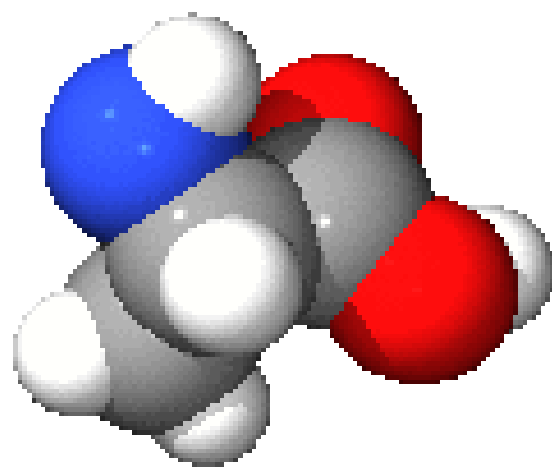
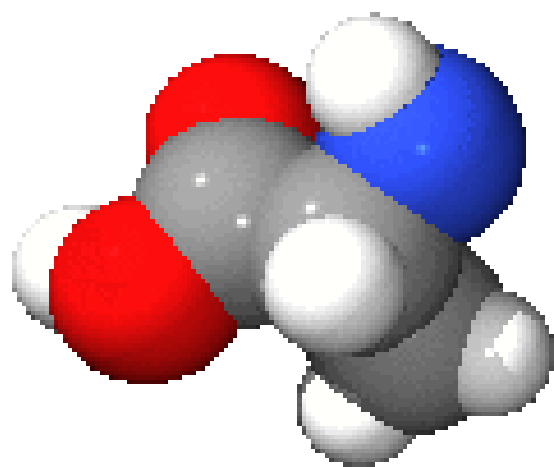
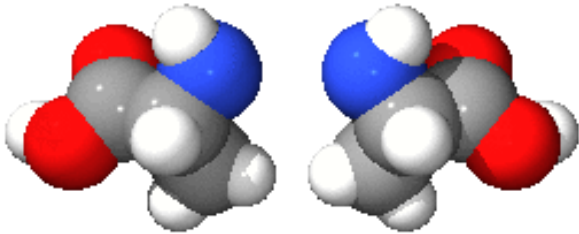




