

# AULA 14 – PTC 3435

04/10/2018

# POTENCIAL DE MEIA CÉLULA

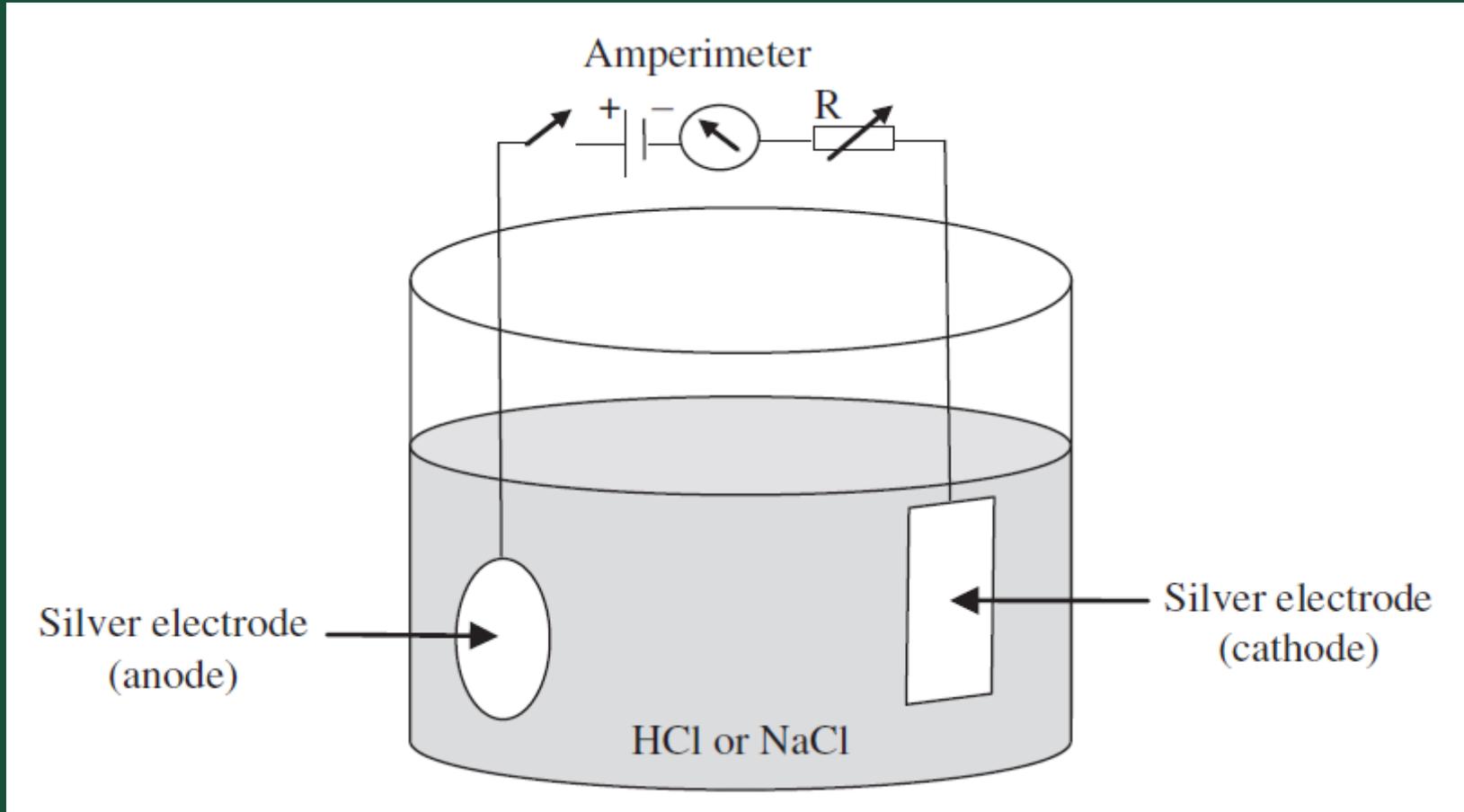
TABLE 10.2 Half-Cell Potentials of Important Metals

