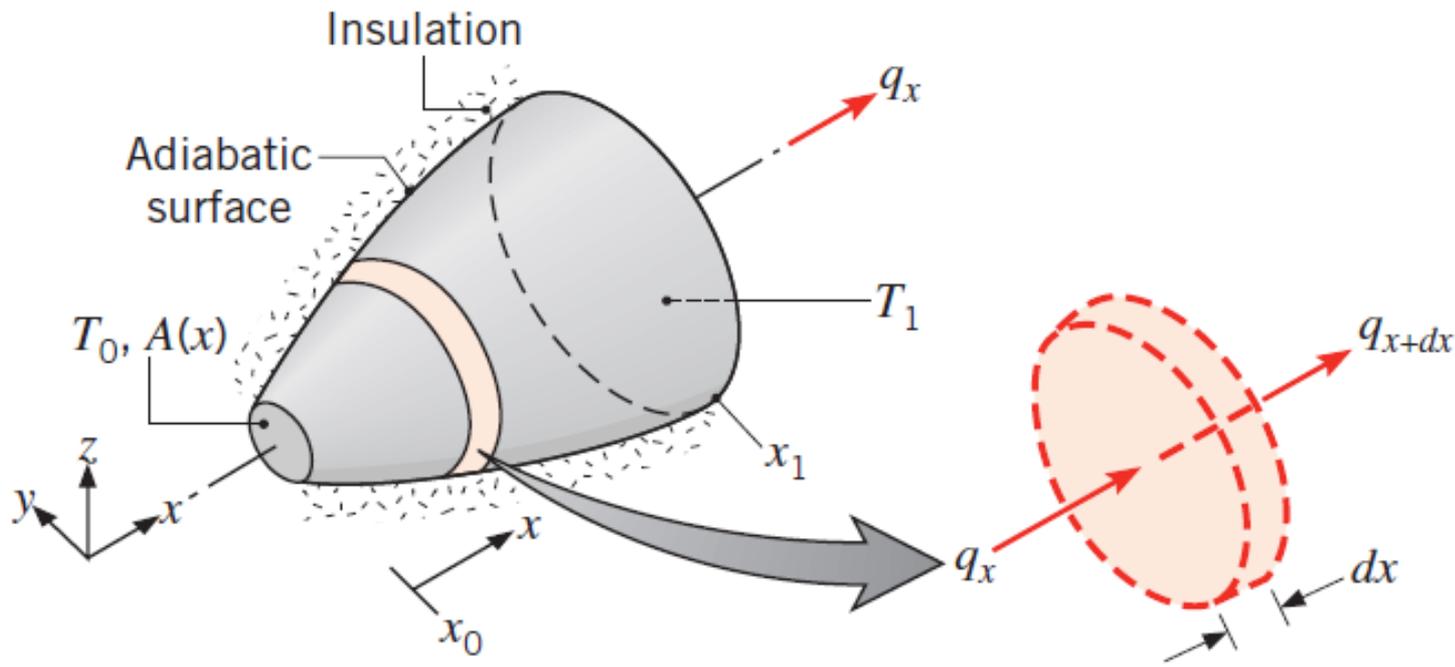


# ALETAS

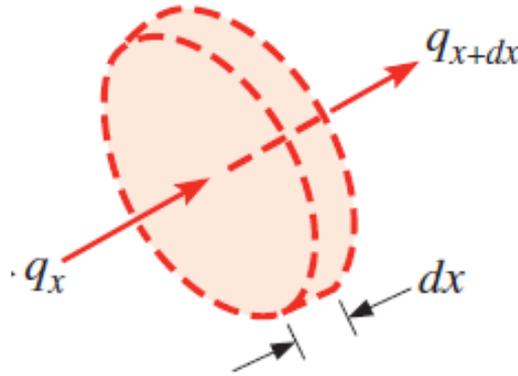
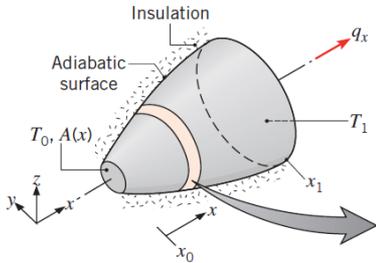
## Alternativa para análise de transmissão de calor

- Regime permanente
- Sem geração de calor no VC
- Sem perdas pelas parede laterais
- $q_x$  é constante e independe de  $x$