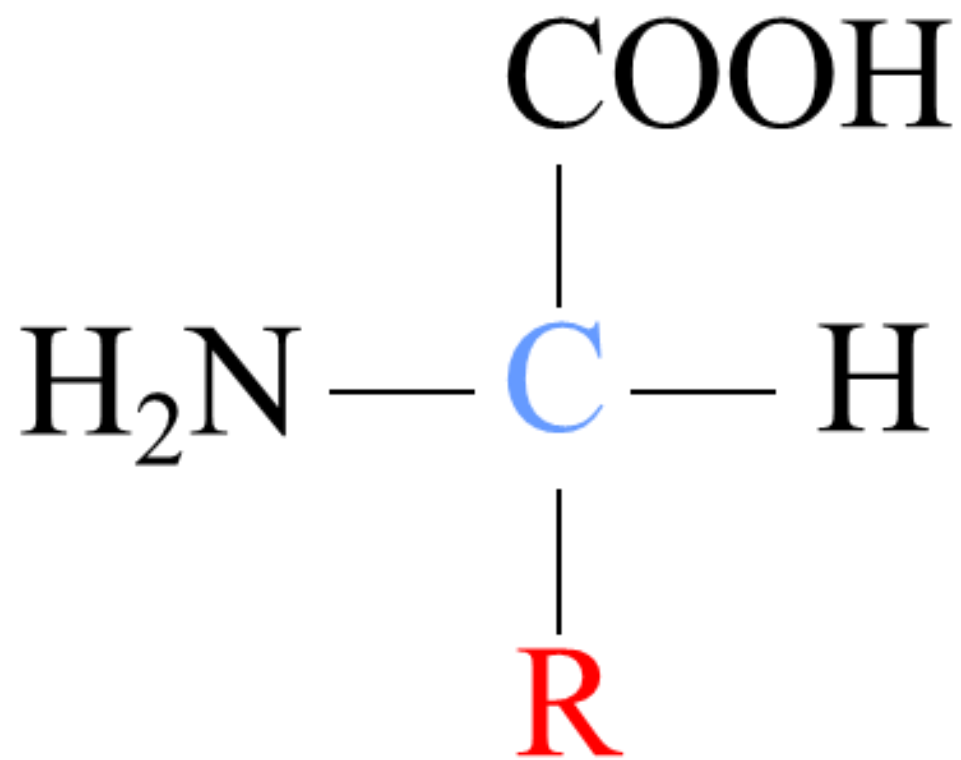
Aminoácidos

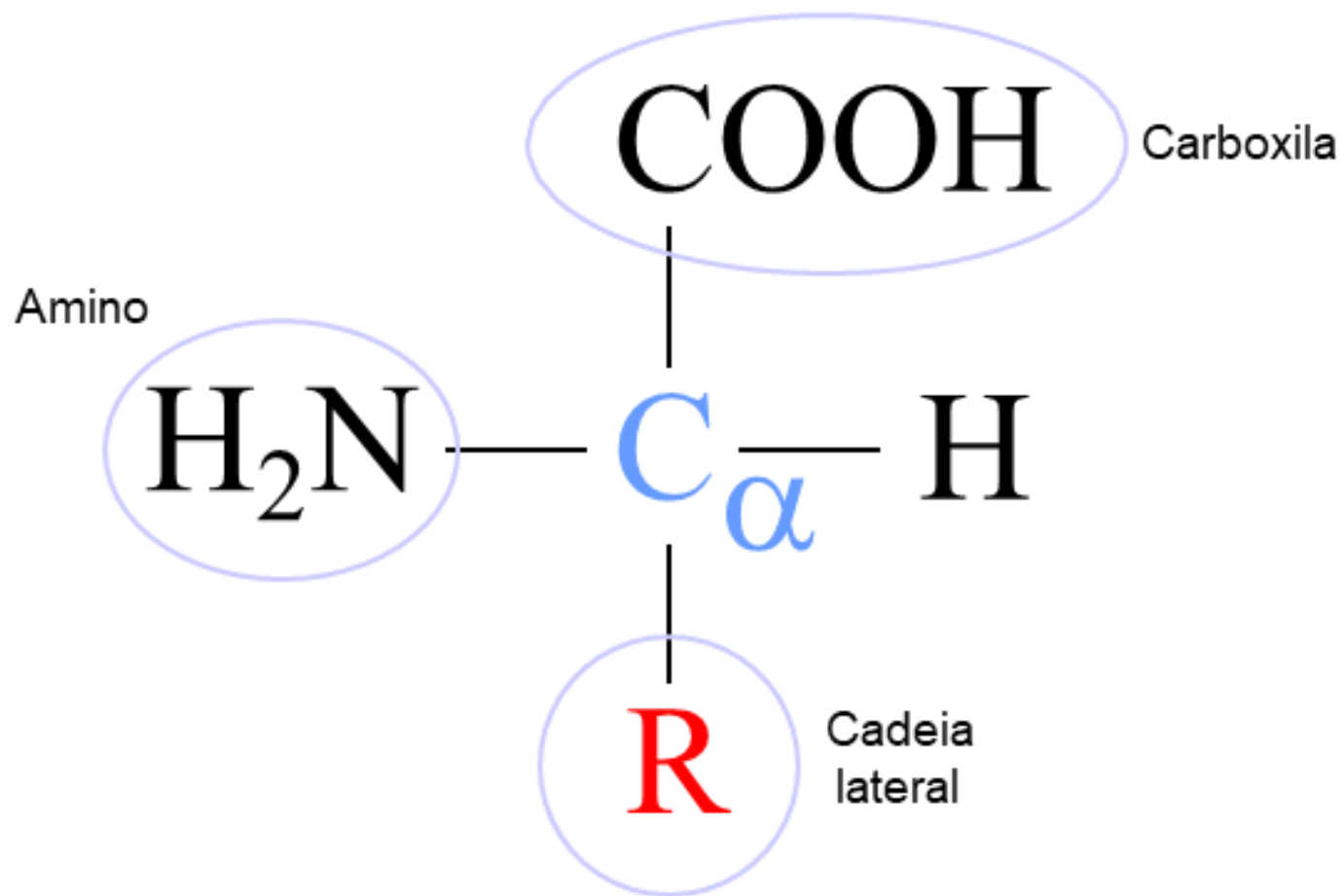


Ronaldo Bento Quaggio

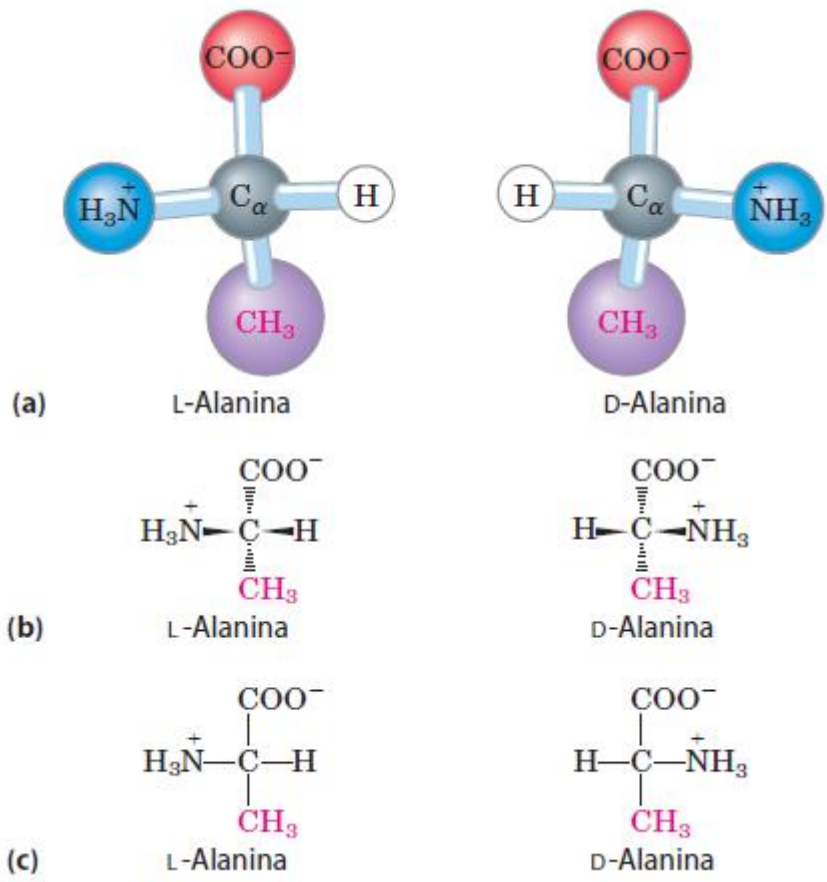
QBQ0204 - Bioquímica

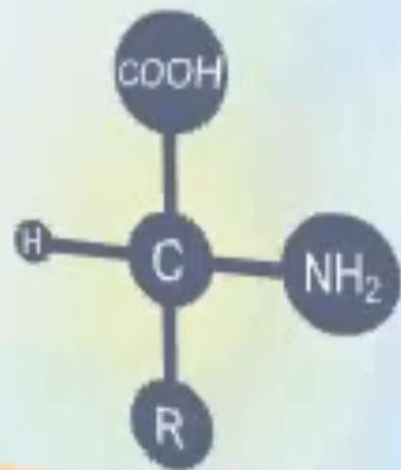
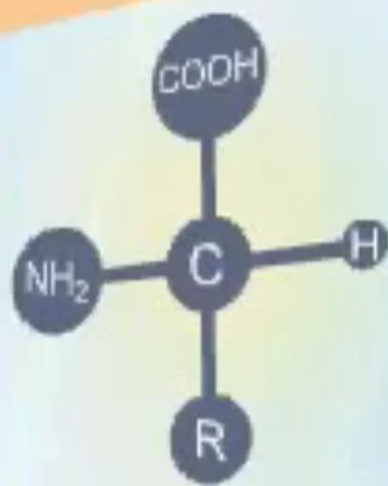
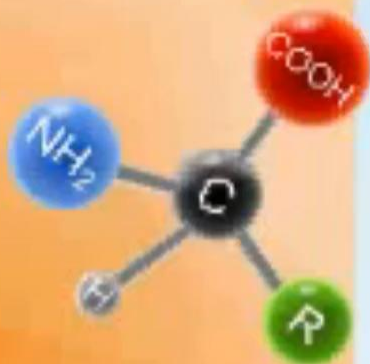
Estrutura de Biomoléculas e Metabolismo





Estereoisomerismo em α -aminoácidos.

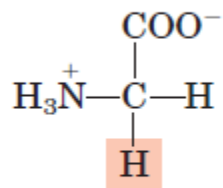




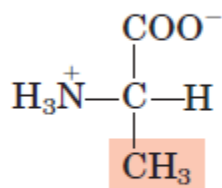
Glicina	Gly	G
Alanina	Ala	A
Valina	Val	V
Leucina	Leu	L
Isoleucina	Ile	I
Prolina	Pro	P
Metionina	Met	M
Fenilalanina	Phe	F
Tirosina	Tyr	Y
Triptofano	Trp	W

Serina	Ser	S
Treonina	Thr	T
Cisteína	Cys	C
Asparagina	Asn	N
Glutamina	Gln	Q
Arginina	Arg	R
Lisina	Lys	K
Histidina	His	H
A. Aspártico	Asp	D
A. Glutâmico	Glu	E