Primary metal and chemical reaction	Half-cell potential (V)
$\text{Al} \rightarrow \text{Al}^{3+} + 3\text{e}^{-}$	-1.706
$\text{Cr} \rightarrow \text{Cr}^{3+} + 3\text{e}^{-}$	-0.744
$\text{Cd} \rightarrow \text{Cd}^{2+} + 2\text{e}^{-}$	-0.401
$\text{Zn} \rightarrow \text{Zn}^{2+} + 2\text{e}^{-}$	-0.763
$\text{Fe} \rightarrow \text{Fe}^{2+} + 2\text{e}^{-}$	-0.409
$\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^{-}$	-0.230
$\text{Pb} \rightarrow \text{Pb}^{2+} + 2\text{e}^{-}$	-0.126
$\text{H}_2 \rightarrow 2\text{H}^{+} + 2\text{e}^{-}$	0.000 (standard by definition)
$\text{Ag} \rightarrow \text{Ag}^{+} + \text{e}^{-}$	+0.799
$\text{Au} \rightarrow \text{Au}^{3+} + 3\text{e}^{-}$	+1.420
$\text{Cu} \rightarrow \text{Cu}^{2+} + 2\text{e}^{-}$	+0.340
$\text{Ag} + \text{Cl}^{-} \rightarrow \text{AgCl} + 2\text{e}^{-}$	+0.223

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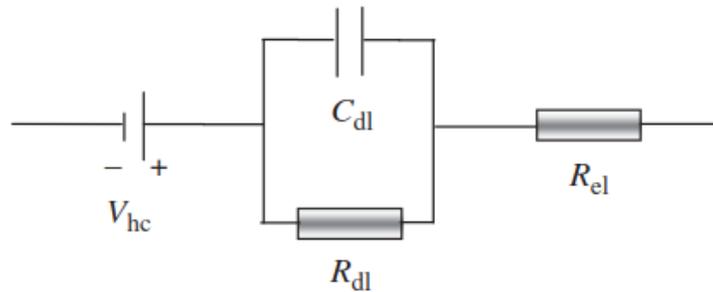
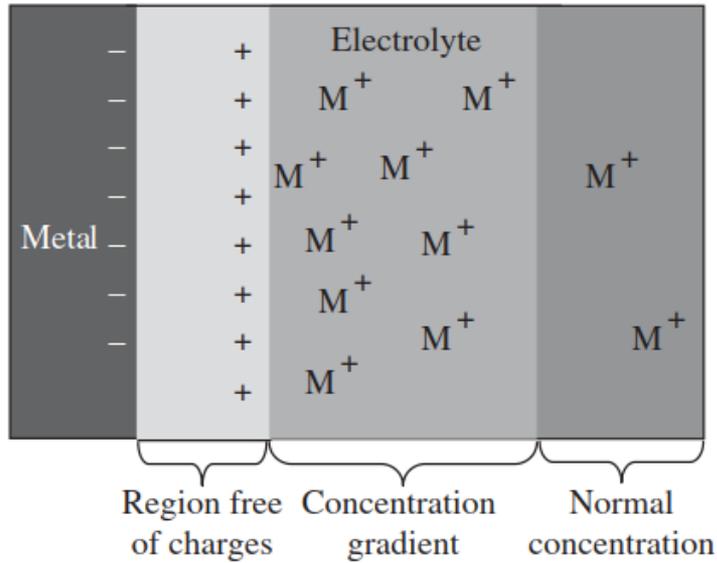