# ALETAS

## Alternativa para análise de transmissão de calor



Equação de Fourier

$$q_x = -KA \frac{dT}{dx}$$

Se  $q_x = -K(T)A(x) \frac{dT}{dx}$

$$\frac{q_x}{A(x)} = -K(T) \frac{dT}{dx}$$

se  $dx \neq 0$

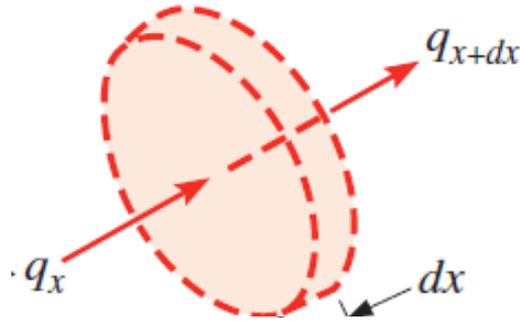
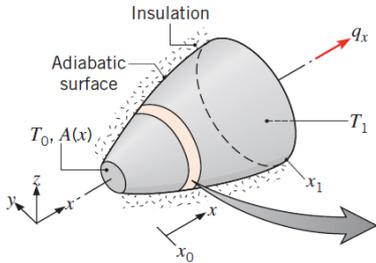
multiplicamos os dois lados por  $dx$

$$\frac{q_x}{A(x)} dx = -K(T) dT \quad \text{Integramos os dois lados}$$

$$\int_{x_0}^x \frac{q_x}{A(x)} dx = \int_{T_0}^T -K(T) dT$$

# ALETAS

## Alternativa para análise de transmissão de calor



should be firmly fixed in our minds: *steady-state* and *one-dimensional* transfer with *no heat generation*.

$$\int_{x_0}^x \frac{q_x}{A(x)} dx = \int_{T_0}^T -K(T) dT$$

$$T = T_0 \text{ quando } x = x_0$$

Se  $A$  e  $K$  são constantes

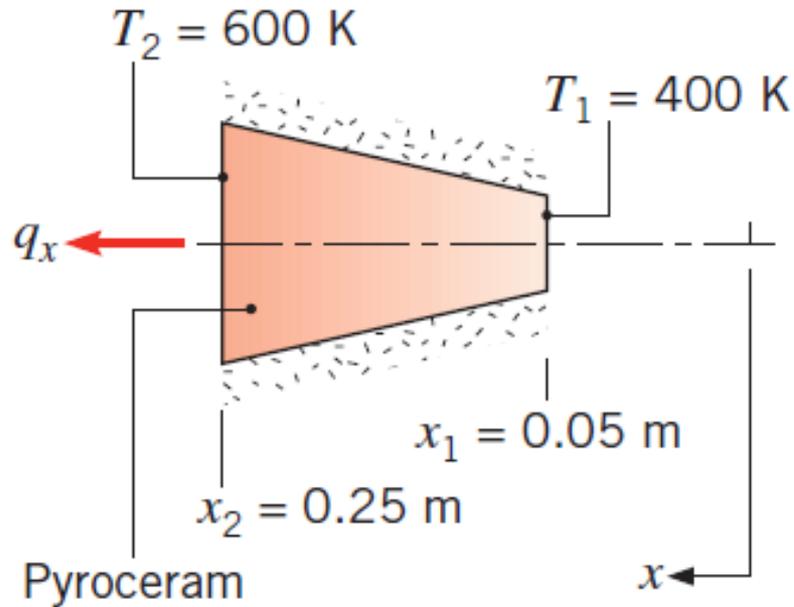
$$\frac{q_x \Delta x}{A} = -k \Delta T$$

**Se e Somente Se**  $\Rightarrow$  Regime permanente, unidimensional e sem geração de calor no VC.

# ALETAS

## Exemplo:

, pyroceram (500 K):  $k = 3.46 \text{ W/m} \cdot \text{K}$ .