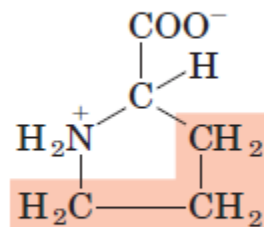
Grupos R apolares, alifáticos



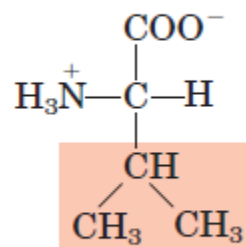
Glicina



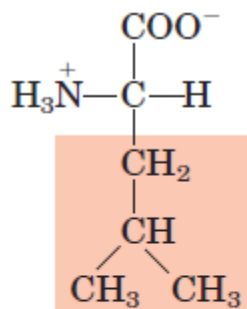
Alanina



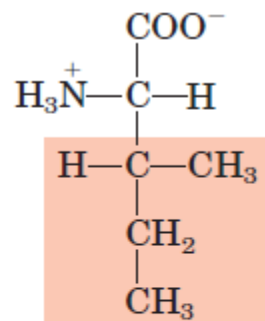
Prolina



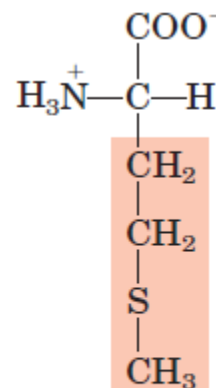
Valina



Leucina

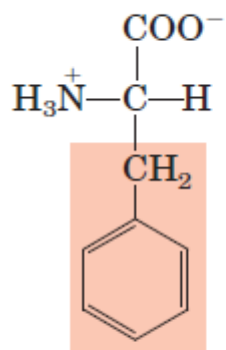


Isoleucina

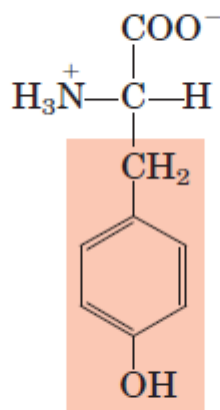


Metionina

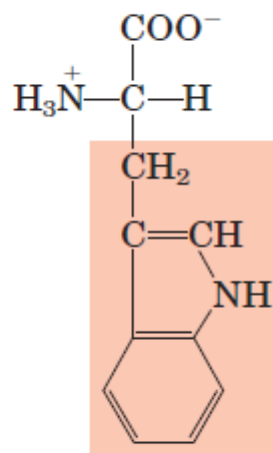
Grupos R aromáticos



Fenilalanina

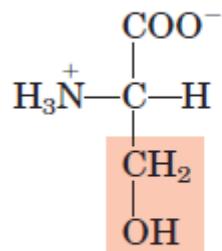


Tirosina

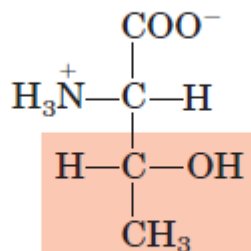


Triptofano

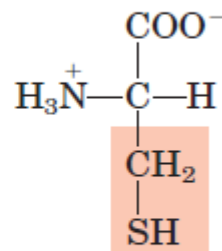
Grupos R polares, não carregados



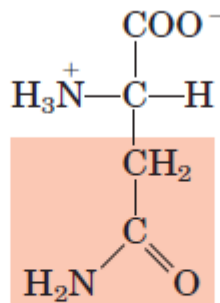
Serina



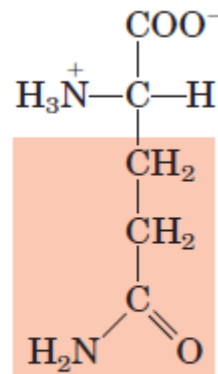
Treonina



Cisteína

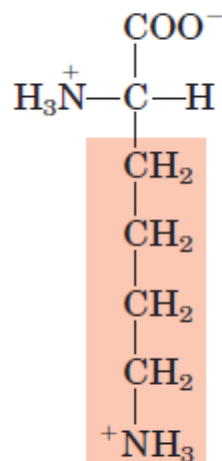


Asparagina

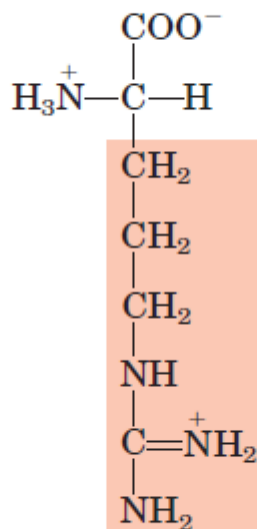


Glutamina

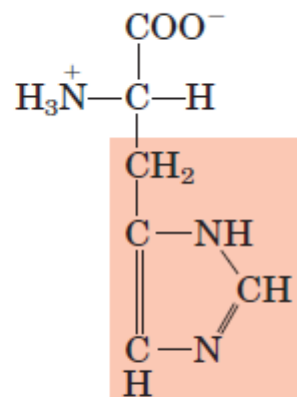
Grupos R cargados positivamente



Lisina

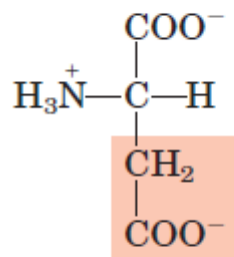


Arginina

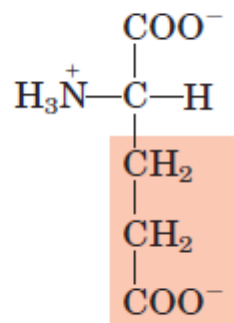


Histidina

Grupos R cargados negativamente



Aspartato



Glutamato





9 ESSENTIAL AMINO ACIDS



Amino acids are the building blocks of human protein, the body uses 20 amino acids, 9 of which are essential*

*Essential amino acids must be part of your diet

FUNCTIONS OF AMINO ACIDS

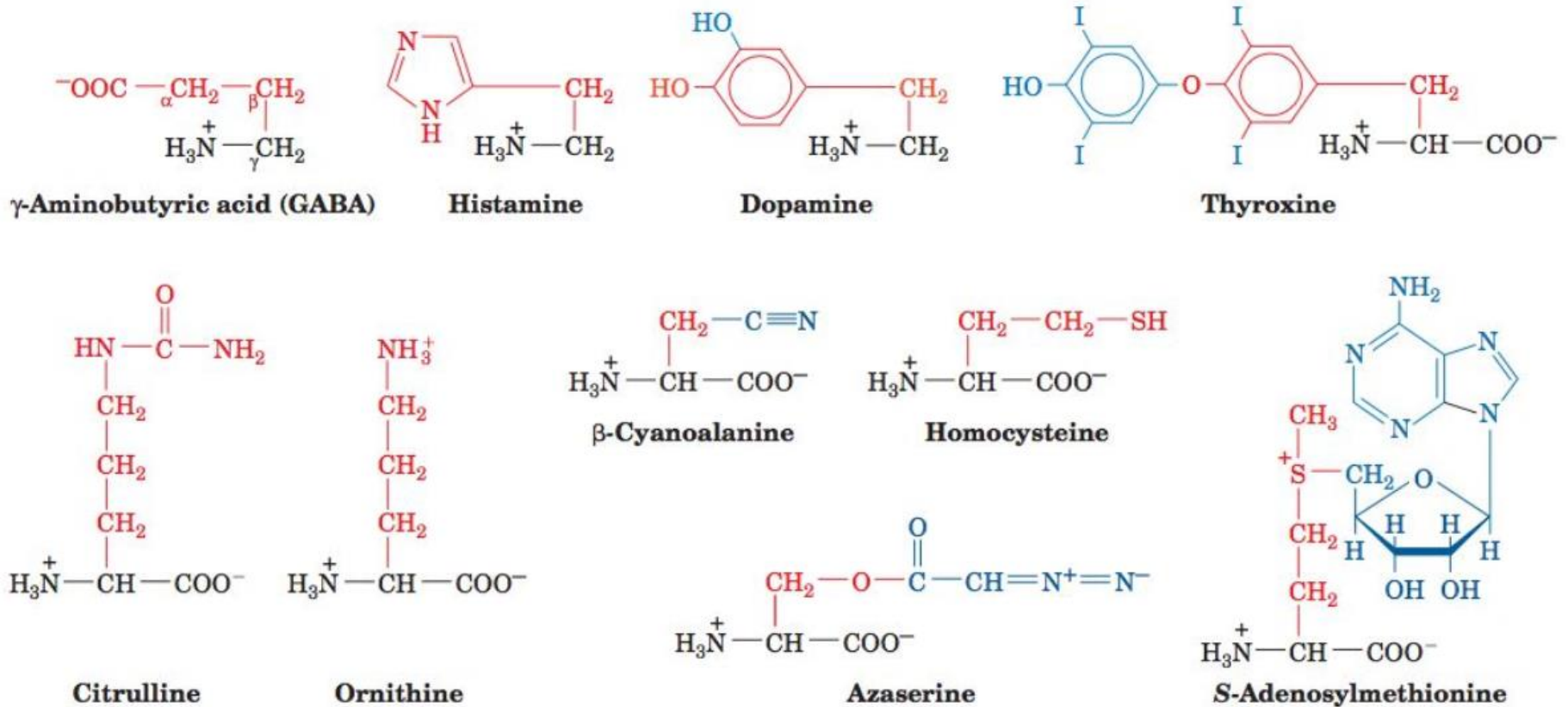
ISOLEUCINE	LEUCINE	LYSINE	METHIONINE	PHENYLALANINE	THREONINE	TRYPTOPHAN	VALINE	HISTIDINE
blood-sugar regulation growth and repair of muscle tissue hemoglobin development energy regulation	blood-sugar regulation growth and repair of muscle tissue hormone production wound healing energy regulation	development and regulation of collagen, antibodies, hormones, and enzymes promotes calcium absorption growth and repair of muscle tissue	may prevent arterial fat buildup may act as an antioxidant promotes collagen synthesis	neurotransmitter function increases blood levels of norepinephrine, epinephrine, dopamine	antibody production	precursor for serotonin aids sleep and combats anxiety niacin production	growth and repair of muscle tissue energy regulation niacin production	growth tissue repair histamine development

