} J_{i}}_{\text{axial}}$

flow



- Mixing flow in the axial direction
 - Account for different phenomena
 - D_A = effective axial dispersion coefficient

$$\frac{dF_i}{dz} = A_c \frac{d}{dz} \left(v_z C_i - D_A \frac{dC_i}{dz} \right) = A_c r_i$$

mixing

flow



Reator tubular com dispersão axial

• Mesmas hipóteses do reator PFR, exceto que agora se considera a ocorrência de mistura na direção z (dispersão axial). O fluxo de mistura axial é modelado como uma difusão: $\vec{r} = \vec{r} \cdot \vec$

$$\frac{\partial C_A}{\partial t} + \vec{\nabla} \bullet (\vec{v} C_A) = \vec{\nabla} \bullet \left(D_a \vec{\nabla} C_A \right) + r_A$$

$$J_A = -D_A \vee C_A$$

$$J_A = -D_A \vee C_A$$

$$J_A = -D_A \vee C_A$$

$$Z = L$$

$$\frac{\partial C_A}{\partial t} = -v_z \frac{\partial C_A}{\partial z} + D_A \frac{\partial^2 C_A}{\partial z^2} + r_i$$

Esta equação diferencial tem, agora, derivadas de 2ª ordem, então requer 2 condições de contorno em z Condições de contorno de Danckwerts

$$z = 0$$
 $c_{A}|_{z=0} = c_{A,e} + \frac{D_{a}}{v_{z}} \frac{dc_{A}}{dz}|_{z=0}$

$$z = L \qquad \frac{dc_{A}}{dz}\Big|_{z=L} = 0$$
33



Axial Dispersion Model





Axial Dispersion Model

Axial Dispersion Model

 $\theta = \frac{t}{T} = \frac{tv}{V}$

Axial dispersion model

Figure 13.17 Experimental findings on dispersion of fluids flowing with mean axial velocity u in packed beds; prepared in part from Bischoff (1961).

Chung & Wen (1968) correlation

$$\varepsilon.Pe_L \frac{d_p}{L} = \varepsilon.\left(\frac{uL}{D_{ea}}\right)\frac{d_p}{L} = \varepsilon.\left(\frac{ud_p}{D_{ea}}\right) = 0.2 + 0.008 \,\mathrm{Re}^{0.48}$$

37

40

F(t)

41

(b)

Figure 14-11 (a) Real system; (b) model system.

$$E(t) = \left(\frac{v_b}{v_0}\right)\delta(t) + \left(\frac{v_a^2}{V_s v_0}\right)\exp\left(-\frac{v_a}{V_s}t\right) = \beta \cdot \delta(t) + \frac{(1-\beta)^2}{\alpha\tau}\exp\left(-\frac{(1-\beta)}{\alpha\tau}t\right)$$
$$F(t) = 1 - \left(\frac{v_a}{v_0}\right)\exp\left(-\frac{v_a}{V_s}t\right) = 1 - (1-\beta)\exp\left(-\frac{(1-\beta)}{\alpha\tau}t\right)$$

Assignment#2:

Draw the RTD curves E(t) for the systems shown bellow.

Explain your results (via mass balances and via "thought experiment")

• (a) a system formed by a CSTR followed by a PFR.

• (b) a system formed by a PFR followed by a CSTR.

Conversion directly from the RTD for Segregated Flow

Segregated Flow

MACROFLUID:

- mixing at the level of macroscopic portions of fluid
- molecules flow together in groups (globules) and they are not mixed until they exit the reactor
- different from a microfluid (where mixing occurs at the molecular level)
- each globule is a closed system (batch reactor)

X

mean conversion of all globules exiting the reactor

$$\begin{array}{c}
\text{conversion}\\
\text{achieved in}\\
\text{a globule}\\
\text{that spend}\\
\text{a time } t\\
\text{in the reactor}
\end{array}$$

 $X_{batch}(t)E(t)dt$

reactor

the rea

 $\begin{cases} fraction of \\ globules that \\ spend between \\ time \\ t and t + dt \\ in the reactor \end{cases}$

Assignments RTD (laboratory activity, gathering real data)

- Assignment #3: Given the experimental data (absorbance versus time) of a pulse RTD experiment (fluid = water) in a real tubular reactor (L=90 cm, D=4 cm) packed with 0,8 cm Raschig rings (void fraction of the packed bed ϕ =0,70).
- (a) Fit the axial dispersion model to the data and determine the value of Peclet number, $Pe = u.L/D_A$ and τ . Compare the obtained Peclet value with literature values (see slide 37). Use the approximated solution for the RTD of the axial dispersion model.

$$E(t) \simeq \frac{1}{\tau} \left(\frac{Pe+1}{4\pi (t/\tau)^3} \right)^{1/2} \exp \left(\frac{-(Pe+1)[1-(t/\tau)]^2}{4(t/\tau)} \right)$$

(a) Fit the N-tanks-in-series model to the exp. data and determine N and τ

$$E(t) = \frac{N\left(\frac{N-\tau}{\tau}\right)}{\tau(N-1)!} \exp\left(-N\frac{t}{\tau}\right) = \frac{t^{N-1}}{(N-1)!(\tau/N)^N} \exp\left(-N\frac{t}{\tau}\right)$$
48

$$E(t) = \frac{t^{N-1}}{(N-1)! (\tau/N)^N} \exp\left(-N\frac{t}{\tau}\right)$$

Assignment #4: Given the experimental data (absorbance versus time) of a pulse RTD experiment in a real tank reactor (fluid = water), fit the compartment model considering a dead volume and a by-pass, and determine the fraction of bypass $\beta = v_{\rm b}/v$ and the fraction of active volume $\alpha = V_s/V$

$$E(t) = \left(\frac{v_b}{v_0}\right)\delta(t) + \left(\frac{v_a^2}{V_s v_0}\right)\exp\left(-\frac{v_a}{V_s}t\right) = \beta \cdot \delta(t) + \frac{(1-\beta)^2}{\alpha\tau}\exp\left(-\frac{(1-\beta)}{\alpha\tau}t\right)$$
$$F(t) = 1 - \left(\frac{v_a}{v_0}\right)\exp\left(-\frac{v_a}{V_s}t\right) = 1 - (1-\beta)\exp\left(-\frac{(1-\beta)}{\alpha\tau}t\right)$$