# ELETRODO Ag-AgCl

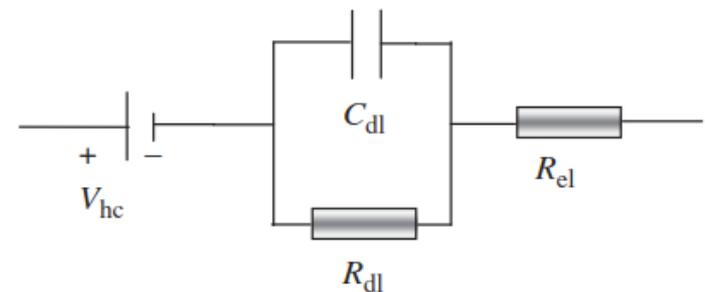
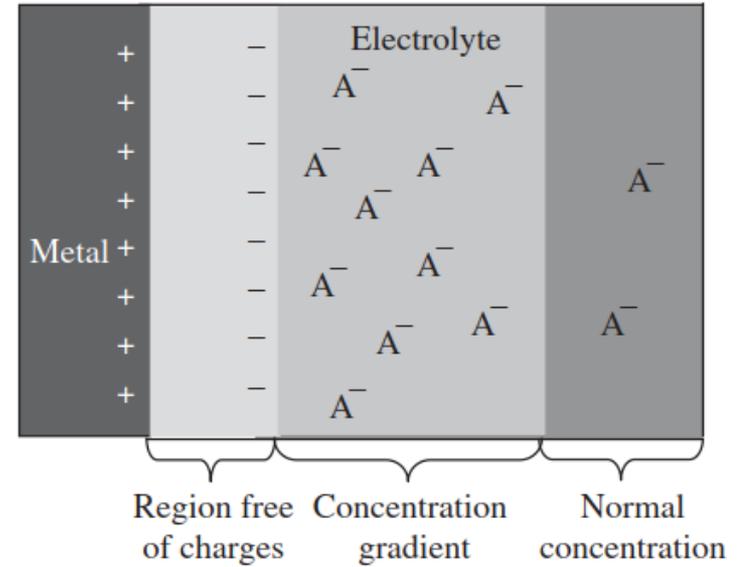