PLANT SOURCES OF EACH AMINO ACID

sacha inchi	sacha inchi	sacha inchi	sacha inchi	sacha inchi	sacha inchi	sacha inchi	sacha inchi	sacha inchi
-	-	beans	beans	-	beans	-	-	-
-	rice	-	-	-	-	rice	-	rice
-	-	-	-	peanuts	peanuts	peanuts	peanuts	-
almonds	almonds	-	-	almonds	almonds	-	almonds	-
soy	soy	soy	-	-	-	-	soy	-
wheat	wheat	-	-	-	-	-	-	wheat

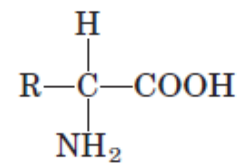
MEAT SOURCES OF EACH AMINO ACID

beef	beef	beef	beef	-	beef	beef	beef	-
dairy	dairy	dairy	dairy	dairy	dairy	dairy	dairy	-
eggs	eggs	eggs	eggs	-	eggs	-	eggs	-
fish	fish	fish	fish	-	-	fish	-	-
poultry	poultry	-	-	-	poultry	poultry	poultry	poultry

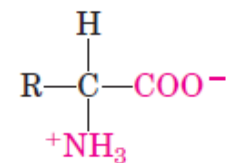


Alguns resíduos incomuns de aminoácidos que são componentes de certas proteínas

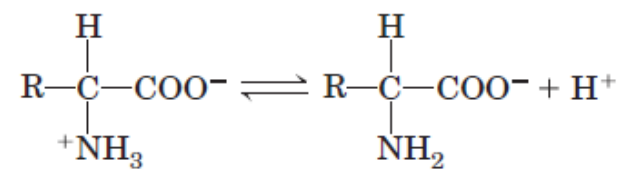
Formas não iônicas e zwitteriônicas de aminoácidos