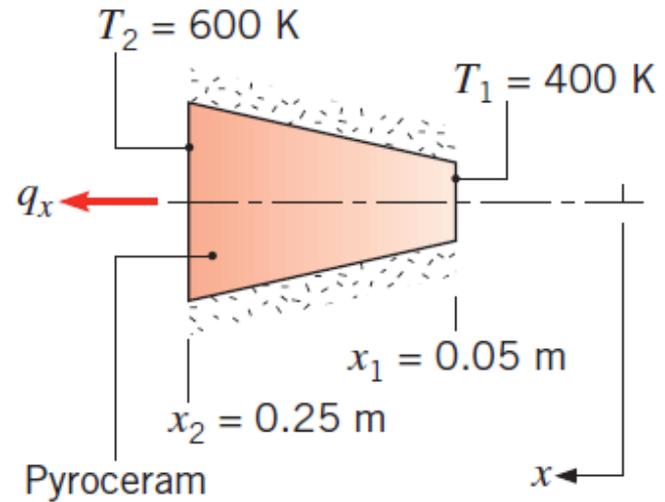


$$q_x = -kA \frac{dT}{dx}$$

# ALETAS

## Exemplo:

$$A = \pi D^2/4 = \pi a^2 x^2/4$$

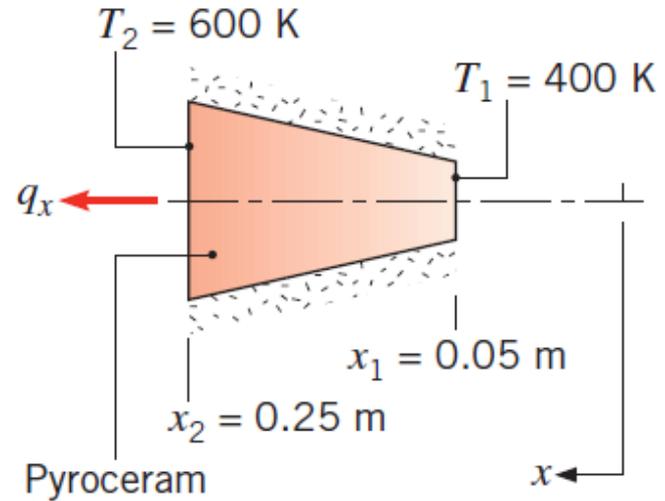


# ALETAS

## Exemplo:

$$A = \pi D^2/4 = \pi a^2 x^2/4$$

$$\frac{4q_x dx}{\pi a^2 x^2} = -kdT$$

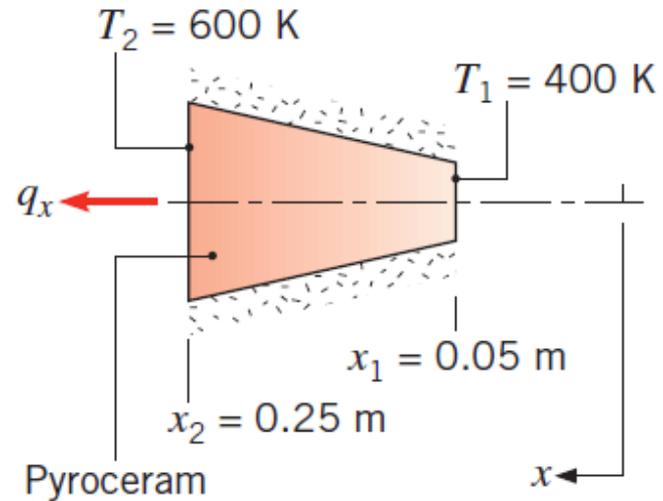


# ALETAS

## Exemplo:

$$A = \pi D^2/4 = \pi a^2 x^2/4$$

$$\frac{4q_x dx}{\pi a^2 x^2} = -k dT$$

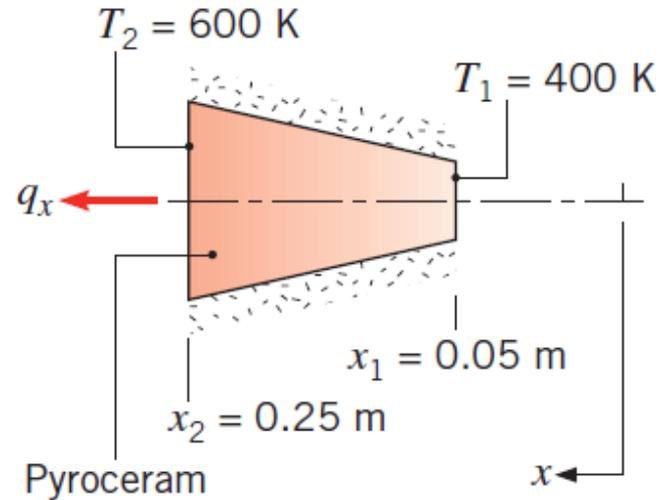


$$\frac{4q_x}{\pi a^2} \int_{x_1}^x \frac{dx}{x^2} = -k \int_{T_1}^T dT$$

# ALETAS

**Exemplo:**

$$\frac{4q_x}{\pi a^2} \int_{x_1}^x \frac{dx}{x^2} = -k \int_{T_1}^T dT$$



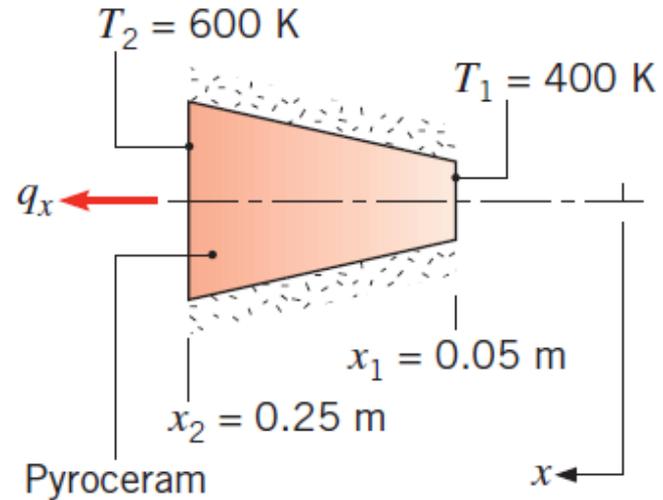
Integrando:

$$\frac{4q_x}{\pi a^2} \left( -\frac{1}{x} + \frac{1}{x_1} \right) = -k(T - T_1)$$