# INTERFACE ELETRODO-ELETRÓLITO

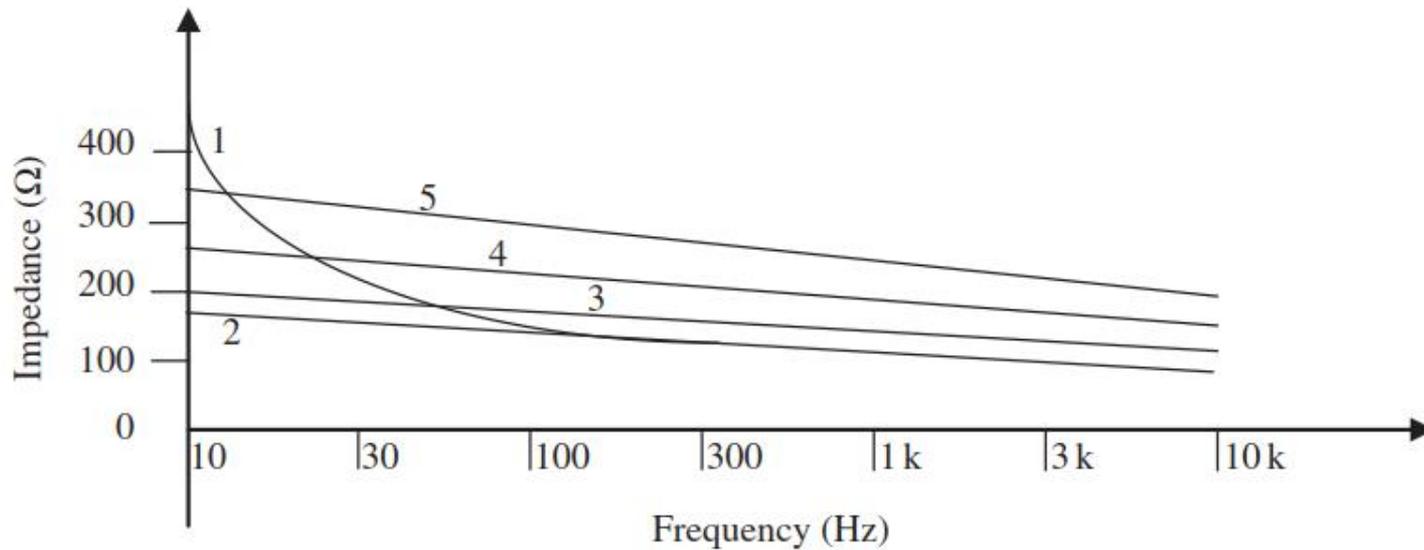
(A)  $V_{hc} < 0$



(B)  $V_{hc} > 0$

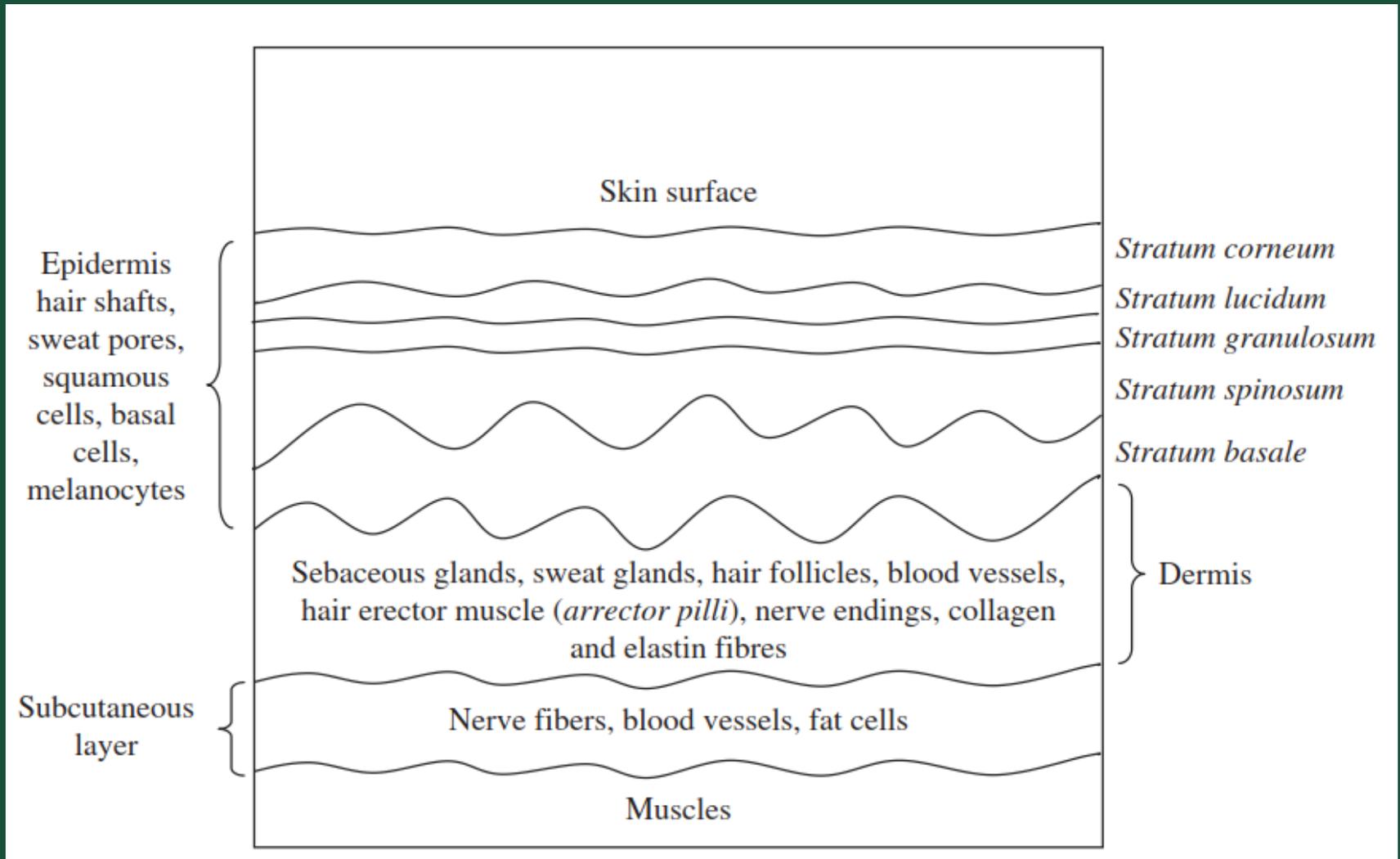


# INTERFACE ELETRODO-ELETRÓLITO

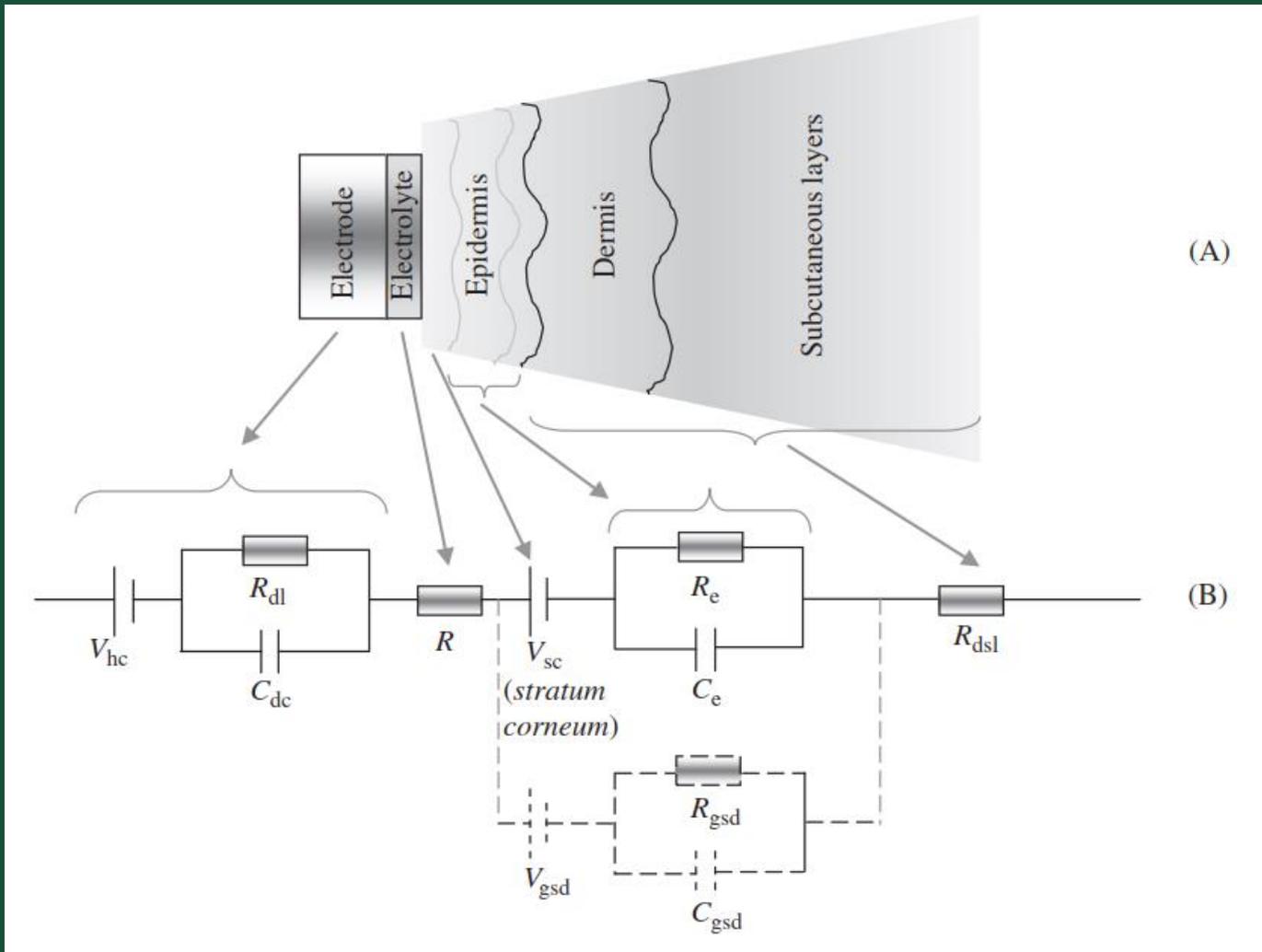


**Figure 2.12** Impedance as a function of frequency for electrodes with an area of  $0.25 \text{ cm}^2$  made of pure silver (1) and silver coated with AgCl layer deposited electrolytically with (2)  $7.5 \text{ mA} \cdot \text{s}$ ; (3)  $125 \text{ mA} \cdot \text{s}$ ; (4)  $275 \text{ mA} \cdot \text{s}$ ; and (5)  $425 \text{ mA} \cdot \text{s}$ .