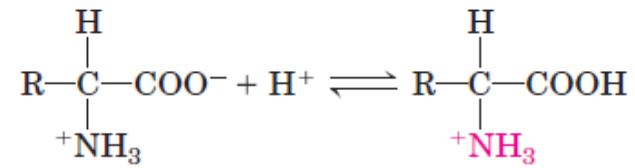
Forma não iônica



Forma zwitteriônica



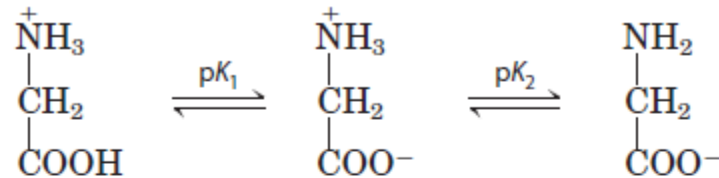
Zwitterión
como ácido



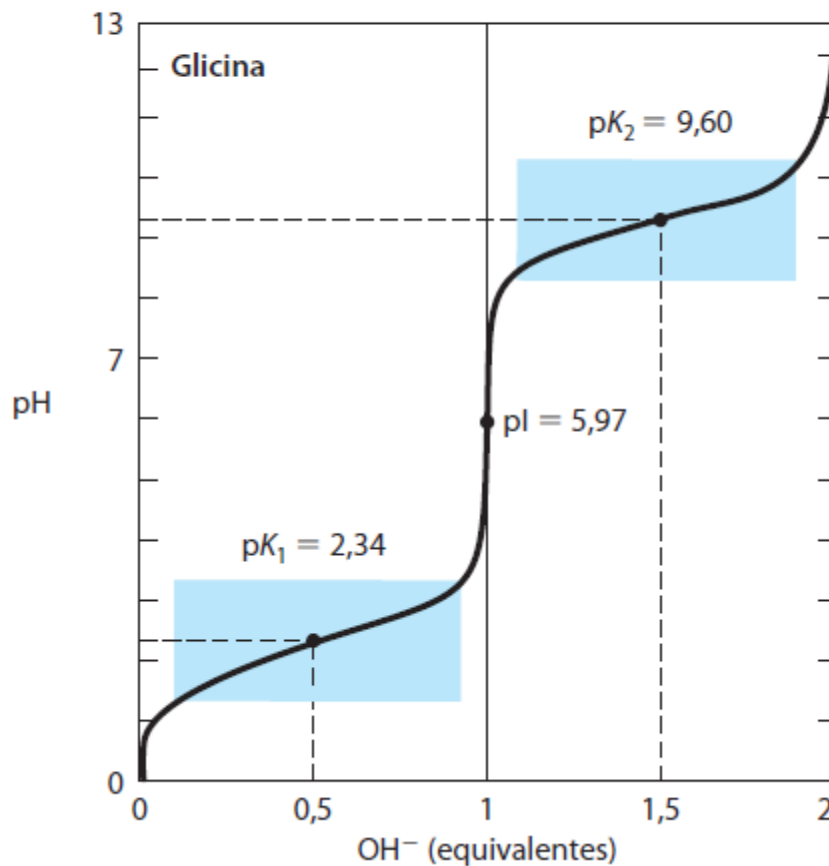
Zwitterión
como base

Titulação de um aminoácido

Glicina



$$pI = \frac{pK_1 + pK_2}{2}$$



pK_a

2

4

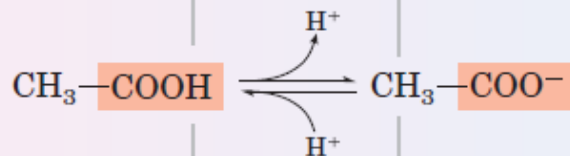
6

8

10

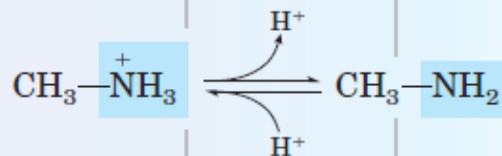
12

Grupos carboxila e amino substituídos por metil



Ácido acético

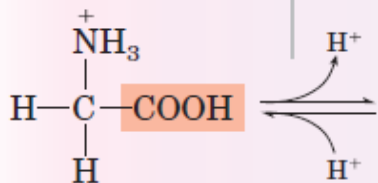
O pK_a normal para um grupo carboxila é de cerca de 4,8.



Metilamina

O pK_a normal para um grupo amino é de cerca de 10,6.

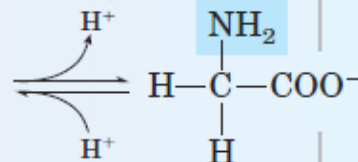
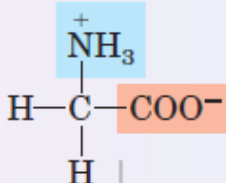
Grupos amino e carboxil na glicina



α-Aminoácido (glicina)

pK_a = 2,34

Grupos com cargas opostas diminuem o pK_a pela estabilização do zwitterion.

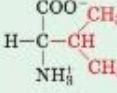
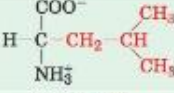
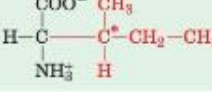
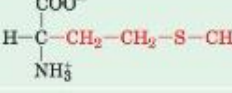
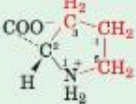
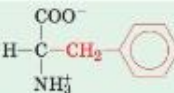
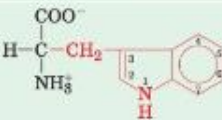


α-Aminoácido (glicina)