# ALETAS

**Exemplo:**

$$\frac{4q_x}{\pi a^2} \int_{x_1}^x \frac{dx}{x^2} = -k \int_{T_1}^T dT$$



Integrando:

$$\frac{4q_x}{\pi a^2} \left( -\frac{1}{x} + \frac{1}{x_1} \right) = -k(T - T_1)$$

$$T(x) = T_1 - \frac{4q_x}{\pi a^2 k} \left( \frac{1}{x_1} - \frac{1}{x} \right)$$

# ALETAS

## Exemplo:

$$\frac{4q_x}{\pi a^2} \left( -\frac{1}{x} + \frac{1}{x_1} \right) = -k(T - T_1)$$

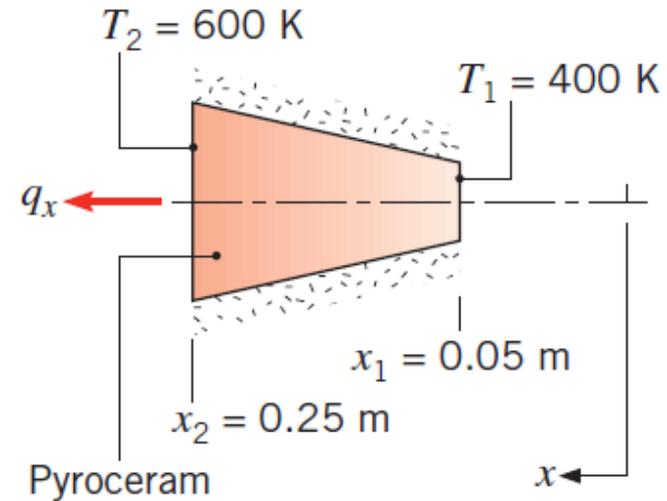
at  $x = x_2$ , where  $T(x_2) = T_2$ .

$$T_2 = T_1 - \frac{4q_x}{\pi a^2 k} \left( \frac{1}{x_1} - \frac{1}{x_2} \right)$$

$$q_x = \frac{\pi a^2 k (T_1 - T_2)}{4 \left[ (1/x_1) - (1/x_2) \right]}$$

Substituindo  $q_x$  em  $T(x)$

$$T(x) = T_1 - \frac{4q_x}{\pi a^2 k} \left( \frac{1}{x_1} - \frac{1}{x} \right)$$