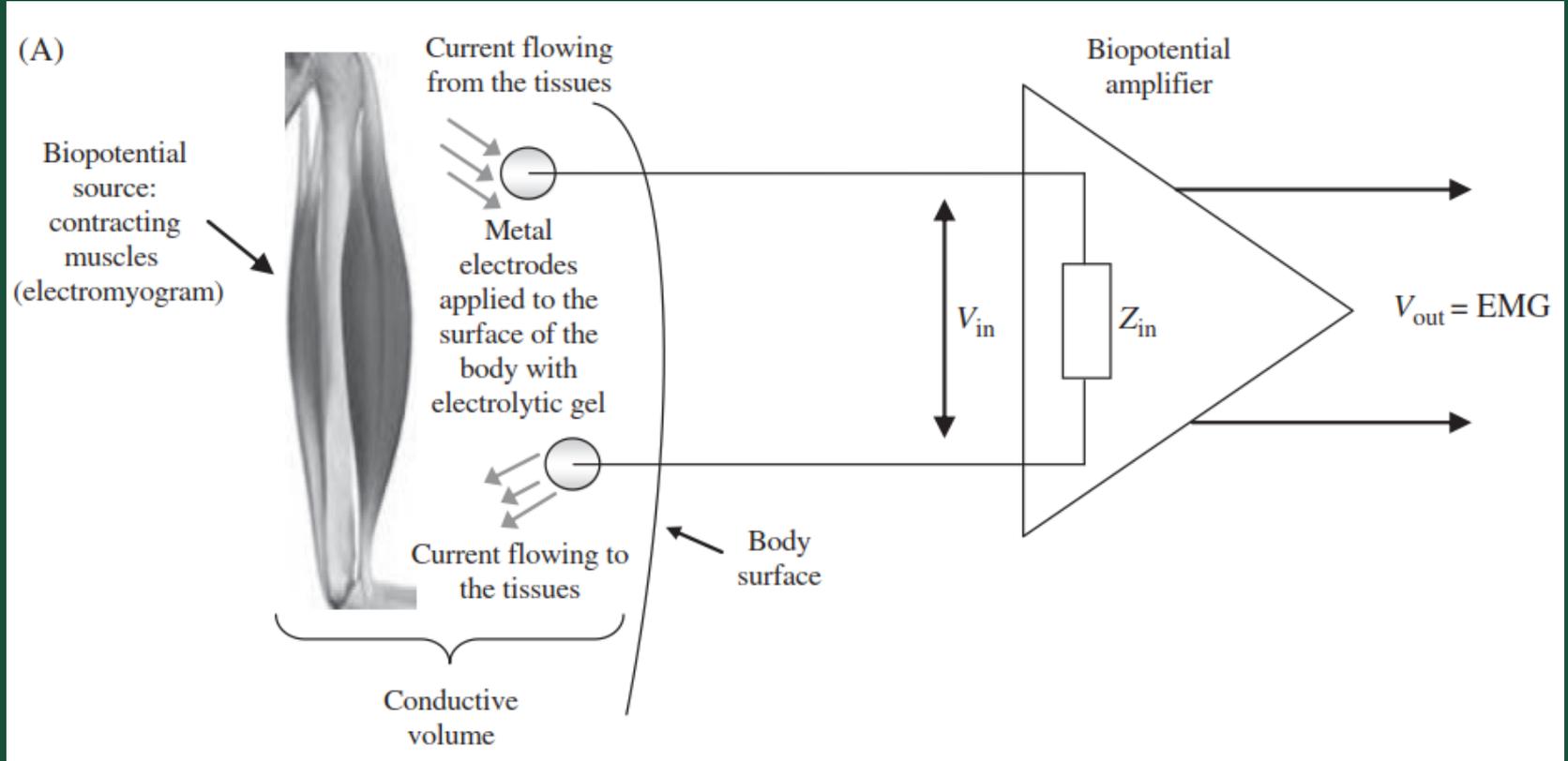
# PELE



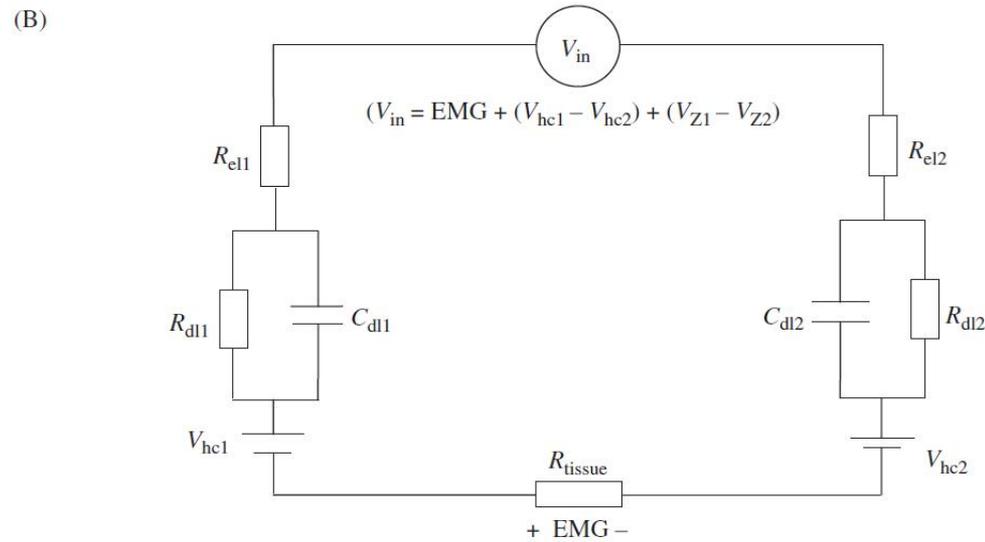
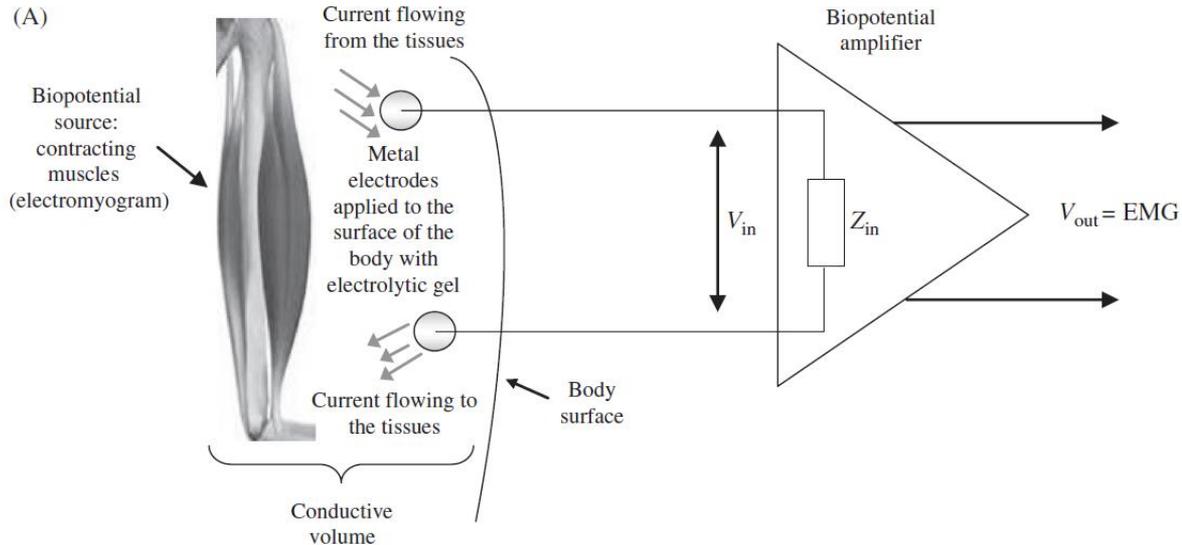
# INTERFACE ELETRODO-ELETRÓLITO-PELE



# ARTEFATOS DE MEDIÇÃO



# ARTEFATOS DE MEDIÇÃO



# ARTEFATO DE MOVIMENTO