pK_a = 9,60

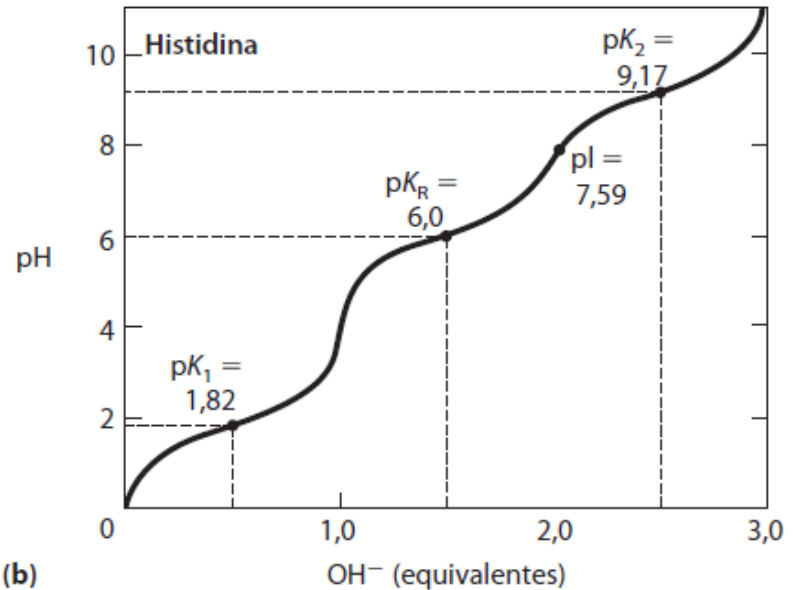
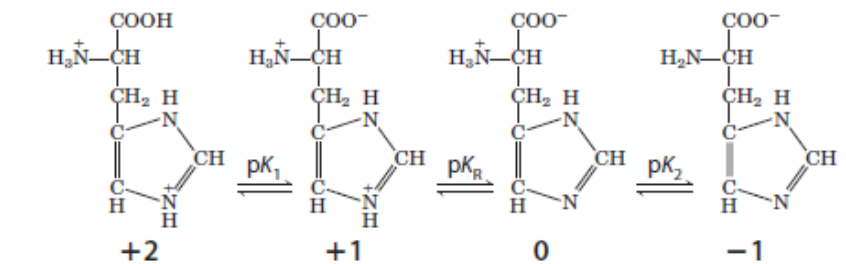
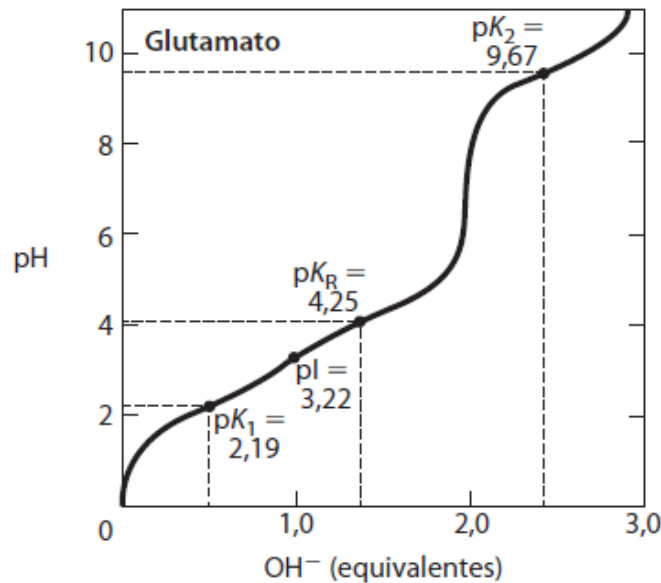
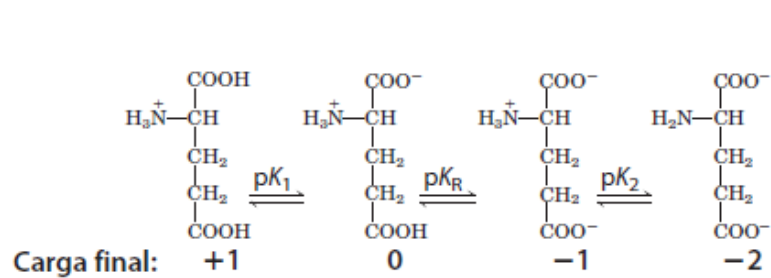
Átomos de oxigênio eletronegativos no grupo carboxila puxam os elétrons para longe do grupo amino, reduzindo seu pK_a.

Efeito do ambiente químico no pKa

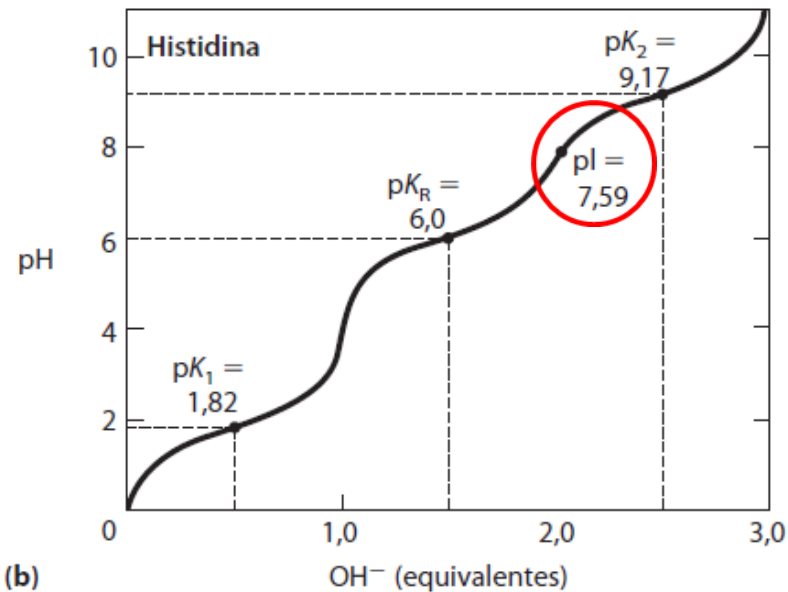
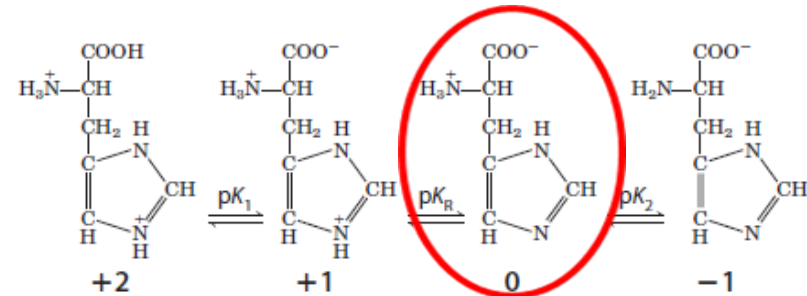
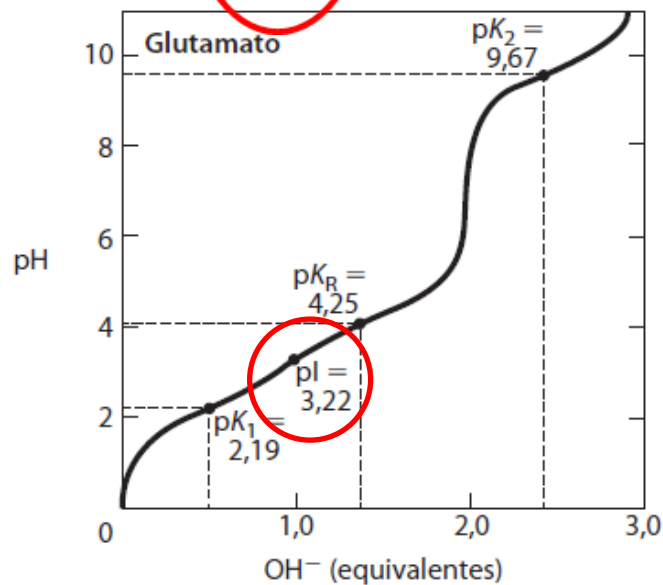
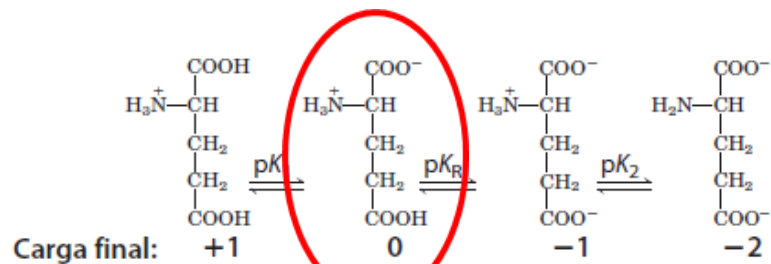
Name, Three-Letter Symbol, and One-Letter Symbol	Structural Formula ^a	Residue Mass (D) ^b	Average Occurrence in Proteins (%) ^c	pK ₁ α-COOH ^d	pK ₂ α-NH ₃ ⁺ ^d	pK _a Side Chain ^d
<i>Amino acids with nonpolar side chains</i>						
Glycine Gly G		57.0	7.1	2.35	9.78	
Alanine Ala A		71.1	8.3	2.35	9.87	
Valine Val V		99.1	6.9	2.29	9.74	
Leucine Leu L		113.2	9.7	2.33	9.74	
Isoleucine Ile I		113.2	6.0	2.32	9.76	
Methionine Met M		131.2	2.4	2.13	9.28	
Proline Pro P		97.1	4.7	1.95	10.64	
Phenylalanine Phe F		147.2	3.9	2.20	9.31	
Tryptophan Trp W		186.2	1.1	2.46	9.41	