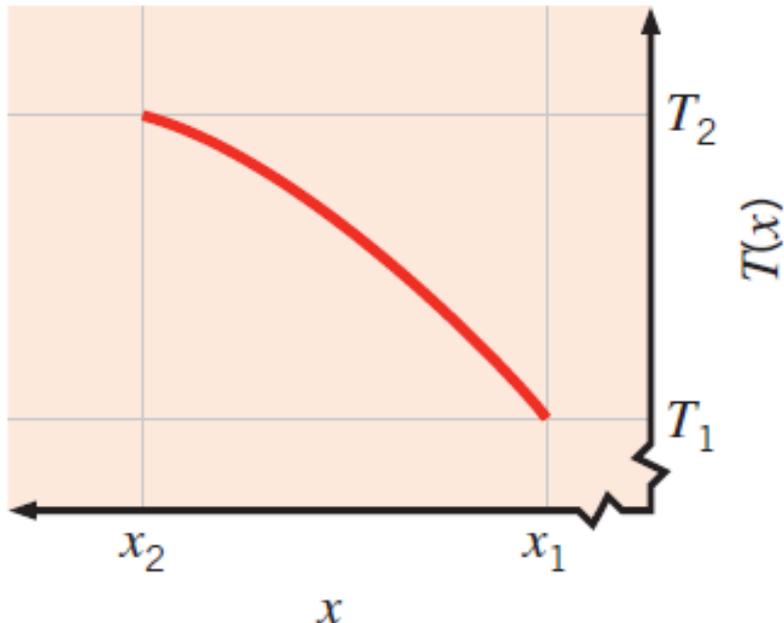
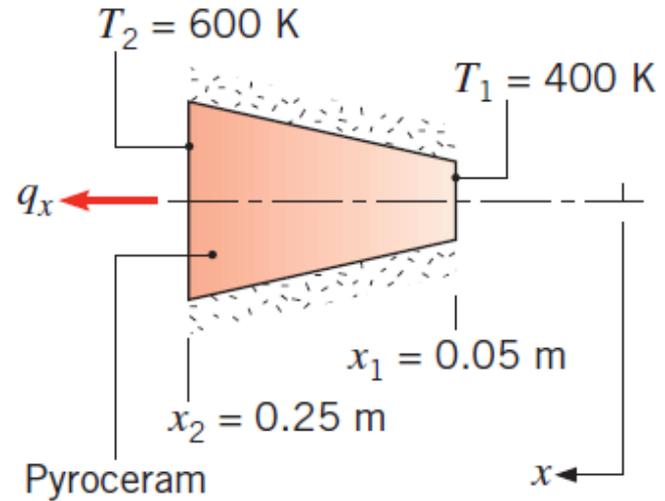


$$T(x) = T_1 + (T_1 - T_2) \left[ \frac{(1/x) - (1/x_1)}{(1/x_1) - (1/x_2)} \right]$$

# ALETAS

**Exemplo:**

$$T(x) = T_1 + (T_1 - T_2) \left[ \frac{(1/x) - (1/x_1)}{(1/x_1) - (1/x_2)} \right]$$

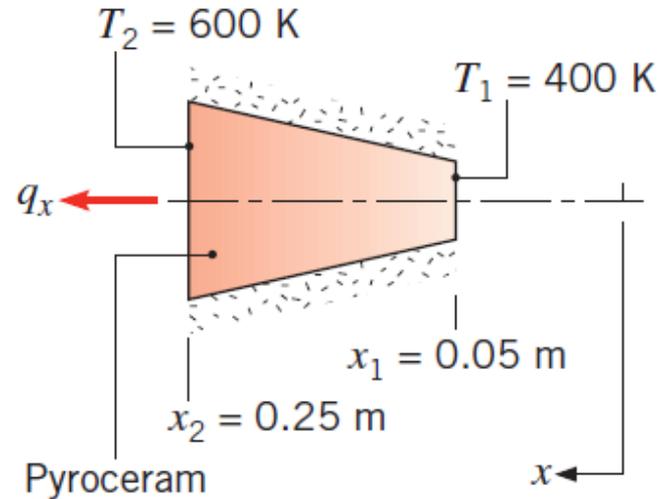


# ALETAS

## Exemplo:

, pyroceram (500 K):  $k = 3.46 \text{ W/m} \cdot \text{K}$ .

$$q_x = \frac{\pi a^2 k (T_1 - T_2)}{4 \left[ \frac{1}{x_1} - \frac{1}{x_2} \right]}$$



Substituindo valores numéricos:

$$q_x = \frac{\pi (0.25)^2 \times 3.46 \text{ W/m} \cdot \text{K} (400 - 600) \text{ K}}{4 (1/0.05 \text{ m} - 1/0.25 \text{ m})} = -2.12 \text{ W}$$