(continued)

Name Three-Letter Symbol, and One-Letter Symbol	Structural Formula ^a	Residue Mass (D) ^b	Average Occurrence in Proteins (%) ^c	pK ₁ α-COOH ^d	pK ₂ α-NH ₃ ⁺ ^d	pK _R Side Chain ^d
Amino acids with uncharged polar side chains						
Serine Ser S		87.1	6.5	2.19	9.21	
Threonine Thr T		101.1	5.3	2.09	9.10	
Asparagine ^f Asn N		114.1	4.0	2.14	8.72	
Glutamine ^f Gln Q		128.1	3.9	2.17	9.13	
Tyrosine Tyr Y		163.2	2.9	2.20	9.21	10.46 (phenol)
Cysteine Cys C		103.1	1.4	1.92	10.70	8.37 (sulfhydryl)
Amino acids with charged polar side chains						
Lysine Lys K		128.2	5.9	2.16	9.06	10.54 (ε-NH ₃ ⁺)
Arginine Arg R		156.2	5.5	1.82	8.99	12.48 (guanidino)
Histidine ^e His H		137.1	2.3	1.80	9.33	6.04 (imidazole)
Aspartic acid ^f Asp D		115.1	5.4	1.99	9.90	3.90 (β-COOH)
Glutamic acid ^f Glu E		129.1	6.8	2.10	9.47	4.07 (γ-COOH)



Curvas de titulação para (a) glutamato e (b) histidina. O grupo R do $\text{p}K_a$ é designado aqui como $\text{p}K_R$.



Curvas de titulação para (a) glutamato e (b) histidina. O grupo R do pKa é designado aqui como pK_R .

Resumo Aminoácidos

- ★ Os 20 aminoácidos comuns como resíduos em proteínas contêm entre outros encontrados um grupo α -carboxila, um grupo α -amino e um grupo R característico substituído no átomo do carbono α . O átomo de carbono α de todos os aminoácidos, exceto a glicina, é assimétrico e, portanto, os aminoácidos podem existir em pelo menos duas formas estereoisoméricas. Apenas os estereoisômeros L, com uma configuração relacionada à configuração absoluta da molécula de referência L gliceraldeído, são encontrados em proteínas.
- ★ Outros aminoácidos menos comuns também ocorrem, tanto como constituintes de proteínas (pela modificação de resíduos de aminoácidos comuns após a síntese proteica) quanto como metabólitos livres.
- ★ Os aminoácidos podem ser classificados em cinco tipos com base na polaridade e carga (em pH 7) de seus grupos R.
- ★ Os aminoácidos variam em suas propriedades acidobásicas e têm curvas de titulação características. Aminoácidos monoamino monocarboxílicos (com grupos R não ionizáveis) são ácidos dipróticos ($\text{H}_3\text{N}^+\text{CH(R)COOH}$) em pH baixo e existem em várias formas iônicas diferentes à medida que o pH aumenta. Aminoácidos com grupos R ionizáveis têm espécies iônicas adicionais, dependendo do pH do meio e do pK_a do grupo R.

